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1 **Assessment of biogas production from MBT waste under different operating**  
2 **conditions**

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11

12 **Abstract**

13 In this work, the influence of different operating conditions on the biogas production from mechanically-  
14 biologically treated (MBT) wastes is investigated. Specifically, different lab-scale anaerobic tests varying the  
15 water content (26-43 %w/w up to 75 %w/w), the temperature (from 20-25 °C up to 55 °C) and the amount  
16 of inoculum have been performed on waste samples collected from a full-scale Italian MBT plant. For each  
17 test, the gas generation yield and, where applicable, the first-order gas generation rates were determined.  
18 Nearly all tests were characterised by a quite long lag-phase. This result was mainly ascribed to the  
19 inhibition effects resulting from the high concentrations of volatile fatty acids (VFAs) and ammonia  
20 detected in the different stages of the experiments. Furthermore, water content was found as one of the  
21 key limiting factor of the anaerobic biological process. Indeed, the experimental results showed that when

22 the moisture was lower than 32 %w/w, the methanogenic microbial activity was completely inhibited. For  
23 the higher water content tested (75 %w/w), high values of accumulated gas volume (up to 150 NI/kgTS) and  
24 a relatively short time period to deplete the MBT waste gas generation capacity were observed. At these  
25 test conditions, the effects of temperature became evident, leading to gas generation rates of 0.007 d<sup>-1</sup> at  
26 room temperature that increased to 0.03-0.05 d<sup>-1</sup> at 37 °C and to 0.04-0.11 d<sup>-1</sup> at 55 °C. Overall, the  
27 obtained results highlighted that the operative conditions can drastically affect the gas production from  
28 MBT wastes. This suggests that particular caution should be paid when using the results of lab-scale tests  
29 for the evaluation of long-term behaviour expected in the field where the boundary conditions change  
30 continuously and vary significantly depending on the climate, the landfill operative management strategies  
31 in place (e.g. leachate recirculation, waste disposal methods), the hydraulic characteristics of disposed  
32 waste, the presence and type of temporary and final cover systems.

### 33 **KEYWORDS**

34 mechanically biologically treated waste; biochemical methane potential test (BMP); anaerobic tests; gas  
35 generation model; first-order kinetic constants

## 36 **1 Introduction**

37 Landfills still represent the dominant option for waste disposal in many parts of the world (Laner et al.  
38 2012). However, as known, this option may pose a threat to groundwater pollution, soil contamination and  
39 global warming effects due to the potential emission of leachate and landfill gas to the surrounding  
40 environment (Pantini et al., 2014; Scaglia et al., 2010; Thomsen et al., 2012; White and Beaven, 2013).  
41 Indeed, landfill has been recognized as one of the main source of anthropogenic methane emission and a  
42 significant contributor to global warming (Bogner et al., 2008). Gas emissions from landfills are mainly  
43 dominated by methane and carbon dioxide that are generated from the anaerobic conversion of organic  
44 matter contained in waste as a result of biological processes naturally occurring in landfill sites. Moreover,  
45 due to the generally high nitrogen content in wastes, there is also a considerable potential for nitrous oxide

46 emissions from municipal solid waste (MSW) landfills that can further enhance the global warming effects  
47 (Harborth et al., 2013).

48 In view of these concerns, throughout the world, new regulations in waste management and treatment  
49 strategies of municipal solid waste (MSW) have been introduced. For instance, in Europe, the Landfill  
50 European Directive 1999/31/EC imposes member states only landfill wastes that have been preliminary  
51 subjected to treatment or incineration. The directive aims at limiting the amount of biodegradable waste in  
52 landfills while encouraging alternative strategies in order to move towards more sustainable waste  
53 management system, according to the waste hierarchy approach (De Gioannis et al., 2009; Sormunen et al.,  
54 2008). To meet the European targets, member states have adopted different options, such as separate  
55 collection and recycling of organic waste stream, MSW incineration with energy recovery, biological  
56 treatments of source separated organic wastes or Mechanical Biological Treatment (MBT) plants of residual  
57 MSW (Lornage et al., 2007; Pantini et al., 2015; Scaglia et al., 2010). Among these, the MBT technology is  
58 playing a key role in the waste management system of unsorted MSW wastes (Adani et al., 2004; Farrell  
59 and Jones, 2009; Pantini et al., 2015; Siddiqui et al., 2013;). All over Europe, MBT facilities can apply  
60 different combinations of mechanical sorting, bio-drying, and biological processes depending on the  
61 specific target, that may be a pre-treatment before incineration or a pre-treatment to produce a bio-  
62 stabilized product that has a lower impact when disposed of in landfills (Adani et al., 2004; Di Maria et al.,  
63 2013; Farrell and Jones, 2009; Montejo et al., 2013). In the latter case, the MBT plant consists of a  
64 mechanical pre-processing stage including crushing, sieving and recovering of recyclable materials (such as  
65 metals, glass and plastics). This stage leads to two distinct flows: the oversize fraction, which is further  
66 processed to produce refuse-derived fuel, and the undersize fraction, rich in organic putrescible matter,  
67 which is biologically treated using an anaerobic/aerobic process in order to stabilize it. The main distinction  
68 between different MBT systems concerns the sequence of process steps and the type and duration of the  
69 biological treatment (Pan and Voulvoulis, 2007; Pantini et al., 2015). The specific technology and process  
70 applied may strongly affect the long-term behaviour of MBT wastes in landfills in terms of both liquid  
71 composition and gas generation (Boldrin et al., 2011; Siddiqui et al., 2013). However, gas emissions from  
72 MBT waste have been rarely measured on full scale MBT landfills (Harborth et al., 2013). Hence, the current

73 state of knowledge on biogas emissions is based either on laboratory tests or on large scale experiments  
74 such as lysimeters (Sormunen et al., 2008). Depending on the specific aim of the test, lab scale studies on  
75 gas emissions from MBT wastes and solid organic wastes are usually carried out using different procedures  
76 and operative conditions (see Table 1). As highlighted by Lornage et al. (2007), the differences in the  
77 experimental procedure adopted may modify the biogas yield and kinetics, thus leading to results that are  
78 not always comparable. The anaerobic process is indeed sensitive to several factors such as pH, water  
79 content, temperature, particle size, as well as by the presence of inhibitors such as of volatile fatty acids  
80 (VFAs), ammonia and heavy metals (Cabbai et al., 2013; Elbeshbishy et al., 2012; Labatut et al., 2011;  
81 Lornage et al., 2007; Raposo et al., 2011). Among these, pH is recognised as the key parameter to be  
82 maintained in an appropriate range (6.4-7.5) in order to enhance the methane yield (Adani et al., 2004;  
83 Argun et al., 2008; Lo et al., 2010). High pH values would result in increased toxicity due to the shift to  
84 higher concentrations of ammonia, which is identified as one of the most toxic agent for methanogenic  
85 bacteria (Chen et al., 2008; Vigneron et al., 2007). In contrast, low pH values are indicative of the  
86 accumulation of VFAs within the system (Bouallagui et al., 2005; Li et al., 2011). VFAs represent the main  
87 intermediate products during the initial acidogenic stage of the anaerobic process that are successively  
88 converted into methane and carbon dioxide. However, VFAs concentrations at high level may result in an  
89 inhibition of the methanogenic activity, as observed by several authors (Argun et al., 2008; Borzacconi et  
90 al., 1997; Cabbai et al., 2013). Regarding the other operative conditions, an increase of temperature has a  
91 positive effect on the microbial growth and activity (Chen et al., 2008) thus leading to a faster gas  
92 generation process. Similarly, increasing the water content of incubated waste is beneficial for methane  
93 yield since it enhances the solute transport of nutrient, the organic matter solubilisation and the  
94 microorganism mobilization within micro-environments, as well as dilutes the concentration of inhibitors  
95 (Donovan et al., 2010; Mora-Naranjo et al., 2004). Finally, the particle size of materials exerts a relevant  
96 influence on the process kinetic; it is well accepted that particle size reduction results in higher methane  
97 generation rate (Esposito et al., 2012; Lesteur et al., 2010, Mata-Alvarez et al., 2000), whereas its effect on  
98 biogas yield is still not completely elucidated (Mshandete et al., 2006; Nopharatana et al., 2007).

99

**Table 1**

100 The objective of this work was to evaluate the effects of temperature, water content and inoculum addition  
101 on biogas generation from mechanically-biologically treated waste by performing anaerobic batch tests at  
102 different operating conditions. Furthermore, in order to determine the potential gas generation capacity  
103 under optimal conditions, biomethane potential tests (BMP) were carried out. All these tests were then  
104 compared in terms of cumulative biogas yield and rates. Besides, where applicable, a first-order kinetic  
105 model was used to compute the biogas rate constants from the cumulative gas generation curves observed  
106 in each experiment. Finally, the obtained results were addressed to assess the possible implications  
107 resulting from the different environmental conditions expected in the field.

## 108 **2 Materials and Methods**

### 109 **2.1 MBT waste material**

110 Mechanically-biologically treated waste samples were collected at the belt discharge point of the secondary  
111 refinement unit of a full-scale MBT plant operating in Italy. This MBT plant receives residual municipal solid  
112 waste (226,000 ton/y in 2013), with the average composition shown in Table 2.

#### 113 **Table 2**

114 In this plant the incoming wastes are subjected to a mechanical pre-processing consisting of pre-sorting of  
115 bulky materials, shredding and size separation. From these processes two flows are obtained: the light  
116 fraction with particle size >80 mm (96,500 ton/y in 2013), which is further processed to produce refuse  
117 derived fuel (55,500 ton/y in 2013), and the undersize fraction (126,000 ton/y in 2013), which is sent to the  
118 biostabilization basins; the remaining flow is represented by recovered metals (3500 ton/y in 2013). In the  
119 biostabilization basins, the aerobic process occurs for 28 days at forced ventilation condition, with daily  
120 water addition and waste turning. Then, the stabilized output goes to a secondary sieving process to  
121 remove improper materials. The undersize fraction (<20 mm) is the organic MBT waste analysed in this  
122 study. The MBT waste sample was collected in May 2014 using standard procedures (UNI 10802:2013). A  
123 final MBT waste sample of about 80 kg was sent to the laboratory and stored at 4 °C for few days until the  
124 physico-chemical analyses were performed. In the laboratory, three representative sub-samples were

125 obtained by the “coning and quartering” method. One sub-sample was analysed to determine the moisture  
126 content (W), total (TS) and volatile solids (VS), pH, total (TC) and organic carbon (TOC) content, total  
127 Kjeldahl (TKN) and soluble nitrogen (NH<sub>4</sub>-N), the Chemical Oxygen Demand (COD) and the water content at  
128 field capacity (FC) of waste (i.e. the water-retaining capacity including both the hygroscopic and capillary  
129 water). All measurements were performed at least in triplicate; initial waste water content, dry matter and  
130 waste field capacity are expressed as percentage of wet weight whereas the other parameters are  
131 computed on dry weight basis. Average values and standard deviations are reported in Table 2. A sub-  
132 sample (3-4 kg) was used for BMP tests. Before to perform the BMP analysis, the sub-sample was dried at  
133 room temperature (25 °C) to avoid losses of volatile organic compounds and then shredded to 1 mm  
134 particle size. The last sub-sample was used in the incubation tests as received since, currently, the MBT  
135 waste is not subjected to further treatment before landfilling it.

## 136 **2.2 Analytical Methods**

137 In order to characterize the MBT waste with regards to its physical and chemical properties, different  
138 analytical methods were applied. These tests provide basic information that are essential for the  
139 interpretation of the biological test results.

