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# Life-cycle assessment for coal-based methanol production in China

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10 Abstract: More methanol is produced and used in China than in any other country. China has a 11 great deal of coal, less oil, and little gas, so the Chinese government is enthusiastically developing 12 the coal-based chemical industry, of which coal-based methanol production is an important part. 13 Coal-based methanol production strongly affects the environment, so the environmental impacts of 14 coal-based methanol production processes must be assessed. Here, two life-cycle assessment models 15 are established using GaBi6 software, and the models and local data for coal-based methanol 16 production are used to establish a life-cycle inventory. The environmental impacts of two typical 17coal-based methanol production techniques are evaluated using the CML 2001 (mid-point level) 18 method and the Eco-indicator 99 (end-point level) models. The results indicated that less 19 environment harm is caused by producing methanol using the coal coking technology than by 20 producing methanol using the coal gasification technology, especially in terms of acidification, 21 global warming, and photochemical oxidation. In particular, significantly less environmental harm 22 in terms of climate change and radiation is caused by the coal coking technology than by the coal 23 gasification technology. Different sub-processes clearly make different contributions to environmental harm. The results indicated that the methanol production process, heating, and 24 25 desalination are the main sources of environmental harm for both the coal gasification technology 26 and coal coking technology. Importantly, the public engineering process rather than the methanol 27 production process itself was found to determine emissions for the different methanol production

28 methods.

29 Keywords:

30 Coal-based methanol production

- 31 Coal gasification technology (CGT)
- 32 Coal coking technology (CCT)
- 33 Life-cycle assessment (LCA)
- 34 China

#### 35 Highlights:

- 36 1. A comparative LCA for coal-based methanol production was conducted.
- 2. The LCI for coal-based methanol based on site-specific investigations was proposed.
- 38 3. The impacts of two coal-based methanol production techniques were analyzed.
- 39 4. Potential policy implications to lower the related impacts were identified.

40 Abbreviations: CGT, Coal gasification technology; CCT, Coal coking technology; COG, Coke Oven Gas; LCA, Life-cycle assessment; LCI, life-cycle inventory; DALY, Disability Adjusted Life Years; 41 PDF, Potentially Disappeared Fraction; ADP, Abiotic depletion potential; AP, Acidification 42 43 potential; EP, Eutrophication potential; FAETP, Freshwater aquatic eco-toxicity potential; GWP, 44 Global warming potential; ODP, Ozone layer depletion potential; POCP, Photochemical ozone 45 creation potential; TETP, Terrestrial eco-toxicity potential; EQ, Ecosystem quality; AC/NC, Acidification/eutrophication; EC, Eco-toxicity; LC, Land conversion, LU, Land utilization; HH, 46 Human health; CE, Carcinogenic effect; CC, Climate change; OLD, Ozone layer depletion; RA, 47 48 Radiation; IR, Inhalable inorganic matter; OR, Inhalable organic matter.

#### 49 **1** Introduction

50 After the oil crisis in the 1970s, the nations of the world refocused on the coal-based 51 chemical technology. A few coal-rich countries has carried out research on coal liquefaction and gasification technologies, and South Africa has reached the 52 industrialization stage. China is rich in coal resources but has little natural gas or oil 53 (Yang et al, 2001; Xie et al, 2010; Li, 2011), making coal the primary source of energy 54 in China. The Chinese government has promoted the development of the coal-based 55 chemical industry to ensure energy security (National Development and Reform 56 Commission, 2017). The coal-based chemical industry in China has developed rapidly, 57 most notably in the production of methanol (Xu et al, 2007; Liu et al, 2008; Xiao, 58 2010; Liu et al, 2015). 59

Methanol is an important intermediate product that is commonly used to produce 60 61 formaldehyde, methyl tert-butyl ether, acetic acid, dimethyl ether, esters, olefins, and other chemicals. Methanol and its derivatives can be used as fuels, pesticides, and 62 63 medicines and in various industrial processes (Yang et al, 2012). More methanol is produced and consumed in China than in any other country (Shi et al, 2010), and the 64 methanol output capacity of China has increased each year for some time, as shown in 65 66 Fig. 1 and Table A.1. China produced almost 60% of all the methanol produced around the world in 2014 (Futures Daily, 2017). The methanol production capacity of China 67 was  $10 \times 10^6$  t in 2014 but  $30 \times 10^6$  t in 2017. The dominant driving force of this increase 68 has been many methanol-to-olefin/propylene projects starting production and 69 consistent growth of 10%–15% in the use of methanol as a fuel (Wei, 2014). More than 70 80% of the methanol produced in China in 2014 was produced from coal (Xiao, 2015). 71





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Coal-based methanol could provide China with a domestic alternative to imported oil and decrease emissions from conventional fuel used in vehicle engines (Zhang, 2017). However, increased coal-based methanol production could lead to high levels of energy consumption and severe pollution problems that could affect ecosystem and human health (Jackson, 1989, Bhopal et al., 1994; Parodi et al., 2005; Chen, 2008).

There is great controversy about the development of the coal-based methanol 80 81 industry. Yang and Jackson stated that water resource availability seriously constrains 82 the development of the coal-based methanol production industry in China because coal 83 resources are concentrated in a few provinces that have severely limited water resources 84 (Yang et al, 2012). The production of 1 t of coal-based methanol requires about 20 m<sup>3</sup> of fresh water and causes large amounts of wastewater to be produced (Lu, 2005; Chen, 85 86 2008). Unsustainable surface water and groundwater extraction and negative impacts 87 on surrounding ecosystems could occur if the coal-based methanol production capacity continues to increase in coal-rich water-limited regions. 88

89 Life-cycle assessment (LCA) is a quantitative tool that is widely used to identify 90 the environmental impacts of industrial production. LCA methods have been used to 91 study energy consumption and greenhouse gas emissions when methanol is produced from natural gas or biomass (Borjesson et al, 2011; Brynolf et al, 2014; Deniz et al, 92 93 **2016**). It has been found in several previous studies that coal-based methanol used as a 94 fuel for transportation has a larger carbon dioxide footprint than gasoline, diesel, and 95 methanol produced from natural gas (Zhang, 2005; Zhu, 2006; Wei et al, 2007; Du 2012; Zhang, 2012). Few studies of coal-based methanol production (particularly of 96 different production technologies) have been published because of the range of energy 97 98 sources used in developed countries (Li et al, 2010; Xia et al, 2015). Local life-cycle inventory (LCI) databases for coal-based methanol are fundamental to environmental
impact analyses and will facilitate the sustainable development of the Chinese coalbased chemical industry.

