



Decolourisations and biodegradations of model azo dye solutions using a sequence batch reactor, followed by ultrafiltration

Korenak, J.; Ploder, J.; Trček, J.; Hélix-Nielsen, Claus; Petrinic, I.

Published in:

International Journal of Environmental Science and Technology

Link to article, DOI:

[10.1007/s13762-017-1406-z](https://doi.org/10.1007/s13762-017-1406-z)

Publication date:

2018

Document Version

Peer reviewed version

[Link back to DTU Orbit](#)

Citation (APA):

Korenak, J., Ploder, J., Trček, J., Hélix-Nielsen, C., & Petrinic, I. (2018). Decolourisations and biodegradations of model azo dye solutions using a sequence batch reactor, followed by ultrafiltration. *International Journal of Environmental Science and Technology*, 15(3), 483-492. <https://doi.org/10.1007/s13762-017-1406-z>

General rights

Copyright and moral rights for the publications made accessible in the public portal are retained by the authors and/or other copyright owners and it is a condition of accessing publications that users recognise and abide by the legal requirements associated with these rights.

- Users may download and print one copy of any publication from the public portal for the purpose of private study or research.
- You may not further distribute the material or use it for any profit-making activity or commercial gain
- You may freely distribute the URL identifying the publication in the public portal

If you believe that this document breaches copyright please contact us providing details, and we will remove access to the work immediately and investigate your claim.

International Journal of Environmental Science and Technology

DECOLOURISATIONS AND BIODEGRADATIONS OF MODEL AZO DYE SOLUTIONS USING A SEQUENCE BATCH REACTOR, FOLLOWED BY ULTRAFILTRATION

--Manuscript Draft--

Manuscript Number:	JEST-D-16-00404R1	
Full Title:	DECOLOURISATIONS AND BIODEGRADATIONS OF MODEL AZO DYE SOLUTIONS USING A SEQUENCE BATCH REACTOR, FOLLOWED BY ULTRAFILTRATION	
Short Title:	DECOLOURISATION AND BIODEGRADATION OF A MODEL AZO DYE SOLUTION	
Article Type:	Original Paper	
Keywords:	Azo dye wastewater; Decolourisation; Microbial characterisation; Sequencing batch reactor; ultrafiltration	
Corresponding Author:	Jasmina Korenak University of Maribor SLOVENIA	
Corresponding Author Secondary Information:		
Corresponding Author's Institution:	University of Maribor	
Corresponding Author's Secondary Institution:		
First Author:	Jasmina Korenak	
First Author Secondary Information:		
Order of Authors:	Jasmina Korenak	
	Jana Ploder	
	Janja Trcek	
	Claus Helix-Nielsen	
	Irena Petrinic	
Order of Authors Secondary Information:		
Funding Information:	EraSME (430-26/2010/40)	Dr Irena Petrinic
Abstract:	<p>The main objective of this study was to investigate the efficiency of biological treatment of azo dye-containing wastewater with a sequencing batch reactor system, followed by ultrafiltration. The performance of the system was quantified by measuring the chemical oxygen demand and azo dye concentration. The biodegradation was carried out under combined alternating anaerobic and aerobic conditions with Nylosan yellow E2RL SGR as a model azo dye contaminant. The bioprocess revealed a maximal reduction of chemical oxygen demand and dye removal efficiency of 91% and 85%, respectively. After ultrafiltration of effluent from the biological treatment, the efficiency increased to 94% for chemical oxygen demand and to 97% for the azo dye decolourisation. Samples of activated sludge from the bioprocess were collected for microbial characterisation. Bacteria and fungi were isolated and identified by 16S rRNA gene and ITS1-5.8S rDNA-ITS2 sequence analysis, respectively. <i>Serratia marcescens</i> and <i>Klebsiella oxytoca</i> were the most common bacteria with the highest number present during the aerobic and anaerobic phases of the bioprocess. In addition, a high number of <i>Elizabethkingia miricola</i>, <i>Morganella morganii</i>, <i>Comamonas testosteroni</i>, <i>Trichosporon</i> sp. and <i>Galactomyces</i> sp. were detected. Taken together, our results demonstrated that the sequencing batch reactor system combined with ultrafiltration is an efficient technique for treatment of wastewater containing azo dye. Moreover, the ultrafiltration effectively removes the microbiota from the final effluent resulting in stable</p>	

product water.

Author Query Form

Author Query Form	
Reviewers #1 comments	Author(s) response*
<p>Comments to "Inactivation of Escherichia coli Planktonic Cells by Multi-walled Carbon Nanotubes in Suspensions: Effect of Surface Functionalization Coupled with Medium Nutrition Level"</p>	<p>The reviewer #1 provided a comments for an article titled "Inactivation of Escherichia coli Planktonic Cells by Multi-walled Carbon Nanotubes in Suspensions: Effect of Surface Functionalization Coupled with Medium Nutrition Level" and has been sent to us. Therefore, the comments are not relevant for our article titled “ DECOLOURISATIONS AND BIODEGRADATIONS OF MODEL AZO DYE SOLUTIONS USING A SEQUENCE BATCH REACTOR, FOLLOWED BY ULTRAFILTRATION “</p>

Author Query Form

Author Query Form	
Reviewers #2 comments	Author(s) response*
Introduction:	
1. Page 4, line 3, it should be "for color removal" or "for dye degradation".	Corrected Please see page 4, line 3.
2. Page 4, line 8, "Cinar et al", not "at al".	Corrected Please see page 4, line 7.
3. Page 4, line 8-15, add a blank between 24 and hour, and check the whole manuscript please.	Corrected Please see whole manuscript.
Materials and methods:	
1. Page 4, line 41, check the unit please.	Corrected Please see page 4, line 30, 31
Results and discussion:	
1. Page 6, line 46, check the first sentence please.	Corrected Please see page 6, line 20, 21
2. Page 7, line 8, it should be "dye degradation".	Corrected Please see page 6, line 25
3. Page 7, line 17, check "determine the SBR process" please.	Corrected Please see page 6, line 30, 31

<p>4. Actually, I'm interested in the re-coloration process. Is it still toxic? Can you do more experiments to prove it? Because we know the degradation products of azo dye under anaerobic condition are more toxic than azo dyes. And will any method solve it?</p>	<p>The reductive cleavage of azo compounds to aromatic amines needs anaerobic conditions and then bacterial biodegradation of the aromatic amines is aerobic process. Therefore, a combined anaerobic–aerobic bacterial process is most effective.</p> <p>Unfortunately, currently we are not able to do some more tests regarding this to proof the detoxified effluent.</p>
<p>5. Page 9, line 33, check "per se" please.</p>	<p>Corrected Please see page 8, line 32</p>
<p>6. Error bars need to be added to all figures.</p>	<p>Error bars are added Please see Fig. 2 – 6.</p>

DECOLOURISATIONS AND BIODEGRADATIONS OF MODEL AZO DYE SOLUTIONS USING A SEQUENCE BATCH REACTOR, FOLLOWED BY ULTRAFILTRATION

DECOLOURISATION AND BIODEGRADATION OF A MODEL AZO DYE SOLUTION

Jasmina Korenak¹, Jana Ploder², Janja Trček^{1,2}, Claus Hélix-Nielsen^{1,3}, Irena Petrinic¹,

¹University of Maribor, Faculty of Chemistry and Chemical Engineering, Smetanova ulica 17, SI-2000 Maribor, Slovenia

²University of Maribor, Faculty of Natural Sciences and Mathematics, Dep. of Biology, Koroska cesta 160, SI-2000 Maribor, Slovenia

³Technical University of Denmark, Department of Environmental Engineering, Bygningstorvet 115, DK2800 Kgs. Lyngby, Denmark.

(jasmina.korenak@um.si, jana.ploder@gmail.com, janja.trcek@um.si, clhe@env.dtu.dk, irena.petrinic@um.si)

Corresponding author: Jasmina Korenak
Address: University of Maribor, Faculty of Chemistry and Chemical Engineering, Smetanova ulica 17, SI-2000 Maribor, Slovenia
E-mail: jasmina.korenak@um.si
Tel: +386 2 2294 474
Fax: +386 2 2527 774

1 **DECOLOURISATIONS AND BIODEGRADATIONS OF MODEL AZO**
2 **DYE SOLUTIONS USING A SEQUENCE BATCH REACTOR,**
3 **FOLLOWED BY ULTRAFILTRATION**

4
5
6 **DECOLOURISATION AND BIODEGRADATION OF A MODEL AZO DYE**
7 **SOLUTION**
8

1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25
26
27
28
29
30
31
32
33
34
35
36
37
38
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

1 **Abstract:**

2 The main objective of this study was to investigate the efficiency of biological treatment of azo
3 dye-containing wastewater with a sequencing batch reactor system, followed by ultrafiltration.
4 The performance of the system was quantified by measuring the chemical oxygen demand and
5 azo dye concentration. The **biodegradation** was carried out under combined alternating
6 anaerobic and aerobic conditions with Nylosan yellow E2RL SGR as a model azo dye
7 contaminant. The bioprocess revealed a maximal reduction of chemical oxygen demand and
8 dye removal efficiency of 91% and 85%, respectively. **After ultrafiltration of effluent from the**
9 **biological treatment**, the efficiency increased to 94% for chemical oxygen demand and to 97%
10 for the azo dye decolourisation. Samples of activated sludge from the **bioprocess** were collected
11 for microbial characterisation. Bacteria and fungi were isolated and identified by 16S rRNA
12 gene and ITS1-5.8S rDNA-ITS2 sequence analysis, respectively. *Serratia marcescens* and
13 *Klebsiella oxytoca* were the most common bacteria with the highest number present during the
14 aerobic and anaerobic phases of the bioprocess. In addition, a high number of *Elizabethkingia*
15 *miricola*, *Morganella morganii*, *Comamonas testosterone*, *Trichosporon* sp. and *Galactomyces*
16 sp. were detected. Taken together, our results demonstrated that the **sequencing batch reactor**
17 **system combined with ultrafiltration** is an efficient technique for treatment of wastewater
18 containing azo dye. Moreover, the ultrafiltration effectively removes the microbiota from the
19 final effluent resulting in stable product water.

20
21 **Keywords:** Azo dye wastewater, Decolourisation, Microbial characterisation, Sequencing
22 batch reactor, Ultrafiltration.
23

1. Introduction

The textile industry is a large water consumer and overall the sector in Europe is suffering due to more stringent wastewater discharge limits, raising water costs and limited water resources. Consequently, the sector is losing competitiveness on international markets, which results in closure of companies and re-location to countries outside of Europe. Thus, in order to ensure a future competitive European textile industry there is a need to provide efficient treatment solutions for textile wastewater with final effluents that can be discharged or reused.

Raw textile wastewaters typically have a high chemical oxygen demand (COD) and contain a wide-variety of contaminants, such as salts, surfactants, soaps, enzymes, oxidizing and reducing agents and reactive dyes (Hao et al., 2000, Solís et al., 2012). Most of the coloured, high molecular-weight dye compounds are resistant to biological degradation, especially reactive dyes which can persist in the environment under natural conditions for 40 years or more (Hao et al., 2000, Brás et al., 2001, Khehra et al., 2006). Conventional biological wastewater processes can reduce COD significantly. However, these processes do not eliminate dyes. The combined ability to minimize COD and colourisation is of paramount importance in order to reuse the final effluent. Thus, a combination of different techniques is required (Hao et al., 2000).

Biotechnological treatment approaches (anaerobic or aerobic digestion or a combination of the two) for decolourising azo-containing wastewaters have been demonstrated (Saratale et al., 2011, Wu et al., 2012). Under aerobic conditions, e.g. in activated sludge systems for wastewater treatment, azo dye decolourisation can be achieved, although with low activity as oxygen is a more effective electron acceptor than the azo dyes (Chengalroyen and Dabbs, 2013). On the other hand, under anaerobic conditions, e.g. by using granular or flocculent sludges, azo dyes are generally the sole terminal electron acceptor and, because of this, a better decolourisation is obtained. The reduced products of anaerobic treatment (aromatic amines) are, in general, recalcitrant which means that complete biodegradation of azo dyes cannot be accomplished under anaerobic conditions only. However, bioconversion of these aromatic compounds under aerobic conditions is relatively quick and the reduction of azo dyes is generally regarded as a rate-limiting step in the overall biodegradation (Chen, 2006, Chengalroyen and Dabbs, 2013, Solís et al., 2012).

