



Environmental Electrokinetics for a sustainable subsurface

Lima, A.T.; Hofmann, A.; Reynolds, D.R.; Ptacek, C.J.; Van Cappellen, P.; Ottosen, Lisbeth M.; Pamukcu, S.; Alshawabekh, A.; O'Carroll, D.M.; Riis, C.

Total number of authors:
17

Published in:
Chemosphere

Link to article, DOI:
[10.1016/j.chemosphere.2017.03.143](https://doi.org/10.1016/j.chemosphere.2017.03.143)

Publication date:
2017

Document Version
Peer reviewed version

[Link back to DTU Orbit](#)

Citation (APA):

Lima, A. T., Hofmann, A., Reynolds, D. R., Ptacek, C. J., Van Cappellen, P., Ottosen, L. M., Pamukcu, S., Alshawabekh, A., O'Carroll, D. M., Riis, C., Cox, E., Gent, D. B., Landis, R., Wang, J., Chowdhury, A. I. A., Secord, E. L., & Sanchez-Hachair, A. (2017). Environmental Electrokinetics for a sustainable subsurface. *Chemosphere*, 181, 122-133. <https://doi.org/10.1016/j.chemosphere.2017.03.143>

General rights

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

- Users may download and print one copy of any publication from the public portal for the purpose of private study or research.
- You may not further distribute the material or use it for any profit-making activity or commercial gain
- You may freely distribute the URL identifying the publication in the public portal

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

1 **Environmental Electrokinetics for a Sustainable Subsurface**

2 A.T. Lima^{1,2}, A. Hofmann³, D. Reynolds⁴, C.J. Ptacek⁵, P. Van Cappellen¹, L.M. Ottosen⁶, S. Pamukcu⁷, A.
3 Alshawabekh⁸, D.M. O’Carroll^{9, 10}, C. Riis¹¹, E. Cox⁴, D. B. Gent¹², R. Landis¹³, J. Wang⁴, A.I.A.
4 Chowdhury⁹, E.L. Secord¹, A. Sanchez-Hachair³

5
6 ¹ Ecohydrology Research Group, Department of Earth and Environmental Sciences and Water
7 Institute, University of Waterloo, Waterloo, Canada

8 ² Department of Environmental Engineering, Universidade Federal do Espírito Santo, Vitória, ES,
9 Brazil

10 ³ University of Lille, CNRS, UMR 8187, LOG, Laboratoire d’Océanologie et de Géosciences, 59655
11 Villeneuve d’Ascq, France

12 ⁴ Geosyntec Consultants, Waterloo, Ontario, Canada

13 ⁵ Department of Earth and Environmental Sciences and Water Institute, University of Waterloo,
14 Waterloo, Canada

15 ⁶ Department of Civil Engineering, Technical University of Denmark, Lyngby, Denmark

16 ⁷ Department of Civil and Environmental Engineering, Lehigh University, Bethlehem, PA, USA

17 ⁸ Department of Civil and Environmental Engineering, Northeastern University, Boston, MA, USA

18 ⁹ Department of Civil and Environmental Engineering, University of Western Ontario, London, ON,
19 Canada

20 ¹⁰ School of Civil and Environmental Engineering, Connected Water Initiative, University of New
21 South Wales, Manly Vale, NSW, 2093, Australia

22 ¹¹ NIRAS, Sortemosevej 19, 3450 Allerød, Denmark

23 ¹² Environmental Laboratory, USACE Engineer Research and Development Center, Vicksburg, MS,
24 USA

25 ¹³ RichLand Consulting, LLC, Wilmington, Delaware

26
27 **Abstract**

28 Soil and groundwater are key components in the sustainable management of the subsurface
29 environment. Source contamination is one of its main threats and is commonly addressed using
30 established remediation techniques such as in-situ chemical oxidation (ISCO), in-situ chemical
31 reduction (ISCR; most notably using zero-valent iron [ZVI]), enhanced in-situ bioremediation (EISB),
32 phytoremediation, soil-washing, pump-and-treat, soil vapour extraction (SVE), thermal treatment,
33 and excavation and disposal. Decades of field applications have shown that these techniques can

34 successfully treat or control contaminants in higher permeability subsurface materials such as sands,
35 but achieve only limited success at sites where low permeability soils, such as silts and clays, prevail.
36 Electrokinetics (EK), a soil remediation technique mostly recognized in in-situ treatment of low
37 permeability soils, has, for the last decade, been combined with more conventional techniques and
38 can significantly enhance the performance of several of these remediation technologies, including
39 ISCO, ISCR, EISB and phytoremediation. Herein, we discuss the use of emerging EK techniques in
40 tandem with conventional remediation techniques, to achieve improved remediation performance.
41 Furthermore, we highlight new EK applications that may come to play a role in the sustainable
42 treatment of the contaminated subsurface.

43 **Keywords:** electrokinetics, remediation, subsurface contamination, plume migration, phyto-
44 remediation, bioremediation, ISCO, nano zero valent iron (nZVI), landfill

46 1. Introduction

47 Soils, sediments and aquifers are fundamental bases for global environmental sustainability and
48 provide essential resources to humans and nature alike (Godfray et al., 2010; Koch et al., 2013;
49 McBratney et al., 2014). Anthropogenic impacts through land-use changes have affected, to differing
50 degrees, the capacity of these geological features to maintain their basic functionality. Nutrient
51 cycling, water retention, provision of physical/chemical stability, storage/filtering/transformation
52 of compounds and sustaining biodiversity are some of the key functions that are an integral part of
53 groundwater and food security (Godfray et al., 2010; Seto et al., 2011; United Nations, 2009). Most
54 fresh water reserves (98-99%) occur in the subsurface; this means that 8-10 million km³ of
55 freshwater is contained or in direct contact with rock and soil (Margat, 2008; Shah et al., 2007). In an
56 increasingly populated and urbanized world, pollution is widely recognized as a significant challenge
57 to soil and groundwater resources management (FAO, 2003). Efforts to quantify groundwater
58 pollution (Giuliano et al., 1998; Zaporozec, 2002) are scarce and the actual scale of the problem is not
59 well known. Point-source pollution (i.e., the source of many groundwater plumes) is often difficult to
60 accurately locate and address, as depicted in Figure 1. The causes of point-source contamination may
61 be varied: industrial leakage (Gent et al., 2004; Lima et al., 2012a), backfill in construction works
62 (Laethem and Legrand, 1993), or overall urbanized environments (Callender and Rice, 2000;
63 Yongming et al., 2006). Source pollution treatment (A in Figure 1) is key to limiting contaminant
64 transport to the vadose zone environment (Dresel et al., 2011) and subsequent migration to the
65 groundwater zone (B in Figure 1). When present, low permeability lenses (Figure 1), initially act as
66 a sink for contaminants however with time can switch to being a source of long-term contamination.

67 Contamination in low permeability materials remains a significant and unresolved remediation
68 challenge. A large contaminant source has the potential to generate a significant plume in
69 groundwater, and presents a risk to potential downgradient receptors (A in Figure 1). A unique form
70 of contaminant transport occurs with dense non-aqueous phase liquids (DNAPLs), including
71 chlorinated solvents, polychlorinated biphenyls (PCBs), coal tars etc. since they tend to sink to the
72 bottom of aquifers due to their density (Macdonald et al., 2000). Contaminants in the context of
73 groundwater exposure pathways are varied, and can consist of organic molecular compounds, nano-
74 substances, pharmaceuticals, immiscible liquids, and are often toxic at very low (part per billion)
75 concentrations. New ingenious approaches are needed for in-situ (and ex-situ) remediation of soils,
76 sediments and aquifers, particularly when they are comprised of low permeability materials.

77
78 Recent successes in electrokinetic (EK) or EK assisted remediation programs have demonstrated that
79 the technique may be a viable and versatile remediation tool for low permeability soils and
80 sediments. EK approaches generally consist of the application of a direct current (DC) electric field
81 to the subsurface through electrodes to move porewater or migrate contaminants or remediation
82 amendments. EK approaches have wide applicability, from soil dewatering (Lockhart and Stickland,
83 1984; Yang et al., 2005) to desalination of built environments (Ottosen and Rorig-Dalgaard, 2009),
84 removal of soil metal contaminants (Jensen et al., 2007; Ottosen et al., 1997; Pamukcu et al., 1997;
85 Pamukcu and Wittle, 1992) or degrading/removing soil organic contaminants (Lima et al., 2012a;
86 Lima et al., 2011; Pamukcu, 1994; Reddy et al., 2011; Saichek and Reddy, 2004, 2003), including
87 immiscible oil constituents (Ghazanfari and Pamukcu, 2014; Pamukcu et al., 2016). First used at the
88 beginning of the last century for dewatering and stabilizing clays (Casagrande, 1949; Reuss, 1809),
89 EK was intensively studied in the 1990s for removing metals from soils (Acar and Alshawabkeh,
90 1993; Lageman, 1993; Ottosen et al., 1997; Ottosen and Hansen, 1992; Schultz, 1997), as well as
91 radionuclides and other inorganic species (Acar et al., 1995). More recently EK has been used in
92 various formulations for the removal of organic contaminants (Lima et al., 2011; Pamukcu, 1994;
93 Ribeiro et al., 2005). The approach has been highly successful at the laboratory scale for both
94 inorganic and organic contaminants and in recent field trials for organic contaminants. Field-scale
95 in-situ applications are becoming more common, and have demonstrated the potential of the
96 approach. In spite of promising early results at a variety of scales, the technology has not yet
97 advanced to a commonly-accepted commercial status.

98

99 EK applications in recent years have integrated chemical surfactants, chemical amendments and
100 chelating agents to enhance metals or organics desorption from soil (Bolan et al., 2014; Lima et al.,
101 2012b; Ottosen et al., 2012), or to degrade contaminants at their source. Migration of these chemical
102 enhancers in low permeability soils is possible due to four key phenomena occurring when a direct
103 current is applied to soil: electromigration (movement of charged ions), electro-osmosis (movement
104 of pore water), electrophoresis (movement of colloids) and electrolysis (water splitting) (Acar and
105 Alshawabkeh, 1993). Coupling these phenomena with the aforementioned conventional remediation
106 techniques has been the focus of recent EK research (Cameselle et al., 2013a; Lageman and
107 Godschalk, 2007; Pamukcu et al., 2004). In addition to coupling with existing approaches, EK has
108 been demonstrated to be capable of stripping micropollutants, especially non-polar compounds
109 including heavy oils, from low permeability soils and sediments, where other treatment methods may
110 fail (Alcántara et al., 2012; Ghazanfari et al., 2012; Lima et al., 2012b; Pamukcu, 2009; Pazos et al.,
111 2010). Other EK applications are currently being researched for instance the off-site removal of soil
112 metals (Jensen et al., 2007; Ottosen et al., 2012) and a stand-alone technique for oil transport and
113 recovery from geological media including soils, sediments and rock formations (Chilingar et al., 1968;
114 Ghazanfari et al., 2014; Ghazanfari and Pamukcu, 2014; Ghosh et al., 2012; Pamukcu, 1994; Pamukcu
115 et al., 2016; Wittle et al., 2011).

116
117 EK approaches face some unique engineering challenges mainly due to the side effects associated
118 with this technique, such as alteration of natural pH levels near the electrodes, potential enhanced
119 weathering of the porous media during long-term applications, hydrogen and chlorine gas generation
120 at the electrodes and/or other unpredicted redox reactions (Acar and Alshawabkeh, 1993; Ottosen
121 et al., 2000). Some of these side-effects (Acar and Alshawabkeh, 1993) have been engineered to
122 advantage as synergistic effects beneficial for other traditional soil techniques, such as ISCO (USEPA,
123 2006), EISB (Mao et al., 2012; Niqui-Arroyo and Ortega-Calvo, 2007) and phytoremediation
124 (Aboughalma et al., 2008; Cameselle et al., 2013a; Kubiak et al., 2012). Field scale application of EK-
125 combined remedies has recently been demonstrated at sites in Denmark and the USA (Mao et al.,
126 2012; Riis et al., 2012).

127
128 EK has been the topic of a number of review papers, each of which focused on one facet or another of
129 this technique. Yeung (Yeung, 2011) takes a historical approach, Cameselle et al. (Cameselle et al.,
130 2013b) focused on the combination of phytoremediation and EK, Pamukcu (Pamukcu, 2009) on
131 electrochemical treatise, others on specific contamination (Gomes et al., 2015; Saichek and Reddy,

132 2005). EK is an approach with the capacity of overcoming traditional problems in soil remediation
133 practices. The technique can be applied in-situ in a manner that avoids soil excavation and minimizes
134 the disturbance of soil texture, porosity and biodiversity (Saichek and Reddy, 2005; Virkutyte and
135 Sillanpaa, 2002). Treatment costs are variable depending on the contaminant, approach, and other
136 site-specific factors, but are in line with other intrusive in-situ approaches (Virkutyte and Sillanpaa,
137 2002). Due to low soil disturbance, low water and energy usage, EK is often less expensive than other
138 remedial techniques (Alshawabkeh and Acar, 1992; Gomes et al., 2015; Higgins et al., 2008; Wittle et
139 al., 2011; Yang et al., 2011). Herein, we discuss the strength of EK as a complement to traditional soil
140 and groundwater remediation techniques and explore future avenues for EK as a sustainable
141 subsurface remediation approach.

