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Published in:
Journal of Magnetism and Magnetic Materials

Link to article, DOI:
10.1016/j.jmmm.2015.03.068

Publication date:
2015

Document Version
Early version, also known as pre-print

Link back to DTU Orbit

Citation (APA):
Magnetocaloric effect and H gradient in bulk La(Fe,Si)$_{13}$H$_y$ magnetic refrigerants obtained by HDSH

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Abstract

Results are reported on the preparation of bulk parts of La(Fe,Si)$_{13}$H$_y$ via the Hydrogen-Decrepitation-Sintering-Hydrogenation (HDSH) process. Net shape parts for application in room-temperature magnetic refrigeration have been produced in only 8 h of heat treatment which is considerably faster than the conventional ingot homogenization heat treatment of 7 days. The samples produced by HDSH showed higher amounts of hydrogen than the parts hydrogenated by the conventional method of thermal homogenization (20 h at 1423 K), milling to fine powder and subsequent hydrogenation. Hydrogenation parameters play an important role for the stability of the desired La(Fe,Si)$_{13}$ phase during the process. Hydrogen desorption was seen to occur at two temperature ranges as a result of internal gradients. Dissimilar amounts of $\alpha$-Fe were precipitated for different hydrogenation times. As a result, parts produced via HDSH with 2 and 4 h of hydrogenation exhibited different magnetocaloric behaviours. For a hydrogenation step of 4 h, parts with a demagnetization factor of 0.49 showed an adiabatic temperature change ($\Delta T_{ad}$) higher than 1 K for a temperature range of 40 K with a maximum value of 1.57 K for an applied magnetic field of 1.75 T. As the duration of the hydrogenation step of the HDSH process decreased to

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2 h, $\Delta T_{ad}$ was larger than 1 K for a temperature range of 24 K. However the maximum value of $\Delta T_{ad}$ at 328 K was 2.2 K, which is 37.5 % larger than the maximum value for a hydrogenation period of 4 h.

**Keywords:** La(Fe,Si)$_{13}$H$_y$ magnetic refrigerants, magnetocaloric materials, HDSH, hydrogenation

1. Introduction

The magnetocaloric effect (MCE) is the thermal response of certain materials when subjected to a changing magnetic field. It has been investigated since the early 19$^{th}$ century [1] and the pioneering work of Giauque and MacDougall [2] on the use of adiabatic magnetization to cool below 1 K was awarded Nobel Prize in Chemistry in 1949. Magnetic refrigeration at room temperature was firstly demonstrated with gadolinium (Gd) as the magnetocaloric material, where a no-load temperature span of 47 K was achieved with approximately 157 g of Gd and a magnetic field of 7 T [3].

Magnetic refrigeration has a great potential for high thermodynamic cycle efficiency due to the reversible nature of the MCE [3, 4, 5]. The discovery of the giant MCE in Gd$_5$Si$_2$Ge$_2$ by Pecharsky and Gschneidner [6] prompted the research for other first-order phase transition (FOPT) materials with similar characteristics [7].

A more recent group of materials with promising properties is the La-based alloys, LaFe$_{13-x}$Si$_x$, which crystallise in the NaZn$_{13}$-type phase structure for values of $x$ between 1.05 and 2.5 [8, 9]. Usually materials based on this phase are produced by a conventional melting, casting and annealing. The latter is necessary in order to achieve nearly 100 % of the desired phase and it is a long-term homogenization step, which may take up seven days in high temperature heat treatments [10]. Faster routes have been proposed, e.g. melt-spinning, for which a nearly monophasic microstructure has been stabilised after a 2 h high-temperature heat treatment [11]. Other routes based on powder metallurgy have been suggested [12, 13].
LaFe_{13-x}Si_{x} compounds have Curie temperatures, $T_C$, between 180 and 230 K, for increasing amounts of Si \[14\]. Therefore, to use this material for magnetic cooling around room temperature, the $T_C$ must be increased. One possible method to increase the $T_C$ of this material is by expanding the lattice parameter, e.g. inserting interstitial atoms such as H or C \[15, 16\]. The hydrogenation step is commonly performed in the powder form \[17\]. Nevertheless, after the hydrogenation process, where the hydrogen is inserted interstitially into the NaZn_{13}-type structure, some difficulties may be pointed out. Firstly, the resulting powder cannot be heat treated in order to shape it, since the hydrogen stability in the microstructure goes only up to around 500 K \[16\]. If one try to sinter under H$_2$ atmosphere, at around 1023 K LaH$_2$ starts to stabilize, destroying the desired phase. Hence, without being able to shape it, its application is now limited since in many cases a functional part with well-defined geometry is demanded \[18\].

