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Hydrochar-Facilitated Anaerobic Digestion: Evidence for Direct Interspecies Electron Transfer Mediated through Surface Oxygen-Containing Functional Groups

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ABSTRACT: Acceleration of the anaerobic digestion (AD) process is crucial to achieving energy-efficient recycling of organic wastes. Hydrochar is produced by hydrothermal liquefaction of biomass, yet its application in the AD process is rarely reported. The present study showed that sewage sludge-derived hydrochar (SH) enhanced the methane production rate of glucose by 37%. SH increased the methane production rate from acetate but did not affect acidification and the methane production rate from H_2/CO_2 . SH enhanced hydrogenotrophic methanogenesis, which could be due to direct interspecies electron transfer (DIET) by converting H^+ , e^- , and CO_2 to methane. *Trichococcus* and *Methanosaeta* were dominant in the AD process with SH. Label-free proteomic analysis showed *Methanosaeta* was involved in DIET as reflected by the up-regulation of proteins involved in hydrogenotrophic methanogenesis. Hydrochars derived from corn straw (CH), *Enteromorpha* algae (EH), and poplar wood (PH), as well as activated carbon (AC), were also tested in the AD process. SH, CH, and EH obviously increased the methane production rates, which were 39%, 15%, and 20% higher than the control experiment, respectively. It was neither electrical conductivity nor the total redox property of hydrochars and AC but the abundances of surface oxygen-containing functional groups that correlated to the methane production rates.

INTRODUCTION

Anaerobic digestion (AD) is an efficient process for mitigation of a broad variety of waste biomass and production of methane at the same time.¹ However, the unstable performance of the AD process due to the fluctuation of characteristics and flow rates of feedstocks is a major issue, which could lead to volatile fatty acid (VFA) accumulation.² The accumulation of large amounts of VFA would result in the failure of the AD process.³ Efficient conversion of VFA (e.g., propionate, butyrate, and ethanol) to methane requires syntrophic interaction of VFA oxidizing bacteria and hydrogenotrophic methanogens, which could assist in maintaining a low H_2 concentration—an interrelation commonly described as interspecies hydrogen transfer (IHT).⁴

Considerable efforts have been made to accelerate the methane production rates.^{5,6} It has been shown that conductive materials such as activated carbon (AC) and magnetite could improve the methane production rate of VFA and thereby increase the AD efficiency.⁷ Most previous studies suggested that the conductive materials could facilitate direct interspecies electron transfer (DIET) in the AD process. DIET is a syntrophic metabolism, and the electrons directly flow from exoelectrogenic bacteria to electrotrophic methanogen

through cell components (cytochrome C and pilin) or conductive materials without the formation of H_2 , which is a more effective method for VFA degradation compared to IHT.⁸ The genus *Geobacter* was the only genus that was demonstrated to achieve DIET until 2018, and only one bacterium (*Syntrophus aciditrophicus*) besides the genus *Geobacter* was recently demonstrated to participate in DIET.⁹ However, the enrichment of the above bacteria was not found in most previous studies with the addition of conductive materials in the AD process.^{4,8,10} The mechanism is still not clear, since AD is a very complex process and involves many different microbes. In addition, researchers are also trying to explore low-cost conductive materials to accelerate the methane production rate. For instance, biochar, which is

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Figure 1. Effects of SH on methane production at different F/M ratios: (a) cumulative methane yield at F/M of 0.5, 1, and 5; (b) pH variation at different F/M ratios; VFA concentrations variation at (c) F/M = 5, (d) F/M = 1, (e) F/M = 0.5.

produced by pyrolysis of biomass, has been shown to raise the methane production rate.^{8,11}

Hydrothermal liquefaction (HTL) of biomass (sewage sludge, corn straw, etc.) is a thermochemical process that converts biomass into liquid and solid fuels by processing in hot and pressurized water environment (200–350 °C and 4–22 MPa) for certain time to degrade the solid biopolymeric structures, and this process is more suitable for the treatment of wet biomass compared to pyrolysis.^{12,13} Hydrochar is the solid product from the HTL process, and it has better performances compared to the raw biomass in terms of higher

mass and energy density, better dewaterability, and improved combustion performance as a solid fuel.¹⁴ In addition, hydrochar can also be applied as soil amendments.¹⁵ Compared to biochar, the application of hydrochar for the enhancement of methane production has been sparsely reported.⁷ A recent study showed that the addition of hydrochar to the AD process treating dead pig carcasses increased the methane yield and also methane production rate, which was mainly attributed to the decrease of free ammonia concentration due to the adsorption of ammonia by hydrochar and the slightly decreased pH.¹⁶ However, a detailed

