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Pyrrrolizidine alkaloids in honey: Quantification with and without standards

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21 **ABSTRACT**

22 Quantification of a large number of compounds in the absence of authentic standards is always a  
23 challenge. More than 600 pyrrolizidine alkaloids (PAs) have been found in plants however only limited  
24 number of PAs standards are commercially available. As PAs are the most widely distributed natural  
25 plant toxins with threat to human health, risk assessment calls for quantitative analytical methods with  
26 a wide scope including PAs without available standards. In this study, a method was developed that  
27 allows simultaneous quantification of 12 PAs in the honey samples by using HPLC-HRMS with  
28 authentic standards. This method was further extended to screen for other potential PAs in the honey  
29 using multi-target screening combined with a quantitative prediction model in the cases that authentic  
30 PAs were not available. The prediction model was subsequently validated by cross-validation and  
31 additional PAs standards which were not included in the model. The maximum concentration  
32 prediction error was 50.8%.

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37 **KEY WORDS:** Pyrrolizidine alkaloids; validation; quantification without standards; screening;  
38 honey; HPLC-HRMS

39

## 1. INTRODUCTION

40

41 Pyrrolizidine alkaloids (PAs) are a group of at least 600 different secondary metabolites that are  
42 produced by more than 6000 plant species worldwide. They are suspected to be hepatotoxic and  
43 genotoxic carcinogens in humans, since these PAs are activated to electrophilic compounds through the  
44 liver's metabolic enzymes (Bolechová, Čáslavský, Pospíchalová, & Kosubová, 2015; Lucchetti et al.,  
45 2016). Single or continuous intake of 1, 2-unsaturated PAs can cause hemorrhagic necrosis and hepatic  
46 venous occlusion, and also increase the incidence of liver haemangiosarcoma for chronic risk  
47 assessment (EFSA, 2017). As approximately 3% of all flowering plants contain PAs compounds  
48 (Boppré, 2011), this PAs exposure occurs worldwide, and intoxication by contaminated herbal teas,  
49 cereals, herbal medicine, salads, and feed, have been reported (Culvenor, Edgar, Smith, Kumana, &  
50 Lin, 1986; Molyneux, Gardner, Colegate, & Edgar, 2011; Tandon, Tandon, & Narndranathan, 1976;  
51 Wiedenfeld, 2011). Honey is one of the food products that contain PAs. Bees collect pollen or floral  
52 nectar from a wide variety of plants and therefore diverse profiles of PAs are observed in honey  
53 (Boppré, Colegate, & Edgar, 2005; Lucchetti et al., 2016). Previous studies have shown a very high  
54 frequency of PA positive honey samples, e.g., up to 94.2% samples contain PAs in a study of Spanish  
55 raw honey (n = 103) (Martinello, Cristofoli, Gallina, & Mutinelli, 2014; Orantes-Bermejo, Serra  
56 Bonvehí, Gómez-Pajuelo, Megías, & Torres, 2013). Hence with regard to the increased awareness of  
57 this potential health concern, it is important to clarify the types and concentration of PAs in honey.

58 Based on the data collected by the European Food Safety Authority (EFSA), a total of 64 PAs were  
59 reported in food products (EFSA, 2011), while the lack of certified standards hampers an unequivocal  
60 quantification. This is, particularly, a problem in PA analysis, where a complex and variable profile of

61 PAs can be expected. To date, very limited PA are commercially available and no more than  
62 approximately 34 different PAs in one method are reported, while isotopic labeling PAs are still not  
63 commercially available (Schenk, Siewert, Toff, 2017). This is unlikely to change in the near future  
64 since there is no facile chemical access, *in vitro* biosynthesis and isolation (EFSA, 2011). Therefore  
65 measuring the concentrations of a wide variety of PAs within a single experimental detection continues  
66 to remain challenging. Many studies about semi-quantification model have been reported, and the  
67 ionization efficiency were reported to be correlated to  $pK_a$ ,  $\log P$ , molecular volume, molecular size,  
68 surface tension, or the WAPS value as developed by Krueve's group, (Alymatiri, Kouskoura, &  
69 Markopoulou, 2015; Hermans, Ongay, Markov, & Bischoff, 2017; Golubović, Birkemeyer, Protić, &  
70 Otašević, 2016; Raji, et al., 2010; Oss, Krueve, Herodes, & Leito, 2010; Krueve & Kaupmees, 2017).  
71 However, their models either used for specific compounds (such as amino acid, steroids, peptides and  
72 sartans) or challenged to calculate the relevant descriptors used in the model. To our knowledge, there  
73 are still no quantitative prediction model developed specific for PAs up to now.

74 Solid-phase extraction (SPE), followed by HPLC-MS analysis is a commonly used sequence to  
75 analyze PAs in honey. However, the recovery of PAs from the samples are highly dependent on the  
76 types of cartridges, sample loading amount, and elutes used during the extraction procedures  
77 (Betteridge, Cao, & Colegate, 2005; Bodi et al., 2014; Chung & Lam, 2018; Colegate, Edgar, Knill, &  
78 Lee, 2005). The determination and quantification are generally carried out by a target analysis using  
79 GC-MS (with and without derivatization), HPLC-MS or HPLC(GC)-MS/MS (Crews, Berthiller, &  
80 Krska, 2010; Lucchetti et al., 2016; Molyneux et al., 2011). Latest LC-MS methods mostly relied on  
81 the high selectivity of mass spectrometry operated in the multiple reaction-monitoring (MRM) mode  
82 (Chung & Lam, 2018; Kowalczyk & Kwiatek, 2018), where the MRM method can only detect

83 preselected PAs within the correct retention time windows and fragmentation patterns (These, Bodi,  
84 Ronczka, Lahrssen-wiederholt, & Preiss-weigert, 2013). Furthermore, the PAs not included in the  
85 specific methods, e.g. if authentic standards are not available, will remain undetected and therefore not  
86 be quantified.

87 A screening approach (multi-target screening) using high resolution / accurate LC-QTOF-MS with  
88 full scan acquisition can achieve both high sensitivity and selectivity. Full-scan MS offer the advantage  
89 to determine a virtually unlimited number of analytes simultaneously without preselection of precursor  
90 ions. The screening approach also allows a retrospective “targeted” evaluation of data by reconstructing  
91 any desired ion chromatogram to explore the data for presence of additional PAs (Heeft, Bolck,  
92 Beumer, Nijrolder, Stolker, & Nielen, 2009). These data are however qualitative in character as  
93 standards with known concentration is needed to determine the concentration of compounds.

94 The objective of this study is to develop a high-resolution analytical method to detect and estimate  
95 the concentrations of a wide range of PAs in honey, even for those without authentic PAs standards.  
96 Firstly, we developed a validated method that allows the simultaneous detection and quantification of  
97 12 PAs in honey using HPLC-QTOF-MS with pure authentic standards. Secondly, the extracts of  
98 different honey samples were analyzed for occurrence of other PAs using a multi-target screening  
99 approach. Suspicious signals were verified by recording of product ions using a broadband  
100 fragmentation. Consequently, we proposed a semi-quantification strategy for those detected peaks  
101 without available standards through a quantitative prediction model (QPM). To evaluate this prediction  
102 model, feasibility and reliability have been validated by leave-one-out (LOO) cross validation. In  
103 addition, as a test, quantification of four unknown PAs, that were not included in the original dataset,

104 has also been carried out. Finally, we estimated the contents of PAs in honey samples combining the  
105 available reference standards and the QPM.

## 106 **2. MATERIALS AND METHODS**

### 107 **2.1 Chemicals and reagents**

108 Milli-Q grade water was used for all analyses (Millipore Corporation, Molsheim, France). Methanol  
109 (LC–MS grade), acetonitrile (LC–MS grade), ammonium hydroxide solution (LC–MS grade), and  
110 formic acid (LC–MS grade) were purchased from Fluka (Steinheim, Germany). Pyrrolizidine alkaloids  
111 (PAs) (7-acetylintermedine, echimidine, heliotrine, heliotrine *N*-oxide, intermedine, intermedine *N*-  
112 oxide, lasiocarpine, lasiocarpine *N*-oxide, lycopsamine, lycopsamine *N*-oxide, retrorsine, retrorsine *N*-  
113 oxide, senecionine, senecionine *N*-oxide, seneciphylline, seneciphylline *N*-oxide) were purchased from  
114 PhytoLab (PhytoLab GmbH & Co. KG, Vestenbergsgreuth, Germany). Sulfuric acid (analytical grade,  
115 95-97%) was obtained from Merck (Darmstadt, Germany). All other solvents used in this study were of  
116 analytical grade.

