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Monje, Vicente; Junicke, Helena; Flores-Alsina, Xavier; Gernaey, Krist V.

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Gradients of pressure in anaerobic digesters

Vicente Monje, Helena Junicke, Xavier Flores-Alsina, Krist V. Gernaey

Process and Systems Engineering Center (PROSYS), Department of Chemical and Biochemical Engineering, Technical University of Denmark, Søltofts Plads 227, 2800 Kgs. Lyngby, Denmark.

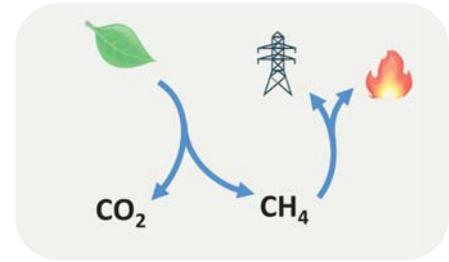


Introduction and problem statement

Anaerobic digestion is a state-of-the-art technology for organic carbon removal from waste water. Among other advantages are the high loading capacity, low sludge production, and biogas generation. Compact bioreactor designs are used to reduce the needs of surface. The bioreactors are tall cylinders up to 30 meters high. The column of water produces differences in hydrostatic pressure along the height of the bioreactor.

Objective: To model the effect of pressure on (i) gas solubility along the reactor, (ii) reaction thermodynamics, and (iii) gas-liquid mass transfer.

From biomass to biogas

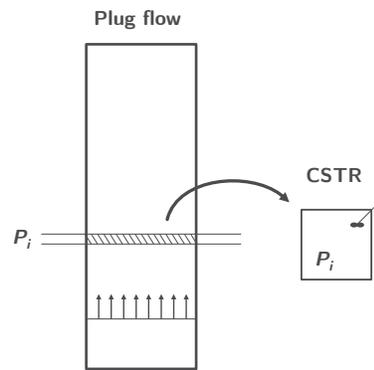


Anaerobic Digester



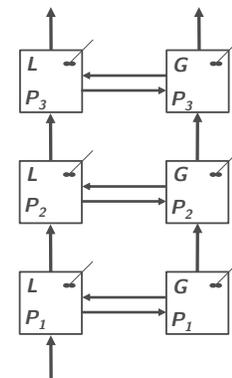
Water with high content of organic carbon dissolved enters at the bottom of the tank. In the bottom compartment, the anaerobic granular biomass converts organic compounds to CH_4 and CO_2 . In the upper compartment, biomass is retained by settling. (Picture: BIOPAQ®IC)

Conceptual representation



The anaerobic digester behaves as a plug flow reactor in terms of pressure. Slices of it can be regarded as a series of Continuous Stirred Tank Reactor (CSTR), each under a different working pressure.

Model implementation



The anaerobic digester is modelled as a chain of CSTRs, each of them under a different working pressure. Different discretizations of the bioreactor will be tested. NOTE: each slice is composed of a liquid and gas phase.

Thermodynamics

Concentration gradients along the vertical axis of the anaerobic digester may affect:

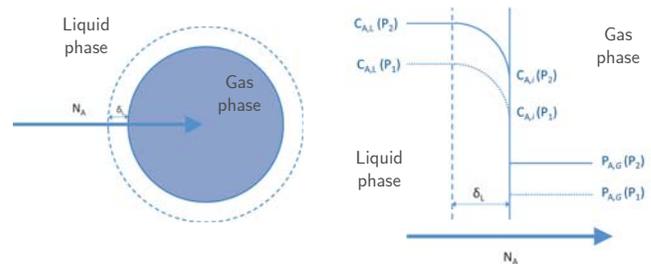
Reaction thermodynamics

The Gibbs free energy change (ΔG) of (bio)chemical reactions may change depending on the vertical position

pH and salt precipitation

Gaseous chemical species such as CO_2 influence the pH, certain zones in the reactor can allow precipitation/solubilization of salts

Gas-Liquid mass transference



P: Total pressure. The subindex stands for the pressure value.
 C_{A_i} : Concentration, i stands for chemical compound, and j for the phase (gas/liquid)
 N_A : Molar flow of compound A
 δ_L : Length of stationary layer in liquid