mechanism is still unknown, especially from the microbial aspects. Although there is still no literature reporting the conductivity of hydrochar, it is known that hydrochar has redox-active properties.¹⁷ Previous studies showed that biochar could facilitate DIET for microbial reduction of ferrihydrite and methanogenesis due to its redox-active properties.^{18,19} Therefore, it was hypothesized that hydrochar could promote methanogenesis through DIET. Previous studies also showed the characteristics of hydrochar (e.g., redox properties) varied with the types of feedstocks, and it might have different effects on AD, which remains to be elucidated.^{15,20} In addition, the commonly used high-throughput sequencing of 16S rRNA genes analysis could only provide information on microbial community compositions but could not provide insights into the reactions taking place in the AD process. It is crucial to reveal the changes of metabolic pathways with the addition of hydrochar to better understand the real functions of hydrochar in the AD process. The proteomic analysis could reveal the changes of enzymes involved in different reactions and thereby provide important metabolic information on the microbial communities.^{4,21} The combination of 16S rRNA genes and proteomic analysis could reveal the microbial mechanism of the AD process with the addition of hydrochar.

On the basis of the above considerations, the present study aimed to investigate the possibility of using hydrochar derived from sewage sludge to enhance the methane production rate from the model compound glucose, and the microbial mechanism was investigated by stable carbon isotope analysis, high-throughput sequencing of 16S rRNA genes analysis, and label-free proteomic analysis. In addition, the commonly used activated carbon (AC) and hydrochars derived from different feedstocks were used as a comparison to identify the main characteristics of hydrochar contributing to the promotion of methane production rate.

MATERIALS AND METHODS

Preparation of Hydrochar. Hydrochar was obtained by HTL of dewatered sewage sludge (SH), which was obtained from a wastewater treatment plant in Shanghai, China. HTL was conducted in a 3 L hydrothermal reactor at 300 °C for 1 h, which was a common operation set up for bio-oil production.¹² The detailed information about the operation of the HTL process was described in our previous study.¹³ The bio-oil attached on hydrochar was removed by washing with tetrahydrofuran (THF). Hydrochar was soaked in THF (10 mL THF/g hydrochar), and the mixture was shaken for 1 h in a shaker at ambient temperature. THF was then removed by centrifugation. This process was repeated several times until THF after centrifugation became transparent, which indicated most bio-oil was removed. The hydrochar was then dried at 60 °C for 24 h for further utilization. In addition, corn straw, *Enteromorpha* algae, and poplar wood were used to prepare three additional hydrochars (CH, EH, and PH, respectively) in order to make a comparison with SH. Activated carbon (AC) purchased from a commercial source was also used as a comparison. All the hydrochars and AC were ground and passed through a 2 mm sieve before usage. The characteristics of hydrochars and AC are summarized in Table S1.

Substrate and Inoculum. The inoculum was obtained from a mesophilic upflow anaerobic sludge blanket (UASB) reactor treating cassava stillage in an ethanol plant. The inoculum was put into a grinder (QX-650, Qinxin, China) and milled for 5 min to make it homogeneous before usage. The

characteristics of the inoculum were as follows: total solids (TS) 37.2 ± 2.8 g/L, volatile solids (VS) 30.1 ± 1.9 g/L, and pH 7.21 ± 0.08 . Glucose was used as a substrate.

Effects of SH on the Methane Production. The effects of SH on the methane production were investigated in batch experiments. The experiments were conducted in 118 mL serum bottles with a working volume of 60 mL. Each bottle contained glucose (4 g/L), sodium bicarbonate (2 g/L), and BA medium. The composition of BA medium was described in a previous study.²² The bottles were inoculated with different amounts of inoculum to achieve glucose to inoculum mass ratios (F/M) of 0.5, 1, and 5, which resulted in different initial organic loading rates. SH was added to the bottles to a final concentration 10 g/L, and the bottles without SH were used as a control. The bottles with only inoculum were used as a blank. In addition, the bottles without glucose but with SH were included to determine potential methane production from SH itself. The initial pH of each bottle was adjusted to 7.5. Bottles were flushed with pure N₂ for 5 min, sealed with butyl rubber stoppers, and then incubated at 37 °C. The gas and liquid samples were collected every 2 days initially for analysis (Figure 1). The liquid samples were taken from the bottles with a syringe without opening the bottles. In order to understand the effects of hydrochar concentrations on methane production, additional experiments were conducted with hydrochar concentrations of 1 g/L, 2.5 g/L, 5 g/L, 10 g/L, 15 g/L, and 20 g/L with F/M 5. All the tests were prepared in triplicate.