142

143 **2. Conventional soil remediation techniques for source control and their limitations**

144 Commonly practiced technologies for soil and groundwater remediation include permeable reactive
145 barriers (PRBs) (Benner et al., 2002), bioremediation (Guerin, 1999; Tromp et al., 2012), i.e. the use
146 of either plants (phytoremediation) or microorganisms (bioremediation) to degrade, accumulate or
147 reduce/oxidize chemical contaminants, soil-washing (Meuser, 2012), pump-and-treat (Meuser,
148 2012), in-situ chemical oxidation (ISCO) and in-situ chemical reduction (ISCR), the use of chemicals
149 to oxidize or reduce redox sensitive contaminants in high porosity soils and sediments, vapour-
150 technologies to volatilize organic contaminants, thermal approaches to vapourize organic
151 contaminants and remove them through vacuum extraction, and “dig-and-dump”. The majority of
152 these approaches (with the exception of thermal and dig-and-dump) have significant limitations
153 when contaminants to be treated reside in low permeability regions. Table 1 summarizes the main
154 characteristics and main challenges of bioremediation, phytoremediation, nano-particles, ISCO,
155 thermal, and landfilling (dig-and-dump). As mentioned previously, a number of these techniques can
156 be combined with EK to increase their suitability for treatment of contaminants in low permeability
157 regions. A summary of the techniques that can be enhanced using EK, and a brief synthesis of the
158 target contaminants and major limitations is presented in the next sections.

159

160 **Bioremediation** is a cost-effective technique for treating a variety of contaminants, including
161 chlorinated solvents, BTEX, selected inorganic substances (e.g., perchlorate and nitrate) and
162 polycyclic aromatic hydrocarbons (PAHs) (Liebeg and Cutright, 1999; Sturman et al., 1995; Vidali,
163 2001). Major bioremediation techniques broadly fall into two categories: biostimulation and
164 bioaugmentation. Biostimulation most commonly refers to the addition of nutrients, such as electron

165 acceptors or electron donors, to promote biodegradation of contaminants by indigenous microbes.
166 Bioaugmentation most commonly refers to the addition of microbes possessing specific capabilities
167 to biodegrade contaminants. A number of factors can hinder bioremediation processes, including
168 limited physical interaction between microorganisms and substances (contaminant bioavailability
169 and/or bioaccessibility) (Semple et al., 2004); adverse site conditions (temperature, high co-
170 contamination (Straube et al., 2003), pH, salts, oxygen, etc.) that may be inhibitory or toxic to
171 microorganisms (Wick et al., 2011); and lastly, absence of bacterial species that can degrade the
172 target contaminant.

173
174 **Phytoremediation** is a low-cost and environmentally acceptable solution for the remediation of
175 shallow soils, soil water and runoff for both metals and organic contaminants e.g., (Chaney et al.,
176 1997; McCutcheon and Schnoor, 2004; Tromp et al., 2012). Phytoremediation is highly suitable for
177 metals contamination as plants are capable of concentrating metals and extracting them from soils
178 (Pilon-Smits, 2005). However, remediation occurs at a shallow soil depth (root zone) and may
179 require extended remediation time. Solutions may include deep planting of trees in boreholes (tree-
180 wells), and the use of polluted groundwater for plant irrigation (Pilon-Smits, 2005).

181
182 **ISCO** for groundwater remediation was first introduced in the 1990s as an aggressive in situ
183 technique to address groundwater contamination without requiring soil excavation (Innocenti et al.,
184 2014; Schnarr et al., 1998; Yukselen-Aksoy and Reddy, 2012). Four oxidants have been commonly
185 used: hydrogen peroxide (Fenton's reagent), ozone, permanganate, and persulfate (USEPA, 2006).
186 Permanganate and persulfate oxidant stability in the subsurface is high, and natural soil oxidant
187 demand for these particular oxidants is potentially low, making ISCO an attractive, cost-effective
188 remediation technique, especially in high hydraulic conductivity material (Ferrarese et al., 2008;
189 O'Mahony et al., 2006; USEPA, 2006). The major limitations of ISCO are related to the hydrogeological
190 conditions: ISCO is more effective in medium to high permeability material, while less effective in low
191 permeability soils (such as clay, loams, glacial tills, hydromorphous soils) as advective transport of
192 oxidants is dramatically decreased (USEPA, 2006). A high content of reduced substances, such as
193 Fe(II), in the soils to be treated may trigger oxidant activation before delivery to the contaminant
194 location (Benner et al., 2002; Petri et al., 2011).

195
196 **Nano zero valent iron (nZVI)** application is a relatively recent remediation technology (Bennett et
197 al., 2010; Elliott and Zhang, 2001; He et al., 2010; Johnson et al., 2013; Kocur et al., 2014) suitable for

198 contaminants that can be treated by chemical reduction. Several laboratory-scale experiments have
199 shown that nZVI can remediate a wide range of contaminants such as nitrate anions (Suzuki et al.,
200 2012; Yang and Lee, 2005), heavy metals (Boparai et al., 2011), pesticides (Satapanajaru et al., 2008;
201 Sayles et al., 1997), PCBs (Wang and Zhang, 1997), chlorinated volatile organic compounds (cVOCs)
202 (Sakulchaicharoen et al., 2010; Song and Carraway, 2005) and radionuclides (Roh et al., 2000). Field-
203 scale studies have used commercially available nZVI (Henn and Waddill, 2006; Krug et al., 2010; Wei
204 et al., 2010) or onsite synthesized nZVI (Bennett et al., 2010; Elliott and Zhang, 2001; He et al., 2010;
205 Kocur et al., 2014; Wei et al., 2010; Zhang, n.d.) that were either unstablized (Elliott and Zhang, 2001;
206 Zhang, n.d.) or polymer coated (0.10%-0.80%, weight/volume) to increase suspension stability
207 (Bennett et al., 2010; He et al., 2010; Henn and Waddill, 2006; Kocur et al., 2014; Wei et al., 2010). A
208 major challenge for nZVI is delivery time: it needs to be delivered to the target treatment zone while
209 reactive and in suspension (e.g. days to a few weeks) (Kocur et al., 2014, 2013; Sakulchaicharoen et
210 al., 2010). As with other chemical in-situ remediation techniques, most successful field studies have
211 been performed in highly permeable soils. Fine grained soils limit nZVI travel distances (Chowdhury
212 et al., 2012).

213
214 **Landfilling**, often referred to as a “dig-and-dump” approach, is considered one of the least
215 sustainable remediation approaches (EEA, 2009). Excavation and landfill disposal of contaminated
216 soil resolves site problems immediately and can be used widely regardless of pollution type and soil
217 characteristics. From a sustainability standpoint, landfills should be considered a last resort for
218 waste. They are expensive in terms of land (area that is allocated for this purpose, with few examples
219 of rehabilitation post-exploitation); fees for hazardous waste disposal are high (US EPA, 2014); and
220 the environmental cost is also considerable, since leachate emission poses an additional threat to
221 groundwater quality. Additionally, landfills constitute a subsurface legacy, transferring waste and
222 groundwater contamination issues to future generations (Bai and Sutanto, 2002; EEA, 2009).

223
224 The following section examines how EK approaches can be used to enhance bioremediation,
225 phytoremediation, ISCO and nZVI, since these are already tried-out combinations. Particular focus is
226 given to bioremediation enhanced by EK (EK-BIO) which has been demonstrated to be highly
227 effective in field-scale studies, and ex-situ soil remediation for metals.

228 229 **3. EK in combination with conventional techniques: How is EK overcoming limitations?**

230 An electric field is applied to a porous media, including saturated soil or sediment, that (i) drives ions
231 present in the media towards one of the electrodes – electromigration, (ii) moves pore water when
232 soil porosity and zeta potential is conducive – electro-osmosis, (iii) mobilizes colloids when soil
233 macropores are sufficiently large to enable their passage – electrophoresis, and that (iv) instigates
234 electrolysis, i.e. generation of an acidic front from the anode and alkaline front from the cathode (Acar
235 and Alshawabkeh, 1993; Lageman, 1993; Mitchell, 1993; Ottosen et al., 2000; Pamukcu and Wittle,
236 1992). Due to the many different phenomena that may occur during the use of an EK approach, EK
237 requires sound engineering to control potential side-effects. Early applications of EK technology did
238 not effectively manage potentially negative processes and hindered early field applications
239 (Alcántara et al., 2012; Gent et al., 2004; Lima et al., 2012b; Reddy and Cameselle, 2009; Simons,
240 1984). More recent research and field experience has overcome early difficulties. EK approaches are
241 significantly favoured over most other in-situ techniques when it comes to low permeability
242 soils (Cameselle and Reddy, 2012; Ghazanfari et al., 2012; Paillat et al., 2000) due to the increased
243 ability to drive remediation amendments to the contaminants, extract contaminants directly, and its
244 limited impacts on soil structure.

245
246 3.1. Sustainability of EK Approaches
247 Sustainability principals are being increasingly recognized as important considerations in the
248 remedy selection process (Hadley and Ellis, 2009). SURF (Sustainable Remediation Forum) defines
249 sustainable remediation as those practices that reduce global impacts at the same time as minimizing
250 local atmospheric effects, potential impacts on worker and community safety, and/or the
251 consumption of natural and energy resources that might be attributable to remediation activities
252 (Hadley and Ellis, 2009; ITRC (Interstate Technology Regulatory Council), 2011). As presented in
253 Table 1, even the most promising techniques present limitations.

254
255 EK inherently uses electrical energy, which can be from fuel sources (Lima et al., 2012a) or from
256 renewable sources (solar, wind) (Zhang et al., 2015). Despite electric energy being at the base of the
257 technique, the sustainability score for EK approaches is high, as the consumption of renewable
258 resources, such as water, is low, use of electricity is small compared to other techniques, it generates
259 minimal waste, and has limited impact on local surface activities. Recent field applications have used
260 a closed water circuit for EK, where only groundwater is used and no external water input is
261 necessary (Mao et al., 2012; Riis et al., 2012).

262

263 3.2. EK Enhancement of Traditional Approaches

264 Bioremediation presents a number of limitations, including the need to control abiotic conditions,
265 mass transfer challenges, bioavailability, bioaugmentation and potentially high operation or long-
266 term re-application costs (Vidali, 2001) (Table 1). EK has been used to surpass some of these
267 challenges, such as control of physico-chemical conditions of soil (Niqui-Arroyo et al., 2006),
268 decrease mass transfer limitations through nutrient transfer and stimulating bioactivity and
269 controlling bacterial population (Mao et al., 2012; Tyagi et al., 2011). DeFlaun & Condee (DeFlaun and
270 Condee, 1997) first enhanced migration of bacteria through soil, while Mao et al. (Mao et al., 2012)
271 applied the same principle as a bioaugmentation tool for the remediation of chlorinated solvents.
272 Different EK principles have been used: electro-osmosis or electrophoresis to mobilize bacteria
273 (Figure 2) (DeFlaun and Condee, 1997; Wick et al., 2004) and electromigration to drive nutrients in
274 low porous materials (Gill et al., 2014; Mao et al., 2012). After successful lab trials, the patented EK-
275 BIO has been applied in-situ (Luo et al., 2006) with considerable success (Riis et al., 2012). Some
276 bioremediation drawbacks (primarily challenges with nutrient distribution in low permeability
277 soils) may thus be addressed remarkably with EK, but evidently target contaminants still need to be
278 biodegradable. EK-BIO has been used primarily on the degradation of organic contaminants, but can
279 also be used for biodegradation of nitrates and perchlorate (Mao et al., 2012; Riis et al., 2012).