117 Each alkaloid standard was dissolved in either acetonitrile or acetonitrile/water (50/50 v/v)  
118 depending on their solubility to obtain stock solutions at a concentration of 0.5 mg/mL. All the stock  
119 solutions were stored at -20 °C. Stock standard mixtures (1 µg/mL) were prepared in acetonitrile/water  
120 (50/50 v/v) from the individual stock solutions, and used set up calibration curves.

### 121 **2.2 Samples**

122 Thirty-two honey samples were included in this study purchased from either local supermarkets or  
123 bee keepers, originating from at least 10 different countries. The honey samples were declared to be  
124 collected from summer flowers, white clover, wild flowers, creamy flowers, heather flowers, acacia,

125 fruit trees (apple, plum, rape), and etc. The information of samples: names, geographical and botanical  
126 origin, and symbols are listed in **Table 1**.

### 127 **2.3 Negative control and quality control samples**

128 Honey sample previously analyzed and found not containing PA compounds, was used as blank  
129 sample (mountain honey coming from wild flowers of Nicaragua) for method validation. Quality  
130 control (QC) samples were made by spiking blank honey samples (25  $\mu$ L stock standard mixtures to  
131 1.0 g blank honey sample). QC samples were analyzed in each series of samples and were controlled to  
132 be in the range of  $\pm 20\%$  of the validation results. If not, the whole set of samples were reanalyzed.

### 133 **2.4 Solid Phase extraction**

134 The strong cation exchange (SCX) SPE was found to be suitable for plant extracts and for a series of  
135 pyrrolizidine alkaloid/N-oxide (PAs/PANOs) representing the major hepatotoxic structural groups  
136 (monoesters, diesters and macrocyclic diesters of retronecine, heliotridine and otonecine bases), and  
137 also provided efficient capture from honey-based matrices and subsequent release into ammoniated  
138 methanol (Betteridge et al., 2005; Colegate et al., 2005). The SPE columns (150 mg/6 mL, Biotage  
139 Evolute Express CX) used in our study combines non-polar and SCX functionality for extraction of  
140 basic analytes from aqueous samples through a dual, mixed-mode mechanism. Also these specific SPE  
141 columns can be used without a preanalysis conditioning and equilibration steps without affecting the  
142 recovery significantly.

143 The extraction of PAs was based on a modified method from German Federal Institute of Risk  
144 Assessment (BfR) (BfR, 2013). One gram of honey was weighed into a 10 ml centrifuge tube and  
145 dissolved in 3 mL of 0.05 M  $H_2SO_4$  in water at room temperature by shaking for 30 min. The samples  
146 were centrifuged at  $2000 \times g$  for 10 min at room temperature. The supernatant was transferred to a SPE

147 cartridge by using a vacuum manifold. After loading the sample, the column was washed with 4 mL of  
148 0.05 M H<sub>2</sub>SO<sub>4</sub> followed by 4 mL of methanol, then the column was eluted with 5 mL of 2.5% (v/v)  
149 ammonium hydroxide in methanol at a rate of 0.5-1 mL/min. The eluate was dried under a stream of  
150 nitrogen at 45 °C. The residue was dissolved in 0.5 mL acetonitrile/water (5/95, v/v), transferred to an  
151 Eppendorf vial, and centrifuged at 10.000 × g for 10 min at room temperature. Then 200 µL of  
152 supernatant was transferred into a HPLC-vial with insert.

### 153 **2.5 HPLC-ESI-QTOF-MS.**

154 Samples were analyzed on an Agilent 1200 HPLC (Agilent Technologies, CA, U.S.) coupled to a  
155 Bruker Daltonics micro-TOFq mass spectrometer (Bruker Daltonics, Bremen, Germany). The  
156 resolution of the Bruker micro-TOFq is approximately 10000. Samples were injected onto a  
157 Nucleoshell C18 reversed-phase column (2.7 µm, 100 × 2 mm, MACHEREY-NAGEL GmbH & Co.  
158 KG, Düren, Germany). The mobile phase consisted of water (A) containing 0.1% formic acid and 2.5  
159 mM ammonium hydroxide, and acetonitrile (B). The analysis was done using a gradient elution at a  
160 flow rate of 0.3 mL/min at 40 °C using the gradient: isocratic at 2% of B for 1.0 min, then increased to  
161 10% of B at 4.0 min, to 40% of B at 9 min, and finally to 95% of B at 10.0 min and maintained to 11  
162 min, where after the gradient returned to starting conditions 2% of B at 11.2 min followed by 3.8 min  
163 equilibration with 2% of B to giving a total runtime of 15 min.

164 An electrospray ion source (ESI) was used in positive ion mode (ESI<sup>+</sup>) with the following settings:  
165 capillary voltage 4 kV, nebulizer pressure 2 bar, dry gas temperature 180 °C, and dry gas flow rate 10  
166 L/min. Spectra were collected from m/z 50 to 800 Da at a rate of 2 Hz. Sodium formate dissolved in  
167 50% of 2-propanol was introduced into the ion source between 0.2–0.4 min for internal calibration of  
168 the data files. Hexakis (1H, 1H, 2H-perfluoroethoxy) phosphazene was used as the lock mass calibrant.

169 To confirm the presence of pyrrolizidine alkaloids, LC-MS/MS analyses were conducted using  
170 broadband collision-induced dissociation (bbCID) with a fragmentation energy of 35 eV and nitrogen  
171 as the collision gas.

## 172 **2.6 Design of validation**

173 Validation parameters such as linearity, specificity, limits of detection (LOD), limit of quantification  
174 (LOQ), measurement uncertainty, recovery, and precision were determined for 12 PAs using the  
175 established method according to Nordic Committee on Food Analysis (NMKL) procedure No. 4 with  
176 minor modification (NMKL, 2009). The spiked samples were prepared according to the procedure  
177 described above and analyzed on three different days. All 12 PA compounds were spiked at 4  
178 concentration levels: 2 ng/mL, 10 ng/mL, 50 ng/mL, and 150 ng/mL corresponding to 1 µg/kg, 5 µg/kg,  
179 25 µg/kg, and 75 µg/kg in the honey samples. At each concentration level at least 10 replicates of the  
180 same spiked blank honey samples were analyzed (13 replicates at 10 ng/mL, and 10 replicates for the  
181 other 3 levels). Method precision was evaluated by intra- and inter-day analyses. The repeatability  
182 (intra-day variability) was tested within one day for each concentration, reproducibility (inter-day  
183 variability) was examined on three different days for each concentration (13 replicates for 5 µg/kg, 10  
184 replicates for 1 µg/kg, 25 µg/kg and 75 µg/kg).

185 LOD and LOQ were calculated by using the standard deviation of 10 replicates at 2 ng/mL (blank  
186 honey spiked before SPE clean-up).  $LOD = 3 \times SD / \text{recovery of each compound}$ ,  $LOQ = 6 \times SD /$   
187  $\text{recovery of each compound}$ . Matrix effects (ME) was estimated by adding different levels of standard  
188 solution to a sample of the blank honey extracts. Matrix match calibration curves were used to correct  
189 for matrix effects (ME), and were made by analysis of each analyte at 7 calibration levels from 2 to 150  
190 ng/mL. The standard curves were the peak area versus the concentration of each analyte.  $ME (\%) =$

191 (Slope of matrix match standard curve / Slope of standard curve without honey matrix)  $\times$  100%. ME of  
192 some other honey samples were also measured at 50 ng/mL spike level to compare the ME difference  
193 between different honey samples. The recovery was determined by comparing the analytical results of  
194 the extracted PAs from spike honey samples with the results of standards added in the same  
195 concentrations to blank matrix (13 replicates for 5  $\mu$ g/kg, 10 replicates for 1  $\mu$ g/kg, 25  $\mu$ g/kg and 75  
196  $\mu$ g/kg).

## 197 **2.7 Quantification**

198 All the quantification was performed by integration of peak area of extracted ion chromatograms  
199 using exact mass  $\pm$  5 mDa. The concentration of PAs in the samples was calculated by QuantAnalysis  
200 2.2 (Bruker Daltonics, Bremen, Germany). All the values of PAs of samples were calculated by using  
201 the average value in duplicate samples of the same sequence.