### 140 *Moisture content (W), total (TS) and volatile solids (VS), total carbon (TC) and organic carbon (TOC)* 141 *content, pH*

142 Moisture (W) and total solid (TS) content were determined according to the standard method UNI EN  
143 14346 (2006). Volatile solids (VS) were measured by loss-on-ignition (LOI) at 550 °C for 8 h (UNI/TS 11184,  
144 2006). Total carbon (TC) and organic carbon content (TOC) were analysed by Shimadzu SSM-5000A  
145 instrument according to UNI EN 13137( 2001). The own pH of MBT waste was determined after elution  
146 following the standard method UNI EN 12457-2 (2004).

### 147 *Total (TKN) and soluble nitrogen (NH<sub>4</sub>-N)*

148 Total Kjeldahl Nitrogen (TKN) was measured on solid waste samples (2.5-3 g) by mineralization with a  
149 strong acid medium (97% sulphuric acid) followed by steam distillation and titrimetric determination, as

150 proposed by Mohajer et al. (2010) and Tremier et al. (2005). Blank and control tests were performed  
151 simultaneously, in triplicate. In blank tests, 4 ml of deionized water were used whereas in control tests, to  
152 evaluate the efficiency of ammonia recovery, 4 ml of L-glutamic acid (1000 mg/l) were utilized. Samples  
153 were digested in FOSS 2020 Digester at 180 °C for 1 h and thereafter at 350 °C for 1-2 hours (warm-up time  
154 excluded). After cooling, samples were distilled using FOSS Kjeltac 8100 distillation unit. In the distillation  
155 method, 30 ml of deionized water and 70 ml of the alkaline solution (32 %w/w NaOH) were added to each  
156 tube. The steam supply was set to 60% and the distillation time was 5 minutes. A solution consisting of 50  
157 ml deionized water, 4 ml boric acid (40 g/l) and 3 drops of Kjeldhal indicator (mixture of methyl red  
158 indicator and Bromocresol green indicator, MERCK KGaA) was used as absorbent solution during  
159 distillation. The ammonia content was determined by titration of distillate with 0.1 M H<sub>2</sub>SO<sub>4</sub>.  
160 Determination analyses of soluble nitrogen (NH<sub>4</sub>-N) were carried out on 2.5-3 g using the same procedure  
161 of TKN. For the distillation method, 30 ml of deionized water and 50 ml of the alkaline solution (32 %w/w  
162 NaOH) were added to each tube. Steam supply and distillation time were the same as mentioned above, as  
163 well as the titration method. The ammonia recovery of the instrument was evaluated by adding 4 ml of a  
164 known solution (1000 mg/l NH<sub>4</sub>-N) to 50 ml of deionized water. An efficiency up to 100% was detected.

#### 165 *Chemical Oxygen Demand (COD)*

166 To determine the COD of wet solid samples, the modified method proposed by Raposo et al. (2008) was  
167 adopted. This method consists of a wet oxidation with potassium dichromate as the oxidant and silver  
168 sulphate as the catalyst in a strong sulphuric acid solution (Raposo et al., 2008). COD measurements were  
169 carried out on 1.0 g of MBT waste sample, adding 6 ml of 97% sulphuric acid and 30 ml of deionized water  
170 to the flask while stirring it for 30 minutes. Then, 2.0 ml of Potassium Dichromate 0.025 M (for high range  
171 detection) and 4.5 ml of silver sulphate sulphuric acid solution were added to each flask containing 3.5 ml  
172 of initial solution. The reaction mixtures were boiled in a Holm & Halby Techne Dri Block at 148 ± 2 °C for  
173 110 minutes. After cooling, 5.0 ml of deionized water and 3 drops of ferroin indicator were added and  
174 samples were titrated with 0.035 M Ferrous Ammonium Sulphate solution. COD measurements were



175 performed in triplicate. Five blanks (3.5 ml of deionized water) and three control tests (3.5 ml of 500 mg  
176 COD/l standard solution) were carried out simultaneously.

### 177 *Volatile Fatty Acids (VFAs)*

178 VFAs were measured in fresh solid waste as well as in waste samples at different incubation time. Samples  
179 were prepared weighing about 5 g of MBT waste, adding 12.5 ml of deionized water and acidifying them  
180 with 0.4 ml of 97% sulphuric acid to ensure  $\text{pH} < 2$ . A magnet was inserted and samples stirred for  
181 approximately 10-15 minutes to homogenize them. Then, 1.5 ml of each sample was placed in an  
182 Eppendorf tube and centrifuged at Eppendorf mini spin table centrifuge at 13,400 rpm for 10 min. After  
183 centrifugation, 1.0 ml sample was transferred to a GC glass vial and 0.1 ml of internal standard (2.2 mM 4-  
184 Methyl valeric acid) was added. Concentrations of acetate, propionate, iso-butyrate, butyrate, iso-valerate,  
185 valerate, hexanoic acids were determined by using GC Shimadzu GC – 2010 equipped with and FID (flame-  
186 ionization-detector). VFA compounds were separated by a capillary column (ZB – FFAP, 30 m, 0,53 mm I.D x  
187 1,0  $\mu\text{m}$ ) and concentrations were computed by means of a linear calibration curve obtained after standards  
188 injection (range: 5-1500 mg/l). All measurements were performed in triplicate.

### 189 *MBT waste field capacity (FC)*

190 Water content of MBT waste at field capacity was determined by performing column test. A Plexiglas  
191 column with an inner diameter of 3.5 cm and a total height of 15 cm was packed with about 70 g of as  
192 received MBT waste (water content at 19.4% w/w), corresponding to initial wet density of  $0.5 \text{ g/cm}^3$  (dry  
193 bulk density of  $0.4 \text{ g/cm}^3$ ). The packed column was weighted ( $M_{\text{in}}$ ) and then saturated from the bottom  
194 section until a water head of few millimeters formed at the top and the pump was stopped. After  
195 saturation, the column was let drain until no significant outgoing flow was detected and weighted again  
196 ( $M_{\text{end}}$ ). The difference in weight ( $M_{\text{end}} - M_{\text{in}}$ ) is the adsorbed water ( $M_{\text{w,ads}}$ ). The ratio between the total  
197 water in the column at the end of the experiment (i.e. sum of adsorbed water and initial moisture water,  
198  $W$ ) and the final mass of MBT material in the column ( $M_{\text{in,MBT}} + M_{\text{w,ads}}$ ), shown in Eq. (1), represents a rough  
199 estimation of the field capacity of MBT waste (expressed as percentage of wet weight):

200 
$$FC(\% w / w) = \left( \frac{M_{w,ads} + W(\%) \cdot M_{in,MBT}}{M_{in,MBT} + M_{w,ads}} \right) * 100 \quad (1)$$

201 **2.3 Gas production tests**

202 ***Biochemical Methane Potential test (BMP)***

203 Different BMP assays, experimental set-ups and employed protocols can be found in literature due to a lack  
204 of harmonization and standardization of biochemical methane potential methods. Indeed, some methods  
205 are designed to evaluate the biodegradability of chemical substances under methanogenic conditions (ISO  
206 14853, 1999; ASTM E2170-01, 2008 (withdrawn 2013); ASTM D5210-92, 2007; DIN 38414-8, 1985) while  
207 others aim at quantifying the ultimate biodegradability and gas generation of complex organic substrates  
208 (ISO 11734, 1995; ISO/DIS 14853, 1999) using different experimental set-ups. Additionally, these methods  
209 were applied differently or modified by researchers (Angelidaki et al., 2009), making the inter-comparison  
210 of BMP test results quite difficult. In this study, the BMP protocol proposed by Hansen et al. (2004) was  
211 adopted. Glass bottles (1 l) with a thick rubber septum were used as reactors. Approximately 1 g of air  
212 dried waste sample (particle size < 1 mm), 80 ml of deionized water and 320 ml of a fresh de-gassed  
213 inoculum were used in the experiments in order to achieve an organic load of 1.4 gVS/l (weight of VS in  
214 substrate per unit volume of inoculum). Tests were carried out for 30 days with six replicates, due to the  
215 relatively high heterogeneity of the MBT material. Thermophilically digested material from a full-scale  
216 biogas plant was used as inoculum. Three blanks with only water and inoculum were run to test the biogas  
217 production from the inoculum itself. Control tests, containing 0.8 g of AVICEL (Fluka, Sigma Aldrich,  
218 Vallensbæk Strand, Denmark) as a standard substrate, were performed to check the quality of the  
219 inoculum. After set-up, the reactors were flushed with N<sub>2</sub> for 10 minutes, to ensure the establishment of  
220 anaerobic conditions in the headspace of the glass bottles, then sealed and placed in the incubator at 55 °C  
221 (± 1 °C). The methane concentration in the reactors was measured every two days during the first two  
222 weeks, and later once per week. Gas samples (0.2 – 0.5 ml) were taken from the headspace of the reactors  
223 by using a syringe with a pressure lock and directly injected into the gas chromatograph for methane  
224 determination (Shimadzu GC 14A) and for qualitative analysis of gas composition in terms of %CH<sub>4</sub> and

225 %CO<sub>2</sub> (Mikrolab GC Aarhus ). In order to avoid build-up of high pressure inside the reactors, the gas was  
226 released during the experiment. Based on the difference of CH<sub>4</sub> concentration before and after release of  
227 excess gas, the generated amount of CH<sub>4</sub> was computed.

### 228 *Anaerobic gas generation tests*

229 In order to evaluate the effects of temperature and water content on the gas generation rate and yield,  
230 anaerobic batch tests were performed at three different temperatures: room temperature (20-25 °C), 37 °C  
231 and 55 °C. In tests at room temperature and 37 °C, a mesophilic inoculum derived from a biogas plant  
232 carrying out mesophilic co-digestion of manure and organic waste was utilized whereas in tests at 55 °C the  
233 same inoculum of BMP tests (thermophilically digested material) was used.

234 Before starting the anaerobic tests, different amounts of water were added to four MBT sub-samples (as  
235 received material with 19.4% water content) in order to achieve four values of the initial water content in  
236 waste: 26%, 32%, 38%, 43% (expressed on wet weight basis). After homogenizing the samples, an aliquot of  
237 each sub-sample at the specific water content was weighted and introduced into reactor and then  
238 incubated at the corresponding temperature. The values set for temperature and water content aimed at  
239 covering the actual ranges generally observed at real scale landfill sites (Mor et al., 2006; Mora-Naranjo et  
240 al., 2004). Note that the operating conditions of anaerobic tests were selected in order to simulate  
241 different disposal scenarios and, hence, they differ from the optimal ranges usually set in biogas plant  
242 treating MSW waste organic fractions. Namely, in these plants, the anaerobic digestion of incoming  
243 feedstock, usually mixed with a large amount of digester effluent or sewage sludge, may be carried out in  
244 dry systems (60-80% water content) or in wet systems (water content >90%) at thermophilic or mesophilic  
245 conditions in continuous or static digesters (Braber 1995; Gunaseelan 1997; Schievano et al., 2010).

