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Mean Inner Potential of Liquid Water

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Improving our experimental and theoretical knowledge of electric potentials at liquid-solid boundaries is essential to achieve a deeper understanding of the driving forces behind interfacial processes. Electron holography has proved successful in probing solid-solid interfaces but requires knowledge of the materials’ mean inner potential (MIP, \( V_0 \)), which is a fundamental bulk material property. Combining off-axis electron holography with liquid phase transmission electron microscopy (LPTEM), we provide the first quantitative MIP determination of liquid water \( V_0 = +4.48 \pm 0.19 \) V. This value is larger than most theoretical predictions, and to explain the disagreement we assess the dominant factors needed in quantum simulations of liquid water. A precise MIP lays the foundations for nanoscale holographic potential measurements in liquids and provides a benchmark to improve quantum mechanical descriptions of aqueous systems and their interfaces in, e.g., electrochemistry, solvation processes, and spectroscopy.

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Introduction.—The mean inner potential (MIP, \( V_0 \)) is the volume-averaged electrostatic potential of a material with respect to a distant vacuum reference region at zero volts. The MIP is an intrinsic material property that depends on its elemental composition, structure, and electronic configuration [1]. The MIP is also known as the Bethe potential since Hans Bethe was the first to derive it [2], and provides an essential benchmark for both classical [3] and quantum mechanical (QM) molecular modeling [4], especially when studying electrodynamics of materials. For liquid water in particular, acquiring a reliable experimental MIP will benefit the accurate quantification of a variety of processes, e.g., for theoretical prediction and data interpretation of photoelectron spectroscopy (PES) [5,6], and will shed light on the nature of surface potentials, which is a source of debate in various scientific communities [7]. Using the MIP as a benchmark will also lead to improvements in models of aqueous systems where electrical potentials or fields at interfaces influence ion solvation processes [3], Stark vibrational spectroscopy [8], and electrocatalysis [9], where QM models preferably should be able to predict a reliable MIP before being used. Hence, rigorous MIP measurements and theoretical representations are essential to improve our understanding of water’s electric properties and thereby our capabilities to predict and rationally design processes involving aqueous systems.

The MIP can also be characterized as an electric potential jump [3] experienced by an electron when crossing the surface from vacuum to the bulk material. It can be influenced by the material’s surface termination, which in the past may have caused some discrepancies [10]. For a well-defined surface and bulk structure, the MIP is also experimentally and quantum mechanically well-defined.

In the simplest quantum approximation, called the independent atom model (IAM), isolated atoms in vacuum are superimposed at the relevant density to estimate an upper bound of the MIP (\( V_{0,MIP} \)) using Dirac-Fock electron scattering factors \( f_i^e(0) \) at zero angle for the \( i \) atoms in a unit cell of volume \( \Omega \) [11–13]

\[
V_{0,i}^{IAM} = \left( \frac{\hbar^2}{2\pi m_e e} \right) \sum_i f_i^e(0). \tag{1}
\]

For liquid water at 25°C, \( V_{0,water}^{IAM} = +4.87 \) V [3,14,15]. Current x-ray and electron diffraction measurements rely on the IAM to obtain atomic electron densities and potentials, respectively [4,15,16], however, a more accurate MIP would be a significant improvement.

In QM density functional theory (DFT), the MIP can be calculated by either averaging the total electric potential from an all-electron simulation, or by summing the contribution of the surface dipole layer and bulk quadrupole contributions [12,17] using maximally localized Wannier centers [18]. Table I summarizes published experimental and theoretical water MIP values, and also density corrected values [3,4,18–22], as the MIP is proportional to mass density via the unit cell volume \( \Omega \) [23,24]. The quantum MIP values for vitreous ice and water vary significantly around +3.72 V (as corrected via the liquid
density), the only previously reported experimental MIP value based on vitrified ice [21]. Some DFT models employ nuclear core corrections [4,19], where the true nuclear \(Z/r\) potential, from the nuclear delta function of charge \(Z\), is used in the actual MIP calculation, rather than the very narrow Gaussian nuclei used for numerical purposes. Thus, an accurate measurement of the MIP of liquid water provides an essential guiding benchmark for improving quantum descriptions of water.

Off-axis electron holography in a transmission electron microscope (TEM) is one of the most accurate methods to measure the MIP as well as mapping out the electric potentials through materials and their interfaces [25]. However, liquids have not been measured reliably, due to evaporation in the TEM vacuum, charging, and radiation damage [26,27], as well as a lack of accurate thickness measurements when using chip-based liquid phase TEM (LPTEM) [28]. To date, the only water MIP \(V^\text{ice}_0 = +3.5 \pm 1.2\) V [21] was measured on an amorphous vitrified ice sample at cryogenic temperature, by measuring the electron wave phase shift through polystyrene spheres with a known MIP relative to that of the surrounding vitrified ice. This value could include systematic errors from charging, thermal expansion [23,24] of both materials, and recrystallization. Recently, an estimated thickness of a single water droplet in LPTEM gave \(V^\text{water}_0 = +3.5 \pm 0.5\) V [22].

