Hot branching dynamics in a lightharvesting iron carbene complex revealed by ultrafast xray emission spectroscopy

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Published in: Angewandte Chemie

Link to article, DOI: 10.1002/ange.201908065

Publication date: 2020

Document Version Peer reviewed version

Link back to DTU Orbit

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Zitierweise: Angew. Chem. Int. Ed. 10.1002/anie.201908065
Angew. Chem. 10.1002/ange.201908065

Link zur VoR: http://dx.doi.org/10.1002/anie.201908065
http://dx.doi.org/10.1002/ange.201908065
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Abstract

Iron nitrogen heterocyclic carbenes (NHC) have received a great deal of attention recently, due to their growing potential as e.g. light sensitizers and photocatalysts. We present a sub-ps x-ray spectroscopy study of a FeII\textsuperscript{I}NHC complex allowing us to identify and quantify the states involved in the deactivation cascade after light absorption. We find that excited molecules relax back to the ground state along two pathways: after population of a hot 3MLCT state, from the initially excited 1MLCT state, £30\% of the molecules undergo ultrafast (150 fs) relaxation to the 3MC state, in competition with vibrational relaxation and cooling to the relaxed 3MLCT state. The relaxed 3MLCT state then decays much more slowly (7.6 ps) to the 5MC state. The 3MC state is rapidly (2.2 ps) deactivated to the ground state. The 5MC state is not involved in the deactivation pathway. The ultrafast partial deactivation of the 3MLCT state constitutes a loss channel from the point of view of photochemical efficiency and highlights the necessity to screen other FeNHC complexes (and perhaps other transition metal complexes) for this ultrafast decay of 3MLCT population, in order to optimize photochemical performance.
1 Introduction

Light harvesting systems based on first-row transition metals like iron are attractive for large-scale application due to their abundance and non-toxic properties. In a light harvesting complex, the absorption of light creates a charge separated state and the study of ultrafast photo-induced dynamics has recently received considerable attention.[1, 2, 3, 4, 5, 6, 7] The ultrafast progression of excited states after the initial light absorption is determined by the shapes and crossings of the excited state potential energy surfaces, which determine the photophysical and photochemical properties of the complexes and how they can be used. A challenge is to manipulate the positions and shapes of these potential energy surfaces, such that the charge separated state becomes sufficiently long lived for reducing or oxidizing catalysts/semiconductor acceptor systems. Traditional low spin Fe$^{II}$ – polypyridyl complexes generally show ultrafast $<100$ fs decay of metal to ligand charge transfer states (MLCT) into metal centered (MC) states.[8] For instance, in the extensively studied prototype Fe$^{II}$ molecule Fe$^{II}$[(bpy)$_3$] (bpy = 2, 2$'$-bipyridine), the charge separated MLCT state deactivates on an ultrafast timescale into high spin MC states.[9, 10] Similar ultrafast transitions into $^3$MC states have been observed and described for ruthenium sensitizer complexes.[11, 12, 13] By de-stabilizing the MC scavenger states we, and other groups, have recently developed iron complexes with significantly slower deactivation of the MLCT states.[14, 15, 16, 17, 18, 19, 20, 21] This makes the iron carbene complexes interesting as a promising new class of photosensitizers [22], with recently demonstrated capability to inject electrons efficiently into a TiO$_2$ substrate, and carry out biomolecular oxidation and reduction processes.[17, 15] Ultrafast optical measurements in combination with quantum chemical calculations have provided significant insight into the excited state structure underlying this remarkable improvement of photochemical properties.[23, 24]