246 Incubation tests at 37 °C and 55 °C consisted of 1 l glass bottle filled with waste sample (0.5-0.7 kg), sealed  
247 with a rubber septum and equipped with a PVC pipe, which connected it to a 3 l SKC Tedlar Sampling Bag  
248 (SKC Inc., Eighty Four, PA, US) for gas collection. At each measurement, 5 ml of gas were sampled with a  
249 syringe and injected into evacuated glass vials fitted with pierceable rubber septa (Exetainer Vail, Labco Ltd,  
250 Lampeter, UK), which were then analysed for determining gas composition. A 490-PRO Micro GC (Agilent

251 Technologies Denmark Aps, Glostrup, Denmark) equipped with two columns (PoraPLOT Q PLOT, 0.25mm,  
252 10m, and Molecular Sieve 5A PLOT, 0.25 mm, 20m) was used to measure CH<sub>4</sub>, CO<sub>2</sub> and O<sub>2</sub> in gas samples  
253 with a detection limit of 0.1% for all gases. In the incubation tests at room temperature, 12 l steel drums  
254 with airtight lids were used as reactors. Drums were filled with MBT waste samples at three different  
255 water contents (26%, 34%, 43%) and flushed with nitrogen for 30 minutes before sealing them. Lids were  
256 equipped with T-shaped sampling ports and connected to 5 l SKC Tedlar Sampling Bags. More information  
257 about tests conditions and experimental activities are reported in Table 3.

258 As shown in this table, the anaerobic gas generation tests were carried out in three sequential stages. In  
259 the first stage, no inoculum was used. During the second stage of experimental activity, a low amount of  
260 mesophilic or thermophilic inoculum (20 g of inoculum, i.e. approximately 5 %w/w of the waste dry matter  
261 used in the test) was introduced into the incubation bottles with lower water contents (T1, T2, T5, T6) to  
262 enhance the microbial activity. In the following stage III, due to the unexpected very low biogas production,  
263 some reactors were opened (R2, T2A, T3A/B, T4B, T6A, T7A/B) in order to partially remove the material,  
264 which was then analysed with regard to pH, VFA, TKN and ammonia content. In this stage, inoculum (30  
265 %w/w of waste dry matter in reactor) and water (230 %w/w of waste dry matter, to obtain a final moisture  
266 content of 75 %w/w) were introduced within these reactors that were successively purged with nitrogen  
267 and incubated again. A triplicate measurement of the biogas production from the added inoculum was  
268 performed on blank experiments and deduced from the biogas yield of waste samples.

269 **Table 3**

270 The gas volume produced by each reactor was computed timing the emptying of gas bags using a Fluid  
271 Metering Inc. laboratory pump (QG, Fluid Metering Inc., Syosset, NY, US). The flow rate of the pump was  
272 tested several times during the experiment and an average flow of 0.5 l/min was measured.

## 273 **2.4 First-order gas generation model**

274 To compute the biogas kinetic constants under different operating conditions, the widely adopted first-  
275 order kinetic model (Gunaseelan 1997; De Gioannis et al., 2009; Lo et al., 2010; Mou et al., 2015) was used

276 for the interpolation of experimental data. The generic formulation, which accounts for the lag-time  
277 observed in the test, is the following:

$$278 \quad L(t) = L_0 \left[ 1 - \exp(-k \cdot (t - t_{lag})) \right] \quad (2)$$

279 where  $L$  is the biogas accumulation (NI/kgTS) at the time  $t$  (d),  $L_0$  the potential biogas production (NI/kgTS)  
280 for the tested conditions (at optimal conditions,  $L_0$  approaches the potential gas generation capacity  
281 measured in the BMP experiments),  $t$  the time over the digestion period,  $t_{lag}$  the lag-phase (d) and  $k$  the  
282 first-order kinetic constant ( $d^{-1}$ ).

## 283 **3 Results and discussions**

### 284 **3.1 MBT waste characterization**

285 Results of the characterization analysis performed on the MBT waste are reported in Table 4. Moisture  
286 content (W), as well as water field capacity, were slightly lower than the values usually measured for this  
287 type of waste (Di Lonardo et al., 2014; Pantini et al., 2015; Zach et al., 2000). Despite the waste underwent  
288 an aerobic treatment process in the MBT plant, the organic matter of waste is still quite high, as confirmed  
289 by VS, TOC and COD contents. As shown in Table 4, the pH was almost neutral and in the optimal range for  
290 the anaerobic process. It is also interesting to point out that values of TKN and  $NH_4$ -N were quite high, close  
291 to the range usually observed for untreated waste or poorly treated waste (Modin 2007; Pognani et al.,  
292 2010).

293 **Table 4**

### 294 **3.2 Biochemical Methane Potentials**

295 Figure 1 shows the cumulative  $CH_4$  generation curve measured in control (red dots) and in MBT waste tests  
296 (green dots) as a function of the incubation time. The results reported in this figure, expressed as  
297 cumulative volume of methane per gTS at standard temperature and pressure (STP) conditions (0 °C, 1  
298 atm), represent the average values measured in the different replicates obtained after subtracting the  $CH_4$   
299 measured in the blank experiments. The solid red line and the dotted green line depict the theoretical

300 methane potential for the cellulose substrate (control) and the MBT waste, respectively. The latter was  
 301 estimated from the total organic carbon (TOC) of the substrate (Table 4), as shown in Eq. (3). Assuming that  
 302 the biogas generated by a complete degradation of organic carbon contains 60% of methane (that  
 303 corresponds to the average CH<sub>4</sub> concentration measured during the BMP test), a theoretical value of 268  
 304 NmlCH<sub>4</sub>/gTS was computed (i.e. 60% of CH<sub>4</sub> in biogas implies that 1 g of TOC generates 1.12 NI of CH<sub>4</sub> at  
 305 STP).

$$306 \quad L_{theoretical,TOC} \left( \frac{NmlCH_4}{g_{TS}} \right) = 60\% CH_4 \cdot \frac{1}{12 \frac{g_C}{mol_C}} \cdot 22.414 \left( \frac{NI}{mol} \right) \cdot 0.239 \frac{g_{TOC}}{g_{TS}} \cdot 1000 \quad (3)$$

307 From Figure 1, it can be noticed that the cumulative CH<sub>4</sub> curves observed for control and waste samples  
 308 showed a rapid increase in the first two weeks and then achieved an asymptotic value. The lag phase was  
 309 absent, confirming that the BMP test was run under optimal conditions. The average cumulative methane  
 310 generation in controls resulted in an average gas generation of 367 NmlCH<sub>4</sub>/gTS on the 13<sup>th</sup> day after  
 311 starting the batch tests and reached the theoretical potential value of 415 NmlCH<sub>4</sub>/gTS in 27 days. Within  
 312 the first two weeks, the methane generation curve observed in control tests appeared linear (R<sup>2</sup>=0.99),  
 313 with an average slope of 27.5 NmlCH<sub>4</sub>/(gTS·d). Similarly, in the first stages of the MBT waste sample test a  
 314 linear methane production rate of 8.4 NmlCH<sub>4</sub>/(gTS·d) was observed. Thereafter, the slope rapidly  
 315 decreased and the cumulative CH<sub>4</sub> generation curve asymptotically approached a constant level of 121  
 316 NmlCH<sub>4</sub>/gTS. Hence, it seems that, on average, only 45% of the theoretical methane generation (i.e. 268  
 317 NmlCH<sub>4</sub>/gTS) was achieved during the BMP experiment of the MBT waste sample, most likely due to the  
 318 presence of non-biodegradable fractions (plastics), recalcitrant organic substances or lower degradable  
 319 compounds. Moreover, it should be noted that the theoretical methane (Eq. 3) was computed neglecting  
 320 the biomass synthesis and, thus, it could be overestimated. Some authors indicate that 5-10 % of organic  
 321 matter is consumed by bacteria growth (Angelidaki and Sanders, 2004; Elbeshbishy et al., 2012; Labatut et  
 322 al., 2011; Raposo et al., 2011) and, hence, does not contribute to CH<sub>4</sub> production. Even though the extent  
 323 of degradation achieved in a BMP test is strongly dependent on the composition of the analysed substrate  
 324 as well as on the test methodology applied (substrate to inoculum ratio, test duration, inoculum

325 characteristics), the 45% degradability estimated for the analysed MBT waste based on TOC content is  
326 consistent with other previous studies on similar MBT materials. For instance, using experimental results  
327 (TOC, BMP, % CH<sub>4</sub> in biogas) presented by Bayard et al. (2010) for different MBT wastes, the degradability  
328 of MBT waste was estimated within the range 10% - 36% in 90-days BMP tests. An organic carbon  
329 degradability of 42.4% was computed from results reported in Barrena et al. (2008) for aerobically treated  
330 MBT waste.

331 Making reference to Figure 1, some variation of accumulated CH<sub>4</sub> volume was observed between the  
332 replicate of MBT waste tests (coefficient of variation in the range of 12.4-23.3 %), probably due to its  
333 relatively high heterogeneity and to the low amount of material tested in the BMP experiment. On the  
334 contrary, both substrate (control) and inoculum (blank) showed a good internal homogeneity with  
335 coefficients of variation in the range of 0.5-7.0% and 1.9-4.9%, respectively.

336 From the qualitative analyses of gas composition, an average value of  $60.8 \pm 1.1$  % and  $62.4 \pm 1.4$  % of  
337 methane concentration in biogas was detected for sample and control, respectively. Assuming this  
338 percentage, a potential CO<sub>2</sub> yield of  $78 \pm 25$  Nm<sup>3</sup>CO<sub>2</sub>/gTS for MBT waste was computed. Hence, a maximum  
339 potential gas production of  $199.2 \pm 63$  Nm<sup>3</sup>/gTS was estimated for the analysed MBT waste. This result is  
340 consistent with some previous BMP studies performed on aerobically treated MBT wastes. For instance,  
341 Barrena et al. (2008) measured a total gas production of  $187 \pm 16$  NI/kgTS, with an average methane content  
342 of 57 %v/v, from MBT wastes after 32 days of aerobic treatment. Bayard et al. (2010) analysed the gas  
343 generation potential of different flows in a French MBT plant; they observed that the intermediate fraction  
344 (< 50 mm), after 6 weeks forced-aerobic treatment, still exhibited high gas generation potential ( $232 \pm 23$   
345 NI/kgTS). Lornage et al. (2007) measured a gas potential of about 160 NI/kgTS from MBT wastes subjected  
346 to 4-weeks aerobic treatment process.

347

**Fig. 1**

### 348 **3.3 Anaerobic gas generation tests**

349 Figs. 2-4 show the cumulative generation curve of methane (red circles) and carbon dioxide (black squares  
350 dots) obtained for the MBT waste samples at room temperature (Fig. 2), 37 °C (Fig. 3) and 55 °C (Fig. 4), and

351 for different initial water contents of waste, as a function of the incubation time. Results are expressed as  
352 cumulative volume of gas per kg of total solids (TS) at STP. For comparison purposes only, the potential CH<sub>4</sub>  
353 and CO<sub>2</sub> values computed in BMP test have also been reported in Figs. 2-4 as dotted lines. From Figs. 2-4 it  
354 can be noticed that in the first stage of experimental activities that were carried out at low water contents  
355 and without inoculum addition, a long lag-phase was detected for all operating conditions. During this  
356 stage, the microbial population needed to get adapted to the micro-environment and to be acclimatized to  
357 the organic substrate in order to be able to grow until a sufficient active population established and the  
358 anaerobic degradation could stably evolve. Results shown in Figs. 2-4 suggested that the duration of the  
359 lag-phase was strongly affected by both the water content of waste and the process temperature.  
360 Regarding the former parameter, it widely documented that water enhances nutrients and substrates  
361 solubilisation in the liquid phase as well as supports bacteria movement and facilitates substrate and  
362 products diffusion through the porous medium (Donovan et al., 2010; Khalid et al., 2011; Liotta et al.,  
363 2014). However, the water content of waste in a landfill disposal scenario could be quite far from the  
364 optimum value for degradation (60% -90%) and, thus, may become limiting for the anaerobic process due  
365 to the accumulation of inhibitors with adverse effects on bacteria population (Donovan et al., 2010).  
366 Indeed, experimental results confirmed that the water content of waste could be considered as one of the  
367 most important factors limiting methane generation and, the probability of achieving a stable  
368 methanogenic stage is significantly reduced for water contents below 32 %w/w (wet weight) at any  
369 temperature for this type of waste.