We present the first quantitative experimental liquid water MIP measurements by developing a novel LPTEM nanochannel liquid cell system [29] [Fig. 1(a), and Supplemental Material S1 [30]] and using electron holography to measure the phase change of the electron wave after passing through liquid water of known thickness (Supplemental Material S2 [30]). Assuming unbiased nonmagnetic materials, the electron wave phase shift (\(\Delta \phi\)), relative to a reference wave passing through vacuum, is proportional to liquid thickness \(t\) and MIP, \(V_0\) as [25]

\[
\Delta \phi = C_E V_0 t, \quad C_E = \frac{2\pi}{\lambda} \frac{E + E_0}{E(E + 2E_0)}.
\]

\(C_E\) is a constant that depends on the TEM beam energy \(E\). \(\lambda\) is the relativistic electron wavelength and \(E_0 = 511\) keV is the rest mass energy of the electron. At 300 kV, \(C_E = 6.53 \times 10^6 \text{ rad}/(\text{V} \cdot \text{m})\) [45]. The holographic interference image in Fig. 1(b) is of a nanochannel filled with liquid water and a stable radiolytic gas bubble. Holographic image reconstruction provides the phase image as in Fig. 1(c).

As shown in Figs. 1(c) and 1(e), Si\(_3\)N\(_4\) has larger MIP than water, therefore, the Si\(_3\)N\(_4\) sidewall gives a larger phase shift. The water MIP is measured from the phase profile plateau relative to outside the channel, Fig. 1(e). The holographic analysis, assumptions, and QM simulations are further described in Supplemental Material S3-S4 [30]. Electron beam current and charging influence were found

\[
\Delta \phi = C_E V_0^\text{water} t + \alpha.
\]
to be negligible, as detailed in Supplemental Material S5 and S6 [30].

Three nanochannel liquid cells were used with different liquid layer thicknesses \( t \), independently verified by chip cross-section images taken by scanning electron microscopy (Supplemental Material S7 [30]). The correlation between liquid thickness and phase change is shown in Fig. 1(f) with each data point representing different regions of a given chip and a linear fit as Eq. (1), \( \Delta \phi = C_E V_0^{\text{water}, \lambda} t + a \), including uncertainty from both phase and thickness. A fitted parameter \( a = 0.38 \pm 0.15 \) rad was included to account for minor local variations in charging and thickness, as well as possible deviations from linearity for very thin layers [46]. The \( a \) parameter is close to zero within the overall error bars, and for a thickness based MIP measurement \( V_0^{\text{water}, \lambda} = +4.48 \pm 0.19 \) V.

A second analysis was done, using the inelastic mean free path of electrons (\( \lambda_{\text{IMFP}} \)), based on the correlation between the phase and amplitude variation [47] in the meniscus with varying water layer thickness. Interference fringe formation [25] is based on coherent electrons and the holographic image analysis removes any inelastic electron scattering contribution. The reconstructed amplitude image is equivalent to a zero loss energy filtered image [47,48] related to \( \lambda_{\text{IMFP}} \) as

\[
\frac{t}{\lambda_{\text{IMFP}}} = -2 \times \ln \frac{A_{\text{obj}}}{A_{\text{ref}}}. \tag{3}
\]

Combined with Eq. (2), a MIP, \( V_{0}^{\lambda} \) can be found without knowing the sample thickness

\[
\frac{\Delta \phi}{-2C_E \ln \frac{A_{\text{obj}}}{A_{\text{ref}}}} = V_{0}^{\lambda} \lambda_{\text{IMFP}}, \tag{4}
\]

An example of the correlation between phase and amplitude in the meniscus is shown in Figs. 2(a)–2(c) and more in the Supplemental Material S8 [30], with a linear fit as expected from Eq. (4) of the phase shift, \( \Delta \phi \), versus \( -\ln(A_{\text{obj}}/A_{\text{ref}}) \), where amplitude in the meniscus \( A_{\text{obj}} \) is normalized relative to the bubble region without liquid \( A_{\text{ref}} \). This relies on knowing the holographic \( \lambda_{\text{IMFP}} \) with high precision which is expected to be shorter than that determined by electron energy loss spectrometry (EELS) TEM [49–52], as EELS cannot resolve the small losses that reduce coherence in holography. Here \( \lambda_{\text{IMFP}} \) is based on the part of the channel with known liquid layer thickness using Eq. (3) as detailed in the Supplemental Material S8 [30], giving a holographic \( \lambda_{\text{IMFP}} = 164 \pm 40 \) nm, as expected smaller than published EELS based water \( \lambda_{\text{IMFP}} \) [53].