On the basis of the results from the above experiments, the bottles with F/M 5 and 10 g/L hydrochar were further used for the second batch cultivation. The previous experiments also showed that there was a significant change of pH (between 6.6 and 7.5) during the batch experiments for methane production, and the bottles with SH generally had higher pH compared to the control. Therefore, the second batch experiments were conducted in two groups with uncontrolled pH and controlled pH at 7.5 by adding 2 M NaOH periodically. For the monitoring of pH in the bottles, liquid samples were taken from the bottles with a syringe without opening the bottles and then measured by pH meter daily. For the controlling of pH, the bottles were opened, and then 2 M NaOH was added to achieve the set value. After that, the bottles were purged with N₂ gas for further incubation. The aim was to elucidate whether SH enhanced methane production by affecting pH. The solid fraction from each bottle of the first experiments at F/M 5 with and without SH was separated by centrifugation and supplemented with fresh medium to a volume of 60 mL, as previously mentioned. The mixture was then distributed equally to two 118 mL bottles, i.e., a 30 mL mixture in each bottle. One of the bottles was used in pH-controlled experiments, and the other one was used in pH uncontrolled experiments. A modified Gompertz model was used to obtain the parameters including the maximum methane potential (P), the maximum methane production rate (R_m), and lag phase (λ) under different conditions (Supporting Information).⁴

The effects of SH on the acidification of glucose were also determined. A 5 mL mixture of each bottle from the second experiment without pH control was added as inoculum in 118 mL bottles. Subsequently, 60 mL of BA medium and SH (10 g/L) was added. Finally, the bottles were amended with glucose (4 g/L) as a substrate, and bromoethanesulfonic acid (BES; 50 mM) was added as a specific methanogen inhibitor.

The concentrations of VFA were then measured and analyzed periodically. The effects of SH on the methanogenesis were also determined. A similar procedure to the last experiments was adopted. However, BES was not added, and either acetate (30 mM) or H_2/CO_2 (80:20, 1 atm) was used as a substrate instead of glucose. All the tests were conducted in triplicate.

Stable carbon isotope analysis was conducted to investigate the methanogenic pathway with and without hydrochar in the second batch cultivation without a pH control, and the biogas was collected periodically for stable carbon isotope analysis.²³ Carbon isotope fractionation (α_c) was calculated to characterize the changes of the methanogenesis pathway. Detailed information is described in the [Supporting Information](#).

Effects of SH on the Microbial and Proteomic Patterns. The enriched mixed cultures in the second batch cultivation with F/M 5 and without pH control as described previously were collected from the bottles. Duplicate samples were retrieved randomly from the triplicate bottles when the methane production was close to half of the maximum value (i.e., around 150 mL of CH_4/g of COD), which corresponded to vigorous microbial metabolisms. C1 and C2 were obtained from the control bottles (without hydrochar), and H1 and H2 were obtained from the bottles with hydrochar. The taxonomic classification of all the sequences was first analyzed based on the high-throughput sequencing of 16S rRNA genes. DNA was extracted from each sample with the QIAamp DNA Stool Mini Kit (51504, QIAGEN) followed by polymerase chain reaction (PCR) amplification with primers 515F/806R. PCR products were then purified, quantified, and sequenced by Illumina Miseq platform. The sequences were submitted to NCBI with the accession number SUB5724470. More detailed information relating to PCR procedure and bioinformatics analysis is described in the [Supporting Information](#).

The above microbial samples were also used for proteomic analysis. Total proteins were extracted by the phenol extraction method, and the extracted protein was enzymatically hydrolyzed by trypsin. The peptides were quantified and then analyzed by liquid chromatography coupled with tandem mass spectrometry (LC-MS/MS; Easy-nLC II nano liquid chromatograph, Thermo, USA; Q-Exactive mass spectrometer, Thermo, USA). Detailed information relating to protein preparation, mass spectrometry analysis, and data analysis is described in the [Supporting Information](#).

Total bacteria, archaea, *Methanobacteriales* (MBT), *Methanomicrobiales* (MMB), *Methanosarcina* (MSC), and *Methanosaeta* (MST) in the samples were also quantified by qPCR with the corresponding primers ([Supporting Information](#)).

Effects of Hydrochars Derived from Different Feedstocks and AC on Methane Production. The effects of hydrochars derived from different feedstocks and AC on methane production were also investigated, and the same procedure was adopted for the methane production experiments as previously described for methane production with SH. The bottles without glucose but with different hydrochars and AC were also set up to determine the possibility of methane production from hydrochars and AC.

Analytical Methods. The analysis of methane content, VFA concentrations, and characteristics of hydrochars and AC is described in the [Supporting Information](#). The analysis of variance (ANOVA) was used to test the significance of results, and $p < 0.05$ was considered to be statistically significant.

