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Mass Spectrometry Guided Discovery and Design of Novel Asperphenamate Analogs From *Penicillium astrolabium* Reveals an Extraordinary NRPS Flexibility

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Asperphenamate is a small peptide natural product that has gained much interest due to its antitumor activity. In the recent years numerous bioactive synthetic asperphenamate analogs have been reported, whereas only a handful of natural analogs either of microbial or plant origin has been discovered. Herein we describe a UHPLC-HRMS/MS and amino acid supplement approach for discovery and design novel asperphenamate analogs. Chemical analysis of *Penicillium astrolabium*, a prolific producer of asperphenamate, revealed three previously described and two novel asperphenamate analogs produced in significant amounts, suggesting a potential for biosynthesis of further asperphenamate analogs by varying the amino acid availability. Subsequent growth on proteogenic and non-proteogenic amino acid enriched media, revealed a series of novel asperphenamate analogs, including single or double amino acid exchange, as well as benzoic acid exchange for nicotinic acid, with the latter observed from a natural source for the first time. In total, 22 new asperphenamate analogs were characterized by HRMS/MS, with one additionally confirmed by isolation and NMR structure elucidation. This study indicates an extraordinary nonribosomal peptide synthetase (NRPS) flexibility based on substrate availability, and therefore the potential for manipulating and designing novel peptide natural products in filamentous fungi.

Keywords: natural product discovery, mass spectrometry, filamentous fungi, asperphenamate, amino acid incorporation, biological activity, NRPS flexibility

INTRODUCTION

Asperphenamate (1) is a linear amino acid (AA) ester, comprised of N-benzoylphenylalanine (2) and N-benzoylphenylalaninol (3). Asperphenamate, first discovered from *Aspergillus flavipes* in 1977 (Clark et al., 1977), was since found to be produced by a wide range of *Aspergillus* (Samson et al., 2011; Zheng et al., 2013; Ratnaweera et al., 2016; Hou et al., 2017) and *Penicillium*

(Frisvad et al., 2004, 2013) species. Additionally, the compound has also been isolated in trace amounts from a number of unrelated plant species (Wu et al., 2004; Dang et al., 2014; Zhou et al., 2017; Bunteang et al., 2018; Caridade et al., 2018), suggesting endophytic fungi being the actual producers, rather than the plants. Although asperphenamate is mainly known for its antitumor activity and immense synthetic chemists interest in asperphenamate backbone modification (Li et al., 2012; Yuan et al., 2012, 2018, 2019, 2020; Liu et al., 2016), recent studies have also shown asperphenamate to be a potential neuroinflammatory inhibitor (Zhou et al., 2017), and to possess anti-HIV (Bunteang et al., 2018) and antidiabetic (Del Valle et al., 2016) properties. In recent years, a handful of new natural analogs have been isolated, namely Asperphenamates B (4) and C (5) (Liu et al., 2018), and 4-OMe-asperphenamate (Zheng et al., 2013; Ratnaweera et al., 2016) (6) from filamentous fungi. Other analogs containing partial structural similarities include: patriscabratine (7), a N-benzoylphenylalanine phenylalaninol acetate ester, aurantiamide (8) and aurantiamide acetate (9) (Zhou et al., 2017), N-benzoylphenylalanine phenylalaninol and phenylalaninol acetate amides, all isolated from plant material; cordyceamides A (10) and B (11) (Jia et al., 2009), a N-benzoyl-L-tyrosinyl-L-phenylalaninol and N-benzoyl-L-tyrosinyl-L-p-hydroxyphenylalaninol acetates, from an insect pathogen fungus (Figure 1); along with a number of tentatively identified related metabolites (Kildgaard et al., 2014; Sica et al., 2016).

Biosynthesis of asperphenamate was first described in the filamentous fungus, *P. brevicompactum* (Li et al., 2018). Here, a two module NRPS system was described, where the first module, ApmA, is responsible for amide bond catalysis between the phenylalanine and benzoic acid moieties, and subsequent reduction to afford N-benzoylphenylalaninol (3), while the second module, ApmB, utilizes the same substrates to produce N-benzoylphenylalanine (2), as well as catalyses the ester bond formation between the two intermediates to release the final product asperphenamate (1). Assuming, that other filamentous fungi may follow the same or a similar biosynthetic pattern, the production of 4–5 in *Penicillium sp.* and 6 in *Aspergillus sp.*, involving tyrosine, 4-OMe-phenylalanine and 4-hydroxybenzoic acid instead of phenylalanine and benzoic acid as substrate molecules, indicates promiscuity of either one or both NRPS modules and provides new insights for production of novel asperphenamate analogs and laying the grounds for molecular biology work to achieve higher production of asperphenamate and related analogs.