280
281 Phytoremediation has shown positive results for the in-situ remediation of both metals and organics.
282 While promoting the degradation of organic contaminants (Kamath et al., 2004; Pilon-Smits, 2005),
283 plants can assimilate and bioaccumulate metals (Ali et al., 2013; Chaney et al., 1997; Weis and Weis,
284 2004). An innovative, however as yet to be field-demonstrated EK combination, is EK-phyto (Figure
285 2) (Aboughalma et al., 2008; Cameselle et al., 2013a). EK can be used to mobilize and redistribute
286 metals in situ, and therefore enhance their transport to the plant roots (Aboughalma et al., 2008). In
287 addition, EK can help stimulate plant growth by mobilizing (micro)nutrients in/to the root zone.
288 Cameselle et al. (Cameselle et al., 2013a) summarized the state-of-the-art regarding the combination
289 of phytoremediation and EK and concluded that laboratory studies yielded the best results with the
290 application of an alternate current (AC) electric field or low direct current (DC) voltage. Therefore,
291 electromigration is used to transfer ions and metals to the root zone while the development of heat
292 (AC electric field) may create ideal assimilation conditions for the plant. Since a low DC field
293 stimulates electroosmosis (Lima et al., 2011), this might explain the improved plant performance
294 under such conditions (Cameselle et al., 2013a).

295

296 Nano-ZVI shows potential to treat redox sensitive contaminants (i.e. organic contaminants in general,
297 or redox sensitive metals) but also stable metal ions, such as divalent Cd^{2+} (Boparai et al., 2011).
298 Polymer stabilized nZVI particles have a net negative surface charge (zeta potential of -48 to -56 mV)
299 (Chowdhury et al., 2012; Kocur et al., 2013) and can adsorb positively charged species. EK has the
300 potential to enhance nZVI transport by electrophoresis. Studies have reported the enhancement of
301 nZVI delivery through coarse and medium grained soils (Chowdhury et al., 2012) while others found
302 that electroosmosis can enhance nZVI delivery through clayey soil (Figure 2) (Gomes et al., 2013;
303 Krishna R. Reddy, 2007). These studies suggest that EK has the potential for enhanced nZVI delivery
304 throughout a number of different soil types to enable subsequent contaminant degradation resulting
305 from nZVI oxidation. EK-nZVI laboratory studies have shown great potential, but field trials are
306 needed.

307
308 Like the other techniques mentioned in this review, ISCO performance in low permeability or highly
309 heterogeneous soils is often a significant unresolved challenge. Commonly used chemical oxidants
310 (permanganate, persulfate) are negatively charged, and highly mobile through electromigration in
311 low permeability soils (Figure 2) (Alshawabkeh, 2009; Nieto Castillo et al., 2012). Electroosmosis or
312 electromigration may also help mobilize specific contaminants (non-charged hydrophobic organics
313 or metals, respectively) by encouraging desorption or transport to different phases (Isosaari et al.,
314 2007; ITRC, 2001). ISCO is a highly commercialized technology, with many different approaches and
315 patented techniques (Table 1; see e.g. (Virkytyte and Sillanpaa, 2002)). EK-TAP (thermal activated
316 persulfate) (Reynolds, 2015) has recently been developed and is currently undergoing field testing
317 at a number of locations in Europe and the USA (ITRC, 2000; Roach and Reddy, 2006; Siegrist et al.,
318 2001). EK-TAP uses a standard DC electric field to migrate persulfate into the contaminated region
319 and then switches to AC to slightly increase the soil and groundwater temperature in the treatment
320 zone ($<40^{\circ}C$) to activate the persulfate.

321
322 Remediation of fine grained soils contaminated with metals has historically been performed through
323 landfilling (Reddy, 2010). Landfilling costs are generally higher and less sustainable in the long-term
324 than any in-situ soil remediation (Table 1), in addition to the environmental costs and sustainability
325 considerations addressed elsewhere in this review. When there is no in-situ solution in sight, an
326 alternative to soil landfilling is off-site treatment of soil. This occurs often where brownfields are
327 considered, either for metal (Merks et al., 2013) or organic pollutants (Gomes et al., 2015; Lima et al.,
328 2012a; Pamukcu, 1994). The electrolytic remediation (EDR) method has been proposed as a fast

329 and continuous in-situ or off-site alternative to landfilling for excavated soil. Promising results
330 through the application of EDR for heavy metal polluted harbour sediment (Nystroem et al., 2005)
331 have been obtained by treating a stirred suspension instead of a stationary matrix. The stirred system
332 for EDR was adapted for the soil remediation, and the soil was suspended in water during the
333 remediation (Gomes et al., 2013; Jensen et al., 2007; Lima et al., 2012b; Ottosen et al., 2012; Sun et
334 al., 2012). Advantages of off-site EDR treatment involve (i) transient and nonlinear changes during
335 remediation that are overcome by the continuous mixing; (ii) the removal rate of heavy metals is
336 faster, as stirring enables mixing and desorption; (iii) easy handling for adding chemical desorbing
337 solutions; and (iv) stirred EDR can be combined with soil washing for minimizing the volume to be
338 treated. In this case the clean coarser fraction and the highly polluted fine fraction are separated
339 during the soil washing and only the fine fraction is treated by EDR. Successful removal rates have
340 been obtained for both metals (Jensen et al., 2007; Ottosen et al., 2012; Sun et al., 2012) and organics
341 (Gomes et al., 2013; Lima et al., 2012b).

342

343 Besides aiding traditional technologies, EK has been applied as:

- 344 - landfill liner enhancement tool, for contaminant confinement (Ouhadi et al., 2010)
- 345 - Extraction of nutrients, like phosphorus, from waste materials (Guedes et al., 2014).
- 346 - Radionuclides control in soils/clays (Maes et al., 1999)
- 347 - EK as enhancement technology for oxidation of emerging contaminants and pesticides (Linley
348 et al., 2014; López-Vizcaíno et al., 2017);
- 349 - And remediating soil by stabilizing/precipitating contaminants in a stable iron-rich band (Cundy
350 and Hopkinson, 2010)

351

352 Landfill-liner enhancements have undergone developments since its first patent (Wittle and Bell,
353 2002). Most recent studies aim at combining chemical stabilisers, such as calcium carbonate, to
354 increase landfill liner contaminant adsorption, with an addition of 28% weight mass of carbonates
355 (Ouhadi et al., 2010). Clay liners are normally used in landfilling. Combining EK with clay liners has
356 been the focus of numerous studies, from predicting the behaviour of swelling clays under such
357 hydro-electric conditions (Lima et al., 2010; Moyné and Murad, 2002) to removing radioactive
358 elements in clays (Kim et al., 2003; Maes et al., 1999; Valdovinos et al., 2016). The latter approach
359 focused on radioactive liquid organic waste (Valdovinos et al., 2016), ^{24}Na (15h) and $^{99\text{m}}\text{Tc}$ (6h) (with
360 71.8% and 61% removal rates respectively) (Valdovinos et al., 2016), and Uranium (23% removal
361 rate) (Kim et al., 2003). Alternatively, more technical aspects have been used regarding EK. For

362 instance, electric fields have been used to generate pH and Eh gradients to induce in situ
363 precipitation of a stable iron-rich band (Cundy and Hopkinson, 2010), or to engineering stabilise soil
364 by dewatering/rewatering soils, (Cundy and Hopkinson, 2010).

365

366 **4. Future avenues for environmental EK**

367 In the decades since EK was first applied to soil with the intention of removing pollutants
368 (Alshawabkeh and Acar, 1992; Hansen et al., 1997; Lageman, 1993; Pamukcu, 1994), EK has
369 developed into a viable alternative for remediation of source contamination, particularly when
370 combined with other conventional techniques for application to low permeable soils. EK approaches
371 have numerous strengths, such as minimizing land disturbance when applied in-situ, reduced costs
372 in terms of energy and transportation, and as an auxiliary tool to a number of new applications
373 (section 3).

374

375 Most site remediation initiatives concern point source pollution. However sources of anthropogenic
376 groundwater pollution are numerous and, many times, diffuse. Plume migration concerns a
377 challenging issue regarding pollution dispersion (B in Figure 1). While the Permeable Reactive
378 Barrier (PRB) concept is currently the leading technology to target downstream pollution effects,
379 previous combination of EK with PRB (EK-PRB) has proven advantageous in terms of treatment, PRB
380 material longevity and cost reduction (Ramírez et al., 2015). The idea was tested at bench and field
381 scale in the 1990ies and coined as the Lasagna Technology (Ho et al., 1995; Sa V. Ho et al., 1999a,
382 1999b). A wealth of experimental studies have been conducted in EK-PRB (Chung and Lee, 2007;
383 Huang and Cheng, 2012; Moon et al., 2005; Vieira dos Santos et al., 2017; Weng et al., 2007; Zhou et
384 al., 2016) with a rekindled interest in the last 2-3 years. The older studies showed that standard PRBs,
385 including zero valent iron filing for the treatment of chlorinated hydrocarbons and chromate in
386 aquifer settings, can be significantly enhanced by coupling to EK. Recent studies show original
387 combinations of pollutants and PRB treatment materials. In particular, the development of biological
388 reactive barriers is proposed (Mena et al., 2016; Ramírez et al., 2015). In these studies bacterial
389 cultures of active sludge from an urban WWTP and coarse mineral soil (kaolinite, gravels) are used
390 as biobarriers. Biological growth was observed in the biobarrier, and under the effect of the electric
391 field, bacteria from the biofilm became detached and were transported through the diesel
392 contaminated soil in both directions (Ramírez et al., 2015). Added surfactant was transported across
393 the treatment zone due to electromigration and electroosmosis, which resulted in diesel
394 emulsification. After two weeks of operation, the combination of biological and EK phenomena

395 resulted in 39% removal of the diesel biodegradable fraction (Ramírez et al., 2015). In fact, EK has
396 proven great applicability in addressing organic contamination, contrary to the attested regarding its
397 first applications for metal contamination. Further studies are expected in addressing LNAPL and
398 DNAPL plumes and exploring solutions for the interface unsaturated-saturated zones (Dresel et al.,
399 2011). There is a need for sustainable solutions to treat these deep locations and EK-PRB can be the
400 answer to these groundwater issues.

401
402 EK techniques are based on the fundamental processes of EM, EO and EP. They had been recognised
403 all by the middle of the XXth century (Casagrande, 1949; Reuss, 1809), but, as pointed out by Yeung
404 (Yeung, 2011), research on the understanding of fundamental phenomena is still needed. Variability
405 in osmotic flow rate, the means of transport of electric current through the soil are still not clearly
406 understood. Here we stress that aspects of soil-contaminant interaction can benefit from further
407 understanding when concerning contaminant removal. It has been shown that, under the influence
408 of an electric field, the phenomena that take place at the interface between matrix and pore fluid are
409 very complex and give rise not only to the dominating processes of EM, EO and EP but to subtle effects
410 related to the complex nature of the solid-liquid and liquid-liquid interfaces present in the porous
411 media. Effects such as diffuse double layer compression, electroosmotic drag or electroosmotically
412 induced displacement give rise to refined or new strategies for soil extraction and remediation. Some
413 new developments emerged that brink between laboratory and pilot scale research:

414
415 - In-situ manipulation of redox-state via EK for redox sensitive metals (e.g. Cr, As, Cu) (Brosky and
416 Pamukcu, 2013; Pamukcu et al., 2004; Sun et al., 2015)

417 In a porous water saturated material, particles typically carry a surface charge that is compensated
418 by the Electric Double Layer (EDL). As electric fields are imposed to this material, a faradaic current
419 is induced which is responsible for electromigration of the unbound aqueous species in the pore
420 water. However, because a conductivity difference exists between the outer part of the EDL and the
421 free pore solution, that outer part – the Diffuse Double Layer (DDL) is compressed while the electric
422 field is applied, leading to a potential difference which adds to the redox potential of the system and
423 can contribute to its reactivity. This EK induced DDL phenomenon has been explored in a laboratory
424 study of Cr(VI) reduction in clay. It showed that a 0.6 mA/cm² applied current was able to increase
425 the reduction rate of the system by a factor of 5 (Sun et al., 2015). While only Cr(VI)/Cr(III) and
426 Cu(II)/Cu(I) redox couples have been studied experimentally up to date (Brosky and Pamukcu,

427 2013; Pamukcu et al., 2004; Sun et al., 2015)), many redox sensitive elements (uranium, selenium,
428 etc) may be candidates for this treatment method.