In microbial consortia, micro-organisms work synergistically, meaning that individual strains may degrade the dye molecule at different positions or may utilise metabolites produced by co-existing strains for further mineralisation (Chengalroyen and Dabbs, 2013). Besides, microbial cultures can have a broad spectrum for dye decolourisation due to variations in the enzymatic reactions of individual types of micro-organisms (Chengalroyen and Dabbs, 2013). It has been reported that azo dyes were almost completely decolourised by *Pseudomonas* sp. (Phugare et al., 2011, Jadhav et al., 2010). Some bacteria, such as *Sphingomonas* use the dye as a carbon source (Ayed et al., 2011), and also *Geobacillus* which removed efficiently dye Orange II from sewage of textile effluent (Evangelista-Barreto et al., 2009). Also other bacterial genera, such as *Acetobacter*, *Bacillus*, *Xanthomonas*, *Morganella*, *Aeromonas*, *Klebsiella* and others are reported as azo dye decolourisers (Solís et al., 2012).

Sequencing batch reactor (SBR) presents a simplistic and flexible unit in which combined aerobic/anaerobic reactions at low costs are possible to be carried out (Al-Amrani et al., 2014, Hosseini Koupaie et al., 2011). It has been widely used in recent studies to achieve dye removal (Rajaguru et al., Lourenço et al., 2000, Sponza and Işık, 2002, Ong et al., 2005, Panswad et al., 2001) and operates by sequential stages of wastewater feeding, reactive phase (anaerobic and/or

1 aerobic digestion), sedimentation, and discharge of the clarified effluent and sludge. The
2 reactive phase can be modified during operation and adjusted to achieve optimal results for dye
3 degradation and the removal of other contaminants. The effluent from the SBR can be then
4 treated with ultrafiltration (UF) or nanofiltration (NF) based on demands for the effluent
5 characteristics (Zuriaga-Agustí et al., 2010).
6

7 Çinar et al. (2008) assessed the effect of a cycle length (composed of anaerobic and aerobic
8 phase) on decolourization of a model textile dye wastewater containing the remazol brilliant
9 violet 5R (RBV-5R) as a model azo dye. They assessed the cycle length of 12, 24 and 48 hours
10 in a SBR system and identified the cycle length of 24 hours (each of anaerobic and aerobic
11 treatment proceeded for 12 hours) as optimal for dye degradation. This study did not, however,
12 investigated the effect of anaerobic and aerobic phase length on the biodegradation, the
13 characteristic which is an important parameter for optimization of SBR-based treatment
14 process.
15

16 The aim of the present study was to optimize the time length of each treatment phase, aerobic
17 and anaerobic, for maximal COD and dye removal in a lab-scale SBR-UF system using the
18 Nylosan Yellow E2RL SGR as a model azo dye. In addition, the microbial consortium was
19 investigated in order to identify microorganisms with the highest capabilities for colour
20 degradation.
21

22 The study was carried out between March 2013 and March 2014, at the Faculty of Chemistry
23 and Chemical Engineering and Faculty of Natural Sciences and Mathematics, University of
24 Maribor, Slovenia.
25

26 **2. Materials and methods**

27 **2.1 Chemicals**

28 The azo dye (C.I. index name “acid yellow 256” and commercial name Nylosan Yellow E2RL
29 SGR) was supplied by Clariant, (Frankfurt am Main, Germany). Nutrient Broth no 3 (Sigma-
30 Aldrich) with composition of meat extract 1.0 g/L, yeast extract 2.0 g/L, peptone 5.0 g/L, and
31 sodium chloride 5.0 g/L was used as micro-organism nutrient. All other chemicals were
32 analytical grade (AR) and were obtained from Sigma Aldrich or Fluka.
33

34 **2.2 Synthetic wastewater**

35 Synthetic textile wastewater was prepared with the azo dye, nutrient broth and D-glucose as
36 indicated in Table 1. D-glucose was added as carbon source as azo dyes by themselves cannot
37 provide a carbon and/or nitrogen source for micro-organisms (Sani and Banerjee, 1999, Saratale
38 et al., 2011).
39
40
41
42
43
44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

1
2 After the process had started the micro-organisms were gradually adapted to high nutrient
3 concentration (13 g/L) by linearly increasing the concentration of azo dye and D-glucose until
4 Day 21. For the remaining time these concentrations were kept constant under low nutrient
5 conditions.
6

7 **2.3 Sequencing batch reactor system**

8 Initially, a 10 L SBR was seeded with 1.5 L of activated sludge from the municipal wastewater
9 treatment plant (biomass concentration of 5.5 g/L), and 0.5 L of synthetic wastewater with dye
10 concentration from 20 to 80 mg/L (see Table 1). The SBR was operated in 24 hour cycles. Each
11 cycle comprised of periods of filling (5 min), reaction & settlement (total 23.5 hour), drawing
12 (10 min) and idling (25 min), where the reaction and settlement times were varied. During
13 filling 0.5 L model wastewater influent was injected into the reactor at a rate of 100 ml/min and
14 the same volume was taken during drawing for analysis. The temperature was between 22 °C
15 and 26 °C. The dissolved oxygen was between 0.02 and 0.18 mg O₂/L and 2.7 mg O₂/L within
16 anaerobic and aerobic phases. Aeration was performed by an aquarium aerator (Tetra WP600,
17 Tetra Werke, Melle, Germany) through porous stones inserted at the bottom of the reactor. The
18 aerator was connected to a timing device allowing for control of aerobic and anoxic periods.
19 During the entire operation (except the idling period) the reactor was stirred at 50 rpm by
20 magnetic stirrer.
21

22 **2.4 Ultrafiltration process**

23 The SBR effluent was filtered through an MN617 cellulose filter paper with 7-12 µm retention
24 (Macherey-Nagel, Düren, Germany) followed by filtration through an ultrafiltration UF PAN
25 membrane (Alfa Laval, Denmark), see Table 2. The operating pressures for the filtration steps
26 were 300 – 400 kPa in dead-end configuration.
27

28 **2.5 Analytical methods**

29 Dyestuff colour, COD concentration, pH, and mixed liquor suspended solids (MLSS)
30 concentrations were measured in supernatants from samples centrifuged at 8500 rpm for 10 min
31 at 26 ± 1 °C. Nylosan yellow concentration was measured using the absorbance peak at 384 nm
32 measured with a UV-vis spectrophotometer (Agilent Instrument, model 61103A, Maribor,
33 Slovenia), see Fig. 1. Absorbance was translated into concentrations (mg/L) using a standard
34 curve (R² = 0.998) thus allowing for determination of $c_i(0)$ and $c_i(t)$ of the dye at time 0 and t ,
35 respectively.
36

37 pH measurements were performed with a pH meter (MA 5740, Iskra, Slovenia). COD analyses
38 were performed by the SIST ISO 6060 method with titrimetric determination and MLSS was
39 measured according to the standard weight method (APHA, 2006).
40

41 The COD and colour removal efficiency of the process was calculated as

$$R_E = \frac{c_{i(0)} - c_{i(t)}}{c_{i(0)}} \times 100\% \quad \text{Eq. 1}$$

42 where $c_i(0)$ and $c_i(t)$ are the initial concentration value of component i (at process start) and after
43 reaction time (t).
44
45

2.6 Isolation and identification of microorganisms

Samples of diluted activated sludge from anaerobic and aerobic phases of the SBR process were plated on a complex NA agar (Nutrient Agar, Sigma) containing acid dye and incubated for four days at room temperature (22 – 26°C). The number of colonies was expressed as CFU/mL. Each type of morphologically different colonies was further purified to isolate the pure strains. The isolates were grown in liquid nutrient media, and harvested for DNA isolation by NucleoSpin Tissue kit (Macherey-Nagel, Düren, Germany). The bacterial 16S rRNA gene sequences were specifically amplified by PCR reaction using primers 16Sfw (5'-AAATTGAAGAGTTTGATCATGGC-3') and rH1542 (5'-AAGGAGGTGATCCAGCCGCA-3') as described previously (Slapšak et al., 2013). Primers 18F1 (5'-AGGTTTCCGTAGGTGAACCT-3') and ITS4 (5'-TCCTCCGCTTATTGATATGC-3') were used for amplification of the ITS1-5.8S rDNA-ITS2 fungal region as described by Štornik et al. (2016). The PCR products were cleaned by NucleoSpin Gel and PCR Clean-up kit (Macherey-Nagel, Düren, Germany) and the sequences were determined at Eurofins Genomics (Ebersberg, Germany). The sequences were analysed by Blastn (NCBI, <http://blast.ncbi.nlm.nih.gov/Blast.cgi>). The isolates were assigned to the genus/species to which the sequences showed the highest identity.

3. Results and discussion

The SBR treatment started under anaerobic conditions and followed by aerobic conditions within a treatment cycle of 24 hours, see Table 3. The entire 24 hour treatment cycle consisted of filling, anaerobic phase, aerobic phase, settling phase, drawing phase and idling phase. The 24 hours cycle with anaerobic and aerobic phases has been used also in other studies (Lourenço et al., 2001, Kapdan and Oztekin, 2006, Hakimelahi et al., 2012). In order to optimise the SBR process for optimal dye degradation and removal of other contaminants, a total of 100 days of SBR operation was divided into three periods, designated A, B and C. For each of the periods the time length of anaerobic and aerobic phases varied (presented in Table 3).

3.1 COD and colour removal efficiency in the SBR process

COD and colour removal efficiency as well as increase of biomass growth and pH values were monitored. Samples were taken at the beginning of the reaction phase (feed), and at the end of the 24 hour cycle. Performance of the SBR process was evaluated based on measurements of the COD and dye concentration at the influent and effluent of the SBR process. The results for the 100 days of operation are shown in Figs. 2a and 2b. During the first 24 days of operation the colour removal efficiency fluctuated around 60% while the COD removal efficiency increased from 70 to 90%. The removal efficiency can be compared with other studies (Kapdan and Oztekin, 2006). During this time the reactor was operated according to the period A described in Table 3. The reason for the low dye removal during period A could be microorganisms that have not yet been sufficiently adapted to the solution containing dye molecules. As dye substances are difficult to be degraded, more time is required to accumulate sufficient micro-organisms for dye degradation.

Azo dyes can be degraded by microbial consortium under anaerobic conditions. The products of these reactions are aromatic amines (Chengalroyen and Dabbs, 2013). A longer anaerobic reaction phase has been shown to result in a higher dye decolourization efficiency (Al-Amrani et al., 2014). Therefore, the anaerobic phase duration was increased from 9 to 11 hours on day 24 in an attempt to increase dye removal efficiency. Thus, the durations of aerobic and settling phases were reduced by one hour each. This operational mode is described as period B in Table 3. However, this change did not increase the colour removal efficiency significantly (Fig. 2a).

1 In fact, it led to a decrease in COD removal efficiency as presented in Fig. 2b. After three days
2 of lag phase, the COD removal efficiency decreased from 90% and stabilised at a value around
3 50% at day 40. As the prolonged anaerobic phase did not lead to an increase in dye removal
4 and also COD removal efficiency was reduced, the SBR operating conditions was changed at
5 day 58 by reducing the anaerobic phase to 9 hours, increasing the aerobic phase to 13 hours and
6 keeping the settling phase at 1.5 hours (presented as period C in Table 3). This resulted in colour
7 removal at around 80% during the remaining experimental period albeit with some fluctuations
8 (Fig. 2a), and COD removal at around 90% after three days of period C (Fig. 2b).

10 In order to gain more insight into the effects of changing the SBR operating conditions, colour
11 degradation and COD removal during the 24 hour cycle were analysed at four selected points:
12 at day 17 of period A, on which COD removal has stabilised, at day 24, immediately after
13 changing from period A to B, at day 39 of period B where the dye removal efficiency reached
14 80% and COD removal had stabilised around 50% and at day 73 of period C where COD
15 removal efficiency has increased to 90% in average while dye removal has stabilized at around
16 80%. These data are presented in Figs. 3a-d and Figs. 4a-d.