To contribute to a better understanding of the diversity of asperphenamate biosynthesis and address the increasing resistance toward anticancer drugs (Vasan et al., 2019), *P. astrolabium* IBT 28865, a distant relative of *P. brevicompactum* from section *Brevicompacta* (Serra and Peterson, 2007), was investigated for production of asperphenamate and related analogs. In this study, we employ an ultra-high performance liquid chromatography diode array detection quadrupole time of flight high-resolution tandem mass spectrometry (UHPLC-DAD-QTOF-HRMS/MS) to dereplicate known and novel asperphenamate analogs. As a result, 22 novel asperphenamate

analogues were characterized by HRMS/MS, of which 21 were designed using proteogenic and non-proteogenic AAs as a supplement to the growth media. This study has further revealed a rare promiscuity of a fungal NRPS, laying the grounds for future NRPS research in filamentous fungi. Altogether, this study demonstrates the HRMS/MS based dereplication and characterization of novel analogs of a known bioactive peptide scaffold to be a powerful strategy in natural product discovery.

MATERIALS AND METHODS

Reagents, Strains, and Media

All solvents and reagents were purchased from Sigma-Aldrich (St. Louis, MO, United States), for the exception of *para*-substituted phenylalanines, which were acquired from Bachema (Bubendorf, Switzerland); ultra-pure water used throughout the study was filtered with a Milli-Q system (Millipore, Burlington, MA, United States).

Penicillium astrolabium (IBT 28865), *Penicillium olsonii* (IBT 28864), *Penicillium bialowiezense* (IBT 28294), and *Penicillium brevicompactum* (IBT 30524) are filamentous fungi from the IBT culture collection at the Department of Biotechnology and Biomedicine, Technical University of Denmark.

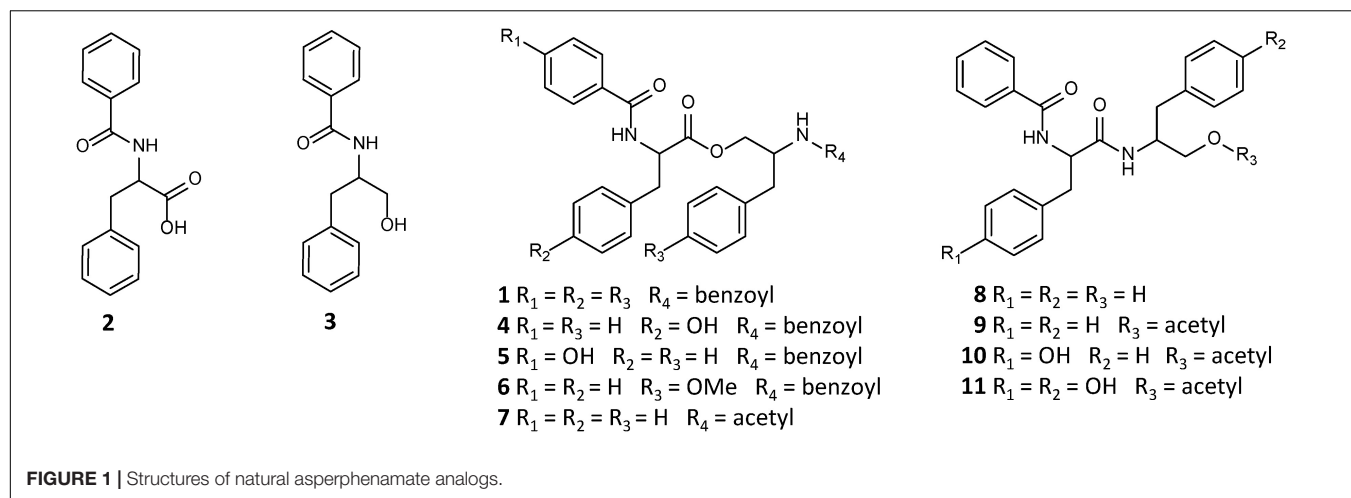
For the chemical profile analysis, *P. astrolabium* was cultivated with 3-point inoculation on Czapek yeast agar (CYA), yeast extract sucrose agar (YES) and malt extract agar (MEA; Oxoid) for 7, 10, and 14 days at 20 and 25°C in the dark. For large scale cultivation, the fungus was cultivated with 3-point inoculation on 200 YES agar plates, and incubated for 10 days at 25°C in the dark. For a proteogenic AA incorporation study, the fungus was cultivated with 3-point inoculation on Czapek (CZ) agar plates (10 cm) for 14 days at 25°C in the dark. Here, triplicates of 24 sets of supplemented media were used: 20 with all proteogenic AAs at 5 g/L; two for anthranilic acid and 4-hydroxybenzoic acid at 2.5 g/L, and two with additional inorganic nitrogen supplement of NaNO₃ at 5 g/L and 10 g/L. For a non-proteogenic AA incorporation study, the fungus was cultivated with 1-point inoculation on CZ agar plates (6 cm) for 14 days at 25°C in the dark. Here, triplicates of four sets of 4-chloro-L-phenylalanine, 4-bromo-L-phenylalanine, 4-amino-L-phenylalanine, and 4-nitro-L-phenylalanine supplemented media were used at 2.5 g/L.