The linear fit as in Fig. 2(c) is only done in the thickness varying meniscus, and passes, within uncertainty, close to the origin, where the slight offset could be due to minor membrane thickness variations, or residual liquid or humid layers and vapor in the assumed empty bubble region, which is another reason for including the \( a \) parameter in Fig. 1(f). The \( \lambda_{\text{IMFP}} \) based MIP is \( V_0^{\text{water}, \lambda} = +4.3 \pm 0.8 \) V from averaging the slopes, agreeing well with the thickness based MIP (Fig. 2). With larger uncertainty and implicit \( \lambda_{\text{IMFP}} \) reliance on thickness, the \( \lambda_{\text{IMFP}} \) based MIP is considered a self-consistency check confirming \( V_0^{\text{water}, \lambda} \).

As shown in Fig. 3, the liquid water MIP values are larger than both the earlier measurement of vitrified ice [21] and the droplet estimated value [22], but partly within their upper uncertainty range. The vitrified ice result increases slightly when density scaled from 0.94 to 1 g/cm\(^3\) [23], and could be influenced by the degree of vitrification [24].

We consider the QM simulations of flexible water molecules (QMF) by Remsing et al. [19]. The QMF give a bulk quadrupole contribution \( V_0^{\text{water}, \text{QMF, bulk}} = 3.84 \) V, 0.08 V higher than the rigid model [4], indicating a negligible effect from flexibility. The QMF surface dipole contribution is \( V_0^{\text{water, QMF, surf}} = +0.48 \) V, which when added to \( V_0^{\text{water, QMF, bulk}} \), gives \( V_0^{\text{water, QMF}} = +4.32 \pm 0.20 \) V as the best matching theoretical MIP from Table I, and is in excellent agreement with our measurements. The main difference between the QMF and the others listed in Table I is a combination of using the correct density, nuclear core correction that adds ca. +0.4 V, and the surface dipole addition. Importantly, the quantum and experimental MIPs are only about –0.5 V below the
IAM upper limit MIP. This means that the total electronic 
redistribution (making OH bonds, lone pairs, and hydrogen 
bonding) in going from the IAM model to liquid water 
molecules only amounts to a MIP decrease of $-0.5$ V. We 
note that our MIP value is also in excellent agreement with 
water’s diamagnetic susceptibility (Supplemental Material 
S9 [30]).

As noted, a reliable water MIP is important when 
comparing condensed bulk phase QM PES calculations 
with measurement, where the measured spectra are with 
respect to a vacuum at zero volts, similar to the reference 
potential for the holographic MIP. Using the MIP, all of the 
computed electronic energy levels must be corrected to the 
vacuum reference level [6]. In particular, the MIP of 
$+4.48$ V makes a large contribution to computed PES 
by shifting the electronic band offsets by more than 45% of 
the threshold ionization energy ($\sim 9.9$ eV). Currently QM 
values around $+3.7$ V are used [20] to predict photo-
electron ionization thresholds [5,6]. Beyond PES, solvation 
processes [3], and field-induced processes such as Stark 
vibrational spectroscopy [8] and electrocatalysis [9] would 
likely benefit from reliable MIPs to benchmark condensed 
phase electric potentials before calculating the fields under-
lying such processes.

In summary, we presented the first quantitative MIP 
measurement of liquid water at $V_0^{\text{water},t}$ = $+4.48 \pm 0.19$ V, 
by combining off-axis electron holography with a nano-
channel liquid phase TEM system, giving direct propor-
tionality between the transmitted electron wave phase shift 
and liquid thickness. A self-consistency check was made 
with a method based on the inelastic mean free path of 
electrons. Both methods’ MIPs are in excellent agreement 
and within error bars of the best available condensed phase 
quantum mechanical simulation including the correct liquid 
density, surface dipole potential, and nuclear core correc-
tions. The values are below the IAM limit and within the 
upper uncertainty of vitrified ice measurements [21]. These 
measurements provide critical benchmarks for quantum 
mechanical simulations of electric potentials inside water 
and at its interfaces to matter. This novel MIP technique is 
fundamental to future electron holographic microscopy of 
electric potentials in nanoscale liquid processes.

All data needed to draw these conclusions are presented 
in the Letter and Supplemental Material [30].

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