18 In general, the majority of the dye removal occurred in the anaerobic phases, as also observed
19 by Hakimelahi et al. (2012). Re-colourisation during the aerobic phase has also been observed
20 in earlier work (Cruz and Buitron, 2001, Libra et al., 2004, Sponza and Işık, 2005) and has been
21 explained by unstable aromatic amines being generated as a by-product of dye degradation
22 under anaerobic conditions. These amines can easily oxidise and re-colour when exposed to
23 oxygen during the aerobic phases. In addition, absorption of colour into the sludge biomass
24 may also influence on the overall removal efficiency (Otero et al., 2003, Wang et al., 2006,
25 Sirianuntapiboon and Srisornsak, 2007).

27 Early in the process (day 17) a pronounced re-colouration in the anaerobic phase was observed.
28 Although dye removal efficiency reached 50% during the anaerobic phase, the efficiency
29 declined during the aerobic phase resulting in an overall poor performance. At day 24, just after
30 entering period B (with an increased anaerobic phase from 9 hours to 11 hours), re-colouration
31 in the anaerobic phase was less pronounced, however, dye removal efficiency never exceeded
32 35%. At day 39, after 15 days under period B conditions, dye removal efficiency increased to
33 43% at the end of the anaerobic phase, reaching almost 60% at the end of the aerobic phase. At
34 day 73 of period B no re-colouration in the anaerobic phase was observed and the removal
35 efficiency reached 65% before entering the aerobic phase. Despite a slight decrease, to about
36 60% at the end of the aerobic phase, the overall performance (reflected in the area under the
37 curve) was optimal, demonstrating the feasibility of SBR-based dye degradation.

39 Figs. 4a-d are presenting the corresponding COD values during the 24 hour cycle. Generally,
40 the COD measurements showed that the COD substantially reduced during the aerobic phase.
41 The average removal efficiency at days 17, 24, and 73 were 34.7%, 63.9%, and 56.8%,
42 respectively. This is consistent with the results of Asadi et al. (2012) who studied the effect of
43 hydraulic retention time and aeration time on COD removal in a bioreactor. The efficiency at
44 day 39 was only 4% which likely reflects the formation of inhibitory products within the
45 microbial consortium. Thus, in conclusion, COD removal was the most efficient within period
46 C, concomitantly with the highest degree of dye removal.

48 Concomitantly with the changes of dye concentrations and COD reduction, we analysed the
49 overall changes in biomass density and pH values. The results are presented in Figs. 5a-b.
50 During the first ten days after the inoculation of the activated sludge, the biomass decreased

1 slightly what is consistent with the fact that the micro-organisms were introduced into the
2 system from an open tank with aerobic conditions, whereas in a closed system the alternating
3 anaerobic and aerobic conditions were provided. In addition, temperature, composition of the
4 nutritional broth, and pH values were different in our SBR system in comparison to the original
5 ecological conditions from where the activated originated. After the lag phase, the sludge
6 concentration increased up to day 34 (first part of period B) indicating the exponential growth
7 phase of biomass. Between days 34 and 40 (second part of period B) the sludge concentration
8 only increased very little. After day 40 the biomass was decaying, indicating the depletion of
9 the nutrients in the medium, and the formation of inhibitory products. After entering period C
10 during which the duration of the aerobic reaction phase was extended, the biomass
11 concentration increased to around 11 g/L. Before biomass decrease in period B, the pH values
12 at the end of the anaerobic reaction phase and pH of the effluent dropped significantly. Finally,
13 at the beginning of period C, pH values were restored to values of period A.

14
15 When comparing COD removal (Fig. 2b) with biomass growth in the SBR (Fig. 5a), one can
16 observe that the COD removal was maximal during the exponential growth of micro-organisms
17 (during period A). The decrease in COD removal efficiency in the effluent is evident after day
18 33 of period B and correlates with the observed effluent pH changes. The decrease in COD
19 removal also manifested itself by changes in odour, higher turbidity, and lower settling ability.
20 The results in Fig. 5b illustrate how the pH values of influent and effluent can be used as one
21 of the indicators of a well-running SBR bioprocess. The pH values of the activated sludge were
22 between 5.5 and 7.5 after the anaerobic reaction phase and between 8 and 9 at the end of the
23 SBR cycle. During period B the pH values of effluent dropped from around 9 to below 6 at day
24 37 and COD removal efficiency was still 70%. Two days later the COD removal efficiency
25 dropped to around 42%.

3.2 UF treatment contribution on SBR effluent

26
27 The effluent from the SBR was subjected to UF treatment. In order to assess the efficiency of
28 the UF treatment two representative effluent samples were selected, one from day 33 of period
29 B where biodegradation was efficient in terms of COD removal and one from day 67 of period
30 C where degradation efficiency was low. This allowed for evaluating the UF treatment as a
31 counter-measure when the biological process **as such** is not efficient enough. The collected SBR
32 effluent was first filtered through cellulose filter paper to prevent membrane fouling and
33 afterwards passed on to the UF cell. The results of the treatments with SBR, UF, and the
34 combined method are shown in Tables 4 and 5 where R_E indicates the removal efficiency in %
35 (Eq. 1).
36

37
38 When the SBR was running efficiently (day 33) the UF treatment reduced effluent dye
39 concentration almost tenfold. When the SBR degradation process in terms of COD removal
40 was impaired (day 67), UF treatment reduced effluent dye concentration only by 40%. As the
41 colour is visible to the naked eye at concentrations as low as 1 mg/L (Öztekın, 2004), UF
42 treatment is not in itself sufficient to obtain colourless effluent. However, UF treatment has the
43 highest efficiency for colour removal when the SBR process is effective in terms of COD
44 removal. The COD concentrations were also decreased by the UF treatment by approximately
45 41% at day 33 and by 48% at day 67 i.e. the effect of UF on COD removal was strongest when
46 the SBR was working under less than optimal conditions. Still when the SBR was working well
47 (day 33) the combined SBR+UF treatment resulted in dye and COD removal efficiencies of
48 96.6% and 94% respectively.

3.3 Characterisation of the SBR microbial consortium

In order to gain insight into the SBR microbial composition, facultative aerobic micro-organisms were isolated from the SBR sludge. The prevalent bacteria were *Serratia marcescens*, *Klebsiella oxytoca*, *Elizabethkingia miricola*, and *Morganella morganii* (presented in Table 6). *Serratia* sp., *Klebsiella* sp. and *Morganella* sp. are classified into the family *Enterobacteriaceae*. They are Gram-negative, facultative aerobic, oxidase negative bacteria. Many of them are part of the intestinal flora, and their typical habitats are soil and water. Their growth is stimulated at aerobic conditions, what happened also during the SBR process. Several studies have reported the strong ability of *Klebsiella oxytoca* to metabolise toxic compounds (Wong and Yuen, 1996, Kao et al., 2003). *Klebsiella oxytoca* isolated from cyanide containing wastewater was also able to use nitriles as the sole source of nitrogen (Kao et al., 2003). Wong and Yuen (Wong and Yuen, 1996) isolated five bacteria from dye contaminated sludge and found that *Klebsiella* sp. and *K. pneumoniae* showed decolourisation ability. The bacteria *Morganella morganii* also effectively decolourise and degrade the multi-sulphonated azo dye RB-B (Kitts et al., 1994) without formation of toxic metabolites. The bacteria identified in here presented work are therefore consistent with the results of previous studies. Besides the bacteria, two types of fungi, *Trichosporon* sp. (1.5×10^3 CFU/mL) and *Galactomyces* sp. ($2.5 \times 10^3 - 3.5 \times 10^3$ CFU/mL) were also detected.

3.3.1 The decolourisation ability of specific micro-organisms

In order to assess the colour degrading ability of single isolates, we cultured individual micro-organisms in nutrient broth containing acid azo dye (60 mg/L) under static conditions at room temperature. During the process the absorbance was followed (reflecting colour removal) (Fig. 6). *Elizabethkingia miricola* showed the highest rate - nearly 70% of decolourisation after 10 days of incubation. The decolourisation efficiency of *Trichosporon* sp., *C. testosteroni* and *M. morganii* were approximately 50%. *Galactomyces* sp. decolourised wastewater by 14%.

4. Conclusions

A stable microbial consortium was cultivated within a single SBR system with alternating anaerobic and aerobic reaction phases. The microbiota performed azo dye colour removal and COD reduction in a model azo dye solution. After UF treatment of the SBR effluent, the removal efficiency was 94% for COD and 97% for the azo dye. In the SBR process, the predominant bacterial species were *Serratia marcescens* and *Klebsiella oxytoca* followed by *Elizabethkingia miricola*, *Morganella morganii* and *Serratia* sp. The microbial consortium also contained fungi *Trichosporon* sp. and *Galactomyces* sp. Our results demonstrated that the alternating anaerobic and aerobic SBR process combined with UF treatment can reach a high decolouration efficiency and COD reduction of azo dye containing wastewater.

Acknowledgements

The study was financed within the EraSME project titled “High-efficient and cost-effective air lift MBR for water reuse in textile finishing”, Project number 430-26/2010/40.

References

- AL-AMRANI, W. A., LIM, P.-E., SENG, C.-E. & WAN NGAH, W. S. 2014. Factors affecting biodecolorization of azo dyes and COD removal in anoxic–aerobic REACT operated sequencing batch reactor. *Journal of the Taiwan Institute of Chemical Engineers*, 45, 609-616.
- APHA (ed.) 2006. *Standard Methods for the Examination of Water and Wastewater*, Washington. DC, USA.
- ASADI, A., ZINATIZADEH, A. A. L. & HASNAIN ISA, M. 2012. Performance of intermittently aerated up-flow sludge bed reactor and sequencing batch reactor treating industrial estate wastewater: A comparative study. *Bioresource Technology*, 123, 495-506.
- AYED, L., MAHDHI, A., CHEREF, A. & BAKHROUF, A. 2011. Decolorization and degradation of azo dye Methyl Red by an isolated *Sphingomonas paucimobilis*: Biototoxicity and metabolites characterization. *Desalination*, 274, 272-277.
- BRÁS, R., ISABEL A. FERRA, M., PINHEIRO, H. M. & GONÇALVES, I. C. 2001. Batch tests for assessing decolourisation of azo dyes by methanogenic and mixed cultures. *Journal of Biotechnology*, 89, 155-162.
- CHEN, H. 2006. Recent Advances in Azo Dye Degrading Enzyme Research. *Current Protein and Peptide Science*, 7, 101-111.
- CHENGALROYEN, M. D. & DABBS, E. R. 2013. The microbial degradation of azo dyes: minireview. *World Journal of Microbiology and Biotechnology*, 29, 389-399.
- ÇINAR, Ö., YAŞAR, S., KERTMEN, M., DEMİRÖZ, K., YIGIT, N. Ö. & KITIS, M. 2008. Effect of cycle time on biodegradation of azo dye in sequencing batch reactor. *Process Safety and Environmental Protection*, 86, 455-460.
- CRUZ, A. & BUITRON, G. 2001. Biodegradation of disperse blue 79 using sequenced anaerobic/aerobic biofilters. *Water Science and Technology* 44, 159-166.
- EVANGELISTA-BARRETO, N. S., ALBUQUERQUE, C. D., VIEIRA, R. H. S. F. & CAMPOS-TAKAKI, G. M. 2009. Cometabolic Decolorization of the Reactive Azo Dye Orange II by *Geobacillus stearothermophilus* UCP 986. *Textile Research Journal*, 79, 1266-1273.
- HAKIMELAHI, M., MOGHADDAM, M. & HASHEMI, S. 2012. Biological treatment of wastewater containing an azo dye using mixed culture in alternating anaerobic/aerobic sequencing batch reactors. *Biotechnology and Bioprocess Engineering*, 17, 875-880.
- HAO, O. J., KIM, H. & CHIANG, P.-C. 2000. Decolorization of Wastewater. *Critical Reviews in Environmental Science and Technology*, 30, 449-505.
- HOSSEINI KOUPEI, E., ALAVI MOGHADDAM, M. R. & HASHEMI, S. H. 2011. Post-treatment of anaerobically degraded azo dye Acid Red 18 using aerobic moving bed biofilm process: Enhanced removal of aromatic amines. *Journal of Hazardous Materials*, 195, 147-154.
- JADHAV, J. P., KALYANI, D. C., TELKE, A. A., PHUGARE, S. S. & GOVINDWAR, S. P. 2010. Evaluation of the efficacy of a bacterial consortium for the removal of color, reduction of heavy metals, and toxicity from textile dye effluent. *Bioresource Technology*, 101, 165-173.
- KAO, C. M., LIU, J. K., LOU, H. R., LIN, C. S. & CHEN, S. C. 2003. Biotransformation of cyanide to methane and ammonia by *Klebsiella oxytoca*. *Chemosphere*, 50, 1055-1061.
- KAPDAN, I. K. & OZTEKIN, R. 2006. The effect of hydraulic residence time and initial COD concentration on color and COD removal performance of the anaerobic–aerobic SBR system. *Journal of Hazardous Materials*, 136, 896-901.
- KHEHRA, M. S., SAINI, H. S., SHARMA, D. K., CHADHA, B. S. & CHIMNI, S. S. 2006. Biodegradation of azo dye C.I. Acid Red 88 by an anoxic–aerobic sequential bioreactor. *Dyes and Pigments*, 70, 1-7.
- KITTS, C. L., CUNNINGHAM, D. P. & UNKEFER, P. J. 1994. Isolation of three hexahydro-1,3,5-trinitro-1,3,5-triazine-degrading species of the family Enterobacteriaceae from nitramine explosive-contaminated soil. *Applied and Environmental Microbiology*, 60, 4608-4611.
- LIBRA, J. A., BORCHERT, M., VIGELAHN, L. & STORM, T. 2004. Two stage biological treatment of a diazo reactive textile dye and the fate of the dye metabolites. *Chemosphere*, 56, 167-180.
- LOURENÇO, N. D., NOVAIS, J. M. & PINHEIRO, H. M. 2000. Reactive textile dye colour removal in a sequencing batch reactor. *Water Science and Technology*, 42, 321-328.
- LOURENÇO, N. D., NOVAIS, J. M. & PINHEIRO, H. M. 2001. Effect of some operational parameters on textile dye biodegradation in a sequential batch reactor. *Journal of Biotechnology*, 89, 163-174.
- ONG, S.-A., TOORISAKA, E., HIRATA, M. & HANO, T. 2005. Decolorization of azo dye (Orange II) in a sequential UASB–SBR system. *Separation and Purification Technology*, 42, 297-302.