For the chemical profile analysis and comparison, all four fungi were cultivated with 3-point inoculation on minimal media (MM), CZ, CYA, and YES 10 cm agar plates for 7 days 25°C in the dark.

Extraction and Isolation

For chemical profiling and the asperphenamate analog design study, five 6 mm diameter plugs taken in triplicates and extracted with acidic (1% formic acid; FA) isopropanol (iPr) – ethyl acetate (EtOAc) (1:3 v/v) as described by Smedsgaard (1997). All samples were re-dissolved ultrasonically for 10 min in 100 μL methanol (MeOH) and centrifuged prior to analysis by LC-MS.

For large-scale extraction, the agar plates were extracted twice with acidic (1% FA) EtOAc. The liquid-liquid partitioning was then performed on the crude extract, by dissolving it with 90%



MeOH:water and treating it with the same amounts of heptane, resulting in two phases. After separating the heptane phase, the 90% MeOH:water fraction was then diluted with water to get 50% MeOH:water, and further treated with dichloromethane (DCM), resulting in three phases overall. The DCM phase was dried before loading onto a 50 g SNAP column (Biotage, Uppsala, Sweden) with diol material (Isolute diol, Biotage). Crude fractionation was performed using an Isolera One automated flash system (Biotage) with stepwise increments of 25% at 50 mL/min in heptane-DCM-EtOAc-MeOH system, starting at 100% heptane, finishing at 100% MeOH, resulting in 13 fractions (i.e., heptane, heptane 3:1 DCM, heptane:DCM, heptane 1:3 DCM, DCM, DCM 3:1 EtOAc, DCM:EtOAc, DCM 1:3 EtOAc, EtOAc, EtOAc 3:1 MeOH, EtOAc:MeOH, EtOAc 1:3 MeOH, and MeOH), with 300 mL each. Selected resulting fractions were further fractionated on a 25 g SNAP column with RP C18 material (Grace, 15 $\mu\text{m}/100 \text{ \AA}$) at a flow rate of 30 mL/min using a stepwise 30–100% MeOH:water (both buffered with 50 ppm TFA) gradient as follows: in 10% increments at 30–50, 5% increments of 50–80, and 10% increments of 80–100%, resulting in 11 fractions (i.e., 30, 40, 50, 55, 60, 65, 70, 75, 80, 90, and 100%). Further separation was achieved on an Agilent Infinity 1290 HPLC-DAD (Agilent Technologies, Santa Clara, CA, United States) system, with UV monitoring at 230 and 280 nm, a flow rate of 4 mL/min and column temperature at 40°C as follows: Asperphenamate (1) and Asperphenamate L (13) were purified on a Gemini C₆-Phenyl column (5 μm , 110 \AA , 250 \times 10 mm, Phenomenex) using a linear gradient of 57 to 64% acetonitrile (MeCN)/water over 30 min; Asperphenamate W (12) on a Kinetex RP C18 column (5 μm , 100 \AA , 250 \times 10 mm, Phenomenex) using a linear gradient of 55 to 65% MeCN/water over 20 min at a flow rate of 4 mL/min; Asperphenamate Y (4) and Asperphenidine F1 (1a) on a Kinetex RP C18 column (5 μm , 100 \AA , 250 mm \times 10 mm, Phenomenex) using a linear gradient of 55 to 67% MeOH/water over 20 min at a flow rate of 4 mL/min. All solvents were buffered with 50 ppm TFA.