429
430 - Application of EK in the field of oil extraction and oil transport in water wet porous media (Amba
431 et al., 1964; Chilingar et al., 1968; Ghazanfari et al., 2014, 2012; Ghazanfari and Pamukcu, 2014;
432 Haroun et al., 2013; Pamukcu et al., 2016; Shalabi et al., 2012; Wittle et al., 2011)

433 Viscous coupling between oil and water phase takes place when an electrical interface, similar to
434 that of clay electric double layer, develops between the two phases, provided that the oil has polarity
435 and possesses some functional groups (i.e., O, N, S compounds, carboxylic acids, amides).
436 Electrokinetic transport of hydrocarbon liquids in water wet porous media is governed by the
437 principles of two-phase flow. As the water phase moves by electro-osmosis through water wet
438 porous media, the neighboring oil phase is also transported, to an extent, depending on the strength
439 of the viscous coupling developed between the two phases. More so, in water wet systems of
440 clay/saline water/oil, the reactions of the electrolysis products of saline water (i.e., hydroxyl ions)
441 with the carboxylic acids of oil result in the formation of surfactants at the water/oil interface. This,
442 in turn, reduces significantly the interfacial tension between oil and water further aiding the oil
443 transport by electroosmosis (Ghazanfari et al., 2012; Pamukcu et al., 2016). Another interesting
444 phenomenon is observed when the water and oil phases are strictly immiscible and viscous coupling
445 does not take place (Pamukcu et al., 2016). Then the electroosmotic flow of the water can pass by
446 the nonconductive liquid ultimately displacing the oil in opposite direction of flow. This liquid
447 separation process is shown to be particularly efficient when pore space is restricted. Also it favors
448 increase of permeability of the porous media.

449
450 Laboratory and field applications research show EK as capable of separate and recover oil from
451 water, aqueous sediments and high clay rock formations, advancing EK to be used as a stand-alone
452 or integrative technique for:

- 453 - remediation of oil contaminated soils and sediments where spills have occurred (i.e., coastal
454 sediments, coal gasification sites, abandoned oil production or refinery sites);
- 455 - change in physical properties can be triggered via electrokinetic methods as the classic soil
456 consolidation (Adamson et al., 1966) with the aid of calcium carbonate producing bacteria
457 (Keykha et al., 2014);
- 458 - product extraction, with special interest for mobilization of crude oil in enhanced oil recovery
459 (EOR) processes (Al Shalabi et al., 2012; Amba et al., 1964; Haroun et al., 2013; Shalabi et al.,

460 2012; Wittle et al., 2011) from formations where other extraction methods (i.e., drilling) may
461 not be feasible or environmentally viable.

462
463 As a final remark, we would like to observe that EK is increasingly used in the remediation of
464 emerging contaminants such as pesticides (López-Vizcaíno et al., 2017; Vieira dos Santos et al., 2017),
465 perfluorinated chemicals and radionuclides (control in soils/clays (Maes et al., 1999)). Novel
466 approaches reside in using new catalytic materials (Linley et al., 2014) or as a means for element
467 recycling in waste materials such as phosphorous (Guedes et al., 2014). Because EK acts as a
468 transport tool for dissolved and colloidal particles, as well as solvent through porous media, the
469 possibilities for futures applications are varied and only limited by inventiveness and possibly
470 limited research funds. Therefore, we highlight how EK can act as a versatile and manifold tool for
471 the sustainable treatment of contaminated soil.

472

473 **Acknowledgements**

474 This manuscript is the outcome of a workshop entitled “Environmental Electrokinetics: Advances in
475 Soil and Water Treatment”, held in Waterloo, Ontario, November 10-11, 2014. The workshop was a
476 joint initiative of the Ecohydrology Research Group of University of Waterloo, Geosyntec Consultants
477 and the Geosciences Research Entity (LOG) of Université de Lille - Sciences et Technologies. It was
478 supported by grants from the France–Canada Research Fund (FCRF) and the National Science and
479 Engineering Council (NSERC) Engage Program, and sponsored by Geosyntec and the Faculty of
480 Science and the Water Institute of the University of Waterloo.

481

482 **References**

- 483 Aboughalma, H., Bi, R., Schlaak, M., 2008. Electrokinetic enhancement on phytoremediation in Zn,
484 Pb, Cu and Cd contaminated soil using potato plants. *J. Environ. Sci. Health. A. Tox. Hazard.*
485 *Subst. Environ. Eng.* 43, 926–33. doi:10.1080/10934520801974459
- 486 Acar, Y.B., Alshawabkeh, A.N., 1993. The principles of Electrokinetics. *Env. Sci Technol* 27, 1–5.
- 487 Acar, Y.B., Gale, R.J., Alshawabkeh, A.N., Marks, R.E., Puppala, S., Bricka, M., Parker, R., 1995.
488 Electrokinetic remediation: Basics and technology status. *J. Hazard. Mater.* 40, 117–137.
489 doi:10.1016/0304-3894(94)00066-P
- 490 Adamson, L.G., Chilingar, G. V., Beeson, C.M., Armstrong, R.A., 1966. Electrokinetic dewatering,
491 consolidation and stabilization of soils. *Eng. Geol.* 1, 291–304. doi:10.1016/0013-
492 7952(66)90011-1

493 Al Shalabi, E.W., Ghosh, B., Haroun, M., Pamukcu, S., 2012. The Application of Direct Current
494 Potential to Enhancing Waterflood Recovery Efficiency. *Pet. Sci. Technol.* 30, 2160–2168.
495 doi:10.1080/10916466.2010.547902

496 Alcántara, M.T., Gómez, J., Pazos, M., Sanromán, M.A., 2012. Electrokinetic remediation of lead and
497 phenanthrene polluted soils. *Geoderma* 173–174, 128–133.
498 doi:10.1016/j.geoderma.2011.12.009

499 Ali, H., Khan, E., Sajad, M.A., 2013. Phytoremediation of heavy metals--concepts and applications.
500 *Chemosphere* 91, 869–81. doi:10.1016/j.chemosphere.2013.01.075

501 Alshawabkeh, A.N., 2009. Electrokinetic Soil Remediation: Challenges and Opportunities. *Sep. Sci.*
502 *Technol.* 44, 2171–2187.

503 Alshawabkeh, A.N., Acar, Y.B., 1992. Removal of contaminants from soils by electrokinetics: A
504 theoretical treatise. *J. Environ. Sci. Heal. . Part A Environ. Sci. Eng. Toxicol.* 27, 1835–1861.
505 doi:10.1080/10934529209375828

506 Amba, S.A., Chilingar, G.V., Beeson, C.M., 1964. Use of Direct Electrical Current for Increasing the
507 Flow Rate of Reservoir Fluids During Petroleum Recovery. *J. Can. Pet. Technol.* 3, 8–14.
508 doi:10.2118/64-01-02

509 Bai, R., Sutanto, M., 2002. The practice and challenges of solid waste management in Singapore.
510 *Waste Manag.* 22, 557–567. doi:10.1016/S0956-053X(02)00014-4

511 Benner, S.G., Blowes, D.W., Ptacek, C.J., Mayer, K.U., 2002. Rates of sulfate reduction and metal
512 sulfide precipitation in a permeable reactive barrier. *Appl. Geochemistry* 17, 301–320.
513 doi:10.1016/S0883-2927(01)00084-1

514 Bennett, P., He, F., Zhao, D., Aiken, B., Feldman, L., 2010. In situ testing of metallic iron nanoparticle
515 mobility and reactivity in a shallow granular aquifer. *J. Contam. Hydrol.* 116, 35–46.
516 doi:10.1016/j.jconhyd.2010.05.006

517 Bolan, N., Kunhikrishnan, A., Thangarajan, R., Kumpiene, J., Park, J., Makino, T., Kirkham, M.B.,
518 Scheckel, K., 2014. Remediation of heavy metal(loid)s contaminated soils--to mobilize or to
519 immobilize? *J. Hazard. Mater.* 266, 141–66. doi:10.1016/j.jhazmat.2013.12.018

520 Boparai, H.K., Joseph, M., O'Carroll, D.M., 2011. Kinetics and thermodynamics of cadmium ion
521 removal by adsorption onto nano zerovalent iron particles. *J. Hazard. Mater.* 186, 458–65.
522 doi:10.1016/j.jhazmat.2010.11.029

523 Brosky, R.T., Pamukcu, S., 2013. Role of DDL processes during electrolytic reduction of Cu(II) in a
524 low oxygen environment. *J. Hazard. Mater.* 262, 878–82. doi:10.1016/j.jhazmat.2013.09.032

525 Callender, E., Rice, K.C., 2000. The Urban Environmental Gradient: Anthropogenic Influences on the

526 Spatial and Temporal Distributions of Lead and Zinc in Sediments. *Environ. Sci. Technol.* 34,
527 232–238. doi:10.1021/es990380s

528 Cameselle, C., Chirakkara, R.A., Reddy, K.R., 2013a. Electrokinetic-enhanced phytoremediation of
529 soils: status and opportunities. *Chemosphere* 93, 626–36.
530 doi:10.1016/j.chemosphere.2013.06.029

531 Cameselle, C., Chirakkara, R.A., Reddy, K.R., 2013b. *Chemosphere* Electrokinetic-enhanced
532 phytoremediation of soils : Status and opportunities. *Chemosphere* 93, 626–636.
533 doi:10.1016/j.chemosphere.2013.06.029

534 Cameselle, C., Reddy, K.R., 2012. Development and enhancement of electro-osmotic flow for the
535 removal of contaminants from soils. *Electrochim. Acta* 86, 10–22.
536 doi:10.1016/j.electacta.2012.06.121

537 Casagrande, L., 1949. Electro-osmosis in soils. *Geotechnique* 1, 159–177.

538 Chaney, R.L., Malik, M., Li, Y.M., Brown, S.L., Brewer, E.P., Angle, J.S., Baker, A.J., 1997.
539 Phytoremediation of soil metals. *Curr. Opin. Biotechnol.* 8, 279–284. doi:10.1016/S0958-
540 1669(97)80004-3

541 Chilingar, G. V., El-Nassir, A., Stevens, R.G., 1968. Effect of Direct Electrical Current on Permeability
542 of Sandstone Cores. *J. Pet. Technol.* 22, 830–836. doi:10.2118/2332-PA

543 Chowdhury, A.I.A., O'Carroll, D.M., Xu, Y., Sleep, B.E., 2012. Electrophoresis enhanced transport of
544 nano-scale zero valent iron. *Adv. Water Resour.* 40, 71–82.
545 doi:10.1016/j.advwatres.2012.01.014

546 Chung, H.I., Lee, M., 2007. A new method for remedial treatment of contaminated clayey soils by
547 electrokinetics coupled with permeable reactive barriers. *Electrochim. Acta* 52, 3427–3431.
548 doi:10.1016/j.electacta.2006.08.074

549 Cundy, A.B., Hopkinson, L.J., 2010. Electrokinetic soil remediation and engineering.

550 DeFlaun, M.F., Condee, C.W., 1997. Electrokinetic transport of bacteria. *J. Hazard. Mater.* 55, 263–
551 277. doi:10.1016/S0304-3894(97)00023-X

552 Dresel, P.E., Wellman, D.M., Cantrell, K.J., Truex, M.J., 2011. Review: Technical and policy challenges
553 in deep vadose zone remediation of metals and radionuclides. *Environ. Sci. Technol.* 45, 4207–
554 16. doi:10.1021/es101211t

555 EEA, 2009. Diverting waste from landfill - Effectiveness of waste management policies in the
556 European Union. Copenhagen, Denmark.