- 1 OTERO, M., ROZADA, F., CALVO, L. F., GARCÍA, A. I. & MORÁN, A. 2003. Kinetic and equilibrium
2 modelling of the methylene blue removal from solution by adsorbent materials produced from
3 sewage sludges. *Biochemical Engineering Journal*, 15, 59-68.
- 4 ÖZTEKIN, R. 2004. *Colour removal from textile wastewater in sequencing batch reactor*. Dokuz Eylül
5 University.
- 6 PANSWAD, T., TECHOVANICH, A. & ANOTAI, J. 2001. Comparison of dye wastewater treatment by
7 normal and anoxic + anaerobic/aerobic SBR activated sludge processes. *Water science and
8 technology : a journal of the International Association on Water Pollution Research*, 43, 355-
9 362.
- 10 PHUGARE, S. S., KALYANI, D. C., SURWASE, S. N. & JADHAV, J. P. 2011. Ecofriendly degradation,
11 decolorization and detoxification of textile effluent by a developed bacterial consortium.
12 *Ecotoxicology and Environmental Safety*, 74, 1288-1296.
- 13 RAJAGURU, P., KALAISELVI, K., PALANIVEL, M. & SUBBURAM, V. 2000. Biodegradation of azo dyes
14 in a sequential anaerobic-aerobic system. *Applied Microbiology and Biotechnology*, 54, 268-
15 273.
- 16 SANI, R. K. & BANERJEE, U. C. 1999. Decolorization of triphenylmethane dyes and textile and dye-
17 stuff effluent by *Kurthia* sp. *Enzyme and Microbial Technology*, 24, 433-437.
- 18 SARATALE, R. G., SARATALE, G. D., CHANG, J. S. & GOVINDWAR, S. P. 2011. Bacterial
19 decolorization and degradation of azo dyes: A review. *Journal of the Taiwan Institute of
20 Chemical Engineers*, 42, 138-157.
- 21 SIRIANUNTAPIBOON, S. & SRISORNSAK, P. 2007. Removal of disperse dyes from textile wastewater
22 using bio-sludge. *Bioresource Technology*, 98, 1057-1066.
- 23 SLAPŠAK, N., CLEENWERCK, I., DE VOS, P. & TRČEK, J. 2013. *Gluconacetobacter maltaceti*, a novel
24 vinegar producing acetic acid bacterium. *Systematic and Applied Microbiology*, 36, 17-21.
- 25 SOLÍS, M., SOLÍS, A., PÉREZ, H. I., MANJARREZ, N. & FLORES, M. 2012. Microbial decolouration of
26 azo dyes: A review. *Process Biochemistry*, 47, 1723-1748.
- 27 SPONZA, D. T. & IŞIK, M. 2002. Decolorization and azo dye degradation by anaerobic/aerobic
28 sequential process. *Enzyme and Microbial Technology*, 31, 102-110.
- 29 SPONZA, D. T. & IŞIK, M. 2005. Reactor performances and fate of aromatic amines through
30 decolorization of Direct Black 38 dye under anaerobic/aerobic sequential. *Process
31 Biochemistry*, 40, 35-44.
- 32 ŠTORNİK, A., SKOK, B. & TRČEK, J. 2016. Comparison of cultivable acetic acid bacterial microbiota
33 in organic and conventional apple cider vinegar. *Food Technology and Biotechnology*, 54, 113-
34 119.
- 35 WANG, Y., MU, Y., ZHAO, Q.-B. & YU, H.-Q. 2006. Isotherms, kinetics and thermodynamics of dye
36 biosorption by anaerobic sludge. *Separation and Purification Technology*, 50, 1-7.
- 37 WONG, P. K. & YUEN, P. Y. 1996. Decolorization and biodegradation of methyl red by *Klebsiella*
38 *pneumoniae* RS-13. *Water Research*, 30, 1736-1744.
- 39 WU, Y., LI, T. & YANG, L. 2012. Mechanisms of removing pollutants from aqueous solutions by
40 microorganisms and their aggregates: A review. *Bioresource Technology*, 107, 10-18.
- 41 ZURIAGA-AGUSTÍ, E., IBORRA-CLAR, M. I., MENDOZA-ROCA, J. A., TANCREDI, M., ALCAINA-
42 MIRANDA, M. I. & IBORRA-CLAR, A. 2010. Sequencing batch reactor technology coupled with
43 nanofiltration for textile wastewater reclamation. *Chemical Engineering Journal*, 161, 122-128.
- 44
45
46
47
48
49
50
51
52
53
54
55
56
57
58
59
60
61
62
63
64
65

1 **List of figures**

2
3 **Fig. 1.** UV/VIS spectrum of Nylosan yellow E2RL SGR.

4 **Fig. 2.** Colour and COD removal efficiency: (a) colour concentration in the SBR effluent, (b)
5 COD removal efficiency.

6 **Fig. 3.** Dye removal efficiency in samples taken within the 24-hour SBR cycle: (a) at day 17;
7 (b) at day 25; (c) at day 39; (d) at day 73.

8 **Fig. 4.** COD removal efficiency measured in samples taken within the 24-hour SBR cycle: (a)
9 at day 17; (b) at day 25; (c) at day 39; (d) at day and 73.

10 **Fig. 5.** a) Biomass growth in the SBR; (b) pH values in the influent, during anoxic phase, and
11 in the effluent.

12 **Fig. 6.** Decolourization rate of specific microorganisms.

Table 1 The composition of the synthetic wastewater (influent to the SBR).

Operating day	Azo dye concentration (mg/L)	Nutrient broth concentration (g/L)	D-glucose concentration (mg/L)
1 – 21	20 linearly increasing to 80	13	4 linearly increasing to 8
22 - 100	60	6.5	8

Table 2 UF membrane specifications.

Parameter	
Material	Polyacrylonitrile (PAN)
MWCO (molecular weight cut off) (kDa)	100 – 200
pH	2 – 10
Contact angle	61.33 – 59.30
Zeta Potential (mV)	-73
Isoelectric pH	2.9

Table 3 SBR operating conditions.

Period	Day	Filling (min)	Anaerobic phase (h)	Aerobic phase(h)	Settling phase (h)	Drawing phase (min)	Idling phase (min)
A	1-24	5	9	12	2.5	10	15
B	24-58	5	11	11	1.5	10	15
C	58-100	5	9	13	1.5	10	15

Table 4 The results for dye removal from wastewater by SBR, UF, and the combined method of SBR and UF.

Operational day	Dye in influent (mg/L)	Dye in SBR effluent (mg/L)	R _E of the SBR (%)	Dye in UF effluent (mg/L)	R _E of combined method (%)
33	69.93	21.18	69.71	2.43	96.6
67	68.61	14.44	78.96	10.42	86.8

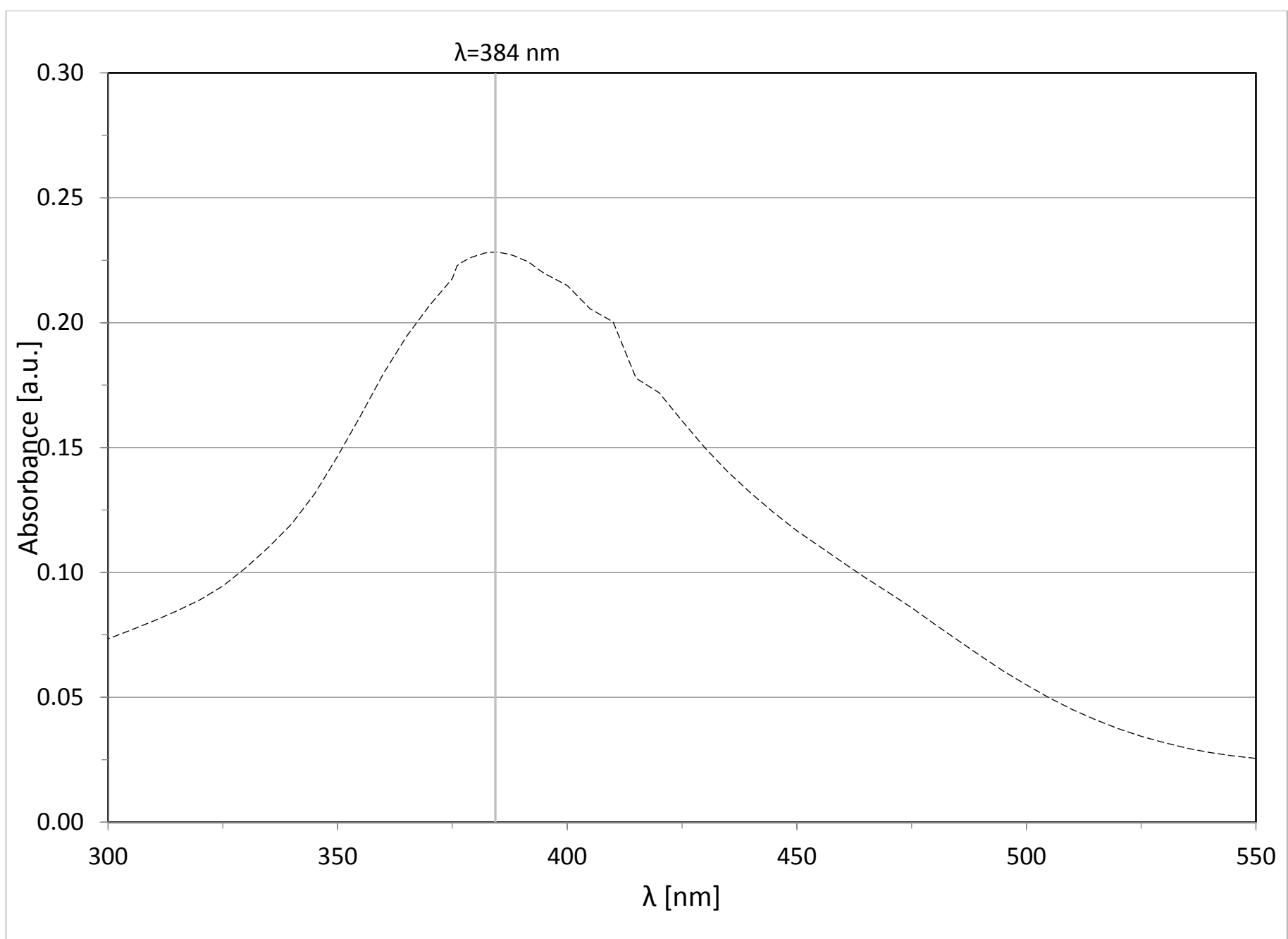
Table 5 The results for COD removal from wastewater by SBR, UF, and the combined method of SBR and UF