Asperphenamate (1): white powder; $[\alpha]_D^{20}$ -25.5° (c 0.11, CHCl_3); UV (MeCN) λ_{max} 238 and 272 sh nm; ^1H and ^{13}C NMR

data, see **Table 1** and **Supplementary Figure S1**; HRESIMS m/z 507.2279 $[\text{M}+\text{H}]^+$ (calculated for $\text{C}_{32}\text{H}_{30}\text{N}_2\text{O}_4$, m/z 507.2278).

Asperphenamate Y (4): white powder; $[\alpha]_D^{20}$ -25.8° (c 0.12, CHCl_3); UV (MeCN) λ_{max} 239 and 275 sh nm; ^1H and ^{13}C NMR data, see **Table 1** and **Supplementary Figure S2**; HRESIMS m/z 523.2227 $[\text{M}+\text{H}]^+$ (calculated for $\text{C}_{32}\text{H}_{30}\text{N}_2\text{O}_5$, m/z 523.2227).

Asperphenamate W (12): white powder; $[\alpha]_D^{20}$ -43.6° (c 0.14, CHCl_3); UV (MeCN) λ_{max} 236 and 278 nm; ^1H and ^{13}C NMR data, see **Table 1** and **Supplementary Figure S3**; HRESIMS m/z 546.2389 $[\text{M}+\text{H}]^+$ (calculated for $\text{C}_{34}\text{H}_{31}\text{N}_3\text{O}_4$, m/z 546.2387).

Asperphenamate L (13): white powder; UV (MeCN) λ_{max} 237 and 272 sh nm; ^1H and ^{13}C NMR data, see **Table 1** and **Supplementary Figure S4**; HRESIMS m/z 473.2436 $[\text{M}+\text{H}]^+$ (calculated for $\text{C}_{29}\text{H}_{32}\text{N}_2\text{O}_4$, m/z 473.2435).

Asperphenidine F1 (1a): white powder; UV (MeCN) λ_{max} 238 and 272sh nm; ^1H NMR data, see **Supplementary Table S5** and **Supplementary Figure S5**; HRESIMS m/z 508.2230 $[\text{M}+\text{H}]^+$ (calculated for $\text{C}_{31}\text{H}_{29}\text{N}_3\text{O}_4$, m/z 508.2231).

UHPLC-DAD-QTOF-MS Analysis

All samples were analyzed on an Agilent Infinity 1290 UHPLC system (Agilent Technologies, Santa Clara, CA, United States) equipped with a diode array detector (DAD), monitoring between 190 and 640 nm. Separation was achieved on an Agilent Poroshell 120 phenyl-hexyl column (150 mm \times 2.1 mm, 1.9 μm particles) with a flow rate of 0.35 mL/min at 40°C, using a linear acetonitrile (MeCN)/water (both buffered with 20 mM FA) gradient of 10 to 100% MeCN in 10 min, followed by 2 min flush at 100% MeCN, return to starting conditions in 0.1 min and equilibration at 10% for 2 min before the following run. It was coupled to an Agilent 6,545 QTOF MS equipped with Dual Jet Stream ESI source with the drying gas temperature of 250°C and gas flow of 8 L/min and sheath gas temperature of 300°C and flow of 12 L/min, capillary voltage 4,000 V and nozzle voltage of 500 V. The mass spectrometer was operated in positive polarity, recording centroid data in m/z range 100 to 1,700 for MS mode, and 30–1,700 for MS/MS mode, with acquisition rate of 10 spectra/s. Automated HRMS/MS was done for ions detected in the full scan above 50,000 counts,

TABLE 1 | ^1H and ^{13}C NMR shifts for asperphenamates F (1), Y (4), and W (12) and L (13) in chloroform (CDCl_3).