557 Elliott, D.W., Zhang, W., 2001. Field Assessment of Nanoscale Bimetallic Particles for Groundwater
558 Treatment. *Environ. Sci. Technol.* 35, 4922–4926. doi:10.1021/es0108584

559 FAO, 2003. Groundwater Management - The search for practical approaches. Rome.

560 Ferrarese, E., Andreottola, G., Oprea, I.A., 2008. Remediation of PAH-contaminated sediments by
561 chemical oxidation. *J. Hazard. Mater.* 152, 128–39. doi:10.1016/j.jhazmat.2007.06.080

562 Gent, D.B., Bricka, R.M., Alshawabkeh, A.N., Larson, S.L., Fabian, G., Granade, S., 2004. Bench- and
563 field-scale evaluation of chromium and cadmium extraction by electrokinetics. *J. Hazard.*
564 *Mater.* 110, 53–62. doi:10.1016/j.jhazmat.2004.02.036

565 Ghazanfari, E., Pamukcu, S., 2014. Mathematical Modeling of Electrokinetic Transport and Enhanced
566 Oil Recovery in Porous Geo-Medi, in: Chilingar, G. V., Haroun, M. (Eds.), *Electrokinetics for*
567 *Petroleum and Environmental Engineers*. John Wiley & Sons, Inc., Hoboken, NJ, USA, pp. 177–
568 236. doi:10.1002/9781118842805

569 Ghazanfari, E., Pamukcu, S., Pervizpour, M., Karpyn, Z., 2014. Investigation of Generalized Relative
570 Permeability Coefficients for Electrically Assisted Oil Recovery in Oil Formations. *Transp.*
571 *Porous Media* 105, 235–253. doi:10.1007/s11242-014-0368-6

572 Ghazanfari, E., Shrestha, R.A., Miroshnik, A., Pamukcu, S., 2012. Electrically assisted liquid
573 hydrocarbon transport in porous media. *Electrochim. Acta* 86, 185–191.
574 doi:10.1016/j.electacta.2012.04.077

575 Ghosh, B., Shalabi, E.W. Al, Haroun, M., 2012. The Effect of DC Electrical Potential on Enhancing
576 Sandstone Reservoir Permeability and Oil Recovery. *Pet. Sci. Technol.* 30, 2148–2159.
577 doi:10.1080/10916466.2010.551233

578 Gill, R.T., Harbottle, M.J., Smith, J.W.N., Thornton, S.F., 2014. Electrokinetic-enhanced bioremediation
579 of organic contaminants: A review of processes and environmental applications. *Chemosphere*
580 107, 31–42. doi:10.1016/j.chemosphere.2014.03.019

581 Giuliano, G., Carone, G., Corazza, A., 1998. Map of Pollution of Groundwater Used for Drinking Water
582 Supply – The BIAS Project. Rome, Italy. doi:GNDCI Publication No. 1875

583 Godfray, H.C.J., Beddington, J.R., Crute, I.R., Haddad, L., Lawrence, D., Muir, J.F., Pretty, J., Robinson, S.,
584 Thomas, S.M., Toulmin, C., 2010. Food security: the challenge of feeding 9 billion people.
585 *Science* 327, 812–8. doi:10.1126/science.1185383

586 Gomes, H.I., Dias-Ferreira, C., Ribeiro, A.B., Pamukcu, S., 2013. Enhanced Transport and
587 Transformation of Zerovalent Nanoiron in Clay Using Direct Electric Current. *Water, Air, Soil*
588 *Pollut.* 224, 1710. doi:10.1007/s11270-013-1710-2

589 Gomes, H.I., Ottosen, L.M., Ribeiro, A.B., Dias-Ferreira, C., 2015. Treatment of a suspension of PCB
590 contaminated soil using iron nanoparticles and electric current. *J. Environ. Manage.* 151, 550–
591 5. doi:10.1016/j.jenvman.2015.01.015

592 Guedes, P., Couto, N., Ottosen, L.M., Ribeiro, A.B., 2014. Phosphorus recovery from sewage sludge
593 ash through an electrodialytic process. *Waste Manag.* 34, 886–92.
594 doi:10.1016/j.wasman.2014.02.021

595 Guerin, T.F., 1999. Bioremediation of phenols and polycyclic aromatic hydrocarbons in creosote
596 contaminated soil using ex-situ landtreatment. *J. Hazard. Mater.* 65, 305–315.

597 Hadley, P., Ellis, D., 2009. Sustainable Remediation White Paper—Integrating Sustainable
598 Principles, Practices, and Metrics Into Remediation Projects. *Remediation*. doi:DOI:
599 10.1002/rem.20210

600 Hansen, H.K., Ottosen, L.M., Kliem, B.K., Villumsen, A., 1997. Electrodialytic remediation of soils
601 polluted with Cu, Cr, Hg, Pb and Zn. *J. Chem. Technol. Biotechnol.* 70, 67–73.

602 Haroun, M.R., Chilingar, G. V., Pamukcu, S., Wittle, J.K., Belhaj, H., Al Bloushi, M.N., 2013. Optimizing
603 Electroosmotic Flow Potential for Electrically Enhanced Oil Recovery (EEORTM) in Carbonate
604 Rock formations of Abu Dhabi Based on Rock Properties and Composition, in: *International
605 Petroleum Technology Conference. International Petroleum Technology Conference.*
606 doi:10.2523/IPTC-13812-MS

607 He, F., Zhao, D., Paul, C., 2010. Field assessment of carboxymethyl cellulose stabilized iron
608 nanoparticles for in situ destruction of chlorinated solvents in source zones. *Water Res.* 44,
609 2360–70. doi:10.1016/j.watres.2009.12.041

610 Henn, K.W., Waddill, D.W., 2006. Utilization of nanoscale zero-valent iron for source remediation—
611 A case study. *Remediat. J.* 16, 57–77. doi:10.1002/rem.20081

612 Higgins, T.E., Halloran, A.R., Petura, J.C., 2008. Traditional and innovative treatment methods for
613 Cr(VI) in soil. *Soil Sediment Contam.* 6, 767–797.

614 Ho, S. V., Sheridan, P.W., Athmer, C.J., Heitkamp, M.A., Brackin, J.M., Weber, D., Brodsky, P.H., 1995.
615 Integrated in situ soil remediation technology: The Lasagna process. *Environ. Sci. Technol.* 29,
616 2528–2534.

617 Huang, Y.-C., Cheng, Y.-W., 2012. Electrokinetic-enhanced nanoscale iron reactive barrier of
618 trichloroethylene solubilized by Triton X-100 from groundwater. *Electrochim. Acta* 86, 177–
619 184. doi:10.1016/j.electacta.2012.03.048

620 Innocenti, I., Verginelli, I., Massetti, F., Piscitelli, D., Gavasci, R., Baciocchi, R., 2014. Pilot-scale ISCO
621 treatment of a MtBE contaminated site using a Fenton-like process. *Sci. Total Environ.* 485–
622 486, 726–38. doi:10.1016/j.scitotenv.2014.01.062

623 Isosaari, P., Piskonen, R., Ojala, P., Voipio, S., Eilola, K., Lehmus, E., Itävaara, M., 2007. Integration of
624 electrokinetics and chemical oxidation for the remediation of creosote-contaminated clay. *J.*

625 Hazard. Mater. 144, 538–48. doi:10.1016/j.jhazmat.2006.10.068

626 ITRC, 2001. Technical and Regulatory Guidance for In Situ Chemical Oxidation of Contaminated Soil
627 and Groundwater. Baton Rouge, LA.

628 ITRC, 2000. Dense Non-Aqueous Phase Liquids (DNAPLs): Review of Emerging Characterization
629 and Remediation Technologies.

630 ITRC (Interstate Technology Regulatory Council), 2011. Technical/Regulatory Guidance Green and
631 Sustainable Remediation: A Practical Framework Prepared by The Interstate Technology
632 & Regulatory Council Green and Sustainable Remediation Team.

633 Jensen, P.E., Ottosen, L.M., Ferreira, C., 2007. Electrodialytic remediation of soil fines (<63 μ m) in
634 suspension—Influence of current strength and L/S. *Electrochim. Acta* 52, 3412–3419.
635 doi:10.1016/j.electacta.2006.03.116

636 Johnson, R.L., Nurmi, J.T., O'Brien Johnson, G.S., Fan, D., O'Brien Johnson, R.L., Shi, Z., Salter-Blanc,
637 A.J., Tratnyek, P.G., Lowry, G. V, 2013. Field-scale transport and transformation of
638 carboxymethylcellulose-stabilized nano zero-valent iron. *Environ. Sci. Technol.* 47, 1573–80.
639 doi:10.1021/es304564q

640 Kamath, R., Rentz, J.A., Schnoor, J.L., Alvarez, P.J.J., 2004. Phytoremediation of hydrocarbon-
641 contaminated soils: principles and applications, in: Vazquez-Duhalt, R., Quintero-Ramirez, R.
642 (Eds.), *Studies in Surface Science and Catalysis* 151. Elsevier B.V., pp. 447–478.

643 Keykha, H.A., Huat, B.B.K., Asadi, A., 2014. Electrokinetic Stabilization of Soft Soil Using Carbonate-
644 Producing Bacteria. *Geotech. Geol. Eng.* 32, 739–747. doi:10.1007/s10706-014-9753-8

645 Kim, K.-H., Kim, S.-O., Lee, C.-W., Lee, M.-H., Kim, K.-W., 2003. Electrokinetic Processing for the
646 Removal of Radionuclides in Soils. *Sep. Sci. Technol.* 38, 2137–2163. doi:10.1081/SS-
647 120021617

648 Koch, A., McBratney, A., Adams, M., Field, D., Hill, R., Crawford, J., Minasny, B., Lal, R., Abbott, L.,
649 O'Donnell, A., Angers, D., Baldock, J., Barbier, E., Binkley, D., Parton, W., Wall, D.H., Bird, M.,
650 Bouma, J., Chenu, C., Flora, C.B., Goulding, K., Grunwald, S., Hempel, J., Jastrow, J., Lehmann, J.,
651 Lorenz, K., Morgan, C.L., Rice, C.W., Whitehead, D., Young, I., Zimmermann, M., 2013. Soil
652 Security: Solving the Global Soil Crisis. *Glob. Policy* 4, 434–441. doi:10.1111/1758-5899.12096

653 Kocur, C.M., Chowdhury, A.I., Sakulchaicharoen, N., Boparai, H.K., Weber, K.P., Sharma, P., Krol, M.M.,
654 Austrins, L., Peace, C., Sleep, B.E., O'Carroll, D.M., 2014. Characterization of nZVI mobility in a
655 field scale test. *Environ. Sci. Technol.* 48, 2862–9. doi:10.1021/es4044209

656 Kocur, C.M., O'Carroll, D.M., Sleep, B.E., 2013. Impact of nZVI stability on mobility in porous media. *J.*
657 *Contam. Hydrol.* 145, 17–25. doi:10.1016/j.jconhyd.2012.11.001

658 Krishna R. Reddy, M.R.K., 2007. Electrokinetic delivery of nanoscale iron particles for insitu
659 remediation of pentachlorophenol-contaminated soils, in: Proceedings of the International
660 Symposium on Geo-Environmental Engineering for Sustainable Development. Proceedings of
661 the International Symposium on Geo-Environmental Engineering for Sustainable
662 Development.

663 Krug, T., O'Hara, S., Watling, M., 2010. Emulsified Zero-Valent Nano-Scale Iron Treatment of
664 Chlorinated Solvent DNAPL Source Areas. Arlington, VA.

665 Kubiak, J.J., Khankhane, P.J., Kleingeld, P.J., Lima, A.T., 2012. An attempt to electrically enhance
666 phytoremediation of arsenic contaminated water. *Chemosphere* 87, 259–264.

667 Laethem, F. Van, Legrand, J., 1993. Impact sur l'environnement des remblais polluants de
668 l'autoroute A22 au nord de Lille. Recherche de solutions. *Bull. liaison des Lab. des Ponts*
669 *Chaussees* 188, 59–66.

670 Lageman, R., 1993. Electroreclamation. *Environ. Sci. Technol.* 27, 2648–2650.

671 Lageman, R., Godschalk, M.S., 2007. Electro-bioreclamation. A combination of in situ remediation
672 techniques proves successful at a site in Zeist, the Netherlands. *Electrochim. Acta* 52, 3449–
673 3453.

674 Liebeg, E.W., Cutright, T.J., 1999. The investigation of enhanced bioremediation through the
675 addition of macro and micro nutrients in a PAH contaminated soil. *Int. Biodeterior.*
676 *Biodegradation* 44, 55–64. doi:10.1016/S0964-8305(99)00060-8

677 Lima, A.T., Kleingeld, P.J., Heister, K., Loch, J.P.G., 2012a. In situ electro-osmotic cleanup of tar
678 contaminated soil - removal of PAHs. *Electrochim. Acta* 86, 142–147.

679 Lima, A.T., Kleingeld, P.J., Heister, K., Loch, J.P.G., 2011. Removal of PAHs from contaminated clayey
680 soil by means of electro-osmosis. *Sep. Purif. Technol.* 79, 221–229.