RESULTS AND DISCUSSION

Effects of SH on Methane Production. The methane production at different F/M ratios is shown in [Figure 1a](#), and the kinetic parameters of methane production are summarized in [Table S2](#). It should be noted that there was no methane production from SH itself. It was obvious that there were no significant differences in the methane production at F/M 0.5 with and without SH ($p > 0.05$), while the reduced length of the lag phase and the increased methane production rate were observed at F/M 1 and 5 with SH. For instance, SH resulted in a significant increase of R_m by about 33% ($p < 0.05$) and 35% ($p < 0.05$) at F/M 1 and 5 compared to control without SH, respectively. The pH varied between 6.5 and 7.5 during the whole periods ([Figure 1b](#)). The produced VFAs were mainly acetate and butyrate. The higher methane production rates with the presence of SH at F/M 1 and 5 were also correlated to the fast consumption of produced VFA as shown in [Figure 1c](#) and [d](#). For F/M 0.5, the presence of abundant bacteria and archaea in the bottles might result in the fast consumption of glucose and its intermediates, and therefore no differences were observed with and without SH. The above results clearly showed that SH enhanced the methane production rate at high F/M. Further experiments also showed that 10 g/L SH was optimal for the enhancement of the methane production rate ([Figure S1](#) and [Table S3](#)). The hydrochar concentration (10 g/L) used in the present study was in the range of previously reported concentrations (0.5–50 g/L) of different materials (e.g., magnetite, GAC, biochar, carbon nanotube, etc.) to enhance the AD process.^{7,8}

A second batch cultivation was further conducted at F/M 5 without pH control ([Figure S2](#) and [Table S2](#)). The lag phase was decreased, and R_m was increased with and without SH compared to those of the first batch experiment, indicating the acclimatization of the mixed culture. R_m in the presence of SH was significantly increased by 37% ($p < 0.05$) in the second batch, which showed the stable promotion effects of SH on the methane production rate. The higher consumption of produced VFA and faster recovery of pH were also observed in the presence of SH. It should be noted that pH at F/M 5 was significantly decreased to 6.6 due to the production of VFA on day 8. It was known that acidic pH had inhibition on methanogens.²⁴ Therefore, an experiment with a constant pH at 7.5 was also conducted in order to see whether SH could still promote methane production at the optimal pH for methane at F/M 5. The results showed that with controlled pH, the lag phases were decreased, and the R_m in the presence of SH was significantly increased by 27% ($p < 0.05$; [Table S2](#) and [Figure S3](#)). The shorter lag phase than in the uncontrolled pH experiment could be due to the fast adaption of methanogens at the optimal pH. For the reactors operated at optimal pH but with high F/M, SH still could promote the methane production rate. However, the increase of R_m (27%) by SH in a pH controlled experiment was lower than that in a pH uncontrolled experiment (37%), which indicated that SH could more efficiently promote methane production at inhibited conditions (e.g., low pH). A previous study showed hydrochar could alleviate ammonia inhibition,¹⁶ which was not the case in the present study where no ammonia inhibition was present. Therefore, further studies were conducted to reveal the role of SH in the AD process.

Effects of SH on the Individual Steps in AD Process. In order to understand how SH affected the AD process, the

Figure 2. Genus level classification of the samples. (a) Archaeal genera and (b) bacterial genera based on 16S rRNA genes analysis; (c) archaeal genera and (d) bacterial genera based on proteomic analysis. Genera making up less than 1% of total composition were classified as “others.”

acidification of glucose with and without SH was compared. As shown in Figure S4a, the addition of SH did not show obvious effects on the concentration and composition of VFA ($p < 0.05$). Therefore, it could be concluded that SH did not affect the acidification step. The methanation of both H_2/CO_2 and acetate was also studied by the enriched cultures. Figure S4b and c show that the methane production rate of acetate by the enriched culture with SH was higher than that by the enriched culture in the control reactor. However, SH did not have an obvious influence on the methane production rate of H_2/CO_2 . Therefore, SH selectively promoted methanation of acetate, while it did not affect methanation of H_2/CO_2 . The methanogenic pathway was further evaluated by stable carbon isotopic analysis. An increase in α_c means a shift to hydrogenotrophic methanogenesis.²⁵ As shown in Figure S5, SH resulted in a significantly higher α_c for each sampling date compared to the control reactor, which suggested SH enhanced hydrogenotrophic methanogenesis. However, Figure S4 did not show an increased methane production rate of H_2/CO_2 by SH. It should be noted that Figure S4 was obtained by using molecular H_2 as a substrate, which indicated the traditional IHT process was not promoted by SH. It was possible that DIET was achieved with the presence of SH, and therefore more H^+ , e^- , and CO_2 were directly utilized for methane production through hydrogenotrophic methanogenesis. A previous study also showed the methanogen *Methanosaeta* (formerly called *Methanothrix*) involved in DIET could only utilize H^+ and e^- but not molecular H_2 for hydrogenotrophic methanogenesis.^{26,27} Therefore, it could be concluded that SH facilitated DIET in the AD process, and further microbial and proteomic analysis also verified this point. It should also be mentioned that the enhancement of the

methane production rate of acetate by SH as shown in Figure S4b was also possible by DIET, since a previous study showed the methane production rate of acetate was higher due to DIET with the addition of magnetite.²⁸ In this case, acetate would not be consumed directly by aceticlastic methanogens, but through syntrophic growth of bacteria and methanogen via DIET. Overall, the above results showed the enhanced methane production rate of glucose by SH was correlated to the promotion of methanogenesis but not acidification.