In this work, the recently reported \[19\] Hydrogen-Decrepitation-Sintering-Hydrogenation (HDSH) process was used to produce bulk parts of La(Fe,Si)$_{13}$H$_y$, varying the H content in order to cover a broader range of working temperatures. Samples from the same ingot were also produced by the conventional process and the resulting samples from both processes were then compared.

## 2. Experimental Procedure

Bulk parts of La(Fe,Si)$_{13}$H$_y$ were produced by HDSH, as described in previous work \[19\]. The process is divided in 3 basic steps:

1. The process uses a La-Fe-Si ingot as raw material, which is hydrogen-decrepitated into small pieces, at 423 K during 1 hour.

2. The material is then milled, pressed into 1 mm thick pellets and sintered at 1423 K, for 6 h under argon atmosphere. During the heating step of the sintering, the hydrogen that was inserted in the material during HD is desorbed due to the high temperatures.
In the cooling step after the sintering, the temperature is reduced to 773 K and the hydrogenation step is, thus, performed during two different time spans: 2 and 4 h; under pure H\textsubscript{2} atmosphere at pressures above 1 atm. In this step the interconnected pores are the free path for hydrogen, so it may reach the whole sample volume. Subsequently, the samples are slowly cooled to room-temperature in the furnace.

In order to compare and evaluate the HDSH process, La(Fe,Si)\textsubscript{13}H\textsubscript{y} powder was prepared by the conventional method, which is divided in 3 basic steps:

1. Cylindrical parts of the same ingot used to produce the HDSH parts, were cut by electro-erosion.
2. The parts were homogenized during 20 h at 1423 K under a 90 kPa Ar atmosphere.
3. The parts were then milled into powder (<100 µm) and a hydrogenation step was carried out, for 2 h at 773 K under pure H\textsubscript{2} atmosphere at pressures above 1 atm.

The phase constitution of the samples was determined via means of powder X-ray diffraction (XRD) on a Phillips X’Pert Plus Diffractometer with Cu K\textsubscript{α} radiation. The quantitative phase analyses and the crystal structure parameters were analysed by the Rietveld method, refining the XRD patterns by GSAS code [20]. The hydrogen content and thermal stability were analysed and calculated by thermo-gravimetric (TG) analysis on a Netzsch simultaneous thermal analyser (STA), model 449 F3 Jupiter.

The MCE was quantified via the adiabatic temperature change, $\Delta T_{ad}$, measured in different synthesised samples using a home-built $\Delta T_{ad}$-ometer device [21]. In this device the sample is rapidly subjected to an applied magnetic field change from 0 to 1.75 T, starting from a stabilised sample temperature between 258 K and 348 K.
3. Results

In order to study the influence of the hydrogenation step on the amount of La(Fe, Si)\textsubscript{13} and on the lattice parameter \(a\) in the HDSH samples, XRD diffraction patterns were refined for both the samples before and after the hydrogenation processes as can be seen in Fig. 1(a). The amount of \(\alpha\)-Fe increases with the hydrogenation time, from 0.97 wt.\% for the as-sintered sample to 2.92 wt.\% and 6.13 wt.\% for 2 h and 4 h of hydrogenation step, respectively, as shown in Fig. 1(b). The increase of \(\alpha\)-Fe content could be related to formation of La oxide or LaH\textsubscript{2}, which even in amounts undetectable by XRD, could lead to significant precipitation of \(\alpha\)-Fe, as the structure La(Fe, Si)\textsubscript{13} is mostly Fe.

The lattice parameter \(a\) after the hydrogenation was increased from 11.48 Å to 11.59 Å, for both 2 h and 4 h of hydrogenation, which in the XRD data may be seen as a shift of the XRD pattern to lower angles, as shown in Fig. 1(b). These results are summarized in Table 1.