681 Lima, A.T., Loch, J.P.G., Kleingeld, P.J., 2010. Bentonite electrical conductivity: a model based on
682 series-parallel transport. *J. Appl. Electrochem.* 40, 1061–1068. doi:10.1007/s10800-009-
683 0060-7

684 Lima, A.T., Ottosen, L.M., Heister, K., Loch, J.P.G., 2012b. Assessing PAH removal from clayey soil by
685 means of electro-osmosis and electrodialysis. *Sci. Total Environ.* 435–436, 1–6.
686 doi:10.1016/j.scitotenv.2012.07.010

687 Linley, S., Liu, Y., Ptacek, C.J., Blowes, D.W., Gu, F.X., 2014. Recyclable graphene oxide-supported
688 titanium dioxide photocatalysts with tunable properties. *ACS Appl. Mater. Interfaces* 6, 4658–
689 68. doi:10.1021/am4039272

690 Lockhart, N.C., Stickland, R.E., 1984. Dewatering coal washery tailings ponds by electroosmosis.

691 Powder Technol. 40, 215–221.

692 López-Vizcaíno, R., Risco, C., Isidro, J., Rodrigo, S., Saez, C., Cañizares, P., Navarro, V., Rodrigo, M.A.,
693 2017. Scale-up of the electrokinetic fence technology for the removal of pesticides. Part II:
694 Does size matter for removal of herbicides? *Chemosphere* 166, 549–555.
695 doi:10.1016/j.chemosphere.2016.09.114

696 Luo, Q., Wang, H., Zhang, X., Fan, X., Qian, Y., 2006. In situ bioelectrokinetic remediation of phenol-
697 contaminated soil by use of an electrode matrix and a rotational operation mode.
698 *Chemosphere* 64, 415–22. doi:10.1016/j.chemosphere.2005.11.064

699 Macdonald, R.W., Barrie, L.A., Bidleman, T.F., Diamond, M.L., Gregor, D.J., Semkin, R.G., Strachan,
700 W.M.J., Li, Y.F., Wania, F., Alaei, M., Alexeeva, L.B., Backus, S.M., Bailey, R., Bowers, J.M., Gobeil,
701 C., Halsall, C.J., Harner, T., Hoff, J.T., Jantunen, L.M.M., Lockhart, W.L., Mackay, D., Muir, D.C.G.,
702 Pudykiewicz, J., Reimer, K.J., Smith, J.N., Stern, G., Schroeder, W.H., Wagemann, R., Yunker, M.B.,
703 2000. Contaminants in the Canadian Arctic: 5 years of progress in understanding sources,
704 occurrence and pathways. *Sci. Total Environ.* 254, 93–234. doi:10.1016/S0048-
705 9697(00)00434-4

706 Maes, N., Moors, H., Dierckx, A., De Cannière, P., Put, M., 1999. The assessment of electromigration as
707 a new technique to study diffusion of radionuclides in clayey soils. *J. Contam. Hydrol.* 36, 231–
708 247. doi:10.1016/S0169-7722(98)00146-6

709 Mao, X., Wang, J., Ciblak, A., Cox, E.E., Riis, C., Terkelsen, M., Gent, D.B., Alshawabkeh, A.N., 2012.
710 Electrokinetic-enhanced bioaugmentation for remediation of chlorinated solvents
711 contaminated clay. *J. Hazard. Mater.* 213–214, 311–7. doi:10.1016/j.jhazmat.2012.02.001

712 Margat, J., 2008. *Les Eaux Souterraines dans le Monde*. Ed. UNESCO/BRGM(Bureau de Recherches
713 Géologiques et Minières).

714 McBratney, A., Field, D.J., Koch, A., 2014. The dimensions of soil security. *Geoderma* 213, 203–213.
715 doi:10.1016/j.geoderma.2013.08.013

716 McCutcheon, S.C., Schnoor, J.L., 2004. *Phytoremediation: Transformation and Control of*
717 *Contaminants*. John Wiley & Sons.

718 Mena, E., Villaseñor, J., Rodrigo, M.A., Cañizares, P., 2016. Electrokinetic remediation of soil polluted
719 with insoluble organics using biological permeable reactive barriers: Effect of periodic polarity
720 reversal and voltage gradient. *Chem. Eng. J.* 299, 30–36. doi:10.1016/j.cej.2016.04.049

721 Merckx, O.K., Loch, J.P.G., Lima, A.T., Dijk, J.A., Kreuk, J.F., Kleingeld, P.J., 2013. The Effectiveness of
722 Electro-Remediation of Aged, Metal-Contaminated Sediment in Relation to Sequential
723 Extraction of Metals. *Water, Air, Soil Pollut.* 224, 1–12. doi:10.1007/s11270-013-1667-1

724 Meuser, H., 2012. Soil Remediation and Rehabilitation: Treatment of Contaminated and Disturbed
725 Land (Google eBook). Springer.

726 Mitchell, J.K., 1993. Fundamentals of Soil Behavior. John Wiley & Sons, New York.

727 Moon, J.-W., Moon, H.-S., Kim, H., Roh, Y., 2005. Remediation of TCE-contaminated groundwater
728 using zero valent iron and direct current: experimental results and electron competition
729 model. *Environ. Geol.* 48, 805–817. doi:10.1007/s00254-005-0023-1

730 Moyne, C., Murad, M.A., 2002. Electro-chemo-mechanical couplings in swelling clays derived from a
731 micro/macro-homogenization procedure. *Int. J. Solids Struct.* 39, 6159–6190.
732 doi:10.1016/S0020-7683(02)00461-4

733 Nieto Castillo, A., García-Delgado, R.A., Cala Rivero, V., 2012. Electrokinetic treatment of soils
734 contaminated by tannery waste. *Electrochim. Acta* 86, 110–114.
735 doi:10.1016/j.electacta.2012.04.132

736 Niqui-Arroyo, J.-L., Bueno-Montes, M., Posada-Baquero, R., Ortega-Calvo, J.-J., 2006. Electrokinetic
737 enhancement of phenanthrene biodegradation in creosote-polluted clay soil. *Environ. Pollut.*
738 142, 326–332. doi:10.1016/j.envpol.2005.10.007

739 Niqui-Arroyo, J.-L., Ortega-Calvo, J.-J., 2007. Integrating biodegradation and electroosmosis for the
740 enhanced removal of polycyclic aromatic hydrocarbons from creosote-polluted soils. *J.*
741 *Environ. Qual.* 36, 1444–1451.

742 Nystroem, G.M., Ottosen, L.M., Villumsen, A., 2005. Electrodialytic removal of Cu, Zn, Pb, and Cd from
743 harbor sediment: Influence of changing experimental conditions. *Environ. Sci. Technol.* 39,
744 2906–2911.

745 O'Mahony, M.M., Dobson, A.D.W., Barnes, J.D., Singleton, I., 2006. The use of ozone in the
746 remediation of polycyclic aromatic hydrocarbon contaminated soil. *Chemosphere* 63, 307–14.
747 doi:10.1016/j.chemosphere.2005.07.018

748 Ottosen, L., Rorig-Dalgaard, I., 2009. Method and device for removing an ionic impurity from
749 building structures. WO2009124890.

750 Ottosen, L.M., Hansen, H.K., 1992. Electrokinetic cleaning of heavy metals polluted soil. Intern.
751 Report, Fys. Inst. Geol. og Geotek.

752 Ottosen, L.M., Hansen, H.K., Hansen, C.B., 2000. Water splitting at ion-exchange membranes and
753 potential differences in soil during electrodialytic soil remediation. *J. Appl. Electrochem.* 30,
754 1199–1207.

755 Ottosen, L.M., Hansen, H.K., Laursen, S., Villumsen, A., 1997. Electrodialytic remediation of soil
756 polluted with copper from wood preservation industry. *Environ. Sci. Technol.* 31, 1711–1715.

757 Ottosen, L.M., Jensen, P.E., Kirkelund, G.M., Dias-Ferreira, C., Hansen, H.K., 2012. Electrodialytic
758 remediation of heavy metal polluted soil - Treatment of water saturated or suspended soil, in:
759 Chemical Engineering Transactions. Italian Association of Chemical Engineering - AIDIC, pp.
760 103–108. doi:10.3303/CET1228018

761 Ouhadi, V.R., Yong, R.N., Shariatmadari, N., Saeidijam, S., Goodarzi, A.R., Safari-Zanjani, M., 2010.
762 Impact of carbonate on the efficiency of heavy metal removal from kaolinite soil by the
763 electrokinetic soil remediation method. *J. Hazard. Mater.* 173, 87–94.
764 doi:10.1016/j.jhazmat.2009.08.052

765 Paillat, T., Moreau, E., Grimaud, P.O., Touchard, G., 2000. Electrokinetic phenomena in porous media
766 applied to soil decontamination. *IEEE Trans. Dielectr. Electr. Insul.* 7, 693–704.
767 doi:10.1109/94.879363

768 Pamukcu, S., 2009. Electrochemical Transport and Transformations, in: Reddy, K.R., Cameselle, C.
769 (Eds.), *Electrochemical Remediation Technologies for Polluted Soils, Sediments and*
770 *Groundwater*. John Wiley & Sons, Inc., Hoboken, NJ, USA, pp. 29–65.
771 doi:10.1002/9780470523650

772 Pamukcu, S., 1994. Electrokinetic removal of coal tar constituents from contaminated soils. Final
773 Report, EPRI TR-103320, Proj. 2879-21.

774 Pamukcu, S., Shrestha, R.A., Ribeiro, A.B., Mateus, E.P., 2016. Electrically induced displacement
775 transport of immiscible oil in saline sediments. *J. Hazard. Mater.* 313, 185–92.
776 doi:10.1016/j.jhazmat.2016.04.005

777 Pamukcu, S., Weeks, A., Wittle, J.K., 2004. Enhanced Reduction of Cr(VI) by Direct Electric Current in
778 a Comminated Clay. *Environ. Sci. Technol.* 38, 1236–1241.

779 Pamukcu, S., Weeks, A., Wittle, J.K., 1997. Electrochemical extraction and stabilization of selected
780 inorganic species in porous media. *J. Hazard. Mater.* 55, 305–318.

781 Pamukcu, S., Wittle, J.K., 1992. Electrokinetic removal of selected heavy metals from soil. *Environ.*
782 *Prog.* 11, 241–250.

783 Pazos, M., Rosales, E., Alcántara, T., Gómez, J., Sanromán, M.A., 2010. Decontamination of soils
784 containing PAHs by electroremediation: a review. *J. Hazard. Mater.* 177, 1–11.
785 doi:10.1016/j.jhazmat.2009.11.055

786 Petri, B.G., Watts, R.J., Teel, A.L., Huling, S.G., Brown, R.A., 2011. Fundamentals of ISCO Using
787 Hydrogen Peroxide, in: Siegrist, R.L. (Ed.), *In Situ Chemical Oxidation for Groundwater*
788 *Remediation*. Springer, Berlin, Heidelberg, pp. 33–88. doi:10.1007/978-1-4419-7826-4_2

789 Pilon-Smits, E., 2005. Phytoremediation. *Annu. Rev. Plant Biol.* 56, 15–39.

790 doi:10.1146/annurev.arplant.56.032604.144214

791 Ramírez, E.M., Jiménez, C.S., Camacho, J.V., Rodrigo, M.A.R., Cañizares, P., 2015. Feasibility Of

792 Coupling Permeable Bio-Barriers And Electrokinetics For The Treatment Of Diesel

793 Hydrocarbons Polluted Soils. *Electrochim. Acta.* doi:10.1016/j.electacta.2015.02.201

794 Reddy, K.R., 2010. Technical challenges to in-situ remediation of polluted sites. *Geotech. Geol. Eng.*

795 28, 211–221.

796 Reddy, K.R., Cameselle, C., 2009. *Electrochemical Remediation Technologies for Polluted Soils,*

797 *Sediments and Groundwater.* John Wiley & Sons.

798 Reddy, K.R., Darko-Kagya, K., Al-Hamdan, A.Z., 2011. Electrokinetic remediation of

799 pentachlorophenol contaminated clay soil. *Water. Air. Soil Pollut.* 221, 35–44.

800 Reuss, F.F., 1809. Sur un nouvel effet de l'électricité galvanique. *Mémoires la Soc. Impériale des Nat.*

801 *Moscou* 2, 327–337.

802 Reynolds, D.A., 2015. In situ remediation of soils and ground water containing organic

803 contaminants. US9004816 B2.

804 Ribeiro, A.B., Rodríguez-Maroto, J.M., Mateus, E.P., Gomes, H., 2005. Removal of organic

805 contaminants from soils by an electrokinetic process: The case of atrazine. *Experimental and*

806 *modeling.* *Chemosphere* 59, 1229–1239.

807 Riis, C., Bymose, M., Cox, E., Wang, J., Gent, D., Terkelsen, M., 2012. Succesfull pilot test of

808 electrokinetic enhanced bioremediation (EK-BIO) as an innovative remedial approach for PCE

809 DNAPL source area, in: *NORDROCS 2012: 4th Nordic Joint Meeting on Remediation of*

810 *Contaminated Sites, International Conference.* Oslo, Norway, p. 4 pp.

811 Roach, N., Reddy, K.R., 2006. Electrokinetic delivery of permanganate into low-permeability soils.

812 *Int. J. Environ. Waste Manag.* 1, 4–19.

813 Roh, Y., Lee, S.Y., Elless, M.P., Cho, K.S., 2000. Electro-enhanced remediation of radionuclide-

814 contaminated groundwater using zero-valent iron. *J. Environ. Sci. Heal. - Part A*

815 *Toxic/Hazardous Subst. Environ. Eng.* 35, 1043–1059.