Effects of SH on the Microbial Community Compositions. The phylum level identification based on high-throughput sequencing of 16S rRNA genes analysis showed that all the samples were dominated by *Firmicutes*, *Euryarchaeota*, *Bacteroidetes*, *Chloroflexi*, *Aminicenantes*, *Proteobacteria*, and *Synergistetes* (Figure S6). SH caused a significant increase in the relative abundance of *Firmicutes* and resulted in obvious decrease in the relative abundances of *Synergistetes*, *Bacteroidetes*, and *Euryarchaeota*. The phylum *Firmicutes* contains acidogenic and acetogenic bacteria capable of converting a wide range of organics to VFA and subsequently to acetate, H_2 , and CO_2 , and the enrichment could be contributed to the higher methane production rate with SH since these bacteria might achieve syntrophic growth with methanogens, which could efficiently convert VFA to acetate and facilitate methane production.²⁹ Although SH increased the methane production rate, strangely it led to the decrease in the relative abundance of *Euryarchaeota*, which contains methanogens.³⁰ It should be remarked that lower relative abundance does not mean low absolute concentration of methanogens, which is discussed in the following part by qPCR analysis.

Identification of both bacteria and archaea was also conducted at the genus level (Figure 2a and b). Four archaeal genera, *Methanobacterium*, *Methanolinea*, *Methanosaeta*, and *Methanosarcina*, were detected. The hydrogenotrophic methanogens *Methanobacterium* and *Methanolinea* accounted for less than 10% of the archaea,³¹ and a slight increase of the relative abundance of *Methanolinea* was observed when SH was present. *Methanosaeta*, the only known strict acetoclastic methanogen,³² increased significantly with the addition of SH (from 24% to 50%). By contrast, *Methanosarcina*, which could use both H_2/CO_2 and acetate for methane production,³³ was greatly reduced from 72–75% to 37–48%. Recent studies showed that the conductive materials could enrich *Methanosaeta*, which could accept e^- via DIET for the reduction of CO_2 to methane,^{26,27} and the results also indicated that SH might stimulate DIET. However, the known bacteria *Geobacter* spp. and *Syntrophus aciditrophicus*, which could perform DIET,⁴ were not found in the present study. The bacterial genus *Trichococcus*, belonging to *Firmicutes*, was obviously enriched with SH. *Trichococcus* could produce lactate, acetate, and ethanol from glucose,³⁴ but the ability to achieve extracellular electron transport has not been reported. It seems that the exact role of *Trichococcus* requires further investigation. The enrichment of *Trichococcus* and *Methanosaeta* was further verified by label-free proteomic analysis (Figure 2c and d). The proteomic analysis also indicated that the genus *Clostridium* played important roles in AD with and without SH, considering its high relative abundances (50–60%) in both cases. The changes of subordinate microbial genera were discussed in the Supporting Information by both 16S rRNA genes and proteomic analysis. Overall, the results showed that SH changed the microbial community structures of AD, which could be related with the higher methane production rate. However, the intrinsic mechanism for enhanced methane production by SH deserves further elucidation in view of the changes in metabolic pathways.

Effects of SH on Metabolic Pathways As Revealed by Proteomic Analysis. In total, 1711 proteins were identified from all the samples, and 262 of the 1711 proteins were identified as differentially expressed proteins with 101 up-regulated and 161 down-regulated proteins by comparing the samples with SH to those without SH (Figure 3). Table S4 shows all the differentially expressed proteins. GO enrichment analysis showed that SH significantly enriched proteins that were involved in cellular component organization or biogenesis

and metabolic process belonging to biological function, cell part and membrane belonging to the cellular component, and binding and structural molecular activity belonging to molecular function (Figure S7a). KEGG pathway enrichment analysis further showed that SH enriched proteins that were involved in various pathways, such as methane production, glycolysis/gluconeogenesis, and carbon metabolism, which were directly related with the efficiency of AD (Figure S7b). The above results provided a general overview of the effects of SH on the metabolic pathways. Detailed analysis based on the KEGG pathway was subsequently conducted to understand the process better.