370 **Fig. 2**

371

372 **Fig. 3**

373

374 **Fig. 4**

375

376 Increasing the operative temperature would enhance both the substrate solubilisation and the microbial  
377 activity (Raposo et al., 2011). It is likely that higher temperature allowed moving from non-equilibrium state



378 towards more suitable conditions for methanogenic bacteria growth, thus favouring a more rapid  
379 establishment of the methanogenic phase (Lesteur et al., 2010; Li et al., 2011; Mata-Alvarez et al., 2000).  
380 Indeed, as shown in Fig. 2 in all the experiments at room temperature methane was not detected within  
381 the first three months but only CO<sub>2</sub> was generated at high levels (80 %v/v, see Fig. S1 in the Supplementary  
382 information). A similar behaviour was observed by Adani et al. (2004) during 90 days incubation tests  
383 carried out on fresh and partially treated wastes (10 days of aerobic treatment). The high CO<sub>2</sub>  
384 concentration without CH<sub>4</sub> generation suggested that the biological process was completely inhibited at  
385 every water content in tests at room temperature, probably due to acidification effects, and revealed the  
386 poor stability degree of the analysed MBT waste. Furthermore, even though the methanogenic activity was  
387 observed in tests at higher temperature and water content, a clear instability associated with the anaerobic  
388 process was still detected.

389 In fact, as reported in Fig. 3, reactors at 37 °C and water content of 38% (T3) - 43% (T4) started producing  
390 methane after 27 days, even though CH<sub>4</sub> concentrations were low (below 20%). A similar trend was  
391 observed for batch tests at 55 °C (Fig. 4) where methane generation started after 22 days and 13 days for  
392 tests at moisture content of 38% (T7) and 43% (T8), respectively. However, in all these reactors (except T8)  
393 just after few days, biogas generation slowed down. Only the experiment at higher water content and  
394 temperature (T8) managed to reach the stable methanogenic phase during the experiment (without  
395 inoculum addition), as confirmed by the CH<sub>4</sub> concentration measured in biogas, which was in the range 45-  
396 60 %v/v (see Fig. S3 in the Supplementary information). Even if a lag phase of 40 days was observed in  
397 these tests, the cumulative gas generation approached an asymptotic value of 29.1±2.2 NI CH<sub>4</sub>/kgTS and  
398 32.6±2.1 NI CO<sub>2</sub>/kgTS within 100 days (Fig. 4). These values are in line with previous studies on treated MBT  
399 waste wetted to water holding capacity in which no inoculum was added (Adani et al., 2004; De Gioannis et  
400 al., 2009). However, compared to the measured BMP value, in this test condition, a very low conversion  
401 degree was achieved (30% of BMP value).

402 Overall, the results obtained during the initial experimental activity indicated that, in most of test  
403 conditions, the anaerobic process was slowed down either due to high levels of inhibiting factors or to a  
404 limited amount of active biomass inside the reactors. Hence, during stage II, a small amount of inoculum

405 (20 g) was introduced into reactors at lower water contents (T1, T2, T5, T6) and the evolution of  
406 degradation was monitored for 20 days. After the inoculum addition, methane started to be produced in all  
407 reactors but its concentration still remained very low (below 10 %v/v) and then decreased again (see Tests  
408 T2B and T6B of Fig. S2 in the Supplementary information). Thus, it seems that the microorganism  
409 population inside the MBT waste was not sufficient to sustain the anaerobic degradation process. For that  
410 reason, during stage III, in reactor T4B (W=43%, T=37 °C) a greater amount of inoculum was introduced (70  
411 g) and, in two weeks, methane concentration achieved the range typical of a stable methanogenic phase  
412 (50-70 %v/v, see Fig. S2 in the Supplementary information).

413 Results shown in Figs. 3-4 also highlight the different behaviour of tests with water content at field capacity  
414 (w=43% at 37 °C and 55 °C, see T4 vs. T8). In fact, even if the water content was the same in these reactors,  
415 only tests at 55 °C (see T8 in Fig. 4) managed to achieve the stable methanogenic phase without the  
416 inoculum addition. Instead, at 37 °C, a very low methane volume was measured in test where no inoculum  
417 was used (see T4A in Fig. 3), whereas methane was stably produced only after the addition of a significant  
418 amount (70 ml) of mesophilic inoculum (T4B). This may be ascribed, on the one hand, to a lower active  
419 mass of mesophilic bacteria in the MBT waste compared to the thermophilic ones, presumably due to the  
420 type of biological process performed in the MBT facility. Indeed, temperatures up to 70 °C were achieved in  
421 the biostabilization basin of the MBT plant during the aerobic treatment. This sanitation process may have  
422 significantly reduced the microorganism population inside the waste mass, especially the mesophilic  
423 bacteria, which are more sensitive to high temperatures than the thermophilic ones. On the other hand,  
424 the lower gas generation measured in test T4A compared to T8 may be explained considering that the  
425 methanogenic mesophilic bacteria could be more vulnerable to unfavourable environmental conditions  
426 (higher toxic effects exerted by VFA and ammonia) and have lower growth rates (van Lier et al., 1997;  
427 Amani et al., 2011) compared to the thermophilic bacteria, that implies the adapting period could last  
428 longer. In conclusion, results obtained during stage I and II suggested that the capability of the MBT waste  
429 to generate methane is drastically limited due to inhibition effects, which are emphasized at lower water  
430 contents, also because the initial bacteria population inside the waste mass could not contain a sufficient  
431 level of methanogens to sustain the anaerobic process under the specific test conditions (high organic

432 load). For a better understanding of these results, during stage III of the experimental activity, reactors  
433 were opened and waste was partially removed in order to measure pH, VFA and  $\text{NH}_4\text{-N}$ . Then, inoculum  
434 (30% of final TS) and water (up to a final moisture content of 75 %w/w) were added in reactors R2, T2A, T3,  
435 T6A, T7 and incubated again. The beneficial effects of water and inoculum supply were evident, resulting in  
436 an immediate growth of the biogas yield, with increasing gas generation rates at higher temperature. On  
437 the one hand, the supplemental water addition may have reduced the inhibitory effect by diluting potential  
438 toxic substances such as heavy metals (copper, chromium or zinc), ammonia and VFAs. (Chen et al., 2008;  
439 Yenigun and Demirel, 2013; Poggi-Varaldo et al., 1997). On the other hand, a proper balance between  
440 acidogens and methanogens could have been achieved by lowering the organic load (through waste  
441 removal) and increasing the active bacteria mass within the anaerobic reactors (through inoculum  
442 addition). Indeed, tests at 55 °C achieved the asymptotical value of  $66.7 \pm 6.3$  NI  $\text{CH}_4/\text{kgTS}$  and  $50.0 \pm 2.3$  NI  
443  $\text{CO}_2/\text{kgTS}$  within 30-40 days after the inoculum and water addition. In tests performed at 37 °C, a  
444 cumulative volume of  $73.1 \pm 2.1$  NI  $\text{CH}_4/\text{kgTS}$  and  $54.9 \pm 1.8$  NI  $\text{CO}_2/\text{kgTS}$  was measured after 60-70 days  
445 from inoculum and water supply. These results showed that the biodegradability of the MBT waste in terms  
446 of methane and carbon dioxide yields did not vary significantly between 37 °C and 55 °C when moisture  
447 conditions were not limiting, as also observed by other authors (Hejnfelt and Angelidaki, 2009; Liu et al.,  
448 2009; Veeken and Hamelers 1999). In fact, in both cases, the total biogas yield at the end of these tests was  
449 in the range of 55–60 % of the methane potential value (66.7 and 73.1 against 121 NI  $\text{CH}_4/\text{kgTS}$ ). However,  
450 from the results obtained at room temperature, it is evident that the temperature surely affects the gas  
451 generation rate but also seems to influence the gas generation capacity. Indeed, in test with water and  
452 inoculum addition (see R2 test in Fig. 2), the gas production achieved a value of  $8.5 \pm 1.0$  NI  $\text{CH}_4/\text{kgTS}$  and  
453  $19.3 \pm 0.3$  NI  $\text{CO}_2/\text{kgTS}$  after 70 days from inoculum addition but was still increasing, indicating that the  
454 stable methanogenic phase has not been reached yet (see Fig. S1 in the Supplementary information).

### 455 **3.4 Estimation of biogas kinetic constants**

456 Fig. 5 shows, for the tests that achieved the stable methanogenic phase, the cumulative biogas production  
457 simulated with the first-order kinetic model (lines) fitted to the measured data (dots). The best-fit

458 parameters used in the model are reported in Table 5. Making reference to Fig. 5, it can be noticed that, in  
459 all cases, the first-order kinetic model accurately reproduces the different shapes of accumulated gas  
460 volume curves, as confirmed by the  $R^2$  values reported in Table 5. Modelling the experimental results  
461 revealed that the biodegradability of the MBT waste, which is expressed as ratio of cumulative gas volume  
462 to potential gas ( $\%L_0/\text{BMP}$ ), ranged between 56% to 75% in tests at 75 %w/w water content and decreased  
463 to 34% in tests with water content at field capacity and without inoculum addition. This range is slightly  
464 lower than the typical values for solid state incubation tests of MBT residues, presumably due to the poor  
465 stability degree of the MBT waste analysed in this study. For example, Binner and Zach (1999) found that  
466 the gas generated within 90 days was about 75% to 90% of the potential gas generation capacity (e.g. gas  
467 volume measured after 240 days) for well treated wastes (duration of pre-treatment >10weeks).

#### 468 Fig. 5

469 As already discussed above, higher operative temperature leads to a faster gas generation since  
470 temperature enhances both microbial growth and activity (Bouallagui et al., 2005; Gavala et al., 2003; Kim  
471 et al., 2002). Specifically, k-values vary from  $0.007 \text{ d}^{-1}$  at room temperature,  $0.03\text{-}0.05 \text{ d}^{-1}$  at  $37 \text{ }^\circ\text{C}$  and  $0.04\text{-}$   
472  $0.11 \text{ d}^{-1}$  at  $55 \text{ }^\circ\text{C}$ . Moreover, a linear correlation of k-values with the operative temperature was observed  
473 for tests at 75 %w/w water content.

474 These k-values were also used to estimate the time required to reach the 99% of the maximum biogas  
475 generation  $L_0$ , as follows:

$$476 T_{99\%} = \frac{\ln(1-0.99)}{k} \quad (4)$$

477 Due to quite high k-values, a relatively short time period  $T_{99\%}$  (Table 5), ranging from few months up to 2  
478 years, was computed.