816 Sa V. Ho, *,†, Christopher Athmer, †, P. Wayne Sheridan, †, B. Mason Hughes, †, Robert Orth, †, David

817 McKenzie, †, Philip H. Brodsky, †, Andrew M. Shapiro, ‡, Timothy M. Sivavec, ‡, Joseph Salvo, ‡,

818 Dale Schultz, §, Richard Landis, §, Ron Griffith, § and, Shoemaker§, S., 1999a. *The Lasagna*

819 *Technology for In Situ Soil Remediation. 2. Large Field Test.* doi:10.1021/ES980414G

820 Sa V. Ho, *,†, Christopher Athmer, †, P. Wayne Sheridan, †, B. Mason Hughes, †, Robert Orth, †, David

821 McKenzie, †, Philip H. Brodsky, †, Andrew Shapiro, ‡, Roy Thornton, ‡, Joseph Salvo, ‡, Dale

822 Schultz, §, Richard Landis, §, Ron Griffith, § and, Shoemaker§, S., 1999b. *The Lasagna*

823 Technology for In Situ Soil Remediation. 1. Small Field Test. doi:10.1021/ES980332S
824 Saichek, R.E., Reddy, K.R., 2005. Electrokinetically enhanced remediation of hydrophobic organic
825 compounds in soils: A review. *Crit. Rev. Environ. Sci. Technol.* 35, 115–192.
826 Saichek, R.E., Reddy, K.R., 2004. Evaluation of surfactants/cosolvents for desorption/solubilization
827 of phenanthrene in clayey soils. *Int. J. Environ. Stud.* 61, 587–604.
828 Saichek, R.E., Reddy, K.R., 2003. Effect of pH control at the anode for the electrokinetic removal of
829 phenanthrene from kaolin soil. *Chemosphere* 51, 273–287.
830 Sakulchaicharoen, N., O’Carroll, D.M., Herrera, J.E., 2010. Enhanced stability and dechlorination
831 activity of pre-synthesis stabilized nanoscale FePd particles. *J. Contam. Hydrol.* 118, 117–27.
832 doi:10.1016/j.jconhyd.2010.09.004
833 Satapanajaru, T., Anurakpongsatorn, P., Pengthamkeerati, P., Boparai, H., 2008. Remediation of
834 Atrazine-contaminated Soil and Water by Nano Zerovalent Iron. *Water. Air. Soil Pollut.* 192,
835 349–359. doi:10.1007/s11270-008-9661-8
836 Sayles, G.D., You, G., Wang, M., Kupferle, M.J., 1997. DDT, DDD, and DDE Dechlorination by Zero-
837 Valent Iron. *Environ. Sci. Technol.* 31, 3448–3454. doi:10.1021/es9701669
838 Schnarr, M., Truax, C., Farquhar, G., Hood, E., Gonullu, T., Stickney, B., 1998. Laboratory and
839 controlled field experiments using potassium permanganate to remediate trichloroethylene
840 and perchloroethylene DNAPLs in porous media. *J. Contam. Hydrol.* 29, 205–224.
841 doi:10.1016/S0169-7722(97)00012-0
842 Schultz, D.S., 1997. Electroosmosis technology for soil remediation: laboratory results, field trial,
843 and economic modeling. *J. Hazard. Mater.* 55, 81–91. doi:10.1016/S0304-3894(97)00014-9
844 Semple, K.T., Doick, K.J., Jones, K.C., Burauel, P., Craven, A., Harms, H., 2004. Peer Reviewed: Defining
845 Bioavailability and Bioaccessibility of Contaminated Soil and Sediment is Complicated.
846 *Environ. Sci. Technol.* 38, 228A–231A. doi:10.1021/es040548w
847 Seto, K.C., Fragkias, M., Güneralp, B., Reilly, M.K., 2011. A meta-analysis of global urban land
848 expansion. *PLoS One* 6, e237-77. doi:10.1371/journal.pone.0023777
849 Shah, T., Burke, J., Villholth, K., Angelica, M., Custodio, E., Daibes, F., Hoogesteger, J., Giordano, M.,
850 Girman, J., van der Gun, J., Kendy, E., Kijne, J., Llamas, R., Masiyandama, M., Margat, J., Marin, L.,
851 Peck, J., Rozelle, S., Sharma, B., Vincent, L., Wang, J., 2007. *Water for Food Water for Life, Water
852 for Food Water for Life: A Comprehensive Assessment of Water Management in Agriculture.*
853 Routledge. doi:10.4324/9781849773799
854 Shalabi, E.W. Al, Haroun, M., Ghosh, B., Pamukcu, S., 2012. The Stimulation of Sandstone Reservoirs
855 Using DC Potential. *Pet. Sci. Technol.* 30, 2137–2147. doi:10.1080/10916466.2010.551811

856 Siegrist, R.L., Urynowicz, M.A., West, O.R., Crimi, M.L., Lowe, K.S., 2001. In-Situ Chemical Oxidation
857 Using Permanganate. Batelle Press, Columbus, OH.

858 Simons, R., 1984. Electric field effects on proton transfer between ionizable groups and water in ion
859 exchange membranes. *Electrochim. Acta* 29, 151–158.

860 Song, H., Carraway, E.R., 2005. Reduction of Chlorinated Ethanes by Nanosized Zero-Valent
861 Iron: Kinetics, Pathways, and Effects of Reaction Conditions. *Environ. Sci. Technol.* 39, 6237–
862 6245. doi:10.1021/es048262e

863 Straube, W.L., Nestler, C.C., Hansen, L.D., Ringleberg, D., Pritchard, P.H., Jones-Meehan, J., 2003.
864 Remediation of Polyaromatic Hydrocarbons (PAHs) through Landfarming with Biostimulation
865 and Bioaugmentation. *Acta Biotechnol.* 23, 179–196. doi:10.1002/abio.200390025

866 Sturman, P.J., Stewart, P.S., Cunningham, A.B., Bouwer, E.J., Wolfram, J.H., 1995. Engineering scale-up
867 of in situ bioremediation processes: a review. *J. Contam. Hydrol.* 19, 171–203.
868 doi:10.1016/0169-7722(95)00017-P

869 Sun, T.R., Ottosen, L.M., Jensen, P.E., Kirkelund, G.M., 2012. Electrodialytic remediation of suspended
870 soil--Comparison of two different soil fractions. *J. Hazard. Mater.* 203–204, 229–35.
871 doi:10.1016/j.jhazmat.2011.12.006

872 Sun, T.R., Pamukcu, S., Ottosen, L.M., Wang, F., 2015. Electrochemically enhanced reduction of
873 hexavalent chromium in contaminated clay: Kinetics, energy consumption, and application of
874 pulse current. *Chem. Eng. J.* 262, 1099–1107. doi:10.1016/j.cej.2014.10.081

875 Suzuki, T., Moribe, M., Oyama, Y., Niinae, M., 2012. Mechanism of nitrate reduction by zero-valent
876 iron: Equilibrium and kinetics studies. *Chem. Eng. J.* 183, 271–277.
877 doi:10.1016/j.cej.2011.12.074

878 Tromp, K., Lima, A.T., Barendregt, A., Verhoeven, J.T., 2012. Retention of heavy metals and poly-
879 aromatic hydrocarbons from road water in a constructed wetland and the effect of de-icing. *J.*
880 *Hazard. Mater.* 203–204, 290–8. doi:10.1016/j.jhazmat.2011.12.024

881 Tyagi, M., da Fonseca, M.M.R., de Carvalho, C.C.C.R., 2011. Bioaugmentation and biostimulation
882 strategies to improve the effectiveness of bioremediation processes. *Biodegradation* 22, 231–
883 41. doi:10.1007/s10532-010-9394-4

884 United Nations, 2009. *World Urbanization Prospects The 2009 Revision Highlights*. New York.

885 US EPA, 2014. *Mid-Atlantic Brownfields & Land Revitalization [WWW Document]*. URL
886 <http://www.epa.gov/reg3hwmd/bf-lr/cleanups.html> (accessed 9.17.15).

887 USEPA, 2006. *In-Situ Chemical Oxidation*. Cincinnati, OH.

888 Valdovinos, V., Monroy-Guzmán, F., Bustos, E., 2016. Electrokinetic removal of radionuclides

889 contained in scintillation liquids absorbed in soil type Phaeozem. *J. Environ. Radioact.* 162, 80–
890 86. doi:10.1016/j.jenvrad.2016.05.017

891 Vidali, M., 2001. Bioremediation. An overview. *Pure Appl. Chem.* 73, 1163–1172.
892 doi:10.1351/pac200173071163

893 Vieira dos Santos, E., Sáez, C., Cañizares, P., Martínez-Huitle, C.A., Rodrigo, M.A., 2017. Reversible
894 electrokinetic adsorption barriers for the removal of atrazine and oxyfluorfen from spiked
895 soils. *J. Hazard. Mater.* 322, 413–420. doi:10.1016/j.jhazmat.2016.10.032

896 Virkutyte, J., Sillanpaa, M., 2002. Electrokinetic soil remediation - critical overview. *Sci. Total*
897 *Environ.* 289, 97–121.

898 Wang, C.-B., Zhang, W., 1997. Synthesizing Nanoscale Iron Particles for Rapid and Complete
899 Dechlorination of TCE and PCBs. *Environ. Sci. Technol.* 31, 2154–2156.
900 doi:10.1021/es970039c

901 Wei, Y.-T., Wu, S.-C., Chou, C.-M., Che, C.-H., Tsai, S.-M., Lien, H.-L., 2010. Influence of nanoscale zero-
902 valent iron on geochemical properties of groundwater and vinyl chloride degradation: A field
903 case study. *Water Res.* 44, 131–40. doi:10.1016/j.watres.2009.09.012

904 Weis, J.S., Weis, P., 2004. Metal uptake, transport and release by wetland plants: Implications for
905 phytoremediation and restoration. *Environ. Int.* 30, 685–700.

906 Weng, C.-H., Lin, Y.-T., Lin, T.Y., Kao, C.M., 2007. Enhancement of electrokinetic remediation of
907 hyper-Cr(VI) contaminated clay by zero-valent iron. *J. Hazard. Mater.* 149, 292–302.
908 doi:10.1016/j.jhazmat.2007.03.076

909 Wick, A.F., Haus, N.W., Sukkariyah, B.F., Haering, K.C., Daniels, W.L., 2011. Remediation of PAH-
910 contaminated soils and sediments: A literature review. Blacksburg, VA.

911 Wick, L.Y., Mattle, P.A., Wattiau, P., Harms, H., 2004. Electrokinetic Transport of PAH-Degrading
912 Bacteria in Model Aquifers and Soil. *Environ. Sci. Technol.* 38, 4596–4602.
913 doi:10.1021/es0354420

914 Wittle, J.K., Bell, C.W., 2002. Electrochemical system and method for rendering contaminated
915 electrically conductive material nonhazardous. US 6486376 B2.

916 Wittle, J.K., Hill, D.G., Chilingar, G. V., 2011. Direct Electric Current Oil Recovery (EEOR)—A New
917 Approach to Enhancing Oil Production. *Energy Sources, Part A Recover. Util. Environ. Eff.* 33,
918 805–822. doi:10.1080/15567036.2010.514843

919 Yang, G.C.C., Chen, M.-C., Yeh, C.-F., 2011. Dewatering of a biological industrial sludge by
920 electrokinetics-assisted filter press. *Sep. Purif. Technol.* 79, 177–182.
921 doi:10.1016/j.seppur.2011.02.012

922 Yang, G.C.C., Lee, H.-L., 2005. Chemical reduction of nitrate by nanosized iron: kinetics and
923 pathways. *Water Res.* 39, 884–94. doi:10.1016/j.watres.2004.11.030

924 Yang, L., Nakhla, G., Bassi, A., 2005. Electro-kinetic dewatering of oily sludges. *J. Hazard. Mater.* 125,
925 130–140.

926 Yeung, A.T., 2011. Milestone developments, myths, and future directions of electrokinetic
927 remediation. *Sep. Purif. Technol.* 79, 124–132. doi:10.1016/j.seppur.2011.01.022

928 Yongming, H., Peixuan, D., Junji, C., Posmentier, E.S., 2006. Multivariate analysis of heavy metal
929 contamination in urban dusts of Xi'an, Central China. *Sci. Total Environ.* 355, 176–186.