As shown in Tables S4 and S5, 60 of the 101 up-regulated proteins were originated from methanogens, which included enzymes involved in both hydrogenotrophic and acetoclastic methanogenesis (e.g., Acetyl-CoA synthetase and methylenetetrahydromethanopterin dehydrogenase),³⁵ most of which (57) were originated from *Methanosaeta*. It was consistent with the enrichment of *Methanosaeta* by microbial community analysis. *Methanosaeta* is strict acetoclastic methanogen. However, the up-regulated protein responsible for hydrogenotrophic methanogenesis (methylenetetrahydromethanopterin dehydrogenase) was found. Although *Methanosaeta* could not use molecular H_2 , it could accept e^- produced by DIET and H^+ to produce methane through hydrogenotrophic methanogenesis.^{26,27} The proteomic analysis demonstrated the presence of DIET with SH in a mixed culture system. Although some up-regulated proteins responsible for hydrogenotrophic methanogenesis were detected from *Methanosarcina* (Table S4), the proteomic analysis could not verify whether *Methanosarcina* was involved in DIET since *Methanosarcina* could achieve hydrogenotrophic methanogenesis through both IHT and DIET.⁹

It should be noted that the presence of bacteria that could achieve DIET was necessary in order to provide H^+ and e^- to *Methanosaeta*. As *Trichococcus* was enriched and a possible partner of *Methanosaeta*, the proteins originated from *Trichococcus* were analyzed. It was found that 17 proteins that originated from *Trichococcus* were enriched, and some of them (e.g., glucose-6-phosphate isomerase and enolase) were involved in glycolysis, which was consistent with its ability for the fermentation of glucose as previously reported.³⁴ However, the proteins relating to e-pilin and cytochrome C, which were involved in DIET,²⁷ were not detected. The absence of such proteins did not imply an absence of DIET,⁹ and they might be substituted by hydrochar for electron transfer. There were several uncharacterized proteins originated from *Trichococcus*, and it is still not known whether they were responsible for DIET. As previously mentioned, *Clostridium* was also dominant in the samples. It was found that a protein correlated to c-type cytochrome was enriched in *Clostridium*. c-type cytochromes are known to be involved in electron transport, and some of them contributed to DIET.¹⁰ However, the cytochrome C551 found in the present study was reported to shuttle electrons within bacterial periplasm,³⁶ while it is still unclear whether it can mediate extracellular electron transfer. Nevertheless, the proteomic analysis evidenced that SH changed the metabolic pathway of *Clostridium*, although its relative abundance was not obviously affected. The above results clearly showed that *Methanosaeta* was involved in DIET. Although there was no direct proteomic evidence for the participation of *Trichococcus* in DIET, its significant enrichment with the addition of hydrochar supported the high

Figure 3. Numbers of up-regulated and down-regulated proteins.

Figure 4. Cumulative methane production from glucose affected by different hydrochars and AC.

likelihood of its involvement in DIET together with *Methanosaeta*. In addition, it was proved that the genus *Trichococcus* contains microbes that could achieve extracellular electron transfer (EET), which also suggested that the genus *Trichococcus* probably participated in DIET in the present study (Supporting Information).

Effects of SH on the Key Microbial Populations As Revealed by qPCR Analysis. qPCR analysis was conducted to quantify the key microbial populations (Figure S8). The concentrations of both bacteria and archaea were higher in the bottles with SH compared to control, which indicated the stimulation of microbial growth by SH. Although the microbial analysis showed that the relative abundance of archaea (*Euryarchaeota*) was decreased in the presence of SH, the qPCR analysis demonstrated that its absolute concentration was higher, which was consistent with the higher methane production rate. The increase of MST (*Methanosaeta*) and decrease of MSC (*Methanosarcina*) in the presence of hydrochar was also demonstrated by qPCR analysis, further demonstrating the important role of *Methanosaeta* in the AD process with SH.

Main Characteristics of Hydrochar Related to the Promotion of the AD Process. The methane production with hydrochars obtained from HTL of different feedstocks as well as AC was compared, and the results are shown in Figure 4 and Table S6. All the tested hydrochars and AC themselves did not produce methane (data not shown). SH, CH, and EH significantly promoted the methane production rate by 39%, 15%, and 20% ($p < 0.05$), respectively, and they also decreased the duration of the lag phase. However, PH and AC did not show obvious enhancement on the methane production rate (Table S6). Moreover, relatively higher pH and faster consumption of VFA were also found for SH, CH, and EH, as seen in Figure S9. Although previous studies suggested that AC could achieve DIET to promote the methane production rate from various substrates,³⁷ the present study did not show any positive effect on the methane production rate, which was consistent with a previous study using glucose as a substrate under mesophilic conditions.³⁸ Nevertheless, the above results showed that hydrochar obtained from the HTL of different feedstocks had varying promotion effects on methane

production. Therefore, the characteristics of hydrochar and AC were determined to better identify potential reasons in different promotion effects on methane production. Previous studies suggested that conductivity was crucial to achieving DIET.^{7,37} However, only AC had notable conductivity ($3732 \mu\text{S cm}^{-1}$) while the conductivity of all the hydrochars could not be detected. The lack of conductivity of hydrochars could be due to the lower hydrothermal temperature. For instance, a previous study showed only a pyrolysis temperature higher than 600°C could form biochar with obvious conductivity that could achieve direct electron transfer.³⁹ AC was produced at a high pyrolysis temperature ($>700^\circ\text{C}$),⁴⁰ and therefore it had high conductivity. The above results showed that conductivity was not correlated to the enhancement of methane production rate.