102 A cradle-to-gate LCA for typical coal-based methanol production processes is presented here by use of Gabi6 software. The LCA is based on data for specific coal-103 based chemical enterprises. The aim was to build a comprehensive LCI for coal-based 104 methanol production in China to allow the environmental impacts of methanol 105 produced using the coal gasification technology (CGT) and coal coking technology 106 (CCT) to be compared. The results are expected to improve the LCI datasets for 107 intermediate industrial products for use in further research and to provide a quantitative 108 basis for stakeholders to improve the decision-making process. 109

# 110 **2 Methods**

- 111 2.1 Coal-based methanol production processes
- 112 The CGT and CCT are currently the main technologies used to produce coal-based 113 methanol (Cao et al, 2006). Here, these two methods are compared using the LCA
- 114 model. The methods are illustrated schematically in Fig. 2.





116 Fig. 2. Schematic diagrams of the coal gasification technology (CGT) and coal coking technology

117

(CCT) used to produce coal-based methanol. COG is the abbreviation of coke oven gas.

118 Coal-based methanol production processes generally have four steps, syngas 119 generation, syngas purification, methanol synthesis, and methanol rectification. CGT 120 adopted Texaco coal-water slurry gasification technology owing to the mobility and 121 stability of water-coal-slurry. An air separation system supplies oxygen for gasification 122 and nitrogen for use in devices in the plant. Coal slurry and oxygen supplied at a high 123 pressure react in a gasifier at a high temperature and pressure, generating raw syngas 124 containing carbon monoxide, hydrogen, carbon dioxide, and other components. The 125 hydrogen concentration is increased using a shift process. The chemical equation for 126 the shift process is shown below.

127

$$\mathrm{CO} + \mathrm{H}_2\mathrm{O} \to \mathrm{CO}_2 + \mathrm{H}_2. \tag{1}$$

The shift gas is then sent to a purification system in which the carbon dioxide, hydrogen sulfide, and carbon oxysulfide are removed through the Rectisol process and the sulfur recycled through the Super-Claus process. Methanol is then synthesized from a mixture of the clean gas and hydrogen in a Lurgi reactor, in which the reactions shown in Equations 2 and 3 occur. The methanol is then rectified using a three-column rectification process.

134

$$2H_2 + CO \rightarrow CH_3OH, \tag{2}$$

135

$$3H_2 + CO_2 \rightarrow CH_3OH + H_2O. \tag{3}$$

The CCT is analogous to the CGT with some small differences. In the CCT, the raw syngas is coke oven gas (COG) produced during the coking process. This syngas contains carbon monoxide, hydrogen, methane, and other chemicals. The coke oven gas is compressed and subjected to incomplete combustion in a reformer (the incomplete combustion reaction is shown in Equation 4). Unlike in the CGT, the purification system uses a polyethylene glycol dimethyl ether technique.

142

$$CH_4 + H_2 O \to CO + 3H_2.$$
 (4)

143 Equipped with purification system such as desulfurization and dedusting equipment, utilities systems (including heating, desalination, and circulating water 144 systems) provide steam, desalinated water, and circulating water for use throughout the 145 production process. These utilities systems also have impacts on the environment. 146 147 Waste gas is reused as fuel, but large amounts of gas are emitted or leaked into the 148 atmosphere according to the research data and environmental impact assessment reports, including carbon dioxide, sulfur dioxide, nitrogen oxide and et al. Wastewater is 149 150 collected through a drainage pipeline and treated using a sequencing batch reactor 151 biochemical treatment system. Waste solids are sent to nearby landfill sites.

152 2.2 System boundaries

The definitions of the system boundaries required by ISO 14040 standards (International Standard Organization, 2006) will determine the LCA results, especially in a comparative study. As shown in Fig. 3, the system boundaries included raw material preparation, transportation, and on-site production, whereas factory buildings and infrastructure was excluded. The use phase and final disposal phase were not considered because methanol is used in a wide range of applications. Methanol was 159treated in this study as an intermediate product, which can supply data for life-cycle studies of downstream products such as dimethyl ether and propylene. To ensure the 160 integrity of the LCI, upstream processes (including hard coal production, natural gas 161 162 production, and auxiliary material production) were considered, extending to the extraction of natural resources. We only considered the transportation of coal from coal 163 mines to the methanol production plant and the transportation of natural gas through 164 pipelines because of the large amounts used and the long distances the materials are 165 transported. Auxiliary materials are often purchased from different suppliers, so it is 166 hard to define the distances these materials are actually transported. However, auxiliary 167 materials are transported much shorter distances than coal and are much lighter than 168 coal. The first iteration life-cycle impact assessment results indicated that coal 169 transportation contributes <5% of the environmental effects of methanol production. 170 We used the cutoff criterion that a flow or a process was excluded from further use if it 171 contributed <1% of the cumulative environmental effects (Liu et al, 2016). We 172 173 assumed that all auxiliary materials were obtained from local suppliers and ignored the transportation of auxiliary materials. No catalysts were considered because catalysts are 174unchanged during the methanol production process and are only replaced once every 175 few years. As well as the production process, on-site auxiliary engineering processes 176 and public engineering processes were within the boundaries. 177



178

179

Fig. 3. System boundaries for coal-based methanol production.