## 202 **2.8 Software**

203 The Chemdraw Professional 15.0 was used for drawing and displaying chemical structures. R  
204 (version 3.3.3), R studio and Origin software were used for all statistical analysis. Jchem version 18.5.0  
205 was used to predict the  $pK_a$ ,  $pK_b$ , logP values, and other properties of molecules.

## 206 **2.9 Statistical model evaluation and validation**

207 Several studies have investigated the relationship between signal response in electrospray ionization-  
208 mass spectrometry (ESI-MS) and molecular descriptors (molecular volume, logP,  $pK_a$ , polarity) of  
209 analytes to estimate analyte concentration (Golubović et al., 2016; Krueve & Kaupmees, 2017; Oss et  
210 al., 2010). Seneciophylline was randomly chosen as the reference standard in the method to correct the  
211 daily variation. Multi-linear regression (MLR) analysis was carried out to obtain the QPM describing

212 the relationship between relative calibration curves' slope (RS) corresponding to seneciophylline in  
213 honey matrix and molecular properties in the following formula:

$$214 \quad \ln RS = \ln\left(\frac{\text{Slope of analytes}}{\text{Slope of Seneciophylline}}\right) = \sum_{i=1}^n \mathbf{a}_i \mathbf{X}_i$$

215 where  $\mathbf{X}_i$  is the physicochemical parameters or their combinations, and  $\mathbf{a}_i$  is the coefficients of these  
216 parameters. Experimental values of several parameters such as logP and  $pK_a$  are not available for all  
217 compounds, therefore these molecular descriptors including Molecular volume, logP,  $pK_a$  (strongest  
218 acidic),  $pK_b$  (strongest basic), and polarizability were computed by using Physico-chemical property  
219 predictors plugin of JChem in excel (Szisz, 2017). All the chemical structures of PAs were transferred  
220 as the SMILES (Simplified molecular input line entry specification) obtained from PubChem. The  
221 molecular descriptors with correlation above 0.9 were removed to avoid the co-linearity in the model.  
222  $\mathbf{X}_i$  and  $\mathbf{a}_i$  were obtained by MLR which was calculated in R. Twelve PAs compounds were used to  
223 develop the MLR model (exact mass is in the range from 299 to 427 Da,  $pK_a$  is in the range from 10.54  
224 to 12.37,  $pK_b$  is in the range from 2.82 to 8.13, logP is in the range from -1.42 to 1.64). In the gradient  
225 elution, all the PAs were eluted from 10% to 37% organic phase, which is considered as a narrow  
226 range, thus the corrections with retention parameters or solvent change was ignored in this part.

227 Cross-validation method 'leave-one-out' (LOO) approach was used to validate the obtained results.  
228 LOO means that each compound was left out from the model fitting process once; thereafter, the model  
229 was used to predict the RS value of the compound not involved in the model development. LOO is  
230 equivalent to k-fold cross-validation where  $k = n$  (the number of observations in the data set), and LOO  
231 for linear regression is exactly equivalent to the PRESS (Predicted Residual Sum of Squares) method  
232 based on the leave-one-out or Jackknife technique, detail explanation can be found in Tarpey's study

233 (Tarpey, 2000). Therefore in this study, PRESS and root mean square error of prediction (RMSEP)  
 234 were used to describe the differences between predicted RS values and measured values.

235 Using the fact of the predicted value for  $y_i$  when the  $i^{th}$  compound is deleted from the MLR model  
 236 is equal to

$$y_i - \hat{y}_{(i)} = \frac{\hat{e}_i}{(1 - h_i)} = \hat{e}_{(i)}$$

237 Here  $\hat{e}_i = usual\ i^{th}\ residual$  and  $h_i = potential\ or\ leverage\ value\ for\ the\ i^{th}\ compound$ ,  
 238  $y_i$  and  $\hat{y}_{(i)}$  are predicted and measured lnRS value in this study, respectively.

$$PRESS = \sum_{i=1}^n (y_i - \hat{y}_{(i)})^2$$

239 
$$RMSEP = \sqrt{\frac{PRESS}{n}}$$

240 The resulting model was also used to predict the slope of the calibration curves for 4 additional PAs  
 241 which were not included in the original dataset but within the model coverage to verify the prediction  
 242 quality. The predicted slopes were used to calculate the concentrations of the other PAs compounds  
 243 detected in the collected samples.

244 
$$C_{predicted} = \frac{Response_{from\ sample}}{Slope_{predicted}}$$

245  $C_{predicted}$  is the predicted PA concentration,  $Response_{from\ sample}$  is the peak area of the tentatively  
 246 identified PAs in honey samples,  $Slope_{predicted}$  is the slope of predicted calibration curve of the  
 247 identified PAs, and it was calculated from the RS value corresponding to the slope value of reference  
 248 standard (seneciophylline) in honey matrix, assuming that the standard curve pass through zero.

249 All statistical tests were performed at 95% significant level. The prediction error for additional 4 PAs  
250 out of the model was calculated with the following formula:

$$251 \quad \text{Prediction error} = \left( \frac{RS_{\text{predicted}} - RS_{\text{measured}}}{RS_{\text{measured}}} \right) \times 100\%$$

## 252 **3 RESULTS AND DISCUSSION**

### 253 **3.1 PART 1. Validation**

#### 254 *3.1.1. Specificity*

255 Blank honey samples and blank honey spiked with 5 ng/ml mix standard solution (corresponding to  
256 2.5 µg/kg) were analysed to examine if other compounds in the samples were interfering with these  
257 analytes. The HPLC shows good separation of all 12 PA compounds, and there are no apparent  
258 interferences observed from the matrix evaluated from the extracted ion chromatograms, thus  
259 confirming the specificity. The structures and chromatogram of 12 PAs are shown in in **Figure 1**.

#### 260 *3.1.2. Linearity*

261 A good linear relationship was observed based on correlation coefficients higher than 0.99 (from 2  
262 ng/mL to 150 ng/mL). The average relative residual is less than 25% at all levels, and the intercepts  
263 obtained are close to the theoretical value zero, demonstrating good consistency. Information about  
264 residues plot can be found in in Supporting Information **Figure S1**.

265 *3.1.3. Precision*

266 The standard deviations of all the PAs at different spiked levels were calculated. Results are shown  
267 in **Table 2**. Coefficient of Variation (CV%) for the reproducibility is in the range from 6.3% to 21.6%,  
268 and CV% for the repeatability is in the range from 2.6% to 18.9%, which is considered acceptable.

269 *3.1.4. Recovery*

270 The average recovery is 89% for all the 12 PA compounds, ranging from 79.2% to 104.4% with a  
271 standard deviation lower than 7.6% (**Table 2**), thus confirming a good reproducibility of the method  
272 and sufficient for reliable analysis.

273 *3.1.5 Limit of detection and limit of quantification*

274 Results are shown in **Table 2**. LOD values for the 12 PA compounds are in the range from 0.2 µg/kg  
275 to 0.6 µg/kg, LOQ values of the 12 PA compounds are in the range from 0.5 µg/kg to 1.3 µg/kg, which  
276 are comparable to those reported in literatures (Bolechová et al., 2015; EFSA, 2012). Although there  
277 are studies which present methods with lower LOQs (BfR, 2013; Kowalczyk & Kwiatek, 2018;  
278 Lucatello, Merlanti, Rossi, Montesissa, Capolongo, 2016), they all rely on MRM mode rather than full  
279 scan detection. These MRM method are very specific however MRM are blind to other PAs that may  
280 occur in samples. **Figure S2** shows samples spiked with 1 ng/mL (corresponding to ~0.5 µg/kg in  
281 honey) with peaks still clearly visible for all analytes, indicating that the calculated limits are realistic  
282 to detect and quantify for the 12 PA compounds.

### 283 3.1.6 Matrix effect

284 The matrix effect (ME) of spiked blank honey sample with 1 g sample load is in the range from  
 285 81.0% to 112.3%, indicating that the analytes' signals are only slightly influenced by the honey matrix  
 286 (**Table 2**). In order to compare the difference of ME between different honey matrix, additional  
 287 experiments were done to compare the ME. Examining these results showed a RSD% in the range from  
 288 4 % to 12% from ME of 8 honey samples, which indicates that it is feasible to use one blank matrix to  
 289 quantify all the other honey samples. Full data are shown in Supporting Information **Table S1**.