Recent studies also suggested that redox property could affect the DIET.^{18,39} Therefore, a cyclic voltammetry analysis was conducted to examine the redox properties of hydrochars and AC (Figure 5a). A pair of redox peaks were observed for each hydrochar, while they were not found for AC. However, PH had higher peaks for the redox property compared to the other hydrochars, while it did not promote methane production. The redox properties of hydrochar were further quantified by I_2/NaBH_4 titration (Figure 5b),¹⁷ and PH still had the highest redox capacity considering the highest electron donating capacity (EDC) and electron accepting capacity (EAC), which was consistent with cyclic voltammetry (CV) results. It seems that the redox properties of hydrochar were also not correlated to the enhancement of the methane production rate.

It should be noted that both CV and I_2/NaBH_4 titration analysis gave the total redox properties of hydrochars. The oxygen-containing functional groups are related to the redox property, while it is possible that not all the oxygen-containing functional groups were accessible to the microbes. For instance, hydrochar and AC contained abundant pores with a size less than 35 nm (contributed to more than 50% of the BET surface area, Figure S10, Table S1), which was much smaller than microbes.⁴¹ Therefore, the surface functional groups were analyzed by FTIR (Figure 6). The oxygen-containing functional groups correlating to redox properties

Figure 5. Redox properties of hydrochars and AC: (a) Cyclic voltammetry curves; (b) EDC and EAC determined by I₂ titration and NaBH₄ titration, respectively.

include C–O, C=O, and C–OH,^{18,42} which were abundant in SH, CH, and EH but not in PH and AC. It is obvious that PH contained less redox functional groups on its surface, which could account for its incapability for the enhancement of methane production rate. Quantitative XPS showed that AC and PH had the lowest surface O/C atomic ratios, indicating they were more reduced than the other hydrochars (Table 1 and Figure S11). It also showed that PH and AC contained lower surface oxygen-containing functional groups compared to the other hydrochars. The highest abundance of surface oxygen-containing functional groups (especially C=O) was found in SH, which was consistent with its highest capacity to promote biogas production. Figure S12a further showed the surface oxygen-containing functional groups were significantly positively correlated to methane production rates with a *p* value of 0.001.

It is noteworthy that CV and quantification of the redox property have been employed to analyze the redox properties of biochar, hydrochar, and AC in previous studies,^{18,19,39} while our results showed that they were not necessarily correlated to

the promotion of biological process. The surface functional groups analysis based on FTIR and XPS provided valuable information to understand the role of hydrochar in AD. The other characteristics of hydrochar were also measured including BET and pH (Table S1), which were shown to be not correlated to the promotion of the methane production rates (Figure S12b and c, *p* > 0.05). Overall, the present study revealed that the surface redox functional groups of hydrochar were more significantly correlated to its promotion on methane production rate. For instance, SH contained more surface redox functional groups compared to CH, EH, and PH, and therefore it performed better in enhancing the methane production rate. The surface redox functional groups are related with quinone and hydroquinone moieties, which are able to donate and accept electrons and thus make hydrochar redox active.^{43,44} Therefore, hydrochar can accept electrons from bacteria by oxidation of organics and also donate electrons to methanogens for methane production, behaving as an electron shuttle for DIET to promote the methane production rate. The result was consistent with a previous

Figure 6. FTIR analysis of hydrochars and AC.

Table 1. Surface O/C Ratios by XPS, C–O Bond Percentages by Deconvolution of XPS C 1s Spectra and Electrical Conductivity (EC; Three Replicates Were Used for the Measurements)

	O/C (by XPS)	C–O bonds (%)				EC ($\mu\text{S cm}^{-1}$)
		C–OH	C=O	O–C=O	total	
AC	0.133 $\pm$ 0.011	33.3 $\pm$ 1.5	4.1 $\pm$ 0.2	5.8 $\pm$ 0.3	43.2 $\pm$ 1.5	3732 $\pm$ 120
SH	0.444 $\pm$ 0.035	56.8 $\pm$ 2.3	22.8 $\pm$ 1.1	5.2 $\pm$ 0.2	84.8 $\pm$ 2.6	
CH	0.286 $\pm$ 0.021	44.9 $\pm$ 1.8	6.6 $\pm$ 0.3	6.6 $\pm$ 0.3	58.2 $\pm$ 1.8	
EH	0.420 $\pm$ 0.015	45.8 $\pm$ 3.1	7.0 $\pm$ 0.2	6.7 $\pm$ 0.5	59.6 $\pm$ 3.1	
PH	0.250 $\pm$ 0.011	32.2 $\pm$ 2.2	9.0 $\pm$ 0.5	3.2 $\pm$ 0.1	44.5 $\pm$ 2.3	

study, which showed redox-active biochar could serve as an electron shuttle and accelerate DIET between *Geobacter metallireducens* and *Methanosarcina* in the coculture system.⁴⁵