180 2.3 Functional unit

181 The functional unit was defined as 1 t of refined methanol, which is up to the 182 national superior product standard described in the publication "Methanol for industrial 183 use" (General Administration of Quality Supervision, Inspection and Quarantine 184 of the P.R. China, 2004) and shown in Table B.2.

### 185 2.4 Data collection

Data on material and energy consumption, amounts of methanol produced, and 186 187 pollutant emissions for the on-site production of coal-based methanol were obtained from site-specific investigations and through consulting staff at the CGT plant with 188 annual capacity of 250,000 tons and CCT plant with annual capacity of 300,000 tons. 189 190 Air pollutant data were determined from on-line monitoring data or through calculating 191 mass balances. We assumed that the regular production processes operated at full capacity under normal working conditions. The materials flows of coal coking 192 193 technology and coal gasification technology can be found in Fig. A.1 and Fig B.2. 194 Emissions mainly calculated according to the formula below:

195 
$$\mathbf{E} = \sum_{i=1}^{n} \sum_{j=1}^{m} C_{ij} \times f_j$$

Where E is the emissions in coal-based methanol production, C<sub>i</sub> is the material or energy consumed in the system i, f<sub>j</sub> is the emission factors of material or energy j, n is the quantity of the systems in coal-based methanol production, and m is the quantity of the inputs in coal-based methanol production.

*Raw material data*: coal was purchased from local coal mines obtaining
 underground mining technology. Primary data for the raw material preparation
 processes were obtained from commercial inventory databases, expert estimates,
 previous publications, and GaBi databases (Weidema and Hischier, 2006; PE
 International, 2012). For example, hard coal extraction and production, natural gas
 production, and chemical production data were obtained from GaBi databases.

*Energy data*: steam used in a methanol production process is mainly supplied by the heating system at the plant. Data for material consumption and emissions were provided by the plant staff. Electricity consumption data were not available, so we assumed that all electricity was supplied by the local power grid. The LCI dataset for electricity in the latest GaBi database for the Chinese electricity grid (2009) was used. This database takes the efficiencies of different techniques and energy carriers into account (Liu et al, 2016).

Transportation data: the distances hard coal and natural gas are transported to the plants was obtained through on-site investigations. Basic LCI data for rail transport of a 100 t payload and for transporting gas 100 km through a pipeline were obtained from the transportation models in the GaBi databases.

217 2.5 Allocations

The CGT and CCT both give by-products that some of the energy inputs and environmental emissions should be allocated to. Three different approaches (mass basis, 220 volume basis, and energy basis) are usually used to allocate energy and emissions for a 221 multi-product system. The different states of the different products meant that it was 222 appropriate to allocate energy inputs and environmental burdens during methanol 223 production on an energy basis (Soam et al, 2015). Net calorific values were used to 224 perform energy calculations. The net calorific value was multiplied by the mass of a 225 product to give the energy content of the product. The allocation factor was defined as 226 the contribution of each product to the total energy content. The allocation factors for 227 the products are shown in Table 1.

#### 228 Table 1

Product	Average net	CGT			CCT		
	calorific value (kJ/kg) <sup>a</sup>	Mass(kg)	Energy content	Allocation factor	Mass (kg)	Energy content	Allocation factor
			(MJ)			(MJ)	
Methanol	20090	1000	20090	80.92%			
Metallurgical coke	28435				17053	484902.06	79.67%
Coke oven gas	44952				1868	83970.34	13.80%
Tar	33453				870	29104.11	4.78%
Crude benzene	41816				244	10203.10	1.68%
Sulfur	9260	6.37	58.99	0.24%	51	472.26	0.08%
Steam	3763	1243	4677.41	18.84%			

229 Allocation factors for the products of coal-based methanol production.

<sup>a</sup> Source: General Administration of Quality Supervision, Inspection and Quarantine of the P.R.

231 China (2008), Grote and Antonsson (2009), Wu et al, (2016).

#### 232 2.6 LCI and life-cycle impact assessment methods

233 The objective was to evaluate and compare the environmental impacts of the two coal-based methanol production methods at the mid-point (MP) and end-point (EP) 234 235 levels. The MP level assessment is problem-oriented, environmental problem targeted, 236 and simulates the environmental mechanisms between the pollutant being emitted and causing damage but neglects the effects of environmental damage on resources, 237 238 ecosystems, and humans (Drever et al, 2003). The EP level assessment is a damageoriented method that simulates the negative impacts on human and ecosystem health 239 240 and on resources.

The CML 2001 method was used to assess MP level environmental impacts (Centre for Environmental Studies, 2001). The characteristics of the coal-based chemical industry and the attention it receives led us to choose eight typical impact categories in the CML 2001 method as indicators at the MP level (Guinee et al, 2001). These impact categories, shown in Table 2, were focused on the environmental 246 problems caused by coal-based methanol production.

247 For the EP level assessment, 10 impact categories, shown in Table 2, were identified using the Eco-indicator 99 (EI99) method for evaluating the impacts on 248 249 ecosystem and human health (Luan, 2004; Goran et al, 2009). Unlike the CML2001 250 method, the EI99 method includes three damage categories, human health, ecosystem quality, and resources. The EI99 method is used to evaluate the EP environmental 251 impacts on human health and ecosystems caused by dust and toxic substance emissions 252 253 from coal-based chemical plants (Goedkoop et al, 1999). We used human health and ecosystem quality to assess the damage caused by coal-based methanol production. 254 Human health was expressed as disability adjusted life years (DALY) that is the sum 255 of years of life lost and years of life disabled, and ecosystem quality was expressed as 256 257 a potentially disappeared fraction (PDF) that is a fraction of species richness that may 258 be potentially lost due to an environmental mechanism.