As an element-specific technique x-ray spectroscopy is a powerful tool to probe the electronic structure of the metal site and surrounding ligands. With the recent advent of x-ray free electron laser (XFEL) facilities with high flux and a temporal resolution better than 30 fs[25] it has become possible to identify the nature of the excited states directly.[10, 26, 27, 28, 29, 30, 31, 32] In these studies the oxidation state, spin and charge transfer characteristics of the molecular state was identified by modeling the measured time resolved spectra with a linear combination of measured or calculated static reference spectra. Here, we use ultrafast $K$$_\alpha$ and $K$$_\beta$ x-ray emission spectroscopy (XES) to investigate the excited state dynamics of one of the recently introduced Fe$^{II}$ carbene complexes, Fe$^{II}$[(bpy)(btz)]$_2$[(PF$_6$)$_2$] where btz is 4,4'-bis(1,2,3-tri-azol-5-yldene) with significantly altered photophysics compared to prototype polypyridyl Fe$^{II}$ complexes like Fe$^{II}$[(bpy)$_3$].[33]

Several of these complexes have been characterized by transient optical spectroscopy to have a $\sim 10$ ps or longer excited state lifetime that manifests itself both in excited state absorption features originating from persistent population of a MLCT excited state, as well as ground state recovery on the same timescale.[33, 34] These timescales are quite similar with recent results from time resolved x-ray measurements for Fe$^{II}$[(bpy)(CN)$_4$] [35] ascribed to a $\sim 20$ ps long-lived population of a $^3$MLCT state. This state decayed to the ground state without undergoing noticeable spin transition to a metal centered state. These times are however noticeably different from the $\sim 260$ ps lifetime obtained in a recent x-ray scattering study and assigned to a prolonged population of a high-spin MC state.[36]

One aspect of particular current interest concerns the experimental distinction between triplet states of MLCT and spectroscopically elusive MC character in the decay cascade on ultrafast to intermediate timescales in the range of 1 ps to 100 ps. This includes systems with intermediate destabilized MC states as the emerging iron carbene photosensitizers.[37, 38] The new results obtained here in particular show that the conventional picture of a consecutive or parallel progression through a number of excited states has to be revised to a branched decay involving hot (vibra-
tionally unrelaxed) and cold (vibrationally relaxed) de-excitation pathways through CT and MC states as has also been recognized by other groups.[39] This significantly enriches our understanding of the factors limiting the photochemical efficiency of this type of complexes.

2 Description of the experiment

We dissolved the molecule into anhydrous acetonitrile to 21 mM/l and measured the dynamics in a round 30 µm jet with the recycled solution in an inert gas atmosphere. The interaction region was pumped with 400 nm 70 fs pulses at a fluence of 0.85 mJ/mm² predominantly within the linear absorption regime and probed near co-linearly in a 10 µm region with the 8.5 keV beam of the LCLS XFEL. The induced non-resonant Kβ emission of the central iron atom with a maximum at 7058.2 eV was detected after different time intervals from the photo-excitation on a 140k Cornell-SLAC PIXEL Array Detector (CSPAD) after dispersion on four cylindrical bent Ge ⟨620⟩ crystals.[40, 41, 42]

The Kα x-ray fluorescence was collected at 6404 eV, corresponding to the maximum intensity, with a single spherically bent Ge ⟨440⟩ crystal in Johann geometry on a second 140k CSPAD. Each pump-probe exposure was recorded separately and binned into a relative arrival time using the timing tool and selected pump-probe delay. The presented short timescale data originates from one run in which we selected a fixed pump-probe delay and used the arrival time jitter of the x-rays to probe a variety of time delays until 1.5 ps after the pump event. During a second run a series of longer time delays were selected to probe the evolving dynamics on a longer timescale up to 300 ps. In both cases the time resolution of ∼130 fs originates from the convolution of the 70 fs FWHM excitation pulse, with the 20fs-50 fs duration of the coherent SASE pulse and the ability of the timing tool to determine the relative arrival time of the two pulses.[42]

During the data reduction the recorded images of the Kβ spectra were compensated for the varying gain mode of the CS-pad, the background rejected by a threshold and the AD values converted into single photon counts. The images were then integrated into energy spectra along the von-Hamos focal axis and crystal misalignment corrected by a previously described procedure.[43]