![Figure 1](image_url)

Figure 1: Colors are available in online version. (a) Rietveld pattern for the XRD data of samples prepared by HDSH before hydrogenation (as-sintered), after 2 h (red line) and 4 h (blue line) of hydrogenation. (b) A zoom in the green dashed box from Fig.1(a) showing the shift on the pattern due to hydrogenation and the \(\alpha\)-Fe peaks.

Thermal analyses of the hydrogenated samples were performed to understand the thermal stability of the interstitial hydrogen as a function of the tem-
temperature. Fig. 2 summarizes the thermo-gravimetric analyses performed in the samples produced by HDSH compared to those produced by the conventional method. The thermal analyses shows two different H desorption ranges for the samples produced by HDSH, which may be related to a gradient of hydrogen, since these parts were hydrogenised in the bulk form. Also, as shown in Fig. 2, the gradient increases with the duration of the hydrogenation. Furthermore, these results indicate that the amount of interstitial H is higher for the parts produced by HDSH.

Although it is expected that longer hydrogenation times should lead to a lower H gradient [22], the different hydrogen contents might be related to the $\alpha$-Fe precipitation. When $\alpha$-Fe precipitates, the ratio $\alpha$-Fe/Si changes, hence changing the hydrogen saturation locally, which could lead to a greater gradient. Other relevant hydrogenation methods are found in the literature [13, 22], which might hydrogenate the sintered samples more efficiently.

Fig. 3 shows the $\Delta T_{ad}$ of the different samples with $T_C$ around room temperature. It should be noted that the demagnetization factor, $N_D$, is different for each sample due to the differences in the shape between each sample. The $N_D$ for the samples from the conventional process, HDSH with 2 h and 4 h of hydrogenation are 0.27, 0.32 and 0.49, respectively [23]. The temperature at which $\Delta T_{ad}$ is maximum, $T_{peak}$, was found to be around 329 K for the samples produced by the conventional process and by HDSH with 2 h of hydrogenation, while $T_{peak}$ was 328 K for the sample produced by HDSH with 4 h of hydrogenation. The sample obtained by the conventional process and tested

<table>
<thead>
<tr>
<th>Description</th>
<th>$a$</th>
<th>La(Fe$<em>{13}$Si)$</em>{13}$H$_y$</th>
<th>$\alpha$-Fe</th>
<th>LaFeSi</th>
</tr>
</thead>
<tbody>
<tr>
<td>As-Sintered</td>
<td>11.48 Å</td>
<td>99.03 wt.%</td>
<td>0.97 wt.%</td>
<td>0.00 wt.%</td>
</tr>
<tr>
<td>HDSH (2 h)</td>
<td>11.59 Å</td>
<td>97.08 wt.%</td>
<td>2.92 wt.%</td>
<td>0.00 wt.%</td>
</tr>
<tr>
<td>HDSH (4 h)</td>
<td>11.59 Å</td>
<td>93.58 wt.%</td>
<td>6.13 wt.%</td>
<td>0.34 wt.%</td>
</tr>
</tbody>
</table>

Table 1: Relative phase amount and lattice parameter $a$, obtained by Rietveld refinement, of the samples produced by HDSH before the hydrogenation step (as-sintered), after 2 h and after 4 h of hydrogenation. The as-sintered sample measurements were taken before hydrogenation.
in powder form still resulted in a larger $\Delta T_{ad}$, with a maximum value of 2.64 K. The maximum values of the $\Delta T_{ad}$ for the bulk parts produced by HDSH process with 2 h and 4 h of hydrogenation were 2.07 K and 1.57 K, respectively. However, the parts produced by HDSH exhibited peaks that were broader than those produced by the conventional process. It is important to point out that the part hydrogenised for 4 h had a $\Delta T_{ad}$ larger than 1 K for a temperature range of 40 K. These range differences can be also related to the H gradients which are observed in Fig. 2. The results are summarized in Table 2. Considering the broader MCE observed in the samples produced by HDSH, this process may be advantageous in the production of bulk materials for magnetic cooling devices, since fabrication of graded FOPT magnetocaloric materials is still in development.

It should be noted that the measured MCE is influenced by the $N_D$ of the samples, which reduces the resulting internal magnetic field. Therefore, the three curves should exhibit larger values of $\Delta T_{ad}$. In this way, the samples produced by HDSH process should have maximum values of $\Delta T_{ad}$ closer to
that of the sample produced by the conventional process. Simple calculations
based on the scaling relationship $\Delta T_{ad} \sim H^{2/3}$ show that the difference between
the maximum value of $\Delta T_{ad}$ is significantly larger only for the sample produced
by HDSH with 4 h of hydrogenation, as its $N_D$ is 0.49.