Operational day	COD in influent (mgO ₂ /L)	COD in SBR effluent (mgO ₂ /L)	R _E of the SBR (%)	COD in UF effluent (mgO ₂ /L)	R _E of combined method (%)
33	10,473	1,057	89.91	628	94.0
67	10,473	3,692	64.75	1,919	81.7

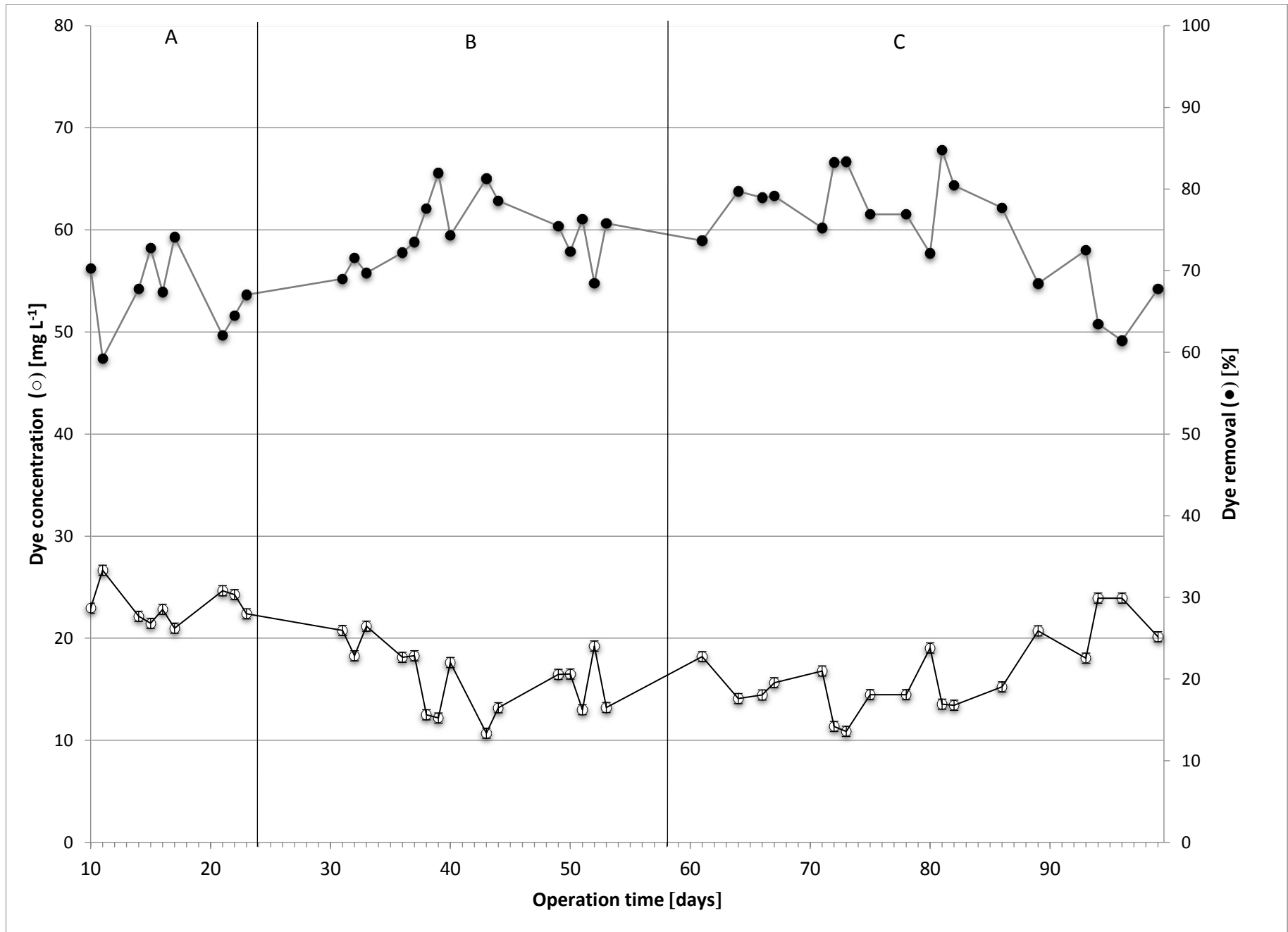
Table 6 Number of different morphological types within the activated sludge from the bioprocess, and identification based on comparisons between sequences of 16S rRNA genes.

CFU/mL after anaerobic phase	CFU/mL after aerobic phase	Identification by 16S rRNA sequence analysis
8,9 x 10 ⁵	9,5 x 10 ⁵	<i>Serratia marcescens</i>
3,2 x 10 ⁵	7,1 x 10 ⁵	<i>Klebsiella oxytoca</i>
2,5 x 10 ⁵	3,5 x 10 ⁵	<i>Elizabethkingia</i> sp.
1,5 x 10 ⁵	3,2 x 10 ⁵	<i>Morganella morganii</i>
3,8 x 10 ⁵	1,7 x 10 ⁵	<i>Serratia</i> sp.
0	1 x 10 ⁵	<i>Serratia</i> sp.
2,5 x 10 ⁵	0	<i>Serratia</i> sp.
1,5 x 10 ³	1,5 10 ³	<i>Trichosporon</i> sp.
2,5 10 ³	3,5 10 ³	<i>Galactomyces</i> sp.