3 Results

Figure 2 shows a series of Kβ spectra measured at selected time-delays after excitation A) together with a series of reference spectra for different spin states B). We use two independent approaches to model the temporally evolving excited state Kβ spectra shown in Figure 2 A). First we followed the previously established approach to deconvolute the spectrum at each time point with a linear combination of the reference spectra shown in Figure 2 B).[44, 31]

![Figure 2: Measured Kβ XES spectra of [FeII(bpy)(btz)2](PF6)2 at a different time scale A, reference Kβ XES spectra for different spin state B.](image)

In this approach no prior assumption is made which dynamic model the molecule is following except for the selection of what states to include in the de-convolution. In this language the MLCT excited states would correspond to a doublet, or quartet after an intersystem crossing and the MC states corresponds to a triplet or quintet reference state (please see SI for further information). As references for this deconvolution we used for a doublet: ([FeIII(bpy)3]3+), a triplet: (FeIIIphtcy, phtcy=phthalocyanine), a quartet: (FeIIIphtcy) and a quintet: (FeIV(Phe)n2(NCS)2, Phen = phenanthroline, NCS = isothiocyanate). The ground state of these ligands is strongly influenced by the σ-donation and π-backbonding character of the ligand field. To achieve a good agreement with the measured ground-state we used as model the linear combination of the singlet FeIII complexes [FeII(bpy)3](PF6)2 (75 %) and [FeII(CN)6] (25 %). Creating similar mixes of reference spectra for the excited states is not feasible due to the lack of isolated states at other time points and the thus limited statistical certainty. The challenge of this approach is thus the use of reference spectra that are potentially slightly different from the actual
spectra of a given state, introducing sources of error.

Modelling of the excitation fraction allows the conclusion that in this experiment \( \approx 8\% \) of the excited fraction of molecules have absorbed a second photon in the excited state, leading to a very long lived state that manifests as an oxidized/ionized state at the iron center. This long lived state was then taken into account in the kinetic and spectral modeling.

![Comparison of population parameters as a function of time gained by the linear combination of reference spectra for each time point and the model formed by global analysis. The fractional population can be understood as the fraction of excited molecules that are in this state.](image)

Figure 3: Comparison of population parameters as a function of time gained by the linear combination of reference spectra for each time point and the model formed by global analysis. The fractional population can be understood as the fraction of excited molecules that are in this state.

The result of forming a linear combination of model spectra for each time point is a set of parameters representing the population in each spin/excitation state. For a good description of the temporal evolution of the measured Fe - K\( \beta \) XES spectrum in Figure 2 the model spectra of the ground state (GS), \(^3\)MLCT and \(^3\)MC states are required. Figure 3 shows the temporal evolution of population of these states, with an assigned standard confidence interval. Linear combinations of model spectra without the ionized state, or with the involvement of the quintet MC state were tested, but led to a worse description of the measured data, as judged from an increased local (larger uncertainty) and total error (\( \chi^2 \) of the fit). The \(^1\)MLCT state was omitted from this plot as its typical lifetime is shorter than the instrument response function and the \(^1\)MLCT and \(^3\)MLCT K\( \beta \) spectra are very similar. The state population dynamics resulting from the model spectra based analysis does not itself provide a characterization of the temporal evolution in terms of exponential time constants. The discussion below of the population dynamics is therefore quantified with the help of the time constants generated by the global analysis method discussed below; as can be seen from Figure 3 there is a very good agreement between the temporal evolution generated by the two methods.