#### 479 Table 5

480 Table 6 reports a brief literature review of the kinetic constants and gas yields experimentally determined  
481 for different types of organic substrates. As shown in Table 6, these parameters vary substantially between  
482 different substrates, experimental procedures and tests conditions. Even if a direct comparison is not  
483 possible, the k-values obtained in this work are in line with most of these studies. For instance, the k-value

484 of  $0.007 \text{ d}^{-1}$  determined at  $T=20-25 \text{ }^{\circ}\text{C}$  and water content of 75 %w/w, is consistent with the results  
485 obtained by Vavilin et al. (2004) for MSW waste at 65% of water content and  $T=30 \text{ }^{\circ}\text{C}$  ( $k=0.007-0.08 \text{ d}^{-1}$ ).  
486 Similarly, the  $k$ -values range ( $0.028-0.054 \text{ d}^{-1}$ ) observed at  $37 \text{ }^{\circ}\text{C}$  appears close to the one reported by Neves  
487 et al. (2006), which refer to a co-digestion of organic waste and sewage sludge ( $0.035-0.063 \text{ d}^{-1}$ ). On the  
488 contrary, significant differences can be observed referring to the results presented by De Gioannis et al.  
489 (2009) and Mou et al. (2015). Indeed, the  $k$ -values reported by those authors are up to one-two orders of  
490 magnitude lower than the ones obtained in this work. This difference can be due to the higher content of  
491 readily degradable organic matter in the analysed MBT waste compared to the low-organic wastes of Mou  
492 et al. (2015) and De Gioannis et al. (2009).  
493 Nevertheless, it should be kept in mind that the high water content (75 %w/w) as well as the inoculum  
494 addition had accelerated the biodegradation process during the anaerobic experiments. Hence, the gas  
495 generation rate and yield listed in Table 5 may be overestimated in comparison to real landfill conditions  
496 where the emplaced MBT wastes will not be able to retain the high water content (75 %w/w) simulated in  
497 lab scale tests. For example, Heyer et al. (2013) stated that the biological conversion process within  
498 lysimeters filled with MBT waste could be accelerated by a factor 3-10 compared to MBT landfills due to  
499 water addition or leachate recirculation.

500

#### Table 6

### 501 **3.5 Inhibition of anaerobic digestion process**

502 Analysis carried out on the MBT waste samples removed from the reactors at the different stages of the  
503 tests revealed that pH was still suitable for the anaerobic digestion (6.6-7.0). Hence, in this specific case, pH  
504 alone did not give a clear indication of process inhibition. However, it should be considered that pH  
505 changes may be very small in highly buffered system even when the process is severely stressed (Ahring et  
506 al., 1995). Thus, it is likely that pH was buffered due to contrasting effects of VFAs accumulation, which  
507 could have led to acidic conditions, and proteins degradation that could have favoured an increase of waste  
508 buffer capacity through the ammonia release (Veeken et al., 2000). In fact, as shown in Table 7, high VFAs  
509 and ammonia concentrations were measured in all MBT samples. Specifically, Table 7 reports the average

510 values of total TKN and soluble nitrogen  $\text{NH}_4\text{-N}$ , the ratio between  $\text{NH}_4\text{-N}$  and TKN, and the total VFAs  
511 concentration measured in the fresh MBT sample and in the MBT samples removed from the anaerobic  
512 batch experiments carried out at room temperature (R2), at 37 °C (T2A and T3) and 55 °C (T6A and T7). As  
513 shown in this table, the ratio of  $\text{NH}_4\text{-N}/\text{TKN}$  exhibited a twofold increase compared to what measured in  
514 the fresh waste. Moreover, VFAs concentrations in all samples were more than one order of magnitude  
515 higher than the ones measured in the fresh sample. These results indicate that the analysed MBT waste still  
516 contains a certain amount of readily and medium-degradable organic matter, which was not expected,  
517 since the easily degradable fraction was supposed to be mineralised during the stabilization process in the  
518 MBT plant. The high biological reactivity of this MBT waste may be in part due to a limited efficiency, during  
519 the waste sampling campaign, of the aerobic decomposition process employed in this specific MBT plant.  
520 However, as also highlighted in previous researches, this poor stabilisation can be mainly ascribed to the  
521 fact that the wastes coming out from this plant are not subjected to a further ripening treatment that might  
522 be necessary in order to obtain a well stabilized waste with lower impacts in landfills (Di Lonardo et al.,  
523 2014). Referring to Table 7, it can also be noticed that total VFAs content, as well as ammonia, exhibited an  
524 increasing trend with temperature ( $T7, T6 > T3, T2 > R2$ ) reflecting the different extent of the biological  
525 process reached at different operating conditions. Namely, a total VFAs content ranging from 5.4 to 7.9 g/l  
526 was measured and the acetate was found as the predominant compound (see Table S1 in supplementary  
527 information). Hence, it seems that the hydrolytic-acidogenic bacteria did not limit the substrate  
528 degradation and the process was held at the acetogenic and methanogenic stage (a similar result was  
529 obtained by Palatsi et al., 2011). Therefore, the inhibition was likely due to an imbalance in the growth rate  
530 of acidogenic bacteria, which led to an accumulation of degradation by-products in reactors, as also  
531 observed by Adani et al. (2004). On the other hand, the inhibition of methanogenic bacteria may be also  
532 due to the high TKN and ammonia content observed in the MBT samples (see Table 7). Thus, it is likely that  
533 the interaction between ammonia, VFA and pH could have led to an “inhibited steady state” condition in  
534 which the process was running stably but a very low methane yield (Chen et al., 2008).

535

#### **Table 7**

### 536 **3.6 Experimental findings and practical implications**

537 The BMP experiments, performed under optimal operating conditions, highlighted that for the analysed  
538 MBT waste, only 45 % of the theoretical gas generation potential can be achieved within 30 days. This may  
539 be due to the presence of less soluble/degradable or more recalcitrant organic substances in the solid  
540 matrix, which cannot be mineralized during the limited duration of the BMP experiment. Hence, BMP tests,  
541 with respect to a simple stoichiometric estimation from the organic carbon measured in the solid matrix,  
542 can provide useful indications on the expected potential gas generation capacity of landfilled waste under  
543 optimal conditions. On the contrary, the results obtained in this study highlighted that particular caution  
544 should be paid when the anaerobic batch tests are carried out under limiting operational conditions, e.g.  
545 low water content and temperature, high organic load, no inoculum addition. This is particularly true when  
546 the material of concern is, as in the present case, a waste with high organic matter content and a poor  
547 stability degree. In fact, although the tests performed under these limiting conditions may better resemble  
548 the environmental conditions expected in the field, the presence of inhibitory substances at high level (such  
549 as ammonia, VFA, heavy metals) may slow down or stop the anaerobic microbial process leading to an  
550 underestimation of the gas yield and generation rate. Indeed, as already shown in Table 7, in all  
551 experiments which exhibited very low gas generation rate and methane content, high concentrations of  
552 VFA and ammonia were measured within the systems, revealing an imbalanced kinetic between acid  
553 forming and acid consuming bacteria. Nevertheless, small pH changes were detected due to high protein  
554 degradation, which have increased buffer capacity of the analysed waste sample as a result of ammonia  
555 release. Hence, it is likely that the interaction between ammonia, VFA and pH lead to an “inhibited steady  
556 state” condition in which the anaerobic process may run stably but at a very low gas yield. In particular, the  
557 experimental results suggested that the microbial activity could be completely inhibited when the water  
558 content of MBT waste was less than 32% (on wet weight) and severely reduced for higher water content  
559 (up to the field capacity of approximately 43%) depending on the operating temperature. These findings  
560 suggest that a stable gas generation process could be delayed for a long time until the environmental  
561 conditions within MBT waste landfills become favourable to the establishment of a stable methanogenic

562 activity. Thus, it is not possible to predict how long the lag-phase can last in a landfill disposal scenario,  
563 where the boundary conditions significantly vary depending on the climate, the landfill geometry (surface,  
564 height), the operative management strategies in place (e.g. leachate recirculation, waste disposal  
565 methods), the presence and type of temporary and final cover systems. Indeed, the experimental results  
566 reported in this study revealed that, as soon as the anaerobic process starts, a relative short time period,  
567 ranging from few months up to two years depending on the water content and temperature, is required to  
568 deplete the gas generation capacity. However, small scale experiments performed under controlled  
569 conditions may not provide a gas generation trend that is completely representative of full-scale landfill  
570 sites. Here, higher heterogeneous and variable conditions are expected due to greater amounts of waste  
571 mass, miscellaneous nature of emplaced waste as well as the heterogeneity of water flow patterns inside  
572 the landfill body that are also affected by operational strategies (such as waste emplacement density,  
573 permeability and thickness of daily cover). Moreover, landfilled waste are subjected to increasing  
574 overloading pressure due to the emplacement of new waste layers. This condition surely influences the  
575 water retention capacity of landfilled MBT waste so that the water content will be surely lower than the  
576 ones simulated in the lab scale tests (e.g. 75 %w/w). This implies that the gas generation, in terms of both  
577 gas yield and rate, measured in anaerobic experiments at high water contents may be significantly higher  
578 than what expected in real scale MBT waste landfills.

#### 579 **4 Conclusions and perspectives**

580 The gas production from MBT wastes was analysed by performing anaerobic batch tests under different  
581 operating conditions. In order to characterize the MBT material regarding its long-term gas emission in  
582 different landfill disposal scenarios, a wide range of water contents (26-43 %w/w up to 75 %w/w) and  
583 temperatures (20-25 °C, 37 °C and 55 °C) were investigated. The obtained results suggest that the analysed  
584 MBT material still contains a large amount of readily degradable organic matter, as confirmed by the long  
585 duration of the lag-phase (several months), the high values of gas production potential ( $199.2 \pm 63$  Nml/gTS),  
586 the gas generation rates (ranging from  $0.007$  d<sup>-1</sup> at room temperature,  $0.03$ - $0.05$  d<sup>-1</sup> at 37°C and  $0.04$ - $0.11$



587  $d^{-1}$  at 55°C), as well as by the strong inhibition effects observed due to high concentrations of VFAs and  
588 soluble ammonia.

589 Based on the results presented in this paper, the following conclusions and perspectives can be drawn:

- 590 – it is very difficult to predict how long the lag-phase can last in MBT waste landfills where the  
591 boundary conditions change continuously and vary significantly depending on the climate, the  
592 landfill geometry (surface, height), the operative management strategies in place (e.g. leachate  
593 recirculation, waste disposal methods) and the presence and type of temporary and final cover  
594 systems.
- 595 – The water content of emplaced MBT waste is the most important factor limiting the anaerobic  
596 biological process. Experimental results showed that when the moisture was lower than 32 %w/w,  
597 the methanogenic microbial activity was completely inhibited whereas for higher values (43% w/w)  
598 only a limited amount of the degradable organic matter was converted to biogas (34% of the  
599 potential gas generation capacity).
- 600 – As soon as the environmental conditions inside the waste mass become favourable to the  
601 establishment of the stable methanogenic phase, a relatively short time period, ranging from few  
602 months up to two years is required to deplete the MBT waste gas generation. However, this result  
603 provides just an indication of the actual lifetime of biogas production from MBT wastes disposed of  
604 in landfills, where much higher amount of waste are emplaced and the environmental conditions  
605 may be quite far from the experimental ones.
- 606 – The benefits of using the MBT technology within a sustainable waste management system strongly  
607 depends on the efficiency of the biostabilization process in reducing the gas generation capacity of  
608 the residual MBT waste. In this specific case, experimental data suggest that the aerobic biological  
609 treatment carried out in this specific MBT facility was not properly managed and did not guarantee  
610 a sufficient degree of stability for the produced MBT waste. Therefore, a further treatment of this  
611 MBT waste might be desirable before landfilling it.

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811 theoretical methane generation of MBT waste, computed according to Eq. (3). Bars: standard deviation. The results is  
812 the average of 6 and 3 bottles for MBT waste and controls, respectively.

813 **Fig. 2.** Cumulative methane (red circles) and carbon dioxide (black square dots) curves as a function of the incubation  
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820 Dotted black line: CO<sub>2</sub> potential from BMP. Solid grey line: starting point of stage II. Dotted grey line: starting point of  
821 stage III.