### 290 3.1.7 Measurement uncertainty

291 The measurement uncertainty (MU) consists of intra-laboratory reproducibility ("CVR"), bias  
 292 ("bias") calculated as 100% recovery, standard deviation on bias ("CVRbias/sqrt(n)") calculated as  
 293 intra-laboratory standard deviation divided by square root of number of results, and standard deviation  
 294 on matrix effect ("CVmatrix effect") listed in **Table S1** in the Supporting Information. The method  
 295 uncertainty was calculated as:

$$296 \text{ MU} = \sqrt{\text{CVR}^2 + \text{bias}^2 + \frac{\text{CVRbias}^2}{n} + \text{CVRmatrixeffect}^2}$$

297 Seneciophylline-N-oxide, with the highest bias (100 – 82.4% = 17.6%) had a MU of 22%

$$298 \left( \sqrt{11.9^2 + 17.6^2 + \frac{11.9^2}{43} + 7.1^2} \right).$$

## 299 3.2 PART 2. PA detection and contents in honey samples

### 300 3.2.1 PA Quantification and identification

301 The 12 PAs included as standards in this study were selected based on the ones reported most often and  
302 in highest concentrations in honey (EFSA, 2011), and they belong to 3 major PAs chemical groups  
303 found in honey:

- 304 • **Senecionine-type** PAs (Retrorsine, retrorsine *N*-oxide, senecionine, senecionine *N*-oxide,  
305 seneciphylline, seneciphylline *N*-oxide), which occur primarily in the *Senecioneae* spp.  
306 (*Asteraceae* family), but are also found in *Crotalaria* spp. (*Fabaceae* family);
- 307 • **Lycopsamine-type** PAs (Intermedine, intermedine *N*-oxide, lycopsamine, echimidine), which  
308 occur in the *Boraginaceae* family and in the *Eupatorieae* spp. (*Asteraceae* family);
- 309 • **Heliotrine-type** PAs (lasiocarpine, lasiocarpine *N*-oxide), which occur in *Heliotropium* spp.  
310 (*Boraginaceae* family) (Mulder, López, These, & Preiss-weigert, 2015).

311 The LC-MS/MS chromatograms of 12 validated PAs in honey samples were characterized by  
312 comparing the fragments pattern with authentic standards. The fragmentation pattern of selected PAs,  
313 C-1 (monoester or open-chained diester), and C-2 (*N*-oxide open-chained diester) are shown in **Figure**  
314 S3 (Avula et al., 2015; These et al., 2013). A compilation of chemical formulas, accurate mass,  
315 fragment ions, retention time, and mass error of HRMS detection for 12 reference PAs are shown in  
316 **Table 3**. The MS<sup>2</sup> broadband fragmentation spectrums show fragment ions at *m/z* 172, 156, 138, 136,  
317 120, 118, and 94 which are the most common and characteristic fragments from these PAs. These  
318 fragment ions are also well known in unsaturated necine ester type PAs and are similar to fragments  
319 reported in the literature (Avula et al., 2015; Beales et al., 2004; Schulz et al., 2015; These et al., 2013).  
320 Detailed information about fragment patterns of PAs found in this study are shown in supplemental  
321 **Figure S4**.

322 The analyses showed that one or more of the 12 PA compounds were found in 60% of these samples.  
323 The combined results are shown in **Table 1**. The concentration of each PA compound in the positive  
324 samples is in the range between 1.4 and 11.1  $\mu\text{g}/\text{kg}$  honey and the total PAs contents are in the range of  
325 1.4 to 14.2  $\mu\text{g}/\text{kg}$  (the sum of the 12 PAs). The highest total PA content is found in honey H27.

326 The most frequently found PA is echimidine. Seven of the tested samples are positive with  
327 concentrations ranging from 1.4 to 3.4  $\mu\text{g}/\text{kg}$ , followed by intermedine and senecionine. These findings  
328 are in accordance with the results reported by Lucatello (et al, 2016), most likely because plants from  
329 the genus *Echium*, namely *Echium italicum* L., *Echium vulgare* L., and *Echium plantagineum* L. are  
330 widely distributed in the European region (Lucatello, Merlanti, Rossi, Montesissa, 2016). For  
331 echimidine in H14, H25 intermedine in H3, H23, retosine *N*-oxide in H12, H23, lycopsamine in H11,  
332 H16, H25, H29, and senecionine in H4 and H9, they are below the LOQ value but are still detected in  
333 LC-MS. Information about extracted ion chromatograms of these detected PAs are shown in **Figure S5**.

334 Although PAs are detected in almost half of the samples studied, they are not found in as high  
335 concentration as in other studies (Avula et al., 2015; Beales et al., 2004). The levels of the 12 validated  
336 and quantified PAs are so low that the honey can be accepted according to both  $\text{BMDL}_{10}$  of EFSA to  
337 assess the carcinogenic risks of PAs (237  $\mu\text{g}/\text{kg}$  bw per day for chronic toxicity) and also BFR  
338 recommendation (maximal intake of 0.007  $\mu\text{g}/\text{kg}$  bw per day) if consumers only consume a limited  
339 quantity of honey (see **Table 1**) (Dusemund et al, 2011; EFSA, 2017; BfR, 2013). However, food  
340 safety issues frequently originate from compounds that are not included or quantified by the analytical  
341 method. Therefore, a multi-target screening method including more PAs compounds is particularly  
342 important to predict potential risk from PAs in honey.

### 343 3.2.2 PAs screening

344 A screening method monitoring 85 different PAs compounds based on elemental composition from  
345 the literature was used to detect if other PAs could be found in the honey samples. The PAs in the  
346 screening list were compiled from literatures and only 1, 2-un-saturated compounds were taken into  
347 consideration. Then the suspicious signals with same molecule weight were further analysed by LC-  
348 MS/MS. Tentative identification of the PAs was done by comparing the fragments with literatures  
349 information (EFSA, 2011). Elemental composition and fragments ion of these tentative identified PAs  
350 from the screening method are shown in **Table 3**.

351 According to the screening results, several other PAs were found in the honey apart from the 12  
352 validated PAs already quantified. Some of these PAs occur in relative high concentration. For example,  
353 a chromatographic peak just next to lycopsamine with the same protonated molecular ions ( $m/z$   
354 300.1805  $[M+H]^+$ ), same fragments pattern, and similar retention time (only 0.1 min difference) is  
355 found in 12 out of 32 honey samples. In order to clarify whether this compound is lycopsamine or not,  
356 we spiked one of these honey samples with this unknown peak. We proved that this large peak is not  
357 lycopsamine, but its peak area is much larger than that of lycopsamine. According to the previous  
358 reports and fragments information we obtained ( $m/z$  94.0655, 138.0932, 156.1019), this unknown  
359 compound is most likely rinderine (other possibilities are echinatine or indicine), and all of them are  
360 belong to lycopsamine-type PA stereoisomers, which include indicine, intermedine, lycopsamine,  
361 rinderine and echinatine (Beales et al., 2004; Kempf et al., 2011). All these five PAs are structurally  
362 closely related, thus a similar retention behaviour and mass spectrometric response can be assumed to  
363 be the same. There are also other large peaks appearing next to senecionine and echimidine again (0.1

364 min retention time difference) with the same molecule weight and fragment pattern, but not the same as  
365 verified by standard addition.

366 Several PAs have the similar molecular structures, therefore we will expect similar properties and  
367 fragments pattern in LC-MS/MS. For these types of compounds, it is difficult to predict the correct  
368 structures without further identification or authentic standard. Therefore, the data can only be seen as  
369 indicative. A list of considered PA compounds included in the screening, and their accurate mass data,  
370 are shown in Supporting Information **Table S2**.

371 To evaluate the concentrations of these other PAs, which were not included in the validation method,  
372 a semi-quantification approach was used to estimate their concentration as discussed in the following  
373 section.