The EI99 method contains three methods, based on culture theory, for reflecting stakeholders attitudes to the environment. These methods are the egalitarian perspective, the hierarchical perspective, and the individualist perspective. We used the hierarchical perspective, which is widely used and the closest to the actual situation (Hofsterter, 2000; Huisman, 2003).

264 **Table 2** 

265	Impact categories	for the CI	ML2001 an	d Eco-indicator	99 (EI99)	models.
-----	-------------------	------------	-----------	-----------------	-----------	---------

Method	Problem/Damage	Category
CML2001	Depletion of abiotic resources	Abiotic depletion potential
		(ADP)
	Acidification	Acidification potential (AP)
	Eutrophication	Eutrophication potential (EP)
	Eco-toxicity	Freshwater aquatic eco-
		toxicity potential (FAETP)
		Terrestrial eco-toxicity
		potential (TETP)
	Climate change	Global warming potential
		(GWP)
	Stratospheric ozone depletion	Ozone layer depletion
		potential (ODP)
	Photo-oxidant formation	Photochemical ozone creation
		potential (POCP)
EI99	Ecosystem quality (EQ)	
	Damage caused by the combined effect of	Acidification/eutrophication
	acidification and eutrophication	(AC/NC)
	Damage caused by eco-toxic	Eco-toxicity (EC)
	Damage caused by land conversion	Land conversion (LC)

Method	Problem/Damage	Category
	Damage caused by land utilization	Land utilization (LU)
	Human health (HH)	
	Carcinogenic effects on humans	Carcinogenic effect (CE)
	Damage caused by climate change	Climate change (CC)
	Effects caused by ozone layer depletion	Ozone layer depletion (OLD)
	Effects caused by lonising radiation	Radiation (RA)
	Respiratory effects caused by inorganic	Inhalable inorganic matter
	substances	(IR)
	Respiratory effects caused by organic substances	Inhalable organic matter (OR)

#### 266 **3 Results and discussion**

#### 267 3.1 LCI results

268 The complete inventory, from cradle to gate, for coal-based methanol production 269 using the CGT and CCT is shown in Table 3. The inventory includes the methanol 270 production process, auxiliary engineering processes, public engineering processes, and upstream and downstream processes, but excludes the consumption of resources or 271 272 pollutant emissions contributing <1% of the total input and output mass. As shown in 273 the LCI, the material types for the CGT and CCT were essentially the same. There were 274 13 types of input materials directly extracted from the environment (including 275nonrenewable energy and resources and renewable resources). There were four types 276 of output materials, stock, discharges to the air, discharges to water, and discharges to 277 soil. Stock includes waste, gangue, degraded products, and surface soil (accumulation). Material discharged to the air was dominated by 13 heavy metals, 19 inorganic 278 279 chemicals, and 18 organic chemicals. Material discharged to water was dominated by 280 14 heavy metals, 31 inorganic chemicals, five organic chemicals, and radioactive matter 281 (Ra). Agricultural and industrial soils are affected by emissions. Material discharged to 282 agricultural soil was dominated by three heavy metals and four inorganic chemicals. Material discharged to industrial soil was dominated by one heavy metal and six 283 284 inorganic chemicals.

285 **Table 3** 

Life-cycle inventory for the production of 1 t of coal-based methanol.

	-				
Subcategory/Unit	CGT	CCT	Subcategory/Unit	CGT	CCT
Unrenewable energy					
Crude oil/kg	20.21	5.54	Lignite/t	8.41	2.28
Hard coal/t	0.50	0.20	Natural gas/kg	73.38	10.22
Unrenewable resource					
Inert rock/t	3.64	0.72	Limestone/kg	232.24	28.05
Renewable resource					

Fresh water/t	64.81	13.15	River water/t	2184.23	276.42
Underground water/t	2.94	0.77	Sea water/t	10.89	1.34
Lake water/t	135.52	18.15	Air/t	14.75	5.45
Rain/t	2.64	0.44			
Waste(stock)					
Highly radioactive waste/g	2.75	0.34	Surface soil (accumulation)/kg	3671.90	721.09
Low radioactive waste/g	33.97	4.16	Degraded product/kg	15.25	3.53
Moderately radioactive waste/g	16.15	1.98	Gangue/kg	2.60	0.29
Radioactive gangue/kg	1.76	0.22	Waste/kg	9.61	2.37
Hazardous waste/g	7.79	1.56			
Heavy metal to air <sup>a</sup>					
Sb/g	1.02	0.47	Mn/g	2.64	1.17
As/g	0.55	0.24	Ni/g	2.98	1.38
Cr/g	1.83	0.84	Se/g	0.55	0.22
Co/g	0.60	0.28	Sn/g	1.57	0.71
Cu/g	0.92	0.42	V/g	3.12	1.40
Pb/g	4.09	1.89	Zn/g	5.26	2.38
Inorganic matter to air <sup>a</sup>					
Ammonia (NH <sub>3</sub> )/kg	0.02	0.04	Hydrogen fluoride (HF)/g	3.87	1.16
Ba/g	12.28	5.59	Hydrogen sulphide (H <sub>2</sub> S)/g	40.78	10.77
Boron compound/g	3.90	1.34	Nitrogen (N <sub>2</sub> )/t	3.45	1.44
Bromine (Br)/g	1.85	0.75	Nitrogen oxide (NO <sub>x</sub> )/kg	32.17	5.21
Carbon dioxide (CO <sub>2</sub> )/t	17.53	2.84	Oxygen (O <sub>2</sub> )/kg	17.54	2.14
Carbon monoxide (CO)/kg	1.66	0.91	Ozone (O <sub>3</sub> )/g	0.11	0.04
Chlorine (Cl)/g	0.42	0.12	Sulfate/g	0.44	0.13
Fluoride/g	0.30	0.04	Sulphur dioxide (SO <sub>2</sub> )/kg	47.86	5.91
Hydrogen (H <sub>2</sub> )/g	32.40	3.51	Steam/t	10.54	4.21
Chlorine hydride (HCl)/g	45.42	6.92			
Organic matter to air <sup>a</sup>					
NMVOC/kg	0.56	0.16	Methane/kg	4.36	1.63
VOC/g	1.33	0.11			
Other emissions to air <sup>a</sup>					
Other emission/t	55.70	3.10			
Heavy metal to water <sup>b</sup>					
As (+V)/g	15.89	6.43	Iron ion/g	2.43	0.72
Cd (+II)/g	57.52	23.33	Pb (+II)/g	5.30	2.10
Cr (+III)/g	6.41	2.59	Mn (+II)/g	16.82	6.68
Cr (+VI)/g	6.40	2.59	Hg (+II)/g	8.95	3.63
Co/g	15.76	6.39	Ni (+II)/g	16.09	6.51
Cu (+II)/g	5.19	2.09	Tl/g	15.76	6.39
Fe/kg	0.34	0.04	Zn (+II)/g	9.16	3.69
Inorganic matter to water <sup>b</sup>					
Al(+III)/g	10.37	2.50	Neutral salt/g	74.21	6.26
Ammonia (NH <sub>3</sub> )/g	23.07	8.61	Nitrate/kg	0.80	0.31