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834  $L_0$ : maximum biogas production (NI/kgTS).  $t_{lag}$ : lag time (d).  $k$ : first-order kinetic constant ( $d^{-1}$ ).  $R^2$ : correlation factor.

835 % $L_0$ /BMP: percentage of gas generated compared to the potential value measured in BMP test.  $T_{99\%}$ : time to reach  
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840 anaerobic batch reactors.

841

842

Table 1 (Line 94)

<i>Reference</i>	<i>Amount</i>	<i>Particle size</i>	<i>Water addition</i>	<i>Inoculum</i>	<i>Temperature (°C)</i>	<i>Test duration (d)</i>	<i>Aim</i>
Binner and Zach (1999)	50 g TS	<20 mm	1 l demineralized water	Used	35	42	Cumulative gas
De Araujo-Morais et al. (2008)	20 g	<20 mm	1.2 l nutrient medium	Used	35	90	Cumulative gas
Sormunen et al. (2008)	2 gVS <sub>w</sub> /gVS <sub>m</sub>	<40 mm	up to 95%	Used	20-22	70-100	Methane potential
Barrena et al. (2008)	200 g	ns	Not added	Used	35	100	Gas production in solid state test
Barrena et al. (2008)	1 g TS	<1 mm	up to 50%	Used	35	75	Gas production in liquid state test
Binner & Zach (1999)	1 kg TS	<20 mm	up to water holding capacity	Not used	40	90	Determining gas generation in landfills
De Gioannis et al. (2009)	500 g	ns	up to water holding capacity	Not used	30	240-390	Modelling gas generation
Sormunen et al (2008)	96 t	<40 mm	ns	Used	40	640	Simulate landfill emissions

ns=not specified

(\*) water added on wet weight basis

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845

**Table 2 (Line 108)**

<i>Categories</i>	<i>Mean</i>	<i>SD</i>
Fines < 20mm	13.4	3.8
Organic waste	28.0	7.3
Paper	16.0	4.6
Cardboard	7.7	3.2
Coupled packaging	1.8	0.5
Textiles	3.0	1.1
Diapers	4.8	2.0
Plastics	14.4	2.8
Rubber	0.4	0.3
Glass	3.7	1.9
Metals	2.9	1.1
Inert materials	1.2	1.0
Hazardous waste	0.4	0.3
Wood	1.9	1.7
Leather	0.1	0.2
Others	0.4	0.4

**Table 3 (Line 250)**

ID TEST	Reactor volume (l)	T°C	MBT weight (kg TS)	Water content (% w/w)	Inoculum addition (g)	Duration (d)
<b>R1A, R1B</b>	12	20-25	4.0	26%	Not used	<b>Stage I:</b> 167
<b>R2A, R2B</b>	12	20-25	4.4	34% ( <b>Stage I</b> ) 75% ( <b>Stage III</b> )	<b>Stage III:</b> 800 g mesophilic inoc. (TS=3.8% wet weight, VS=60%TS) + 6.1 kg water <sup>(a)</sup>	<b>Stage I:</b> 94 <b>Stage III:</b> 73
<b>R3A, R3B</b>	12	20-25	4.5	43%	Not used	<b>Stage I:</b> 167
<b>T1A, T1B</b>	1	37	0.37	26%	<b>Stage II:</b> 20 g mesophilic inoc. (TS=3.8% wet weight, VS=60%TS)	<b>Stage I:</b> 38 <b>Stage II:</b> 94
<b>T2A</b>	1	37	0.36	32% ( <b>Stage I</b> ) 75% ( <b>Stage III</b> )	<b>Stage II:</b> 20 g mesop. inoc. <b>Stage III:</b> 70 g mesop. inoc. + 530 g water <sup>(b)</sup>	<b>Stage I:</b> 38 <b>Stage II:</b> 20 <b>Stage III:</b> 74
<b>T2B</b>	1	37	0.37	32%	<b>Stage II:</b> 20 g mesop. inoc. (TS=3.8% wet weight, VS=60%TS)	<b>Stage I:</b> 38 <b>Stage II:</b> 94
<b>T3A, T3B</b>	1	37	0.38	38% ( <b>Stage I</b> ) 75% ( <b>Stage III</b> )	<b>Stage III:</b> 70 g mesop. inoc. + 530 g water <sup>(b)</sup>	<b>Stage I:</b> 58 <b>Stage III:</b> 74
<b>T4A</b>	1	37	0.38	43%	Not used	<b>Stage I:</b> 132
<b>T4B</b>	1	37	0.38	43% ( <b>Stage I</b> ) 48% ( <b>Stage III</b> )	<b>Stage III:</b> 70 g mesop. Inoc. (same), no water addition <sup>(c)</sup>	<b>Stage I:</b> 50 <b>Stage III:</b> 82
<b>T5A, T5B</b>	1	55	0.36	26%	<b>Stage II:</b> 20 g thermophilic inoc. (TS=3% wet weight, VS=65%TS)	<b>Stage I:</b> 24 <b>Stage II:</b> 76
<b>T6A</b>	1	55	0.36	32% ( <b>Stage I</b> ) 75% ( <b>Stage III</b> )	<b>Stage II:</b> 20 g thermop.inoc. <b>Stage III:</b> 70 g thermop. inoc. + 530 g water <sup>(d)</sup>	<b>Stage I:</b> 24 <b>Stage II:</b> 19 <b>Stage III:</b> 57
<b>T6B</b>	1	55	0.37	32%	<b>Stage II:</b> 20 g thermop. inoc.	<b>Stage I:</b> 24 <b>Stage II:</b> 76
<b>T7A, T7B</b>	1	55	0.39	38% ( <b>Stage I</b> ) 75% ( <b>Stage III</b> )	<b>Stage III:</b> 70 g thermop. inoc. + 530 g water <sup>(d)</sup>	<b>Stage I:</b> 43 <b>Stage III:</b> 57
<b>T8A, T8B</b>	1	55	0.38	43%	Not used	<b>Stage I:</b> 103

(a) Waste removed from reactor before inoculum and water addition. TS content Stage III: 2.66 kgTS

(b) Waste removed from reactor before inoculum and water addition. TS content Stage III: 0.21 kgTS (T3A/B)- 0.24 kgTS (T2A)

(c) Waste removed from reactor before inoculum addition. TS content Stage III: 0.36 kgTS

(d) Waste removed from reactor before inoculum and water addition. TS content Stage III: 0.22 kgTS (T7A/B) - 0.24 kgTS (T6A)

**Table 4 (Line 273)**

	<i>Mean</i>	<i>Unit</i>
<i>Initial water content, W<sup>(*)</sup></i>	19.4 ± 1.1	(%w/w)
<i>Water content at field capacity, FC<sup>(*)</sup></i>	41 ± 5	(%w/w)
<i>Total solids, TS<sup>(*)</sup></i>	80.6 ± 1.0	(%w/w)
<i>Volatile solids, VS</i>	47.3 ± 1.0	(%TS)
<i>Organic carbon, TOC</i>	23.9 ± 0.3	(%TS)
<i>pH</i>	6.7	--
<i>TKN</i>	14.1 ± 2.0	g/kgTS
<i>NH<sub>4</sub>-N</i>	2.0 ± 0.2	g/kgTS
<i>COD</i>	520 ± 40	g/kgTS
<i>Total VFA</i>	0.18 ± 0.05	g/l

<sup>(\*)</sup> expressed on wet weight basis



**Table 5 (Line 427)**

	T= 20-25°C	T= 37°C			T= 55°C		
	R2 (W=34%-75%)	T4B (W=43%-50%)	T2A (W=32%-75%)	T3 (W=39%-75%)	T8 (W=43%)	T6A (W=32%-75%)	T7 (W=39%-75%)
L <sub>0</sub> (NI/kgTS)	75	140	150	129	68	130	111
t <sub>lag</sub> (d)	103	75	73	69	39	59	52
k (d <sup>-1</sup> )	0.007	0.028	0.044	0.054	0.038	0.110	0.110
R <sup>2</sup>	0.993	0.991	0.972	0.996	0.992	0.981	0.986
% L <sub>0</sub> /BMP	37.7	70.3	75.3	64.8	34.1	65.3	55.7
T <sub>99%</sub> (d)	658	165	104	85	121	42	42

**Table 6 (Line 447)**

<i>Gas yield</i>	<i>k-value(d<sup>-1</sup>)</i>	<i>Substrate</i>	<i>Experimental assay</i>	<i>Reference</i>
75 NI/kgTS	0.007 (T=20-25°C)			
129-150 NI/kgTS	0.03-0.06 (T=37°C)	MBT waste (different water contents)	anaerobic batch digester	<i>This work</i>
68 <sup>(*)</sup> -130 NI/kgTS	0.04 <sup>(*)</sup> -0.11 (T=55°C)			
52 -70 gCH <sub>4</sub> /kg	4.3·10 <sup>-4</sup> -5.2·10 <sup>-4</sup>	sludge	anaerobic batch digester	Mou et al. (2015)
107-117 NI/kg	6.6·10 <sup>-5</sup> -6.8·10 <sup>-5</sup>	combustible waste		
37.6 ICH <sub>4</sub> /kg	0.013 (T=55°C)	lignocellulosic biomass	batch reactor (BMP)	Ghatak and Mahanta (2014)
26.6 ICH <sub>4</sub> /kg	0.004 (T=35°C)			
350 NICH <sub>4</sub> /kgTS	0.068 (T=35°C)	mixed waste from landfill	reactor with recirculation	Bilgili et al. (2009)
425 NICH <sub>4</sub> /kgTS	0.056 (T=35°C)	(44% w/w organic)	reactor without recirculation	
19 NI/kg	9.6·10 <sup>-5</sup> (T=30°C)	MBT waste (water=50% w/w)	anaerobic batch digester	De Gioannis et al. (2009)
550 NI/kgVS	0.1	hay (no comminution)	anaerobic batch digester	Vavilin et al. (2008)
590 NI/kgVS	0.15	hay (comminution)		
240-280 NICH <sub>4</sub> /kgVS	0.035-0.063 (T=37°C)	co-digested organic waste and sewage sludge	anaerobic batch digester	Neves et al. (2006)
163 NICH <sub>4</sub> /kgVS	0.0311 (T=35°C)	grey waste (residual MSW, 41% w/w biodegradable)	batch reactor (BMP)	Vavilin et al. (2004)
---	0.007 - 0.08 (T=30°C)	MSW (water=65% w/w)	landfill reactors	Vavilin et al. (2004)
---	0.06-0.245 (T=28°C)	biowaste	continuous anaerobic digester	Veeken et al. (2000)
---	0.03-0.15 (T=20°C)	selected biowaste components	anaerobic batch digester	Veeken and Hamelers (1999)
---	0.24-0.47 (T=40°C)			
140 ICH <sub>4</sub> /kgVS	0.405 (T=35°C)	mechanically separated OF-MSW	continuous anaerobic digester	Mata-Alvarez et al. (1990)

<sup>(\*)</sup> values refer to 103-days experiments performed at 55 °C and 43% (wet weight) water content, without inoculum addition.