The population dynamics resulting from model spectra fitting (Figure 3) exhibits an ultrafast "bleach" of the ground state and a delayed ps-scale exponential recovery. We observe an immediate formation of the \(^3\)MLCT state, which has a single-exponential decay rate identical to the rate of the GS recovery. A fraction (\( \approx 30\% \)) of the \(^3\)MC population forms within the time resolution of the experiment. The \( \sim 0.7 \) fractional population of the \(^3\)MLCT state and only 0.3 of the ultrafast \(^3\)MC formation suggests another relaxation pathway from the \(^3\)MLCT state. The nature of this path becomes clear when we consider the decay of the \(^3\)MC population – it decays with the same delayed single exponential kinetics as the \(^3\)MLCT decay and GS recovery. Since the \(^3\)MC state is expected to be populated from the \(^3\)MLCT state (due to its significantly lower energy[33]) and both states decay with the same time constant (7.6 ps), this strongly suggests a relaxation pattern where the \(^3\)MLCT \( \rightarrow \) \(^3\)MC process is the rate limiting step in the sequence \(^3\)MLCT \( \rightarrow \) \(^3\)MC \( \rightarrow \) GS. For \(^3\)MC and \(^3\)MLCT decays (and GS recovery) to be the same, the \(^3\)MC \( \rightarrow \) GS decay must be much faster than the \(^3\)MLCT \( \rightarrow \) \(^3\)MC process. This so called “inverted kinetics” shows up as a delayed rise in the \(^3\)MC and GS kinetics, with a fitted time constant of 2.2 ps for the \(^3\)MC to GS step. The faster depopulation then population (2.2 vs 7.6 ps) of the \(^3\)MC state implies that no visible population builds up through this channel, explaining the “missing” fractional population. This behavior was already observed in optical TA measurements on the first FeNHC synthesized.[21] The partial ultrafast formation of the \(^3\)MC state implies that there will be a 2.2 ps decay of the state with the same amplitude, as well as the same partial recovery of the ground state. This decay will to a certain extent cancel the 2.2 ps rise in the inverted kinetics of the \(^3\)MC majority population, but because this is the dominating part the rise is nevertheless clearly observable.

The global analysis fitting returns the \(^3\)MC formation time constant as 150 fs, on the limit of the time resolution of the experiment, and the \(^1\)MLCT to \(^3\)MLCT transition cannot be resolved with any certainty because of very high rate and similar spectra.
Therefore, it is difficult to judge whether the ultrafast population of the $^{3}_{\text{MC}}$ state occurs directly from the $^{1}_{\text{MLCT}}$ state or from the $^{3}_{\text{MLCT}}$ state following $^{1}_{\text{MLCT}}$ to $^{3}_{\text{MLCT}}$ ISC. In the latter case, the ISC from the $^{1}_{\text{MLCT}}$ state would have to occur to a hot $^{3}_{\text{MLCT}}$ state, followed by a branched decay to the $^{3}_{\text{MC}}$ and cold $^{3}_{\text{MLCT}}$ states, in order to be compatible with the observed branching ratio.

This is consistent with the generally fast cooling of charge transfer states. Optical TA measurements on similar FeNHC complexes suggesting $\ll$100 fs $^{1}_{\text{MLCT}}$ $\rightarrow ^{3}_{\text{MLCT}}$ ISC[3, 32, 39, 43, 45, 46] would support this scenario. Time constants and associated amplitudes of the processes are summarized in the conclusions.

A to our knowledge novel approach in the field of time resolved XES is a reference free global analysis scheme which is often used in the field of transient optical absorption spectroscopy.[47, 48] In the global analysis approach used here, a parametric model based upon a set of differential equations is created for all time points. In a separate step the optimal spectral representation is calculated by solving the equation system with the observed intensity as a goal (see SI for further details). This approach avoids the use of rigid reference spectra, which may be difficult to obtain for these complexes with strong metal-ligand interaction. The optimal spectral representation is calculated anew for each step of the optimization. In the progress of the analysis a number of potential models are generated, refined with a matrix of starting parameters, and evaluated against measured spectra.