Ammonium/g	12.42	1.39	Nitrogen/kg	0.68	0.28
Ba/g	4.51	0.56	Phosphate/kg	0.08	0.03
Boron/g	4.11	1.48	Phosphorus/kg	0.08	0.03
Bromate/g	1.60	0.50	Kalium/g	5.07	0.85
Brmine/g	3.07	0.34	Potassium-ion/g	0.60	4.99
Carbonate/g	27.79	7.36	Na(+I)/kg	0.69	0.14
Chlorate/g	12.24	3.82	Na <sub>2</sub> SO <sub>4</sub> /g	12.48	1.54
Chloride/kg	9.45	3.05	Sodion/kg	0.09	0.09
Cl (dissolved)/g	8.22	1.00	Strontium/g	3.49	1.17
Fluoride/kg	1.79	0.65	Sulfate/kg	2.46	0.69
Lithium/g	7.70	0.65	Sulfide/g	4.48	1.25
Lithium-ion/g	7.31	0.98	Sulfite/g	1.22	0.44
Magnesium/kg	0.15	0.05	Sulphur/g	0.05	1.65
Magnesium-ion/g	4.55	0.38			
Organic matter to water <sup>b</sup>					
Hydrocarbon/g	17.41	2.47	Propylene/g	8.46	0.70
Methyl alcohol/g	1.82	0.19	Suspended solids/g	0.71	0.16
Petroleum/g	5.71	1.42			
Radioactive matter to water <sup>b</sup>					
Radium(Ra226)/t	136.30	16.84			
Other emission					
Water/g	91.54	29.64	Rainwater/kg	0.08	0.01
Clear water/g	3.87	1.44	Turbine drainage/t	2253.54	279.29
Cooling water/g	58.92	13.56	Process waste water/t	6.87	2.44
Heavy metal to agricultural soil <sup>c</sup>					
Fe/mg	256.09	93.80	Zn(+II)/mg	143.73	32.05
Pb(+II)/mg	55.33	12.33			
Inorganic matter to agricultural soil <sup>c</sup>					
Al/mg	37.51	13.99	Si/mg	91.57	30.02
Ca/mg	225.09	76.10	S/mg	32.93	12.14
Heavy metal to industrial soil <sup>c</sup>					
Fe/g	0.52	0.15			
Inorganic matter to industrial soil <sup>c</sup>					
Al/mg	67.95	15.53	Cl/mg	32.07	51.24
Ca/mg	268.25	56.72	Mg/mg	54.61	11.44
Chloride/g	5.75	0.73	Na/mg	147.46	32.62

<sup>a</sup> Source: CO<sub>2</sub> and CH<sub>4</sub> emissions are derived from the Ecoinvent Database v2.2 (Weidema and

288 Hischier, 2006). The emissions to air of heavy metals, other inorganic matters and organic matters

289 come from the companies that represent the current domain coal-based methanol production in

290 **China**.

<sup>b</sup> Source: COD, ammonia, sulphide, oil and SS are based on the Emission Factor Manual for the 1<sup>st</sup>

292 National Census of Industrial Pollution Sources (The State Council of China).

<sup>293</sup> <sup>c</sup> Source: The emissions to soil come from the companies that represent the current domain coal-

294 based methanol production in China.

- 3.2. MP level analysis
- 3.2.1. Comparative total impact analysis

The life-cycle impact assessment results at the MP level for the CGT and CCT are 297 298 shown in Table 4. The environmental impacts in all impact categories were much greater for the CGT than for the CCT, especially for the acidification potential (AP), 299 300 global warming potential (GWP), and photochemical ozone creation potential (POCP). 301 The CGT to CCT method ratios for AP, GWP, and POCP were 6.7, 6.1, and 5.6, respectively. The smallest environmental burden gap between the two methods was for 302 303 the terrestrial eco-toxicity potential (TETP), the CGT to CCT method ratio for which was 2.5. The main reason for the difference in impacts was that the coke oven gas is 304 305 only one of numerous products of the coke production system. We distributed the 306 environmental impacts of the upstream processes among the products and by-products on an energy basis. For the coking process, the energy content of coke oven gas 307 308 accounted for 13.8% of the total energy content. Using 1 t of coal-based methanol as the functional unit, we determined that methanol production contributed 13.8% of total 309 310 impacts of the coking process, making the environmental performance better for the CCT than for the CGT. 311

- 312 **Table 4**
- 313 Integral environmental impact assessment result at the mid-point level.