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**Table 7 (Line 479)**

<i>SAMPLE</i>	<i>TKN (g/kgTS)</i>	<i>NH<sub>3</sub>-N (g/kgTS)</i>	<i>NH<sub>3</sub>-N/TKN (%)</i>	<i>VFA (g/l)</i>
FRESH WASTE	14.1 ± 2.0	2.0 ± 0.2	14.4	0.18 ± 0.05
R2 (20-25 °C, W= 34 %w/w)	15.2 ± 1.2	3.4 ± 0.1	22.7	5.42 ± 0.24
T2A (37 °C, W=32 %w/w)	15.2 ± 0.6	4.2 ± 0.2	27.4	6.28 ± 0.58
T3 (37 °C, W= 38 %w/w)	14.1 ± 0.9	4.2 ± 0.3	30.1	7.57 ± 0.31
T6A (55 °C, W= 32 %w/w)	14.1 ± 1.4	4.6 ± 0.3	32.4	7.94 ± 0.93
T7 (55 °C, W= 38 %w/w)	15.0 ± 1.3	5.0 ± 0.2	33.6	7.56 ± 0.83

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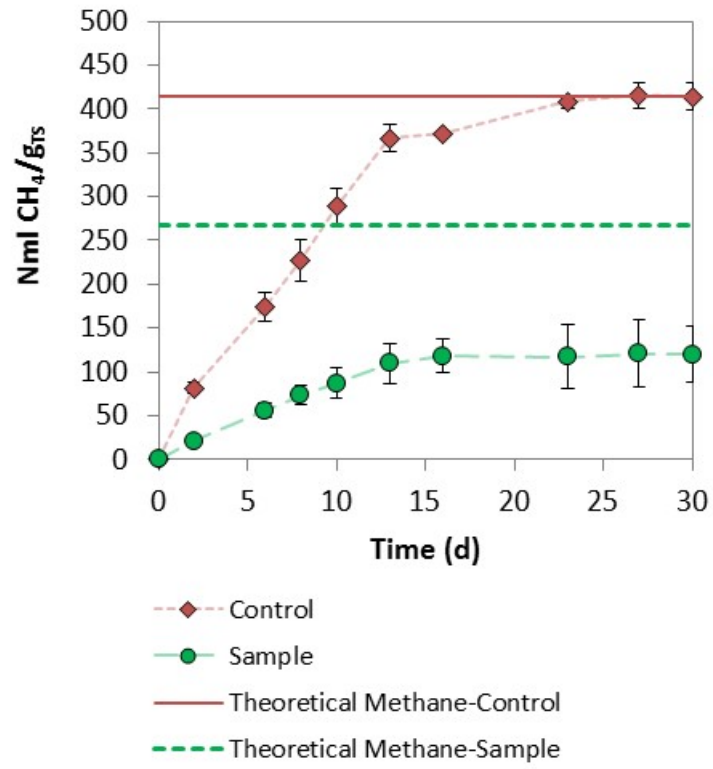
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Figure 1 (Line 317)



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Figure 2 (Line 336)

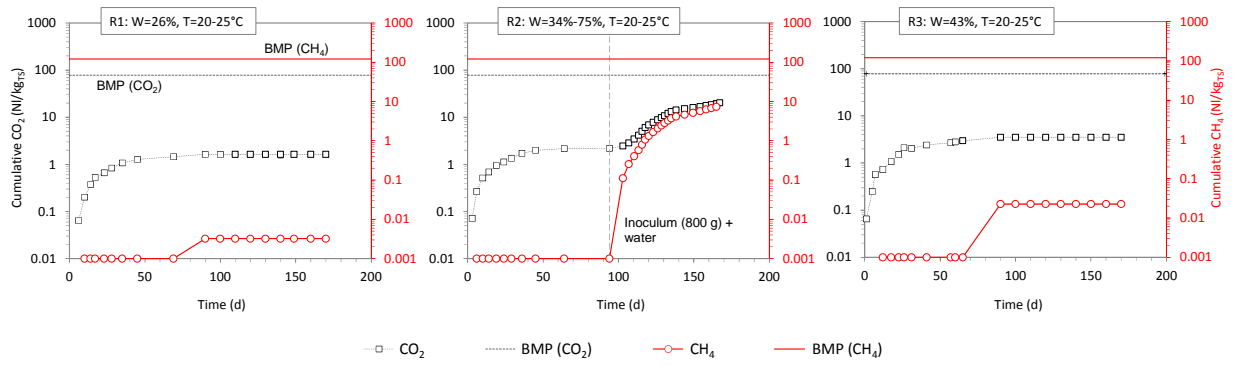


Figure 3 (Line 338)

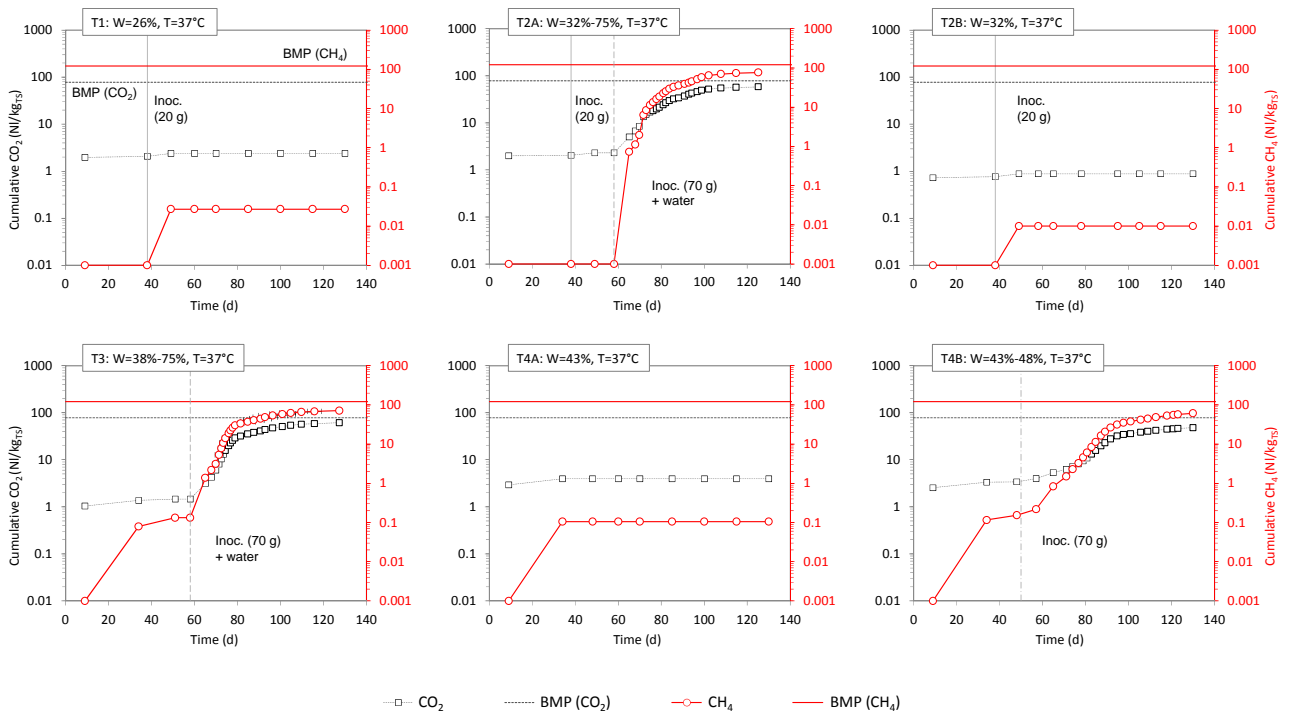
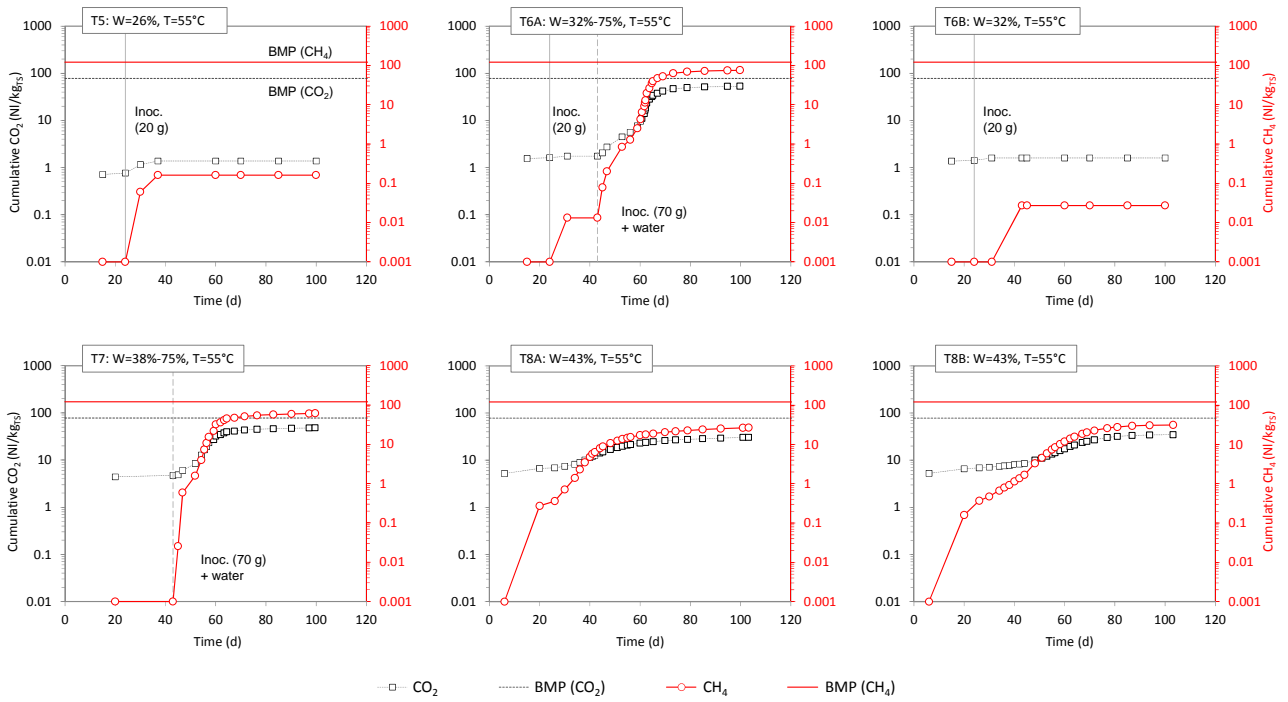
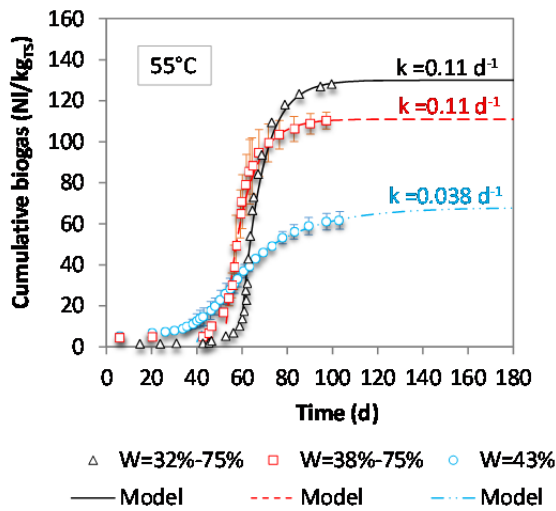
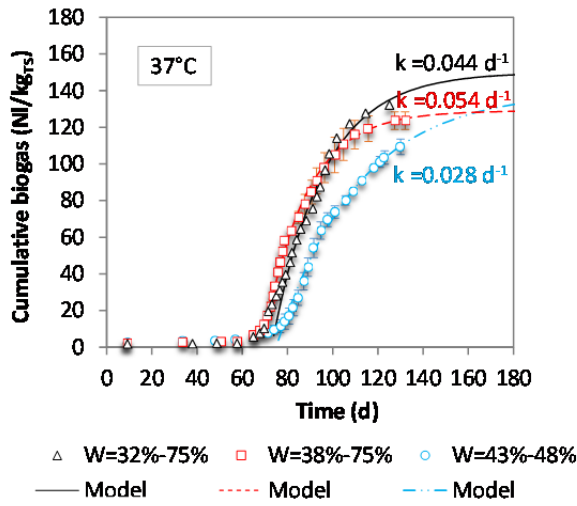
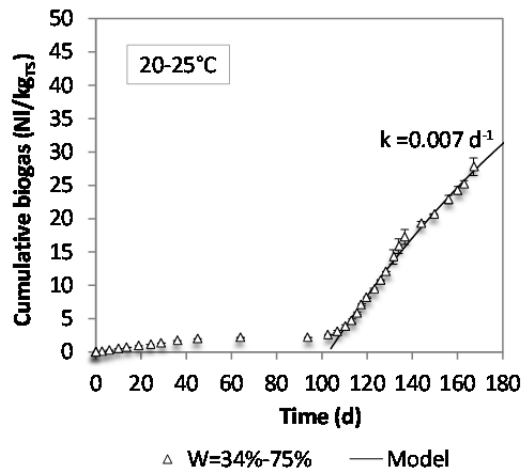