### 374 **3.3 PART 3. QPM and estimation of concentration of other PAs in honey samples**

375 Estimation of the concentration for PAs where standards are unavailable are based on all the  
376 information we can obtain from direct measurement of available standards and honey samples, also  
377 model calculation. In the initial model, all the previously mentioned molecular properties (molecule  
378 volume,  $pK_a$ ,  $pK_b$ ,  $\log P$ , relative slope, exact mass, and retention time of 12 PAs used in the model and  
379 tentative identified PAs in samples are shown in **Table S3a**) were included. While by backward MLR  
380 modelling, the best fit for data was achieved with a model containing  $pK_a$ ,  $pK_b$ ,  $\log P$ , and exact mass.  
381 Several authors have found that hydrophobicity ( $\log P$ , usually expressed as octanol/water partition  
382 coefficient) are important characteristics correlated to ESI ionization efficiency, and increasing the  
383 extent of nonpolar character in a compound leads to enhanced affinity for the surface phase and higher  
384 ESI response (Cech & Enke, 2000; Leito et al., 2008).  $pK_a$  can be expressed as the ability to be

385 protonated and become a cation, and it has been proved that pKa is highly correlated with ionization  
386 efficiency in ESI (Oss et al., 2010). Exact mass was characterized by the molecular size or molecular  
387 volume, which also has been reported as a parameter that can influence ionization efficiency (Hermans,  
388 Ongay, Markov, & Bischoff, 2017; Oss, Krueve, Herodes, & Leito, 2010).

389 The formula calculated in R is shown in the following formula (Procedures of calculation is shown in  
390 Supporting Information).

$$\ln(RS_{predicted}) = (27.0965 \pm 7.6792) + (1.0240 \pm 0.3192) \times \log P - (1.4075 \pm 0.4115) \times pK_a - (0.3048 \pm 0.1010) \times pK_b - (0.0285 \pm 0.0073) \times \text{exact mass}$$

393 Regarding the calculation, eg., rinderine,  $\ln(RS_{predicted})$  of rinderine =  $27.0965 + 1.0240 \times (-0.2900) -$   
394  $1.4075 \times 11.3400 - 0.3048 \times 7.8200 - 0.0285 \times 299.1733 = -0.07149$ , therefore predicted relative slope  
395 of rinderine corresponding to seneciophylline is 0.9310.

396 The correlation between predicted RS and measured RS is shown in **Figure 2-a**. The  $R^2$  for this model  
397 was 0.75 and the residual standard error was 0.21 logarithmic units. By LOO validation, the accuracy  
398 of the models can also be described with RMSEP,  $RMSEP = 0.28 \ln RS$  units. This value shows that on  
399 average the mismatch between the predicted and measured RS from LOO is lower than 1.3 times. The  
400 prediction errors are in the range from -0.3% to 27.1% for the 12 PA compounds used to develop the  
401 model (**Table S3-b** in Supporting Information). To validate the method, the model was also used to  
402 quantify 4 additional PAs compounds available as pure standards but not included in the model above  
403 in a spiking experiment. The concentrations of these 4 authentic standard compounds were estimated  
404 using the model developed by the 12 PAs giving a prediction error in the range between 10.2% and  
405 50.8%. **Figure 2-b** shows the difference in standard curve slope using the predicted concentration and

406 measured concentration of the PAs not included in the model (from 5 to 100 ng/mL). Therefore, a  
407 predicting model based on a 12 PAs calibration sets provides acceptable quantification results for a risk  
408 assessment. Despite some PAs being of different chemical structures, they share quite similar  
409 physicochemical properties, hence this will not have large effect on the predicted concentration. We  
410 also calculated the model based on the data we obtained from two different days; and the coefficients in  
411 the formula are quite similar as shown in **Table S3-b**. The difference between the two formulas  
412 obtained from two different experimental days was compared; the variation for all compounds is within  
413 a range of -24.5% to 10.9%. This indicates the feasibility for the future study, to use only one model  
414 and one reference PA standard (e.g., Seneciophylline in this study) to predict the concentrations of the  
415 other PAs within maximum 50.8% prediction error; as long as we have the structures information of  
416 these PAs.

417 In many cases, studies about PAs estimation may choose calibration alkaloids of similar structure  
418 using different calibration alkaloids (often an isomer) (Hoogenboom et al., 2011; These et al., 2013). Or  
419 just simply reported all alkaloids as equivalent of one alkaloid such as echimidine in Beales' study  
420 (Beales et al., 2004). In the approach of PAs estimation with different calibration alkaloids, several  
421 PAs standards are still necessary. They cover various PAs types, while in the approach of PAs  
422 estimation, with only one alkaloid, it is more unpersuasive due to the variation of PAs response which  
423 can be nearly 3 times based on our results. Hereby, the difference between the approach of PAs  
424 estimation, with only one alkaloid and our prediction model, were compared. It is shown in **Table S4** in  
425 Supporting Information, 3 concentrations (5 ng/ml, 10 ng/ml, 25 ng/ml) were performed, and Max25%  
426 (the sum of 25% biggest absolute prediction error, e.g., absolute prediction error of 4 PAs were chosen  
427 among 16 PAs in our case) was used to estimate the performance of different approaches. In the

428 approach simply reported every alkaloid as equivalent of one alkaloid, the Max25% values are in the  
429 range of 232.2% ~ 603.1%, 202.5% ~ 502.2%, 170.3% ~ 621.2% in the concentration of 5, 10, 25  
430 ng/ml, respectively. However, when we use the prediction model, the Max25% values are 196.50%,  
431 142.16%, 162.26% separately, which is more accurate and controllable. It has been reported by the  
432 EFSA Scientific Panel on Contaminants in the Food Chain (CONTAM Panel) that they identify mainly  
433 4 types of PAs of particular importance for food and feed (Senecionine-types, Lycopsamine-type,  
434 Heliotrine-type, Monocrotaline-type) (EFSA, 2011). Although there were only a limited number of  
435 PAs in our dataset when we developed the QPM, it covers 3 types of most common PAs which exist in  
436 honey. Therefore, this model can make relative accurate estimation of PAs covered by these 3 types, at  
437 least (only monocrotaline-type PAs was excluded). To make this model more generic, further study can  
438 be carried out to include more PAs, such as monocrotaline-type PAs in the model development.

439 The QPM was used for the estimation of other PAs based on the tentative identification, and  
440 prediction error of RS between measured and predicted value are shown in supplemental **Table S3-b**.  
441 The results from semi-quantification of these PAs are shown in **Table 1**. Riderine (or Echinatine  
442 /Indicine), which elutes just next to lycopsamine and shows the highest level among all the quantified  
443 PAs reaching an estimated concentration of about 56  $\mu\text{g}/\text{kg}$  in H15. Crotananine shows the  
444 concentration of about 51  $\mu\text{g}/\text{kg}$  in H8, followed by acetyl-lycopsamine and 3'-acetylindicine, with a  
445 concentration of 47  $\mu\text{g}/\text{kg}$  and 45  $\mu\text{g}/\text{kg}$  both detected in H9.

446 Although according to assessment of chronic risks referring to a BMDL<sub>10</sub> of 237  $\mu\text{g}/\text{kg}$  bw per day  
447 established by EFSA (EFSA, 2017) or acute toxicity data reported by Dusemund et al., (2011), the  
448 detected PAs levels are acceptable. However, following the recommendations from the BFR as

449 mentioned before, the oral intake of PAs should not exceed a daily dose of 0.007  $\mu\text{g}/\text{kg}$  b.w (BfR,  
450 2011). Assuming that a 60 kg body weight and a daily consumption of honey of 20 g, a honey sample  
451 should not contain more than 21  $\mu\text{g}/\text{kg}$  of PAs. Our estimated concentration of PAs, including all PAs  
452 even those where standards are not available, show the total PAs content in honey samples may be as  
453 high as: H6 (ca. 26  $\mu\text{g}/\text{kg}$ ), H7 (ca. 36  $\mu\text{g}/\text{kg}$ ), H5 (ca. 36  $\mu\text{g}/\text{kg}$ ), H11 (ca. 36  $\mu\text{g}/\text{kg}$ ), H18 (ca. 41  
454  $\mu\text{g}/\text{kg}$ ), H8 (ca. 51  $\mu\text{g}/\text{kg}$ ), H15 (ca. 67  $\mu\text{g}/\text{kg}$ ) and H9 (ca.117  $\mu\text{g}/\text{kg}$ ) all exceeding 21  $\mu\text{g}/\text{kg}$ . Even if  
455 we consider a quantification error of 51% the levels will still be of concern.