Categories	Unit	CGT	CCT	CGT/CCT
ADP	MJ	8.68E+04	2.45E+04	3.5
AP	kg SO <sub>2</sub> -eq	7.37E+01	1.10E+01	6.7
EP	kg Phosphate-eq	4.98E+00	1.33E+00	3.7
FAETP	kg DCB-eq	3.65E+02	1.47E+02	2.5
GWP	kg CO <sub>2</sub> -eq	1.77E+04	2.89E+03	6.1
ODP	kg R11-eq	1.37E-05	4.15E-06	3.3
POCP	kg Ethene-eq	3.53E+00	6.33E-01	5.6
TETP	kg DCB-eq	1.89E+01	7.98E+00	2.4

314 3.2.2. Comparative process impact analysis

We used eight environmental impact categories for the CCT and CGT for five processes (raw material preparation, transportation, auxiliary engineering processes, public engineering processes, and methanol production). The relative contributions of the processes to each impact category for the two methods are shown in Fig. 4.



319 320

Fig. 4. Environmental impact assessment for specific processes at the mid-point level.

For raw material preparation, the CGT impacts were about 73% higher than the 321 322 CCT impacts because more energy and resources are consumed by the CGT. For 323 methanol production, the abiotic depletion potential (ADP), AP, GWP, and POCP impacts were lower for the CCT than for the CGT, but not by >10% except for GWP, 324 which was 36% lower. The eutrophication potential (EP), freshwater aquatic eco-325 toxicity potential (FAETP), and ozone layer depletion potential (ODP) impacts were 326 25%-42% higher for the CCT than for the CGT. TETP emissions were only 6% higher 327 328 for the CCT than for the CGT. Unlike for the other four processes, there were no clear differences in the impacts of methanol production process using the two methods. 329 Auxiliary engineering processes had 53% higher impacts in all categories for the CGT 330 331 than for the CCT because coal gasification consumes large amounts of oxygen (to react 332 with the coal-water slurry) whereas the CCT uses a product of coking to produce

333 methanol and does not require large amounts of additional oxygen. Transportation had approximately 73% lower impacts in all categories for the CCT than for the CGT 334 because less coal is used in the CCT than in the CGT and no natural gas is used in the 335 CCT. Public engineering processes had 90% lower impacts in all categories for the 336 CCT than for the CGT. The public engineering processes for both methods and 337 environmental impacts per product unit were the same, so the difference was mainly 338 caused by steam, desalinated water, and circulating water, each of which are used in 339 smaller quantities in the CCT than in the CGT. 340

- 341 3.2.3. Comparative sub-process impact analysis
- The pollution sources were explored further by splitting the CGT and CCT into eight sub-processes, as shown in Fig. 5.



344

Fig. 5. Sub-processes in the coal gasification technology (CGT) and coal coking technology
 (CCT).

The contributions of the sub-processes to the environmental impact categories for 347 the CGT are shown in Fig. 6. Methanol production, heating, and desalination were 348 found to be the main contributors to the environmental burden. Methanol production 349 350 contributed 5.26%-30.32% of the impact categories, contributing least to ADP and most to TETP. Heating contributed 3.54%–84.90% of the impact categories (>60% to 351 AP, EP, GWP, and POCP). Desalination contributed 5.39%-73.64% of the impact 352 categories, and strongly contributed to FAETP, ODP, and TETP. In addition to 353 transportation itself, natural gas transportation included gas exploitation and processing. 354 Natural gas is mainly transported by pipeline, meaning only electricity is consumed, 355 and gas exploitation and processing have lower environmental impacts than do coal 356 mining and processing. Natural gas is a "clean energy" that maintain the operation of 357 steam superhearter in the CGT, so its demands are rather small, so the contributions of 358 natural gas transportation to the impact categories were very small. 359





Fig. 6. Contributions of the coal gasification technology sub-processes to the environmental
 impacts.

The contributions of the sub-processes to the impacts of the CCT are shown in Fig. 7. The coking process contributed >20% of the impacts except for ADP. Methanol production contributed 13.00%–54.98% of the impact categories, contributing most to FAETP and least to ADP. Heating contributed considerably more to AP, EP, GWP, and POCP than to the other categories.



368



370 **3.3.** EP level

371 3.3.1. Comparative total impact analysis

The first iteration indicated that land transformation and ozone layer depletion (indicating ecosystem quality and human health, respectively) made much smaller contributions than the other categories, so they were excluded from further analysis. The results of the impact assessment at the EP level are shown in Table 5.

376 **Table 5** 

377 Environmental performance at the end-point level

Category	Damage	Unit	CGT	CCT	CGT/CCT
EQ	AC/NC	PDF*m <sup>2</sup> *a	2.34E+02	5.10E+01	4.6
	EC	PDF*m <sup>2</sup> *a	9.08E+01	4.07E+01	2.2

Category	Damage	Unit	CGT	CCT	CGT/CCT
	LU	PDF*m <sup>2</sup> *a	9.05E+00	1.67E+00	5.4
HH	CE	DALY	5.18E-03	2.10E-03	2.5
	CC	DALY	3.70E-03	6.05E-04	6.1
	RA	DALY	3.00E-06	3.80E-07	7.9
	IR	DALY	7.06E-03	1.59E-03	4.4
	OR	DALY	7.09E-07	2.67E-07	2.7
EQ		PDF*m <sup>2</sup> *a	3.34E+02	9.33E+01	3.6
HH		DALY	1.59E-02	4.29E-03	3.7