930 Yukselen-Aksoy, Y., Reddy, K.R., 2012. Effect of soil composition on electrokinetically enhanced
931 persulfate oxidation of polychlorobiphenyls. *Electrochim. Acta* 86, 164–169.
932 doi:10.1016/j.electacta.2012.03.049

933 Zaporozec, A., 2002. *Groundwater contamination inventory - A methodological guide.* Madison.

934 Zhang, S., Zhang, J., Cheng, X., Mei, Y., Hu, C., Wang, M., Li, J., 2015. Electrokinetic remediation of soil
935 containing Cr(VI) by photovoltaic solar panels and a DC-DC converter. *J. Chem. Technol.*
936 *Biotechnol.* 90, 693–700. doi:10.1002/jctb.4359

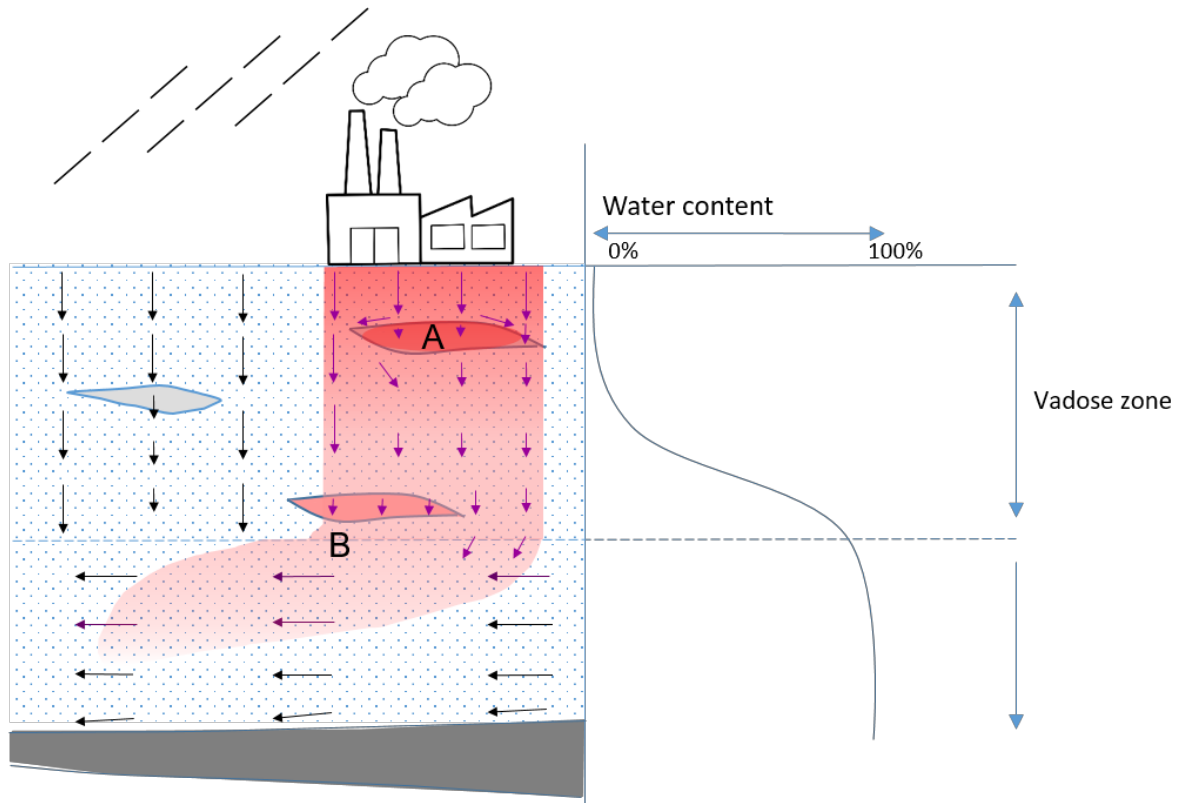
937 Zhang, W., n.d. *Nanoscale Iron Particles for Environmental Remediation: An Overview.* *J.*
938 *Nanoparticle Res.* 5, 323–332. doi:10.1023/A:1025520116015

939 Zhou, M., Zhu, S., Yi, Y., Zhang, T., 2016. An electrokinetic/activated alumina permeable reactive
940 barrier-system for the treatment of fluorine-contaminated soil. *Clean Technol. Environ. Policy*
941 18, 2691–2699. doi:10.1007/s10098-016-1156-5

942

943

944



Legend:

Low permeability layer or lenses

Clay

← Flow direction

Permeable layer

Contamination

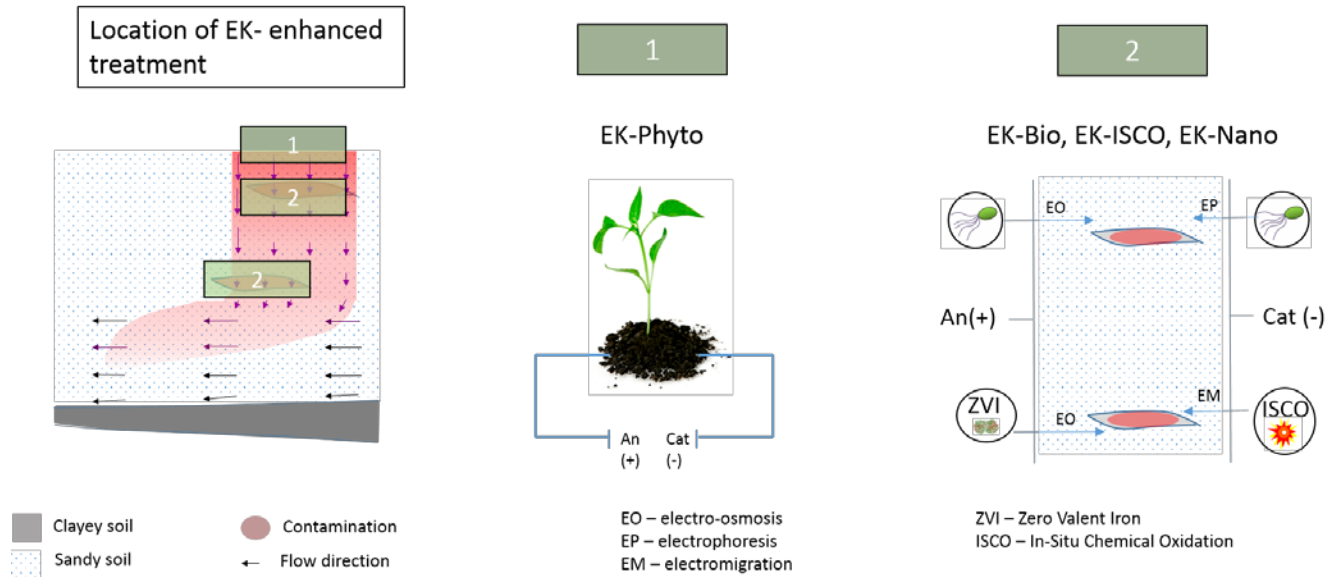
945

946

Figure 1 – Schematics of a plume. A – source pollution; B – plume migration

947

948



949

950 Figure 2 – Schematic representation of electrokinetically enhanced remediation. The left most scheme
 951 shows where, within the subsurface or underground, which EK-enhancement would be better suited.
 952 Scheme 1 (middle frame) represents the combination of EK with phytoremediation. Scheme 2 (right
 953 frame) summarizes how EK would enhance/transport/aid bioremediation, ISCO and nZVI applications.

954

Table 1 – Summary of the main defining characteristics of soil remediation techniques, the main application challenges and how EK can overcome them

	In-Situ Bioremediation	Phytoremediation	Nano-scale Zero Valent Iron (nZVI)	ISCO	Landfilling	Thermal (Lebrón et al., 2013; McGuire et al., 2016)
Target pollutants	Organic contaminants (chlorinated solvents, BTEX and other aromatic compounds, pesticides); some metals(Vidali, 2001)	Metals, metalloids (Ali et al., 2013) and organic contaminants (Seeger et al., 2013)	Organochlorines, nitroaromatics, dyes, phenols, heavy metals, pesticides, and anions (e.g., NO ₃ ⁻¹)	All oxidizable pollutants like organic pollutants (petroleum hydrocarbons, organochlorines, pesticides) and metals(Boparai et al., 2011; Nieto Castillo et al., 2012)	Metals, hazardous contaminants unamenable to in-situ remediation	Volatile organic compounds
Energy requirements	Low; depends on contaminant, bacteria, and technique. Laboratory scale experiments prior to field applications are often required, as well as treatability studies	Low. Plants are placed in-situ and some maintenance is required. Harvesting and replacement of plants	Low to moderate. It requires pumping and delivery of nano-particles to pollutant	Moderate to high; higher than other in-situ (non-thermal) remediation technologies. Electricity application from <0.22 – 44 kWh/ton of ozone activation; 33.33 kWh/m ³ for persulfate activation (USEPA, 2006; Yan et al.)	Moderate to high. Mainly for excavation and transportation	High. The approach requires heating of impacted soil and groundwater to (at a minimum) the boiling point of water)
Water usage	Low to moderate; depends on nutrient requirements	Low to moderate; depends on vegetative cover.	Moderate. ZVI needs to be diluted in water (e.g. 68 m ³ of water with a concentration of 2 g/L of nZVI(US	Moderate. Oxidant needs to be diluted in water (e.g. 8 – 14.5 m ³ of water for persulfate	Low. Landfill leachate needs to be treated, but does not add to water requirements	Low. Water may be used to increase contact with the soil to

		Continued irrigation is required	EPA and USEPA, 2000)). Intensive at the beginning stages	dilution(Yan et al.)). Intensive at the beginning stages		be treated or to cool electrodes.	
Maintenance requirements	Low to moderate (nutrient additions); pH, oxygen content, nutrient regulation(Vidali, 2001)	Low to moderate. Maintenance requirements	Low	Low to moderate depending on oxidant choice, soil/site conditions	Moderate; landfill maintenance and leachate management	Moderate.	
Duration	Long (can last years); depends on microorganisms, contaminant availability (US EPA and USEPA, 2000), and soil heterogeneity	Long (up to 30 years or more(Kamath et al., 2004))	Short (depends on transport of ZVI to target area). However recent studies suggest that nZVI can stimulate bioremediation (Kocur et al., 2015)	Short to moderate. Variable depending on nature of contaminant (sorbed, free phase, solubility, etc.)	Long. Waste is deposited in landfill, contamination transference	Short. Remediation programs are often on the order of weeks to months.	
Disturbance of the subsurface	Mechanical	Low	Low	Low	Low	High	Moderate. Potential for changes to soil structure.
	Biological/chemical	Moderate	Low	Moderate. Change of geochemical conditions	Moderate. Change of geochemical conditions	High	Moderate. Biological polishing following thermal treatment has been demonstrated as effective
Cost (all values in USD)	Wide range of low to high. From \$30 – 100/m ³ (aerobic degradation field demonstration);	Low. \$15 -25/m ³ (calculated based on the treatment of upper 1 m	High. \$255,000 to \$1,400,000 for 2987 ha (numbers based on a specific site)(Gavaskar et al.,	Moderate. Median cost of \$123/ m ³ based on 33 case studies (Krembs et al., 2010) but can	Low to moderate. Landfill disposal costs between \$250 – 350/m ³ (US EPA, 2014), and with possible total costs	High.	

	\$323,000 for a specific site of 480 m ² . (based on project costs) (US EPA and USEPA, 2000; USEPA and US EPA, 1998)	layer)(McCutcheon and Schnoor, 2004)	2005). Nano-particles cost is \$50/kg of nZVI	amount to \$527/m ³ (Innocenti et al., 2014)	(excavation, transport, disposal, labour) amounting to \$650/m ³ (personal communication, Geosyntec)	
Main challenge	Limited to biodegradable compounds, where degradation products may be more persistent or toxic than parent compounds. Long remediation times for some compounds (Singh and Ward, 2004)	Long remediation times. Address contaminant mainly at the relatively shallow subsurface (root zone)	nZVI particles need to be delivered to contaminant zone. Limited reactive lifetime of nZVI	Slow mass transfer, transport, and less control over oxidant due to limiting hydrogeological settings (USEPA, 2006). Rapid oxidant reaction rates (esp. H ₂ O ₂ , Fe ²⁺ , and O ₃) where oxidant demand may be high in some soils/aquifers	Contamination transference: environmental issues concerning soil are merely leachate transferred to landfill site	High energy costs and potential difficulties due to surface infrastructure requirements. Not suitable for inorganic contaminants
How EK can help	Electromigration and/or electro-osmosis can deliver nutrients, microorganisms to target zone	Electromigration can bring pollutants to roots for easier phytostabilisation, rhizofiltration or rhizodegradation(A boughalma et al., 2008; Cameselle et al., 2013a)	Electrophoresis (if sandy soil) and/or electro-osmosis (if clayey soil) can enhance nanoparticle delivery to contaminated regions; faster nanoparticle transport than natural hydraulic conductivity	Oxidant delivery to contaminated regions; faster oxidant delivery than natural hydraulic conductivity	Electro-osmosis can be used to dewater soil/sediment/waste and reduce total residue weight/volume for landfilling; metals or organic contaminants can be extracted ex-situ through the electro-dialytic method and avoid landfill all-together (Pernille E. Jensen et al., 2007; Ana T Lima et al., 2012;	Not amenable to improvement through EK approaches

					Nystroem et al., 2005; Ottoen et al., 2012; Sun et al., 2012)	
--	--	--	--	--	---	--