In transient optical spectroscopy an initial approach often used to get a first qualitative picture of the processes and time scales involved, is to perform single color kinetic measurements, i.e. to measure the excitation pulse induced dynamics at a few different probe wavelengths, rather than measuring the broad-band full spectral response. The analysis of single kinetic traces is also less demanding than that of the full spectral temporal evolution. The kinetic traces in Figure 4 are the single color kinetics taken at two different energies in the Fe – $K_\beta$ XES spectrum in Figure 2, 7057.6 eV and 7060.5 eV, on the low- and high-energy side of the spectrum, respectively. Kinetics measured at 7057.6 eV exhibits a decreasing signal as time progresses, while kinetics at 7060.5 eV shows an increasing signal. The reference spectra now show that the decrease/increase of signal at 7057.6/7060.5 eV is a result of increasing spin density on the iron atom with a rise time of $\sim$150 fs, indicating the presence of an ultrafast intersystem crossing, in agreement with many previous studies on iron complexes.[3, 32, 39, 43, 45, 46] At later times we observe a recovery of the excitation-induced spin momentum, but it does not fully recover to the ground state level within the 1 ns experimental window, due to the long lived oxidized/ionized state formed through excited state absorption.

![Figure 4: Kinetic spectrum of two single energetic slices extracted from the measured Fe – $K_\beta$ spectra.](image)

Left: Focusing on fast kinetics. Dots represent the observed intensity in a single temporal bin as defined by the timing tool. Right: Focusing on temporal development over multiple picoseconds. Error bars represent the variance of measurements within a temporal bin.

The model of the involved states for the analysis of single color kinetic data (Figure 4) as well as the full spectral evolution (Figure 5) is the same emerging from the model spectra based analysis (Figure 5) of the measured Fe – $K_\beta$ XES spectrum: An ultrafast, unresolved $^{3}_{\text{MLCT}}$ to a hot (not relaxed) $^{3}_{\text{MLCT}}$ ISC, followed by a branched decay of the hot $^{3}_{\text{MLCT}}$ state into the $^{3}_{\text{MC}}$ state and relaxed (cold) $^{3}_{\text{MLCT}}$ state, which decays into the $^{3}_{\text{MC}}$ state; the $^{3}_{\text{MC}}$ state finally decays back to ground state. The formation of the permanently (on this time scale) oxidized state through excited state absorption from the $^{1}_{\text{MLCT}}$ state was also included in this modeling. As already mentioned above, the two ultrafast processes $^{1}_{\text{MLCT}}$ to $^{3}_{\text{MLCT}}$ and partial decay to the $^{3}_{\text{MC}}$ state cannot clearly be distinguished as two separate processes, because they both are similar or faster than the measurement response function. In principle, the ultrafast formation of part of the $^{3}_{\text{MC}}$ population could occur from the $^{1}_{\text{MLCT}}$ state, but we choose the relaxation scheme summarized in the conclusions, based on results from optical TA on similar molecules.[3, 32, 39, 43, 45, 46] A comprehensive discussion of the other, less successful, tested models
as well as the used rate equations are given in the supplementary information.

Free parameters in the global analysis fitting of single color kinetics and XES spectral evolution are the lifetimes and the branching ratio from the hot $^3$MLCT state to the $^3$MC and relaxed $^3$MLCT states. The result of this modelling is a full 2D map of spectral intensity vs. time that was formed without imposing a particular spectrum and that used the statistical information of the whole dataset. Figure 5 shows the measured Fe $- K\beta$ XES spectrum as a function of time, the time dependent spectrum obtained from the global analysis fitting and residuals of the analysis with a feature-free error matrix. The solid lines in Figure 4 labeled “GA”, and in Figure 3 are a narrow band spectral integration of the full 2D map. It is encouraging to see that the global analysis approach generates kinetics that almost perfectly traces the temporal evolution of the population of states obtained from the model spectra based analysis (Figure 3).

It should be noted that all analysis and model optimization was performed on the absolute recorded $K\beta$ spectra. This reduces noise and avoids the introduction of spectral errors due to changes of the ground state over time, e.g. due to sample degradation. The calculated spectra can be understood as species associated spectra and be compared to reference spectra and identified with spin/charge transfer states as has been done in Figure 6. The small deviations between the global analysis spectra and reference spectra can originate from the different bonding environments in the studied molecule as compared to the molecules used for reference spectra, experimental conditions like the spectral deconvolution, or statistical limitations of minority species. During the analysis we do not enforce the use of a specific spectrum and are thus robust against influences that are present for all spectra (like misalignment of the spectrometer), or insufficient reference spectra quality. It thus appears as a useful strategy to use both analysis methods in parallel and ensure that one receives good agreement between the two methods like in Figure 3.