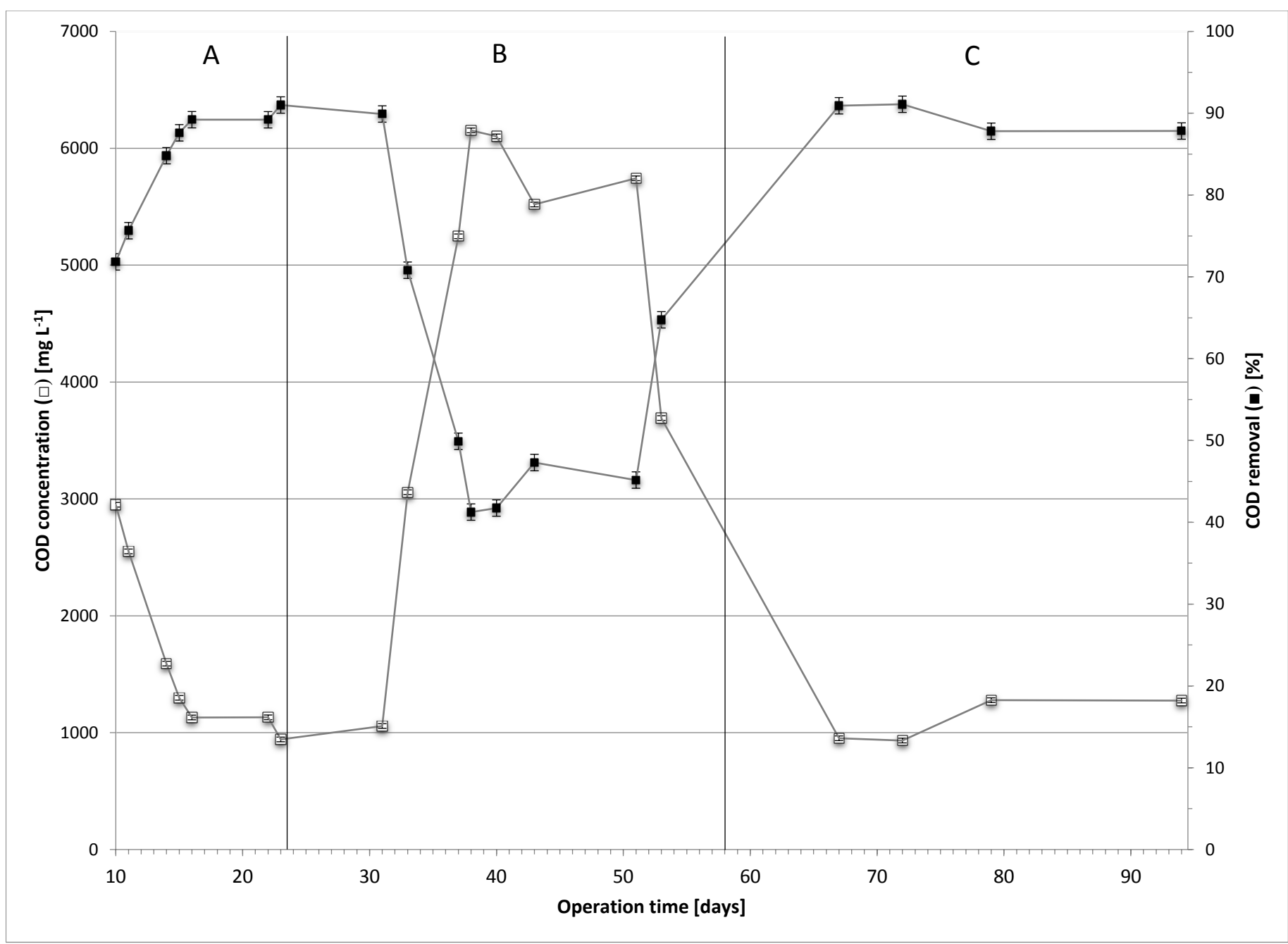
Figure



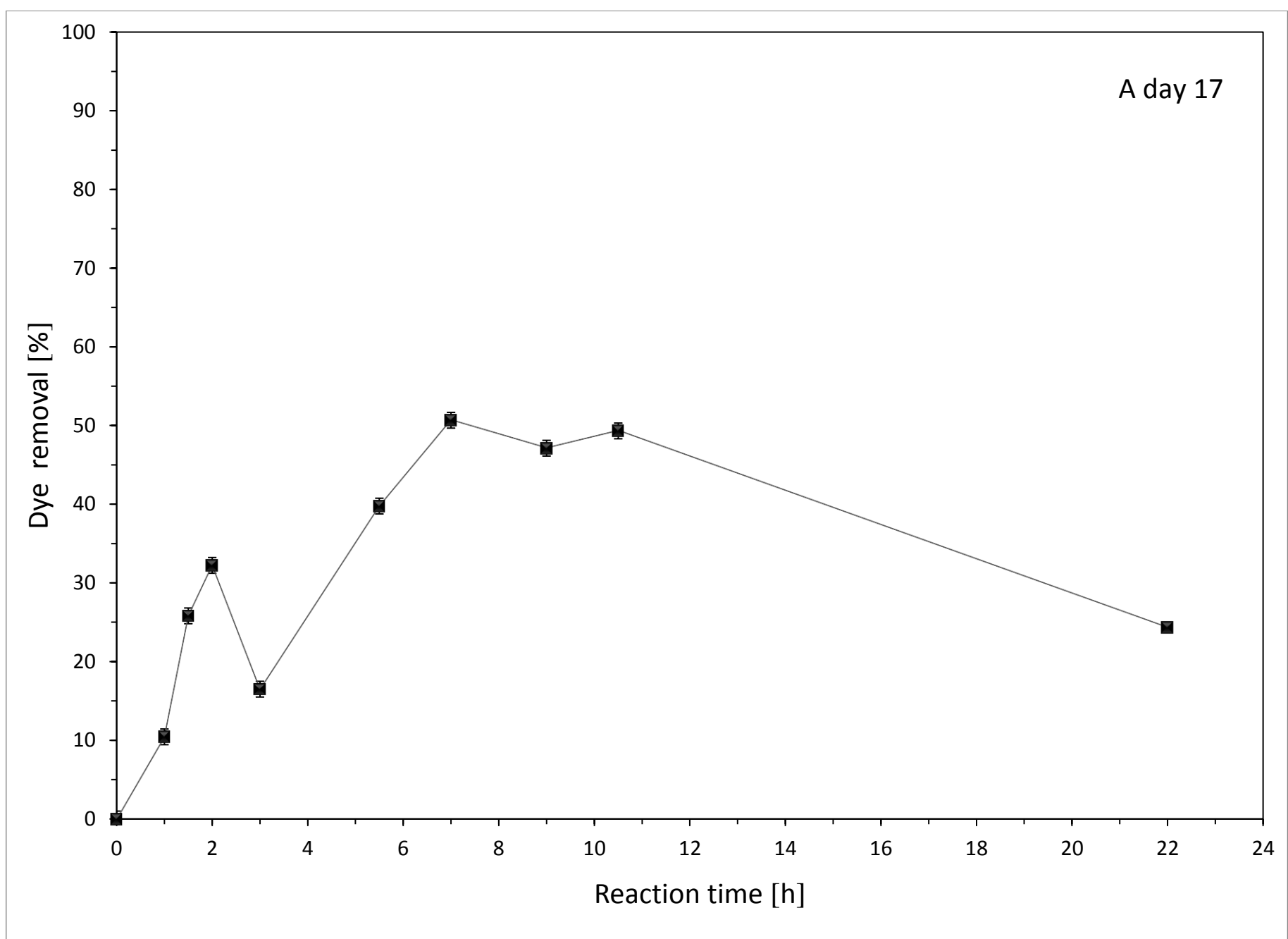
Figure



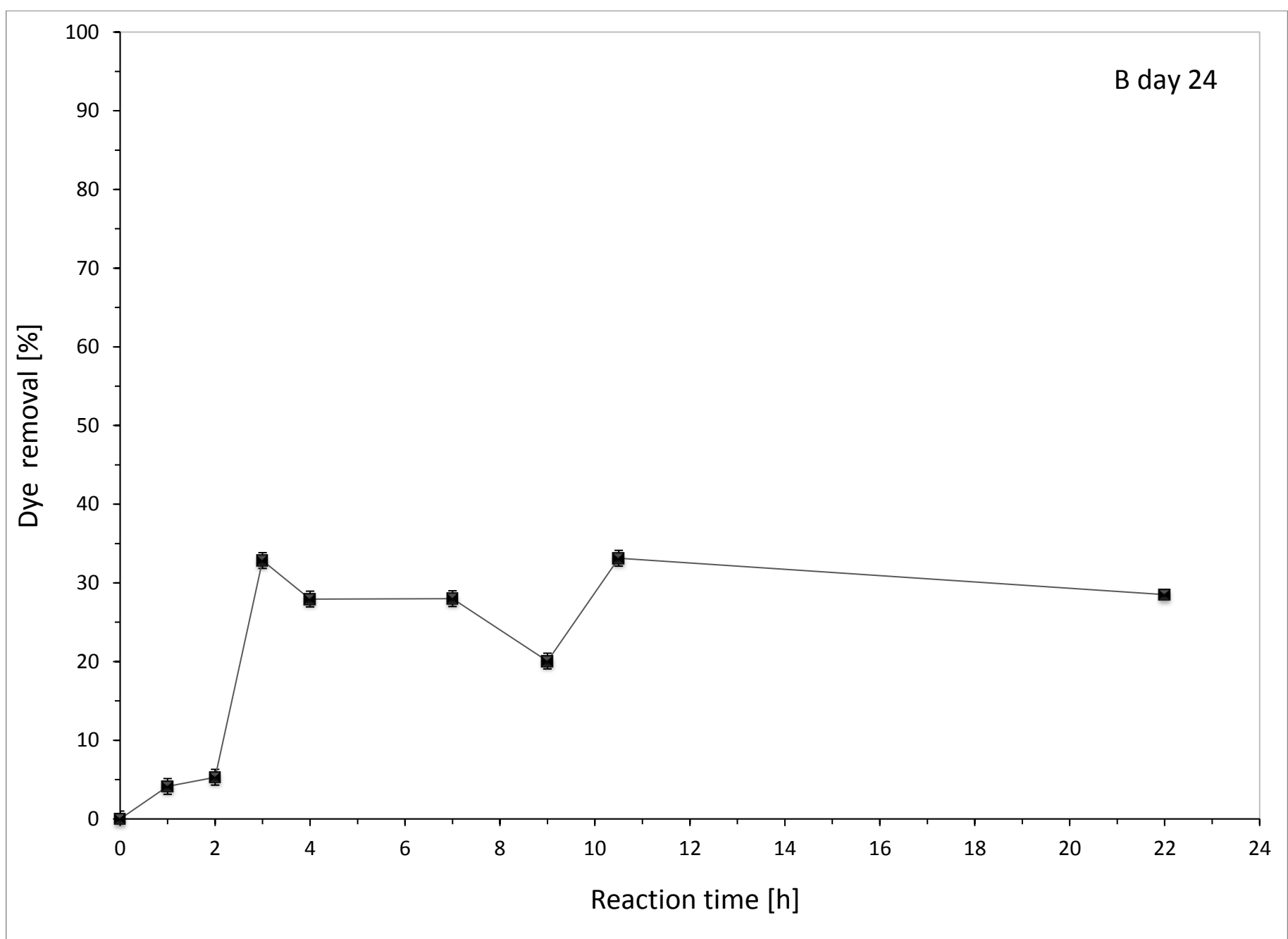
Figure



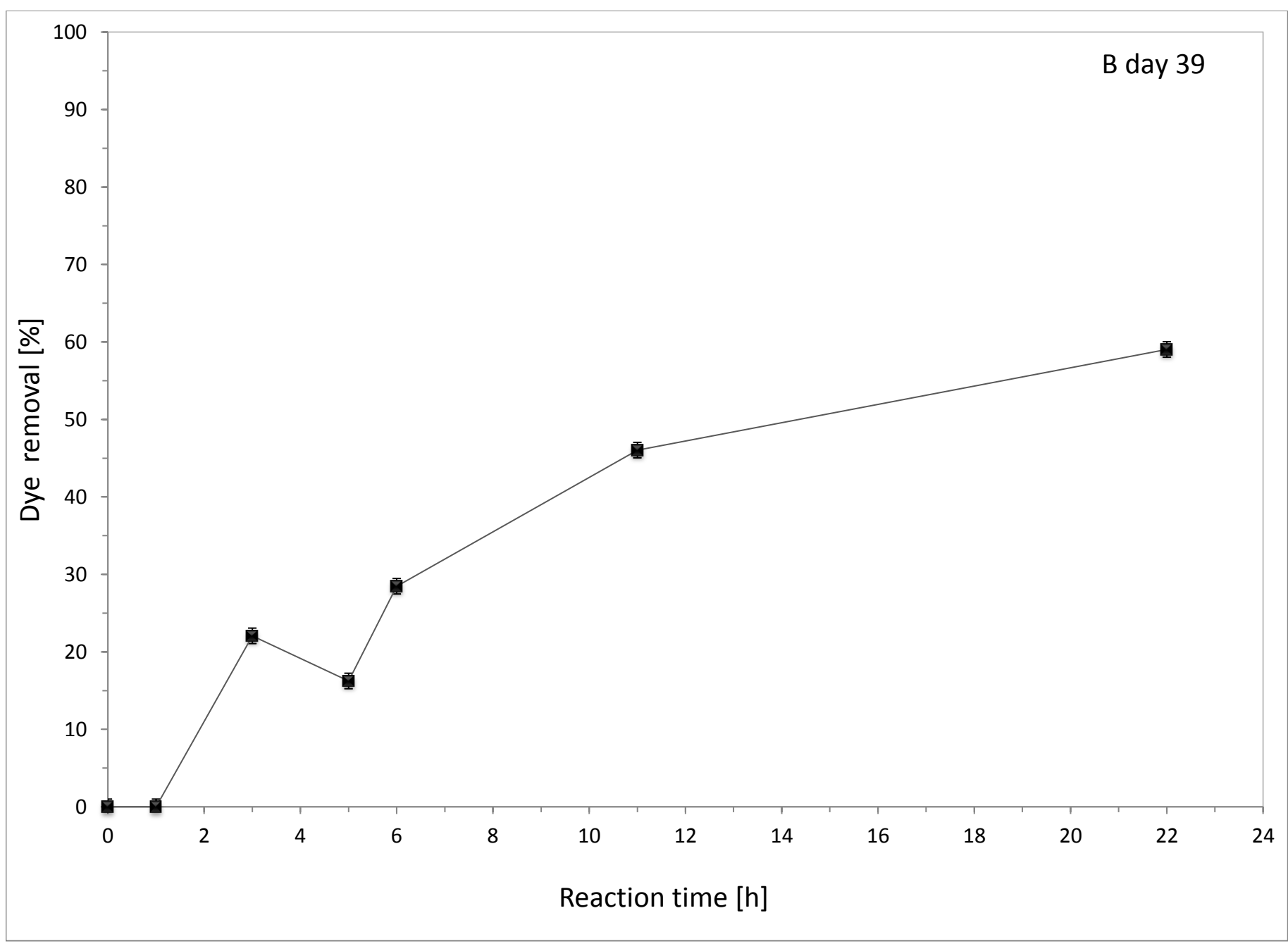