456 In conclusion, we proposed a method that can be extended to a general strategy to achieve estimation  
457 of quantitative data where authentic standards are not available. By using a prediction model based on  
458 physio-chemical parameters, analysis of similar standards, and literature relevant quantitative data for  
459 risk assessment can be estimated using HPLC-HR-MS/MS techniques. This approach was  
460 demonstrated by the simultaneous quantification of 12 PAs and also the further analysis of honey  
461 containing other high content PAs without standards. Future applications of this method could be the  
462 quantification or semi-quantification of PAs in other honey samples, and also other PA containing  
463 products, such as herbal tea, cereal, feed, and milk.

## 464 **ABBREVIATIONS USED**

465 PAs: Pyrrolizidine alkaloids; GC-MS: Gas chromatography–mass spectrometry; LC-MS: Liquid  
466 chromatography–mass spectrometry; MRM: multiple reaction-monitoring; LC-QTOF: liquid  
467 chromatography-quadrupole-time-of-flight-mass spectrometry; QC: Quality control; LOD: limits of  
468 detection; LOQ: limit of quantification;  $\text{p}K_{\text{a}}$ :  $\text{p}K_{\text{a}}$  value of strongest acidic;  $\text{p}K_{\text{b}}$ :  $\text{p}K_{\text{a}}$  value of strongest  
469 basic; SMILE: simplified molecular-input line-entry system; CV%: Coefficient of Variation; ME:

470 matrix effect; ESI: electrospray ion source; BMDL<sub>10</sub>: the lower confidence limit on the benchmark  
471 dose associated with a 10% response

## 472 **SUPPORTING INFORMATION DESCRIPTION**

473 Supplementary data related to this article can be found in the Supporting Information file.

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- 587

## TABLES

**Table 1.** Information of all the tested honey samples: places of production, origins and symbols used in the text.

No	Geographical origin	Botanical origin	PAs ( $\mu\text{g}/\text{kg}$ )
H1	Spain	Mandarin honey	-
H2	Romania	Creamy flowers honey	Echimidine (1.4)
H3	Mixture from outside the EU	Not clear	Intermedine (<LOQ), Rinderine//Echinatine/Indicine (ca. 6.1)
H4	Argentina and Ukraine	Not clear	Echimidine (3.4), Senecionine (<LOQ)
H5	China	Mountain honey (many wild flowers)	Integerrimine/ Senecivernine (ca.1.6), Trichodesmine (ca. 33.9)
H6	Mexico	Not clear	Intermedine (3.5), Lycopsamine (1.7), Retrorsine (4.3), Rinderine//Echinatine/Indicine (ca.16.4)
H7	Brazil	Not clear	Echimidine (1.6), Intermedine (2.3), Lycopsamine (2.2), Retrorsine (1.6), Senecionine (2.4), Rinderine//Echinatine/Indicine (ca.16.5), Integerrimine/ Senecivernine (ca.9.5)
H8	Denmark	Clover and rape	Crotananine (~51.0)
H9	Denmark	Heather flowers honey	Senecionine (<LOQ), Crotananine (ca.24.7), 3'-acetyлиндicine (ca.45.4), Acetyl-lycopsamine (ca.46.8)
H10	Denmark	Fruit trees (apple, plum, rape)	Crotananine (~14.2)
H11	Denmark	Summer flowers and white clover honey	Lycopsamine (<LOQ), Myoscorpine <i>N</i> -oxide/Echiumine <i>N</i> -oxide/ Echiupinine- <i>N</i> -oxide (ca. 2.7-3.2), Rinderine//Echinatine/ Indicine (ca.10.3), Crotananine (ca.7.8), 3'-acetyлиндicine (ca.8.8), Acetyl-lycopsamine (ca.6.7)
H12	Brazil	Not clear	Retrorsine- <i>N</i> -oxide (<LOQ)
H13	Nicaragua	Mountain honey (many	Crotananine (ca.5.2)

			wild flowers)
H14	Ukraine	Fruit trees	Echimidine (<LOQ)
H15	Mixture from countries inside and outside the EU	Not clear	Intermedine (8.4), Lycopsamine (2.0), Rinderine//Echinatine/Indicine (ca.56.4),
H16	Ukraine	Not clear	Echimidine (2.2)
H17	Mixture from outside the EU	Not clear	Crotananine (ca.3.4)
H18	South America	Not clear	Intermedine (8.0), Lycopsamine (0.8), Rinderine//Echinatine/Indicine (ca.31.7)
H19	Denmark	Fruit trees	Crotananine (ca.7.7), Rinderine//Echinatine/Indicine (ca.1.1)
H20	Ukraine	Not clear	Echimidine (2.0), Crotananine (~2.0)
H21	Mixture from countries inside and outside the EU	Acacia	Riddelline/Jacozine (ca.1.5-4.9)
H22	Mixture from outside the EU	Not clear	Retrorsine (8.2), Crotananine (ca. 2.6), Rinderine//Echinatine/Indicine (ca. 2.5)
H23	Mixture from outside the EU	Not clear	Intermedine (<LOQ), Retrorsine (8.2), Retrorsine- <i>N</i> -oxide (<LOQ), Crotananine (ca. 4.4), Rinderine//Echinatine/Indicine (ca. 3.6)
H24	Mixture from outside the EU	Acacia honey with hazelnuts	Crotananine (ca. 1.8)
H25	Argentina and Ukraine	Not clear	Echimidine (<LOQ), Lycopsamine (<LOQ),
H26	China	Acacia	-
H27	Mixture from outside the EU	Not clear	Intermedine (11.1), Lycopsamine (3.1)
H28	Denmark	Not clear	Crotananine (ca. 5.3), Riddelline/ Jacozine (ca. 2.3-7.5), Rinderine//Echinatine/Indicine (ca. 1.5)

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H29	Mixture from countries inside and outside the EU	Wild flower	Echimidine (2.7), Lycopsamine (<LOQ), Croatanine (ca. 2.1)
H30	Mixture from countries inside and outside the EU	Acacia	-
H31	Mixture from outside the EU	Not clear	Echimidine (1.6), Intermedine (1.4), Lycopsamine (0.7), Croatanine (ca. 12.8), Rinderine//Echinatine/Indicine (ca. 3.3)
H32	Mixture from outside the EU	Not clear	-

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**Table 2.** Validation results.

Compounds	Conc		Inter-day reproducibility			Intra-day Repeatability			Recovery %	LOD μg/kg	LOQ μg/kg	ME
			mean	std	CV %	mean	std	CV %				
Echimidine	1.00	μg/kg	0.89	0.19	21.6	0.89	0.17	18.8	93.5			
	5.00	μg/kg	5.22	0.64	12.3	5.22	0.52	10.0	104.4	0.54	1.07	101.92
	25.00	μg/kg	24.20	2.88	11.9	24.20	2.08	8.6	96.8			
	75.00	μg/kg	67.96	5.80	8.5	67.96	1.80	2.6	90.6			
Intermedine	1.00	μg/kg	0.81	0.08	10.3	0.81	0.07	8.0	89.4			
	5.00	μg/kg	4.06	0.69	17.0	4.06	0.46	11.3	81.2	0.41	0.81	90.50
	25.00	μg/kg	24.21	2.12	8.8	24.21	0.81	3.4	96.8			
	75.00	μg/kg	72.45	7.74	10.7	72.45	3.38	4.7	96.6			
Intermedine- <i>N</i> -oxide	1.00	μg/kg	0.96	0.16	16.9	0.96	0.16	16.9	95.6			
	5.00	μg/kg	4.01	0.76	18.9	4.01	0.76	18.9	80.2	0.52	1.04	94.05
	25.00	μg/kg	22.77	2.53	11.1	22.77	1.48	6.5	91.1			
	75.00	μg/kg	72.71	7.53	10.4	72.71	4.54	6.2	96.9			
Lasiocarpine	1.00	μg/kg	0.81	0.11	13.4	0.81	0.07	8.6	80.5			
	5.00	μg/kg	4.19	0.59	14.1	4.19	0.24	5.8	83.8	0.22	0.45	112.28
	25.00	μg/kg	23.57	3.20	13.6	23.57	1.30	5.5	94.3			
	75.00	μg/kg	68.47	8.95	13.1	68.47	3.65	5.3	91.3			
Lasiocarpine- <i>N</i> -oxide	1.00	μg/kg	0.81	0.15	18.6	0.81	0.15	18.6	81.4			
	5.00	μg/kg	4.23	0.57	13.5	4.23	0.41	9.6	84.7	0.60	1.20	97.10
	25.00	μg/kg	22.42	2.14	9.5	22.42	1.16	5.2	89.7			
	75.00	μg/kg	68.65	4.58	6.7	68.65	2.91	4.2	91.5			
Lycopsamine	1.00	μg/kg	0.83	0.10	12.3	0.83	0.08	9.4	82.7			
	5.00	μg/kg	4.65	0.61	13.1	4.65	0.54	11.5	93.0	0.28	0.56	90.68
	25.00	μg/kg	22.68	3.86	17.0	22.68	0.81	3.6	90.7			
	75.00	μg/kg	71.06	8.94	12.6	71.06	1.87	2.6	94.7			
Retrorsine	1.00	μg/kg	0.96	0.14	15.1	0.96	0.14	15.1	95.5			
	5.00	μg/kg	4.66	0.93	20.0	4.66	0.62	13.3	93.1	0.34	0.67	91.55
	25.00	μg/kg	22.94	3.05	13.3	22.94	2.59	11.3	91.8			