An independent verification of this modelling can be achieved by using the simultaneously collected $K\alpha$ line. It has been shown by us and others previously that the intensity and width of the Fe/Co $K\alpha$ emission spectrum is a good marker of the total spin momentum at the metal center.[29, 49, 50] We measured the $K\alpha$ emission at 6404 eV and can use this correlation to monitor the evolution of the total spin momentum in the molecule. Figure 7 shows the measured intensity changes and we analyze the signal independently of the previous analysis in the left-hand panel and then overlay the results we gained from the $K\beta$ analysis with the data in the middle and right-hand panel.
In the left-hand panel the data was modeled with one and two consecutive exponential decay functions, respectively, given arbitrary intensity variation. In order to superimpose the previously obtained population factors we convert the population of the excited states into a change of the total spin momentum at the iron. The state $^{1/3}\text{MLCT}$ thus corresponds to a change of $\Delta S = \frac{1}{2}$ vs. the GS, the $^3\text{MC}$ with $\Delta S = 1$ and the $^3\text{MC}$ state with $\Delta S = 2$ at the iron center.

In the independent fit two exponential functions are required to explain the measured kinetics, reproducing in general the same time constants. The model created by optimizing the measured dynamics with the $K_\beta$ spectra also shows a good agreement with the here measured spectrum. This approach is particularly valuable to test for the involvement of high spin states. If e.g. only a small fraction of the complex would decay over a quintet state, the subtle changes to the total $K_\beta$ spectrum might be difficult to identify. As the difference of the total spin momentum change is $\times 2$ for the $^5\text{MC}$ compared to the $^3\text{MC}$ the signal is significantly clearer in the $K_\alpha$ spectrum, allowing exclusion of models involving high spin states.

4 Discussion

Both optically bright and dark states are important players in the photochemical behavior of transition metal complexes, and optical spectroscopy only provides part of the complete picture. Here we have used ultrafast x-ray emission spectroscopy to monitor the relaxation pathways and identify the nature (spin state) of involved excited states following photoexcitation. Through the $K_\beta$ emission spectrum we observed an ultrafast and strong rise of the spin momentum on the iron atom (Figure 4). This spin increase is analyzed with two different methods, reference spectra based analysis of the temporal evolution of the $K_\beta$ spectrum, and a global analysis approach “inspired” by optical TA spectroscopy. In very good mutual agreement, both methods show that $\text{Fe}^{2+}(\text{bpy})(\text{btz})_2$, following photoexcitation and ultrafast ISC to a hot $^3\text{MLCT}$ state, relaxes along a branched pathway back to the ground state (Figure 8). This relaxation pattern is in addition supported by the temporal evolution of the $K_\alpha$ spectrum. In accordance with recent results from optical TA spectroscopy and quantum chemistry calculations a majority ($\sim 70\%$) of the population of excited molecules return to the ground state via inverted kinetics through a relaxed $^3\text{MLCT}$ state and the $^3\text{MC}$ state, where the transition $^3\text{MLCT} \rightarrow ^3\text{MC}$ is rate limiting (7.6 ps) and the transition $^3\text{MC} \rightarrow ^3\text{MC}$ is much faster (2.2 ps). Along a previously unknown pathway $\sim 30\%$ of the $^3\text{MLCT}$ state population is “short circuited” by an ultrafast ($\sim 150$ fs) decay to the $^3\text{MC}$ state. From a photochemical and application point of view this implies that $\sim 30\%$ of the excited molecules are dumped through an ultrafast process into a photochemically inactive state, which in addition is very short lived (2.2 ps).