Figure



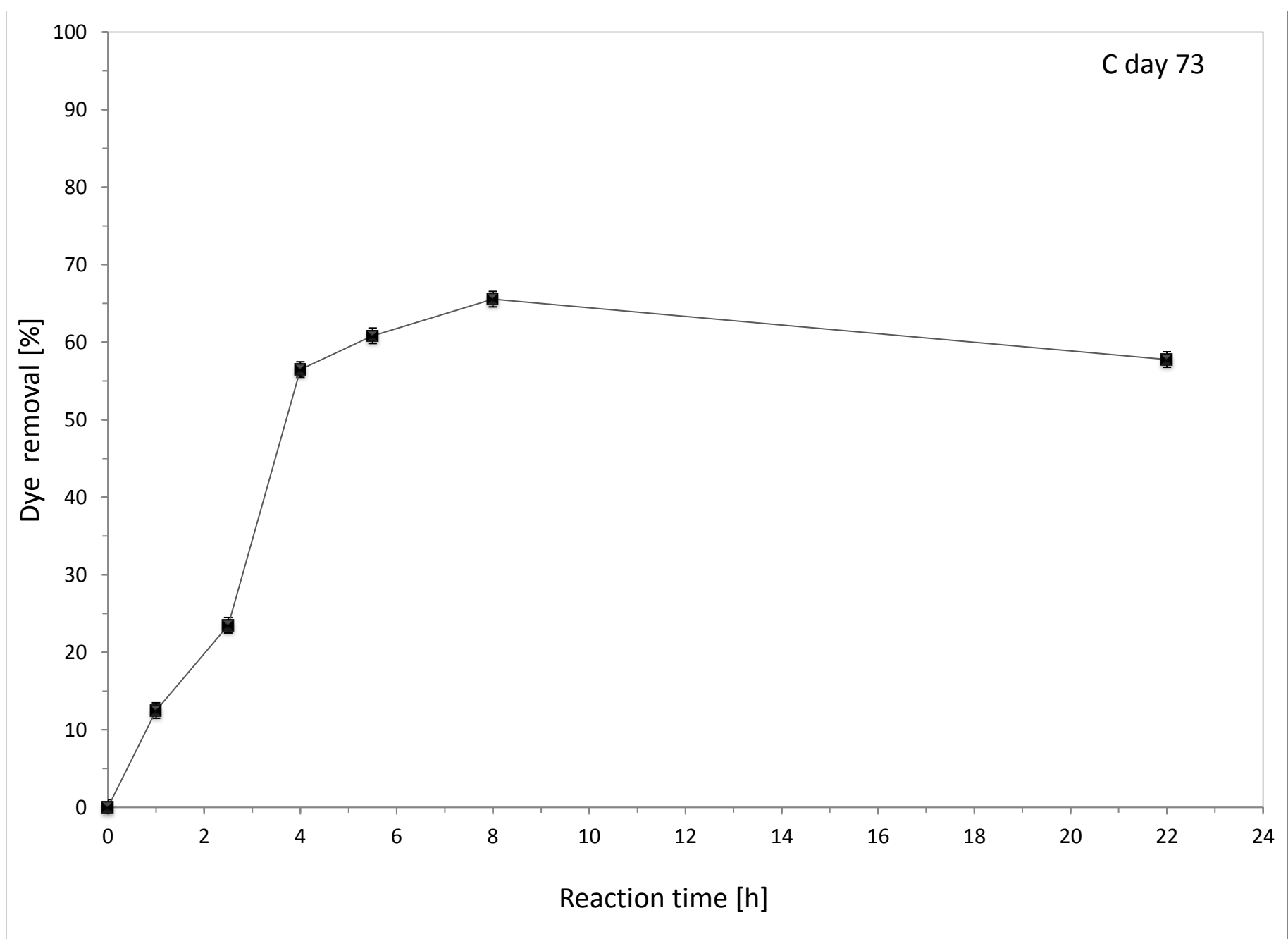
Figure



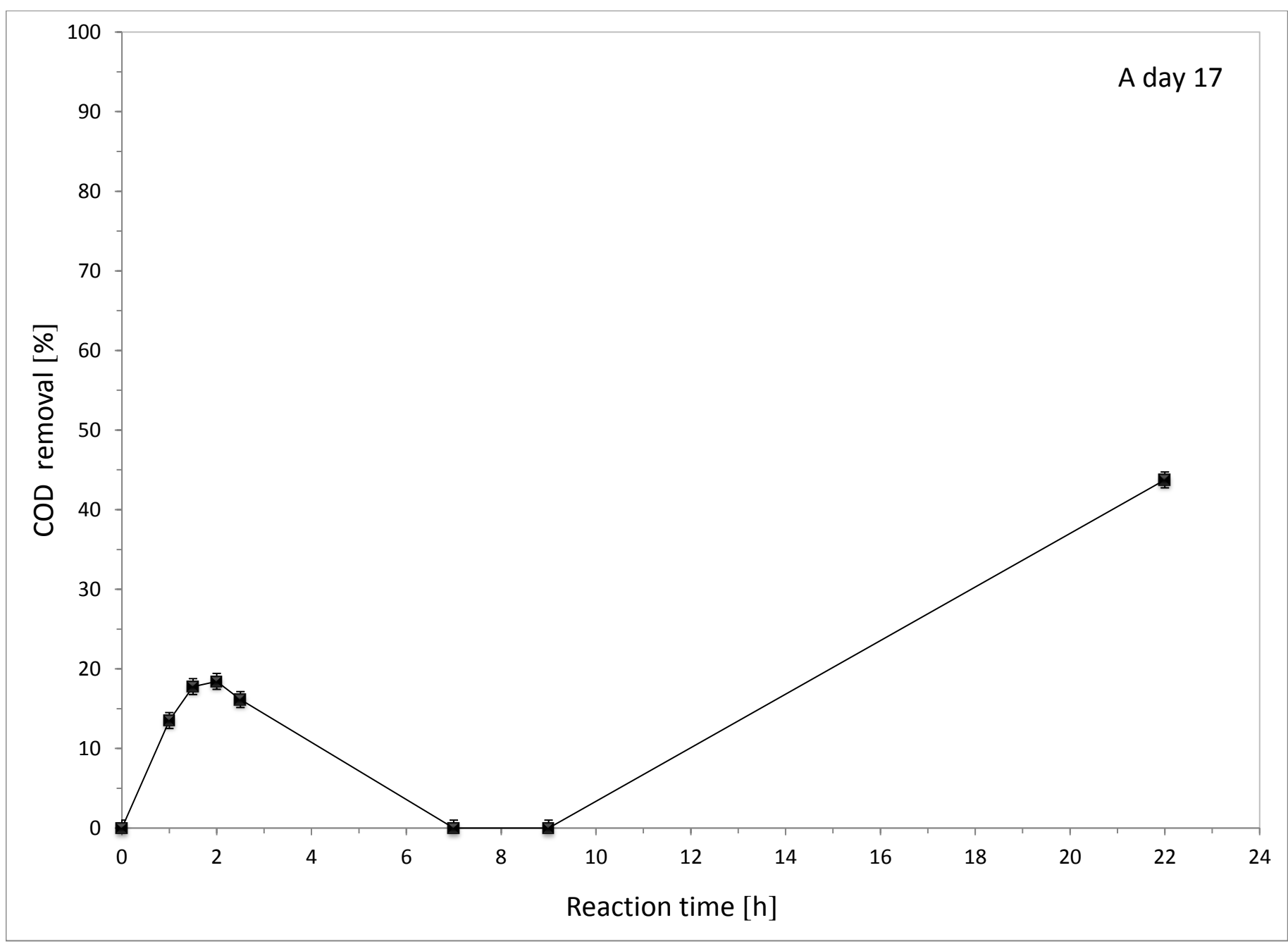
Figure



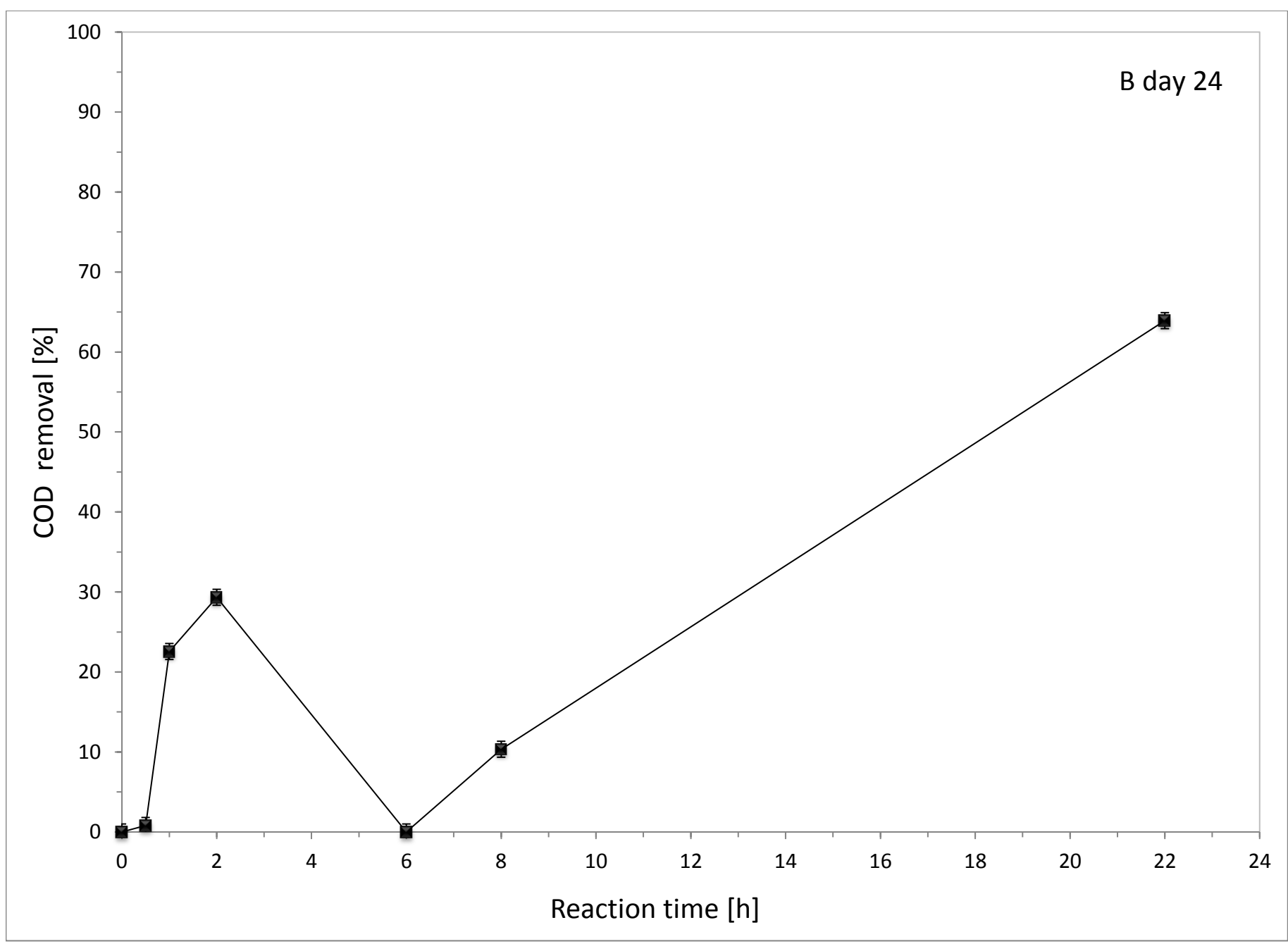
Figure



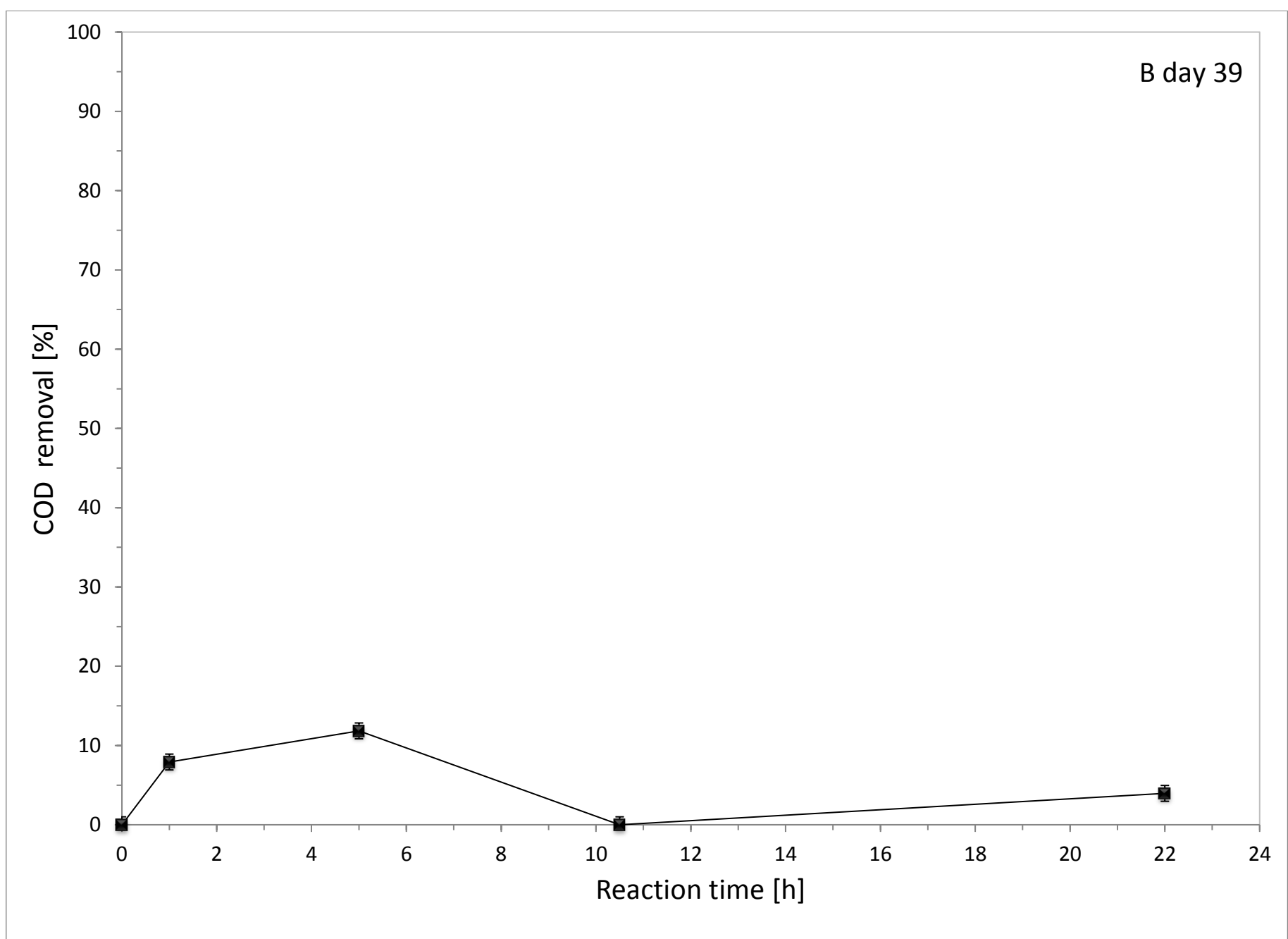