	75.00	µg/kg	69.08	6.10	8.8	69.08	2.35	3.4	92.1			
Retrorsine- <i>N</i> -oxide	1.00	µg/kg	0.79	0.10	12.9	0.79	0.10	12.9	79.2			
	5.00	µg/kg	4.31	0.77	17.9	4.31	0.77	17.9	86.2	0.64	1.27	81.01
	25.00	µg/kg	23.69	3.42	14.4	23.69	1.47	6.2	94.8			
	75.00	µg/kg	70.80	9.91	14.0	70.80	3.47	4.9	94.4			
Senecionine	1.00	µg/kg	0.81	0.13	16.5	0.81	0.13	16.5	81.4			
	5.00	µg/kg	4.19	0.83	19.9	4.19	0.76	18.2	83.9	0.45	0.91	82.95
	25.00	µg/kg	22.91	2.92	12.7	22.91	1.03	4.5	91.6			
	75.00	µg/kg	69.81	11.96	17.1	69.81	5.01	7.2	93.1			
Senecionine- <i>N</i> -oxide	1.00	µg/kg	0.80	0.14	17.8	0.80	0.14	17.8	79.8			
	5.00	µg/kg	4.18	0.81	19.5	4.18	0.52	12.5	83.5	0.42	0.84	99.93
	25.00	µg/kg	22.58	2.66	11.8	22.58	1.98	8.8	90.3			
	75.00	µg/kg	68.89	6.05	8.8	68.89	3.41	5.0	91.9			
Seneciphylline	1.00	µg/kg	0.79	0.05	6.5	0.79	0.05	6.5	79.2			
	5.00	µg/kg	4.12	0.63	15.2	4.12	0.52	12.7	82.4	0.23	0.45	81.55
	25.00	µg/kg	22.20	2.18	9.8	22.20	1.46	6.6	88.8			
	75.00	µg/kg	67.64	4.50	6.6	67.64	2.57	3.8	90.2			
Seneciphylline- <i>N</i> -oxide	1.00	µg/kg	0.84	0.15	17.5	0.84	0.13	15.4	84.4			
	5.00	µg/kg	4.17	0.52	12.5	4.17	0.52	12.5	83.5	0.60	1.19	88.53
	25.00	µg/kg	19.80	2.26	11.4	19.80	1.31	6.6	79.2			
	75.00	µg/kg	61.75	3.90	6.3	61.75	2.70	4.4	82.3			

**Table 3.** Molecular ions and selected MS/MS fragment ions of major PAs.

RT	PAs ( $\mu\text{g}/\text{kg}$ )	Formula	$[\text{M}+\text{H}]^+$ calculated	$[\text{M}+\text{H}]^+$ experimental	Fragment pattern and abundance	Error (ppm)
7.6	Echimidine	$\text{C}_{20}\text{H}_{31}\text{NO}_7$	398.2173	398.2170	120.0810 (48), 220.1301 (2)	-0.75
4.3	Intermedine	$\text{C}_{15}\text{H}_{25}\text{NO}_5$	300.1805	300.1800	138.0916 (33), 156.1021(100)	-1.66
5.2	Intermedine <i>N</i> - oxide	$\text{C}_{15}\text{H}_{25}\text{NO}_6$	316.1755	316.1751	138.0914(42), 172.0961(100)	-1.27
8.2	Lasiocarpine	$\text{C}_{21}\text{H}_{33}\text{NO}_7$	412.2330	412.2333	120.0815(100), 238.1427(5)	-0.73
8.5	Lasiocarpine <i>N</i> - oxide	$\text{C}_{21}\text{H}_{33}\text{NO}_8$	428.2279	428.2262	94.0651(19), 120.0806 (73), 138.0908 (51)	-3.96
4.5	Lycopsamine	$\text{C}_{15}\text{H}_{25}\text{NO}_5$	300.1805	300.1805	138.0917(28), 156.1020(100)	0
6.1	Retrorsine	$\text{C}_{18}\text{H}_{25}\text{NO}_6$	352.1755	352.1744	120.0822 (100), 138.0894 (58)	-3.12
6.3	Retrorsine- <i>N</i> - oxide	$\text{C}_{18}\text{H}_{25}\text{NO}_7$	368.1704	368.1707	118.0648(100), 136.0765 (52)	0.81
7.2	Senecionine	$\text{C}_{18}\text{H}_{25}\text{NO}_5$	336.1805	336.1792	120.0810 (98) 138,0906 (33)	-3.87
7.4	Senecionine <i>N</i> - oxide	$\text{C}_{18}\text{H}_{25}\text{NO}_6$	352.1755	352.1753	94.0645(26), 118.0649 (94), 136.0741(77)	-0.56
6.5	Seneciphylline	$\text{C}_{18}\text{H}_{23}\text{NO}_5$	334.1649	334.1655	120.1818(100), 138.0911(66)	1.79
6.8	Seneciphylline <i>N</i> - oxide	$\text{C}_{18}\text{H}_{23}\text{NO}_6$	350.1598	350.1595	120.0827 (100), 136.0748 (41)	-0.85
<b>Tentative identified PAs</b>						
7.8	3'-acetylindicine	$\text{C}_{17}\text{H}_{27}\text{NO}_6$	342.1911	342.1910	94.0699 (7), 121.0680	-0.29

					(81), 120.0813 (25)	
8.1	Acetyl-lycopsamine	$C_{17}H_{27}NO_6$	342.1911	342.1911	120.0791 (46), 156.0903 (8)	0
7.1	Crotananine	$C_{17}H_{25}NO_5$	324.1805	324.1793	120.0768(92), 138.0907 (18)	-3.70
4.6	Echinatine /Indicine/Rinderine	$C_{15}H_{25}NO_5$	300.1805	300.1795	138.0912 (100), 156.0986 (28)	-1.67
7.1	Integerrimine/ Senecivernine	$C_{18}H_{25}NO_5$	336.1805	336.1813	120.0796 (100), 138.0904 (54)	2.38
6.1	Jacozine /Riddelline	$C_{18}H_{23}NO_6$	350.1598	350.1605	120.0772 (33.9), 138.0963 (6)	2
7.5	Myoscorpine <i>N</i> - oxide/Echiumine <i>N</i> -oxide	$C_{20}H_{31}NO_7$	398.2173	398.2184	120.0807 (56), 136.0693(2), 172.1412 (15)	3.26
5.2	Trichodesmine	$C_{18}H_{27}NO_6$	354.1911	354.1911	94.0700(17), 120.0810 (100)	0

Note: the corresponding chromatograms of LC-MS/MS are shown in Figure S1 in supporting information.

**FIGURE CAPTIONS**

**Figure 1.** Structures and chromatogram of 12 PAs

(a) Chemical structures of 12 PAs.