As shown in Fig. 8, somewhat more environmental damage was found for the CGT 378 than for the CCT, so the CCT will be more acceptable. The results matched the results 379 380 at the MP level. In general, the values for the damage caused by the CGT to ecosystem quality and human health were  $334 \times \text{potentially}$  disappeared fraction  $\times \text{m}^2 \times \text{a}$  and 381 0.016 disability adjusted life years, respectively, which were about 3.6 and 3.7 times 382 383 higher, respectively, than the values for the CCT. Radiation and climate change (both 384 affecting human health) were affected very differently by the different methods, and 385 land utilization, acidification/eutrophication, and inhalable inorganic matter were 386 affected somewhat differently by the different methods. Eco-toxicity, carcinogenic effects, and inhalable organic matter were affected similarly by the different methods. 387



388 389

390

**Fig. 8.** Environment performances for the coal gasification technology (CGT) and coal coking technology (CCT).

In terms of ecosystem quality (see Fig. 9), acidification/eutrophication and ecotoxicity were the main categories affected by the CGT (accounting for 70.06% and 27.19% of the impact categories, respectively). The construction of factory buildings and infrastructure was not included, and the land-use value was relatively small (2.71%). The buildings and infrastructure of coal-based chemical production cover large areas of land. When the impacts of coal-based chemical production on ecosystem and human health were taken into account, much larger areas of land around the plants were found to be damaged. Further research is required to acquire more data and incorporate factory building construction into the system. For the CCT, the relative contributions were similar. The contribution of acidification/eutrophication (54.66%) was less than for the CGT (70.06%). Eco-toxicity was 16% higher for the CCT than for the CGT.



Fig. 9. Comparative damage to ecosystem quality for the (a) coal gasification
 technology (CGT) and (b) coal coking technology (CCT).

402

413

In terms of human health (see Fig. 10), different impact categories were affected 405 to different degrees, but carcinogenic effects, climate change, and inhalable inorganic 406 matter were the main factors affected by both methods. For example, the CGT affected 407 408 radiation and inhalable organic matter very little, but inhalable inorganic matter was affected most (44.40% of the total impact), followed by carcinogenic effects and 409 inhalable inorganic matter. The CCT affected carcinogenic effects the most (48.95% of 410 the total impact), followed by inhalable inorganic matter. The impact on climate change 411 was relatively small. 412



Fig. 10. Comparative results for damage to human health from the (a) coal gasification technology
 (CGT) and (b) coal coking technology (CCT).



different degrees. The two methods caused damage to different categories affecting
human health. Therefore, different priorities will need to be used when designing
mitigation measures for the two methods.

420 3.3.2. Comparative process impact analysis

In the same way as for the MP level, the impacts of the two methods were 421 422 calculated for five processes (raw material preparation, transportation, auxiliary engineering processes, public engineering processes, and methanol production). As 423 424 shown in Fig. 11, the CGT had about 73% stronger impacts than the CCT on all the raw material preparation categories. This was because more energy and resources are 425 426 consumed during the CGT than during the CCT. The environmental impacts (except 427 for climate change and inhalable organic matter) were lower for the CGT methanol 428 production process than for the CCT methanol production process. The auxiliary 429 engineering process, transportation processes, and public engineering processes all had stronger impacts for the CGT than for the CCT. The main reason for this may have been 430 431 that more environmental damage is caused by the CGT than the CCT. The same conclusions were drawn from the MP level results. The results for these supporting 432 433 processes were determined directly from the demand and cyclic use rates for raw 434 materials and accessories for the entire methanol production process.





438 3.3.3. Comparative sub-process impact analysis

435

The contributions of the CGT sub-processes to the environmental impacts at the 439 440 EP level are shown in Fig. 12. The environmental damage caused by methanol production, heating, and desalination were all comparatively serious. Ecosystem quality 441 damage was caused mainly by heating, which accounted for 62.31% of the total impact. 442 443 Desalination and methanol production were the next biggest contributors, contributing 19.90% and 14.41%, respectively, of the total impact. Human health damage was 444 caused mainly by heating, followed by desalination and methanol production. The other 445 sub-processes made smaller contributions to these two impact categories. Natural gas 446 transportation contributed <0.01% of almost all the impact categories, so could be 447 448 considered to be negligible.





450 **Fig. 12.** Contributions of the coal gasification technology sub-processes to environmental damage.

The contributions of the CCT sub-processes to the environmental impact 451 452 categories at the EP level are shown in Fig. 13. Coking, methanol production, and 453 heating were the main contributors to ecosystem quality and human health damage. Methanol production had the strongest impacts, followed by coking and heating. 454 Methanol production contributed 62.31% and 52.38% of the total ecosystem quality 455 and human health impacts, respectively, because of direct and indirect emissions during 456 457 production. About 28% of the CCT contribution to the ecosystem quality and human 458 health impacts came from coking and heating, but the effects were stronger on ecosystem quality than human health. The impacts of desalination and circulating water 459 460 were relatively small but could not be neglected. Desalination affected human health 461 more, and circulating water affected ecosystem quality more.



463 **Fig. 13.** Contributions of the coal coking technology sub-processes to environmental damage.

464 **3.4. Uncertainty analysis** 

The accuracy and comprehensiveness of the data used are the important 465 foundations for a LCA model and for LCA research, and therefore important 466 467 prerequisites for reliable conclusions to be drawn. Sulfur emission data for different sources have been used in previous comparative studies, and the results indicated that 468 469 the survey data were accurate. We preferred to use data from spot investigations, but 470 the businesses required some data to be kept secret. The data therefore contained some uncertainty, and this will have caused uncertainty in the results. We performed a 471 472 qualitative uncertainty analysis of the system boundary, distribution coefficients, and unit dataset, and our conclusions allowed the direction further research should take to 473 474 be identified.