A relationship has been identified between the reduction of the MCE and the
phase content fraction as well. Firstly, due to $\alpha$–Fe precipitation, some distri-
bution on $x$ and $y$ of LaFe$_{11.3-x}$Si$_{1.7+y}$H$_y$ may be found, instead of one singular
value. For the 2 h hydrogenation HDSH sample, the total amount of $\alpha$–Fe was
2.92 wt.%, while for the 4 h hydrogenation HDSH sample the $\alpha$–Fe amount was
6.13 wt.%. The difference of $\alpha$–Fe content, not only decreases the MCE due to
the lower global amount of La(Fe,Si)$_{13}H_y$, but it may also create preferential
pathways for the magnetic flux through the $\alpha$–Fe impurity grains rather than
through the surrounding La(Fe,Si)$_{13}H_y$ matrix, because of the higher magnetic
susceptibility of the former. A simple 2D magnetostatic model was set up to
analyse the effect of the magnetic susceptibility. In the model a magnetic field
is applied to rectangular sample with the susceptibility of La(Fe,Si)$_{13}H_y$, and
the average internal field is calculated. Then areas equivalent to 2.92 and 6.13
wt.% of $\alpha$–Fe were substituted as randomly positioned fine grains in the matrix
of La(Fe,Si)$_{13}H_y$, equivalent to the samples produced by HDSH with 2 and 4 h
of hydrogenation, respectively. In both situations the average internal field in
the matrix around the $\alpha$–Fe grains was observed to be within 1% of the value
for the pure sample. Thus, the effect of the $\alpha$–Fe impurities on the internal

Table 2: Summary of the $\Delta T_{ad}$ measurements. In this case, $\Delta T_{ad}^{Max}$ is the maximum value
of $\Delta T_{ad}$ at $T_{peak}$.

<table>
<thead>
<tr>
<th>Sample</th>
<th>$N_D$</th>
<th>$T_{peak}$</th>
<th>$\Delta T_{ad}^{Max}$</th>
<th>Range of $\Delta T_{ad} \geq 1$ K</th>
</tr>
</thead>
<tbody>
<tr>
<td>Conventional</td>
<td>0.27</td>
<td>329 K</td>
<td>2.64 K</td>
<td>18 K</td>
</tr>
<tr>
<td>HDSH (2 h)</td>
<td>0.32</td>
<td>329 K</td>
<td>2.07 K</td>
<td>24 K</td>
</tr>
<tr>
<td>HDSH (4 h)</td>
<td>0.49</td>
<td>328 K</td>
<td>1.57 K</td>
<td>40 K</td>
</tr>
</tbody>
</table>
Figure 3: Colors are available in online version. Adiabatic temperature changes, measured during heating and magnetization, for the \( \text{La(Fe,Si)}_{13}\text{H}_y \) samples produced by the conventional process (black line), HDSH with 2 h (red line) and 4 h (blue line) of hydrogenation.

4. Conclusions

In this paper, a comparison between the hydrogenation process of \( \text{La(Fe,Si)}_{13}\text{H}_y \) compound in: i) powder and ii) sintered porous samples was carried out, from which the following conclusions could be withdrawn. Firstly, XRD patterns have shown an increasing intensity of \( \alpha\text{–Fe} \) after the hydrogenation, in the case of samples prepared by HDSH. Thermal analysis revealed a gradient of H desorption ranges in the samples produced by HDSH. This gradient was also identified in the \( \Delta T_{ad} \) measurements, where the sample produced by HDSH with 4 h of hydrogenation showed a \( \Delta T_{ad} \) larger than 1 K in a range of 40 K. The maximum \( \Delta T_{ad} \) of the samples produced by the conventional process, HDSH with 2 h and 4 h of hydrogenation were 2.64, 2.07 and 1.57, respectively. 2D modelling showed that the effect of \( \alpha\text{–Fe} \) on the surrounding magnetocaloric material is negligible.
5. Acknowledgements

The authors would like to express their gratitude to Paulo V. Trevizoli and M. S. Capovilla for the fruitful discussions and measurements regarding the thermal and magnetic analyses.

References