The here observed lifetime of the relaxed $^3\text{MLCT}$ state is shorter than previously measured with optical transient absorption spectroscopy ($\sim 13$ ps). Part of the reason is the different excitation wavelength (705 nm vs. 400 nm used here) and analysis method used in the two experiments. Another reason may be difficulties in state assignment. We performed new optical TA measurements using 400 nm excitation and the branched relaxation model of Figure 8 to analyze the response. This gives a 9.2 ps lifetime of the rate limiting $^3\text{MLCT} \rightarrow ^3\text{MC}$ step, close to the 7.6 ps determined with the x-ray probe. The $^3\text{MC}$ to
5 Conclusion

We have used x-ray spectroscopic techniques to investigate a Fe(II) transition metal complex of the recently developed nitrogen heterocyclic carbene type, which have drastically extended CT state lifetimes as compared to traditional polypyridyl complexes. Measurement and analysis of the temporal evolution of the $K_\beta$ and $K_\alpha$ spectra following photoexcitation helped us to identify and quantify the population dynamics of the different excited states involved in the relaxation cascade. The analysis involved two methods for extracting excited state dynamics and spectra, a model spectra based and global analysis of the time dependent $K_\beta$ spectrum. This analysis was further supported by recording the temporal evolution of the $K_\alpha$ spectrum, whose peak intensity is a sensitive marker of the total spin momentum of the molecule. The results of this analysis can be summarized as three major findings: 1) Substantiating previous conclusions from optical TA measurements and quantum chemical calculations we conclude that the $^5\text{MC}$ state is never populated in the relaxation from the $^3\text{MLCT}$ state back to ground state. Thus, a major effect of the increased electron density on the Fe-center accomplished by the NHC ligands is to move the $^5\text{MC}$ state out of the relaxation pathway; 2) A previously unknown relaxation channel is identified: an ultrafast sub-ps decay from a hot $^3\text{MLCT}$ state to the $^3\text{MC}$ state, in competition with vibrational cooling of the $^3\text{MLCT}$ state. This process constitutes a partial “short circuit” of the $^3\text{MLCT}$ state and reduces the yield of molecules available for photochemical functions. 3) The ultrafast deactivation of the $^3\text{MLCT}$ state of Fe-polypyridyl complexes, and Fe(II)(bpy)$_3$ in particular, has been explained as a sequential process involving the $^5\text{MC}$ and $^3\text{MC}$ states[43, 51, 52, 26, 53, 54], or a direct one-step process[3]. The present results, highlighting a branched decay of a hot $^3\text{MLCT}$ state, may perhaps offer a possibility to reconcile these apparently disagreeing results: also in the Fe-polypyridyl complexes there is perhaps a branched decay of the $^3\text{MLCT}$ state, but various time resolved methods (e.g. optical and x-ray) used to monitor the relaxation processes have different sensitivity to the two pathways and states.

Acknowledgments

PP acknowledges support from the Crafoord Foundation, the Knut and Alice Wallenberg Foundation (KAW), the Swedish Research Council (VR), the Swedish Energy Agency (Energimyndigheten), and the Swedish Foundation for Strategic Research (SSF). JU acknowledges support from the Crafoord Foundation, the Knut and Alice Wallenberg Foundation (KAW) and from the Trygger Foundation for this research. The DTU-affiliated authors gratefully acknowledge DANSCATT for supporting the beamtime efforts. M. M. N., M. G. L., E. B., A. O. D., and K. B. M. thank the Independent Research Fund Denmark for financial support under Grants No. DFF-4002-00272 and No. DFF-8021-00347B. AOD
acknowledges support from the Icelandic Research Fund, Grant 196279-051. S. E. C. acknowledges funding from the Helmholtz Recognition Award. The ELI-ALPS project (GINOP-2.3.6-15-2015-00001) is financed by the European Union and co-financed by the European Regional Development Fund. This work is in part financed by the ‘Lendület’ (Momentum) Program of the Hungarian Academy of Sciences (LP2013-59), the Government of Hungary and the European Regional Development Fund under Grant VEKOP-2.3.2-16-2017-00015, and the National Research, Development and Innovation Fund (NKFIH FK 124460). Our thanks to the SLAC/LCLS facility and the staff from XPP for granting us beamtime and enabling this study.

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