Figure 4 (Line 340)



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Figure 5 (Line 418)



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8 *Supplementary Material for:*

9 ***Assessment of biogas production from MBT waste under different operating conditions***

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11

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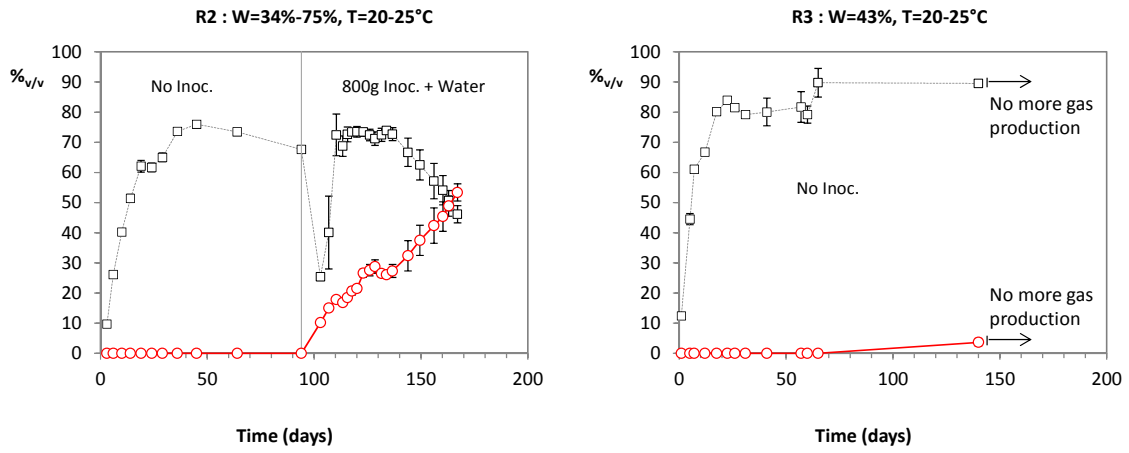
28 Figure S3. Gas composition (percentage by volume) measured in tests at 55°C and initial water content of  
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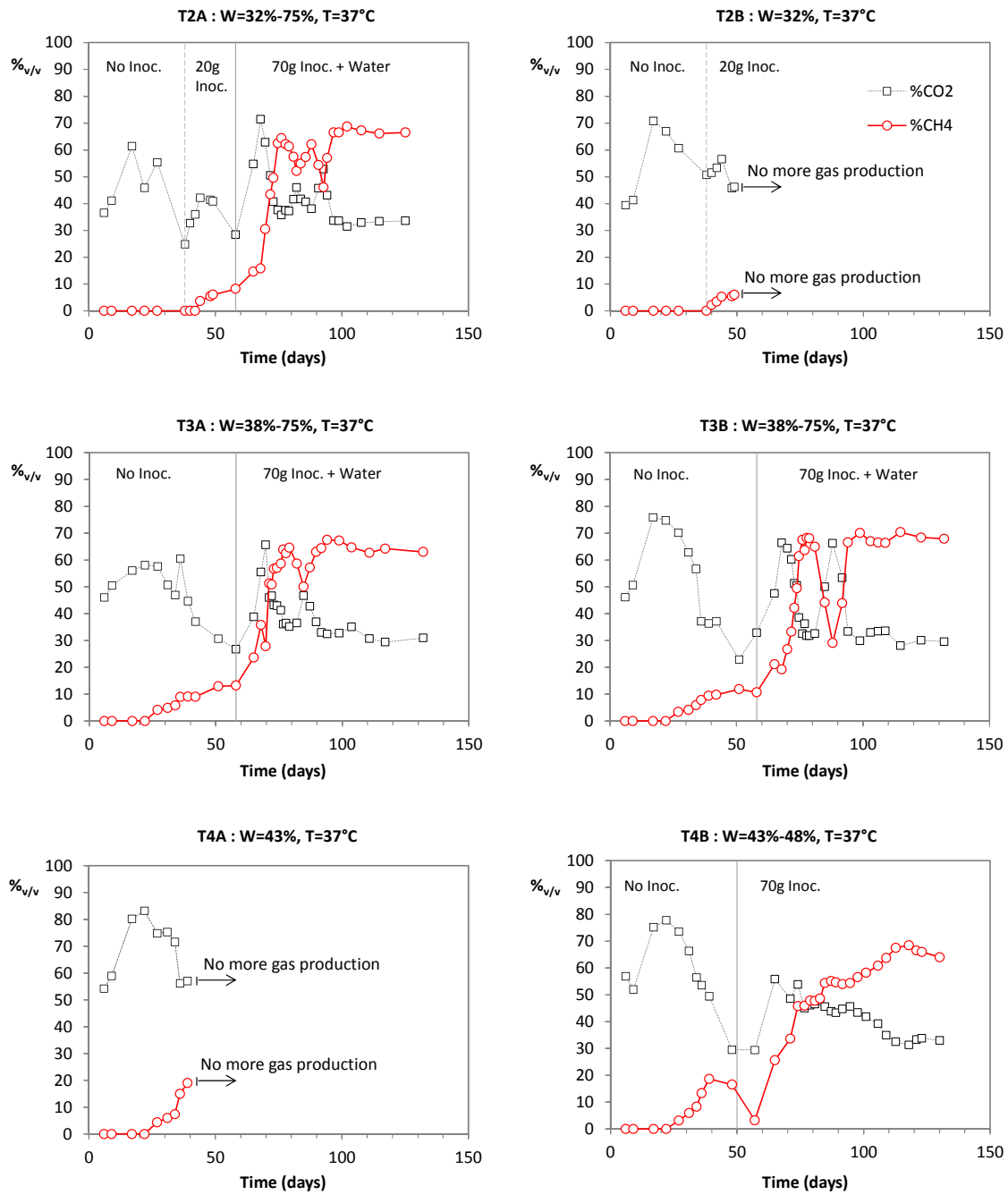
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7 **Figure S1.** Gas composition (percentage by volume) measured in tests at room temperature (20-25°C) and initial water content of 32%-75% (R2) and 43% (R3). Solid grey line: starting point of stage III.

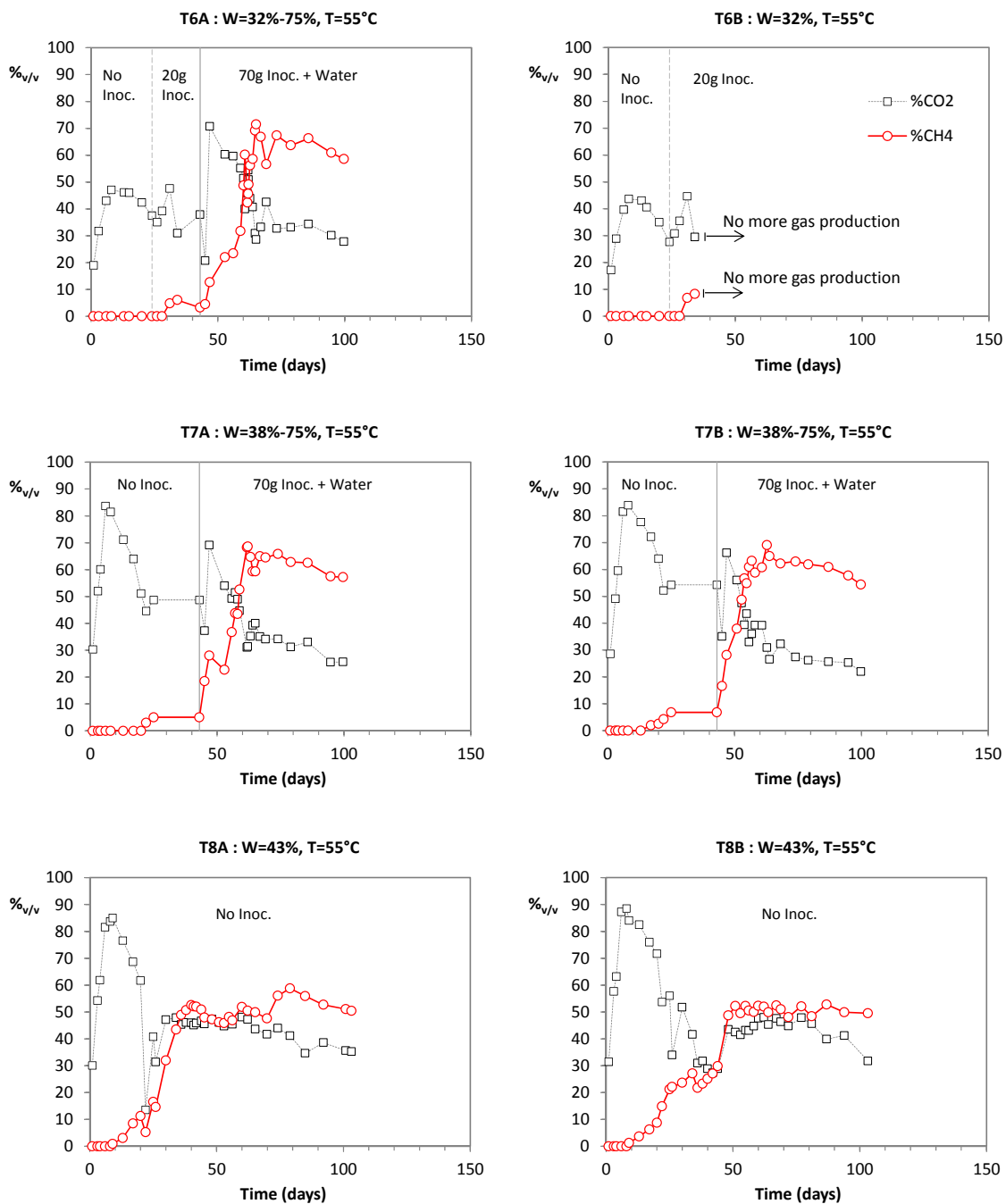


1 8 **Figure S2.** Gas composition (percentage by volume) measured in tests at 37°C and initial water content  
 2 of 32%-75% (T2A), 32% (T2B), 38%-75% (T3A/B), 43% (T4A) and 43%-48% (T4B). Dotted grey line:  
 3 starting point of stage II. Solid grey line: starting point of stage III.



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8 9 **Figure S3.** Gas composition (percentage by volume) measured in tests at 55°C and initial water content  
 9 of 32%-75% (T6A), 32% (T6B), 38%-75% (T7A/B), 43% (T8A/B). Dotted grey line: starting point of stage  
 10 II. Solid grey line: starting point of stage III.



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