Figure



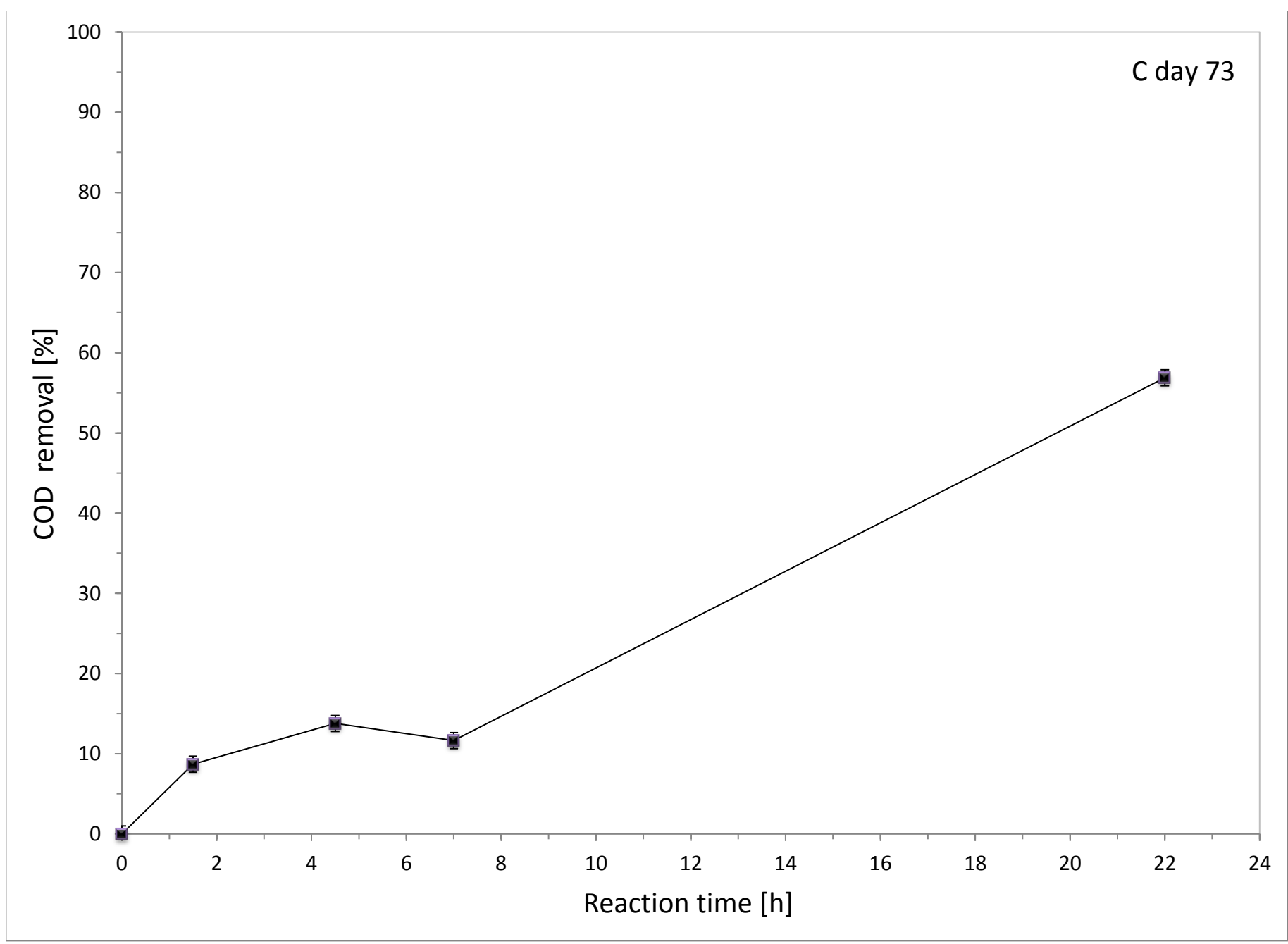
Figure



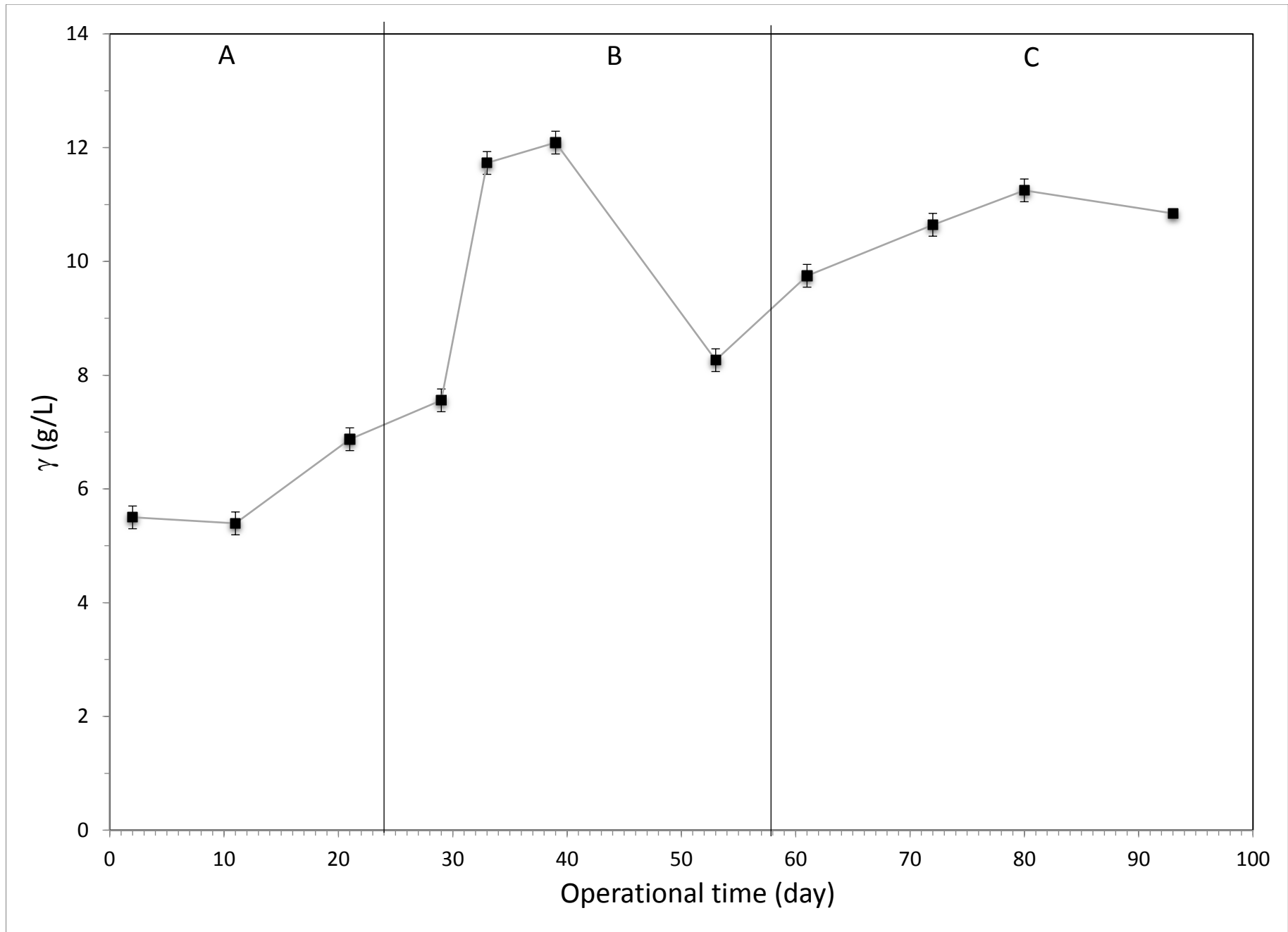
Figure



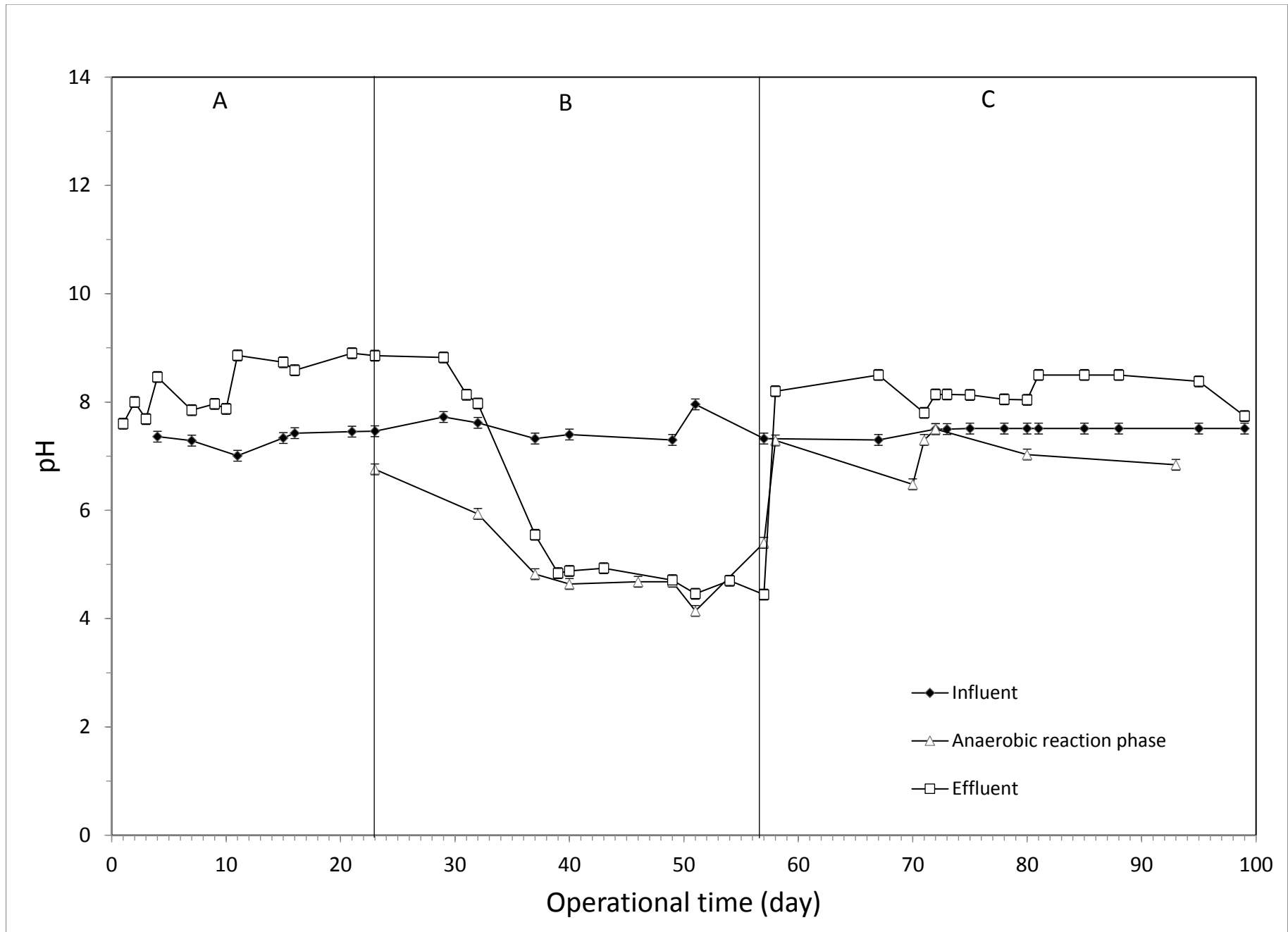
Figure



Figure



Figure



Figure