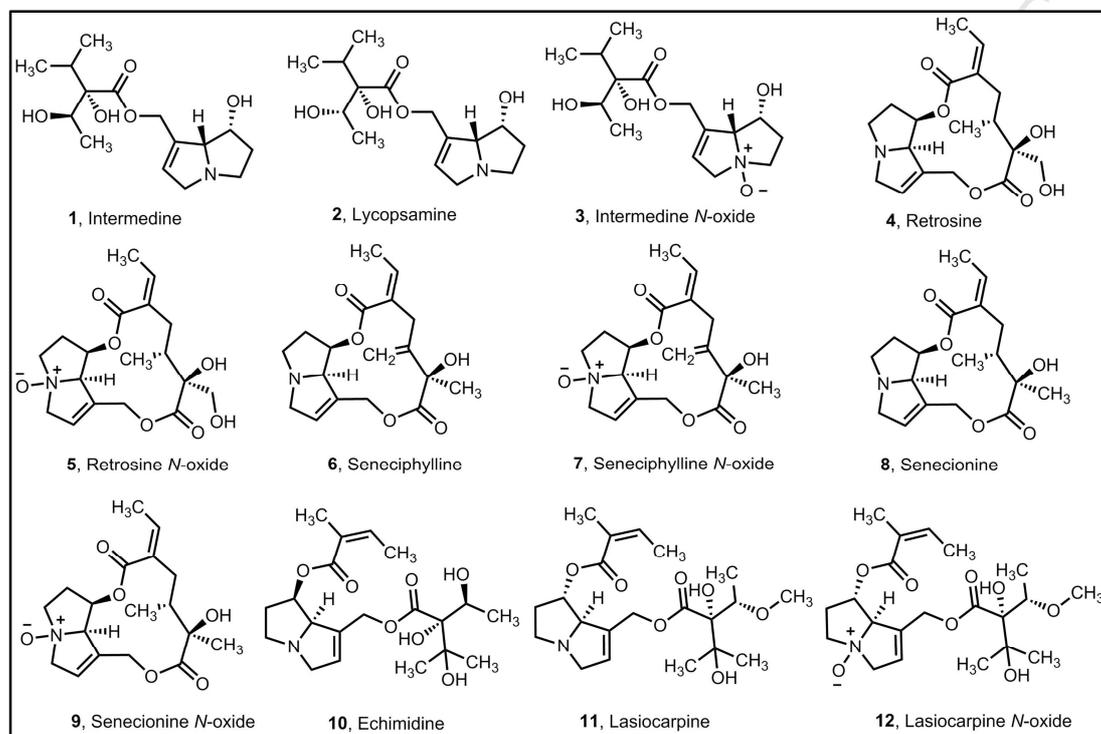
(b) Chromatogram of 12 PAs in 5 ng/ml spiked blank honey sample: 1. Intermedine, 2. Lycopsamine, 3. Intermedine *N*-oxide, 4. Retrosine, 5. Retrosine *N*-oxide, 6. Seneciphylline, 7. Seneciphylline *N*-oxide, 8. Senecionine, 9. Senecionine *N*-oxide, 10. Echimidine, 11. Lasiocarpine, 12. Lasiocarpine *N*-oxide.

**Figure 2.** (a) Correlation between relative predicted slope and measured slope in all PAs corresponding to Seneciphylline. (b) Correlation between predicted concentration and measured concentration in PAs (from 5 to 100 ng/mL) out of the model. Note: Lycopsamine *N*-oxide, heliotrine, heliotrine *N*-oxide, 7-acetylintermedine are 4 other PAs out of the 12 PAs in the model that used for verification of the prediction model.

## FIGURE GRAPHICS

Figure 1.

(a)



(b)

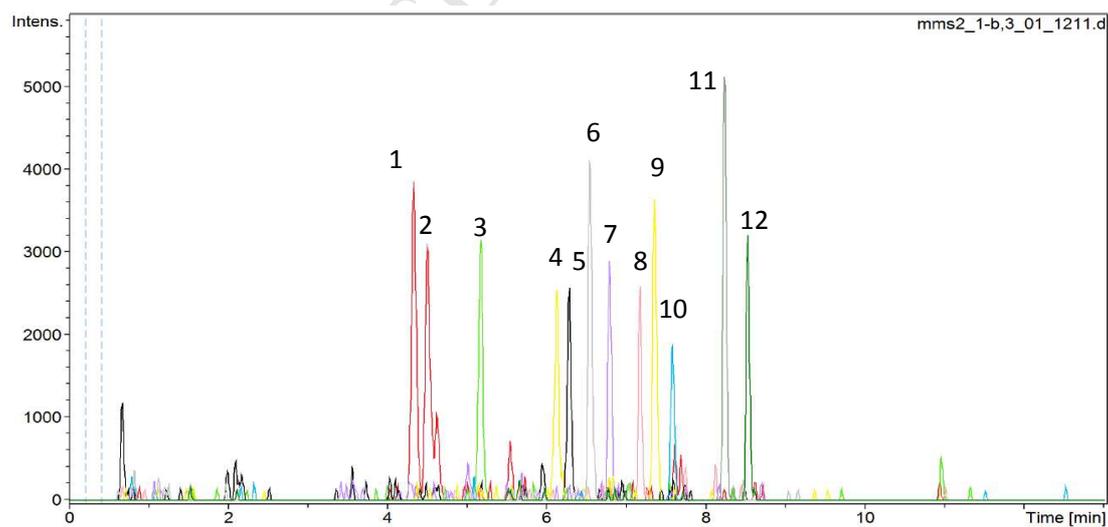
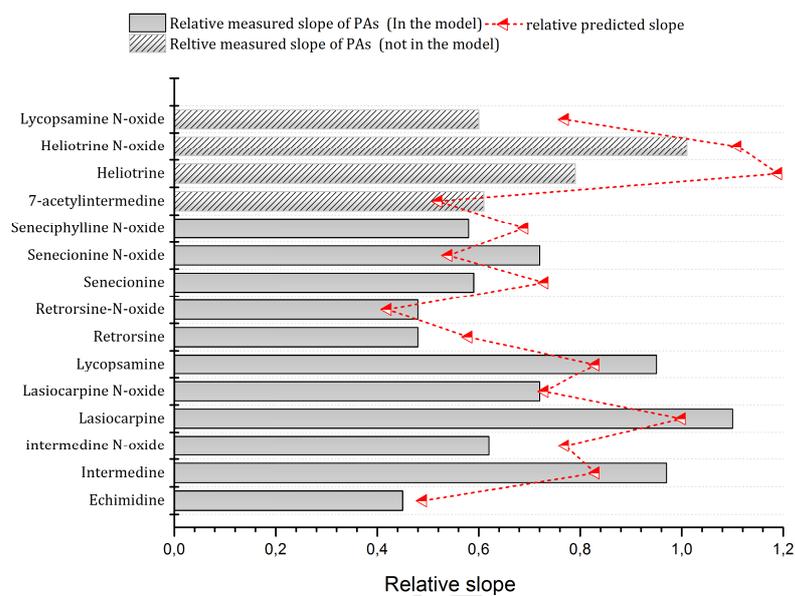
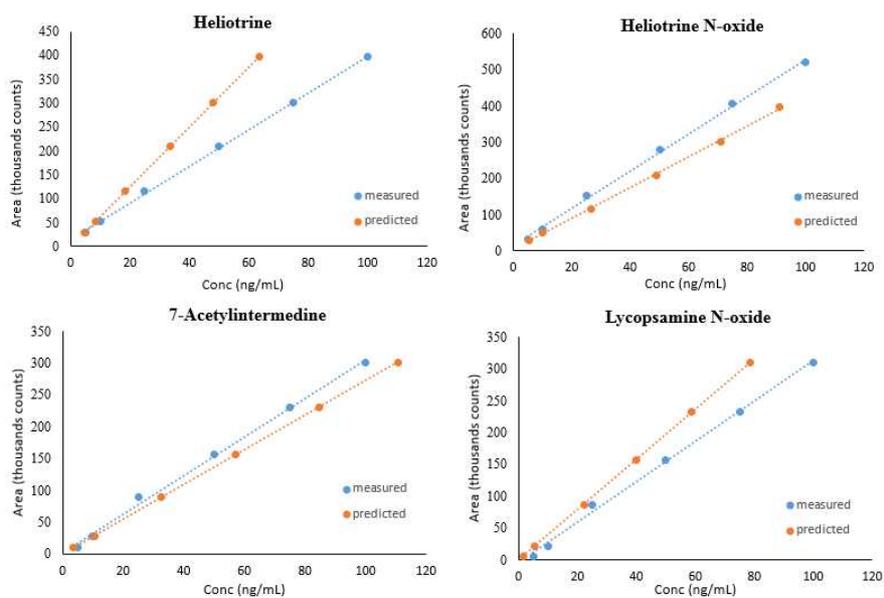


Figure 2.

(a)



(b)



- Semi-quantification of other PAs by quantitative prediction model without standards
- Method validation for the analysis 12 PAs in honey
- Multi-target screening of potential PAs in honey samples

## GRAPHIC ABSTRACT