475 3.4.1. System boundary

The system boundary definition is the basis of a LCA, and different system boundaries will strongly affect the results obtained. Our data did not include infrastructure construction, which may have led to the environmental impacts, especially on terrestrial habitats, the soil, and land use, to be underestimated. The construction process should be within the system boundaries of future studies if appropriate data are available.

482 3.4.2. Distribution coefficients/ratios

483 An important characteristic of the coal-based chemical industry is that not only products but various by-products are produced. The by-products should share the 484 environmental burden or damage caused by the whole process. Different methods of 485 486 distributing the environmental burden affect the assessment results. The environmental 487 burden or damage is usually distributed according to mass, volume, or energy. The 488 diverse products and by-products led us to distribute the environmental burden or 489 damage on the basis of energy. The methanol distribution coefficients for the CGT and 490 CCT were 80.92% and 13.80%, respectively. Under this distribution principle, the overall assessment result was better for the CCT than for the CGT at both the MP and 491 492 EP levels. This means that, from the environmental perspective, coal-based methanol 493 is better produced from coke oven gas than using a modern coal gasification system.

We determined the effect of the distribution coefficients used on the results treating other products or by-products as incidental emissions of the major product (methanol), without adding additional value. We assumed that both methods took no account of the environmental burden or damage distribution for other products and byproducts, and compared these worst-case values. That is, the methanol distribution coefficients for both methods were both 100%. The CCT distribution only took the coking process into consideration, so the CCT values were smaller than the 501 characteristic values shown in Table 6. At the MP level, the CGT offered advantages for all the impact categories except AP and GWP. The CGT offered advantages for all 502 the impact categories except for climate change and radiation at the EP level. Overall, 503 504 however, the impacts of both methods were in the same order of magnitude, so the environmental burdens and damages for both methods were the same in the worse-case 505 scenario but the CGT was better than the CCT for some impact categories. Therefore, 506 507 the comprehensive utilization for by-products was an important factor for coal-based methanol production. 508

#### 509 **Table 6**

511

510 Characteristic values for the worst-case values for the coal gasification technology (CGT) and coal

coking technology (CCT) at the mid-point (MP) and end-point (EP) levels

Catagory	MP level		Catagory	EP level	
Category	CGT	CCT	Category	CGT	CCT
ADP	1.07E+05	1.78E+05	AC/NC	2.89E+02	3.70E+02
AP	9.11E+01	7.91E+01	EC	1.12E+02	2.95E+02
EP	6.15E+00	9.64E+00	LU	1.12E+01	1.21E+01
FAETP	4.51E+02	1.07E+03	CE	6.40E-03	1.52E-02
GWP	2.19E+04	2.09E+04	CC	4.57E-03	4.38E-03
ODP	1.69E-05	3.01E-05	RA	3.71E-06	2.75E-06
POCP	4.36E+00	4.59E+00	IR	8.72E-03	1.15E-02
TETP	2.34E+01	5.78E+01	OR	8.76E-07	1.93E-06

512 **3.4.3. Unit dataset** 

The data were mostly obtained from the investigations at the plants, but some 513 514 processes were substituted with other similar processes included in the GaBi6 or Ecoinvent databases. For example, for desalination, high-salinity water was replaced 515 with production wastewater because high-salinity water was not present in the 516 517 databases. However, it is more complex and more chemicals are required to dispose of 518 production wastewater than high-salinity water, causing the environmental impacts of desalination to be overestimated. Since the average desalting water consumption for 519 520 CGT and CCT were 5.12 t per ton and 0.41 t per ton, respectively, the environmental 521 impacts of desalination of CGT were much higher that of CCT. The replacement of 522 dataset had a limited impact on the analysis results. We will reexamine the dataset to 523 improve the reliability of the results.

#### 524 **4. Conclusions**

The Chinese coal-based chemical industry is growing rapidly, and methanol production in China will soon account for more than 50% of global production. Coalbased methanol production is an important part of the Chinese coal-based chemical industry, but has important environmental impacts. The LCA method was used to evaluate the environmental impacts of producing 1 t of coal-based methanol using the 530 CGT and CCT based on the companies that represent the current domain coal-based 531 methanol technologies in China. The environmental performances of the two different 532 methods were examined at the MP and EP levels. We improved the LCI database for 533 intermediate industrial products for further research. The comparative LCA results have 534 the implications for policy described below, which may prompt companies to upgrade 535 their technologies and industrial structure.

(1) The CGT is generally seen as being more advanced than the CCT in terms of 536 production. Surprisingly, however, the environmental burden is more severe for the 537 538 CGT than the CCT. Environmental damage analysis based on the LCI indicated the 539 CCT has a relatively small burden on coal resources and water resources, and produces smaller amounts of typical air pollutants than the CGT. From the coal-based methanol 540 life-cycle perspective, CCT is a more environment-friendly technology for two main 541 reasons. The by-products of CCT share most of environmental impacts while the 542 products of CGT bear most of those impacts. Moreover, the environmental impacts of 543 544 heating system and desalination of CGT are much stronger than that of CCT.

(2) The environmental impacts of auxiliary engineering processes and public 545 engineering processes should not be ignored. Public engineering processes are the 546 dominant contributors to the marked differences between the environmental impacts of 547 548 the two methods. Heating and desalination contribute more than the production process 549 to the total impact. The CCT methanol production process has a stronger impact than the CGT methanol production process, but the environmental impacts of other 550 processes involved in the CGT are more important than the CGT methanol production 551 process. Essentially, the CGT shifts the environmental impacts from the methanol 552production process to other processes, so the CGT cannot be seen as a cleaner method 553 554 than the CCT from the viewpoint of environmental impacts.

(3) The different sub-processes made very different contributions to the different environmental impact categories. We found that there is great potential for decreasing the environmental impacts of the CGT by improving the heating and desalination systems. The CCT could be improved by decreasing the impacts of the coking and methanol production processes.

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