Implications. Hydrochar is attracting much attention in recent years due to its potential application in environmental remediation, energy storage, and climate change mitigation.¹⁵ Regarding the environmental applications of hydrochar, previous studies mainly investigated its utilization as adsorbents for inorganic and organic pollutants.⁴⁶ The application of hydrochar in biological processes is an emerging field that has tremendous potentials. The present study demonstrated that SH facilitated the AD of glucose by surface oxygen-containing functional group-mediated DIET. The demonstration of DIET in the AD process is challenging since it is a mixed culture system with various microbes. Although recent studies investigated the addition of various materials to the AD process to achieve DIET for increasing methane production rate, only indirect evidence of DIET is available in most studies.⁹ A recent review suggested that the combination of different methods is necessary to confirm the presence of DIET,⁹ which was the case in the present study. Stable carbon isotopic analysis showed that SH increased hydrogenotrophic methanogenesis but did not affect IHT considering unchanged H_2 consumption rate, which indicated the presence of DIET. Further microbial and proteomic analysis showed SH enriched the strict acetoclastic methanogen *Methanosaeta* with up-regulated proteins involved in hydrogenotrophic methanogenesis, also demonstrating the presence of DIET considering the ability of *Methanosaeta* to use H^+ and e^- through hydrogenotrophic methanogenesis. The

present study provided evidence for DIET facilitated by SH in a mixed culture system and also showed that the genus *Trichococcus* contains microbes that could achieve EET. However, there was no direct evidence for the involvement of *Trichococcus* in DIET. Further study by metagenomic analysis deserves to be carried out to extract the genomes of enriched bacteria and reveal their potential metabolic pathways, which would possibly discover the bacteria involved in DIET.^{47,48} It is also deserved to attempt to isolate the enriched bacteria and investigate their roles in DIET. However, it might be difficult since only a few well characterized model microbes (*Geobacter* spp. and *Syntrophus aciditrophicus*) were demonstrated to be involved in DIET, and there are no studies reporting on which microbes have been recovered in culture with DIET as the enrichment/isolation procedure.¹⁰ In addition, several methods to identify electrochemically active bacteria, including U-tube microbial fuel cells and voltage-based screening assays designed using MFC engineering principles, could also be used to demonstrate the potential involvement of *Trichococcus* in DIET.^{49,50} Furthermore, the present study mainly focused on the demonstration of DIET in the hydrochar-facilitated AD process, and the role of adsorption in such a process was not considered. The role of adsorption in the present study might be ignored since hydrochar had a much lower BET surface area ($<100\text{ m}^2/\text{g}$) compared with that of GAC ($\sim 1000\text{ m}^2/\text{g}$) and might have very low adsorptivity for glucose and acetate with high polarity.

The higher methane production rate with SH means the organics can be converted to methane faster. Therefore, lower hydraulic retention time (HRT) or higher organic loading rate

(OLR) can be applied in the continuous experiments with SH in order to get the same organic removal efficiency, which would decrease the cost for wastewater treatment. Further study should be conducted in continuous experiments with various operational conditions (HRT, OLR, substrate type, reactor configuration, etc.) to investigate the promotion effects of SH on reactor performances. Furthermore, it was known that acidification was the common problem of anaerobic reactors, and it generally takes a long time for the recovery of the reactor due to the low pH and the presence of high concentrations of VFA.^{4,24} The addition of SH could be a simple yet effective solution to accelerate the recovery of acidified anaerobic reactors, since SH accelerated the degradation of VFA as seen in Figure 1c and d and Figure S3b. In addition, hydrochar could be potentially applied for the biodegradation of organic contaminants such as phenols and azo dyes in wastewater because their degradation has been reported to be enhanced by additive materials with redox properties.^{51,52} For the real organic wastes/wastewater, the mechanisms for the application of hydrochar to enhance methane production might be more complex considering its potential to adsorb some inhibitory compounds such as ammonia.⁵³ More research advancing the application of hydrochar in the AD process should be conducted. It should be noted that not all the hydrochars from different feedstocks could promote the methane production rate, and the abundant surface oxygen-containing functional groups of hydrochar are crucial. The methods (e.g., HNO₃ and H₂O₂ pretreatment) to modify the hydrochar to increase surface oxygen-containing functional groups could be explored in future studies to further enhance AD efficiency.^{18,20,54}

■ ASSOCIATED CONTENT

SI Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/acs.est.0c00112>.

Description of stable carbon isotope analysis, high-throughput sequencing of 16S rRNA genes, label-free proteomic analysis, qPCR analysis, evaluation of the extracellular electron transfer capacity of *Trichococcus* spp., analytical methods, Figures S1–S13, and Tables S1–S7 (PDF)

Upregulated and downregulated proteins (XLSX)

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Notes

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