Position	1		4		12		13	
	δ_{C}^*	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)
1			167.7		167.6		167.8	
2			133.4		133.5		133.3	
3	127.0	7.70 dd (8.3, 1.1)	127.2	7.65 m	127.3	7.63 m	127.1	7.72 m
4	128.6	7.39 m	128.8	7.39 m	128.6	7.29 m	128.6	7.41 m
5	131.9	7.50 tt (7.5, 1.1)	132.2	7.50 tt (7.4, 1.2)	132.1	7.48 t (7.4)	132.2	7.52 m
1'			172.2		172.5		173.2	
2'	54.4	4.92, q (6.6)	54.8	4.87 d (6.7)	54.3	5.04 q (6.5)	52.1	4.71 m
3' NH		6.58 d (6.6)		6.59 d (6.6)		6.69 d (6.4)		6.46 d (6.8)
4'	37.5	3.29 dd (14.0, 6.6)	36.9	3.20 dd (14.0, 6.5)	27.7	3.43 d (6.0)	40.8	1.79 m
		3.21 dd (14.0, 7.0)		3.14 dd (14.0, 6.9)				1.69 m
5'			127.6		110.1		25.1	1.75 m
6'	129.1	7.21 m	130.5	7.05 m	127.5		22.2	0.99 d (6.5)
7'	128.8	7.29 m	116	6.76 m	118.7	7.64 m	22.8	1.02 d (6.5)
8'	126.7	7.24 m	155.3		120.1	7.12 t (7.4)		
9'					122.7	7.20 m		
10'					111.6	7.33 d (7.9)		
11'					136.4			
12' NH						8.06 s		
13'					123.1	7.06 d (2.2)		
1''	65.3	4.54 dd (11.4, 3.4)	65.5	4.50 dd (11.4, 3.6)	65.4	4.46 dd (11.6, 3.6)	65.1	4.59 dd (11.5, 3.3)
		4.04 dd (11.4, 4.4)		4.04 dd (11.4, 4.5)		4.06 dd (11.6, 4.6)		4.08 dd (11.5, 4.6)
2''	50.2	4.62 m	50.6	4.60 m	50.6	4.55 m	50.5	4.69 m
3'' NH		6.67 d (8.4)		6.71 d (8.4)		6.59 d (8.4)		6.73 d (8.2)
4''	37.2	3.00 dd (13.7, 6.4)	37.4	3.01 dd (13.9, 6.5)	37.3	2.94 dd (13.6, 6.7)	37.3	3.10 dd (13.6, 6.5)
		2.89 dd (13.8, 8.5)		2.91 dd (13.9, 8.2)		2.81 dd (13.6, 8.4)		3.01 dd (13.8, 8.2)
5''			137.1		137.4		137.3	
6''	129.2	7.23 m	129.5	7.22 m	129.5	7.18 s (7.6)	129.3	7.29 m
7''	128.3	7.32 m	128.9	7.31 m	128.7	7.36 t (7.8)	128.7	7.32 m
8'''	127.3	7.25 m	127	7.24 m	126.9	7.23 m	126.8	7.25 m
1'''			167.8		167.4		167.3	
2'''			134.1		134.4		134.3	
3'''	126.9	7.65 dd (8.3, 1.1)	127.3	7.68 m	127.2	7.63 m	127.1	7.70 m
4'''	128.6	7.31 m	128.6	7.31 m	128.8	7.29 m	128.4	7.28 m
5'''	131.3	7.43 tt (7.5, 1.1)	131.7	7.43 tt (7.4, 1.1)	131.5	7.43 t (7.4)	131.3	7.42 m

* ^{13}C NMR data available only from HSQC experiment.

with a cycle time of 0.5 s, quadrupole width of $m/z \pm 0.65$ using fixed CID energies of 10, 20, and 40 eV with maximum three precursor ions per cycle. A lock mass solution of 70% MeOH was infused in the second sprayer, with an extra LC pump at a flow of 15 $\mu\text{L}/\text{min}$ using a 1:100 splitter. The solution contained 1 μM tributylamine (Sigma-Aldrich) and 10 μM hexakis(2,2,3,3-tetrafluoropropoxy)phosphazene (Apollo Scientific Ltd., Cheshire, United Kingdom) as lock masses. The $[\text{M} + \text{H}]^+$ ions of both compounds (m/z 186.2216 and 922.0098, respectively) were used.

In-house fungal metabolite library search was done as described by Kildgaard et al. (2014). Data files were processed in MassHunter workstation B.07.00 with "Find by Auto MS/MS function" with a processing limit to 200 largest peaks and mass match tolerance m/z 0.05. HRMS/MS library search using parent and fragment ion accuracy of

20 ppm + 2 mDa, with minimal forward score of 50 and reverse score of 80.

Targeted analysis for the asperphenamate analog design study was performed using expected masses of individual AA and benzoic acid analogs, for all potential precursors, intermediates and final products, see **Supplementary Table S1**. Relative amounts of asperphenamate analogs were quantified by direct integration of peak area of target compounds, normalized to xanthoepocin water loss adduct peak area ($[\text{M}-\text{H}_2\text{O}+\text{H}]^+$ m/z 589.0975) in control samples. All analyses were performed in triplicates. All MS/MS spectra reported were at 20 eV, unless stated otherwise.

General Experimental Procedures

1D and 2D NMR analyses were performed on a Bruker Avance 800 MHz spectrometer (Bruker, Billerica, MA, United States),

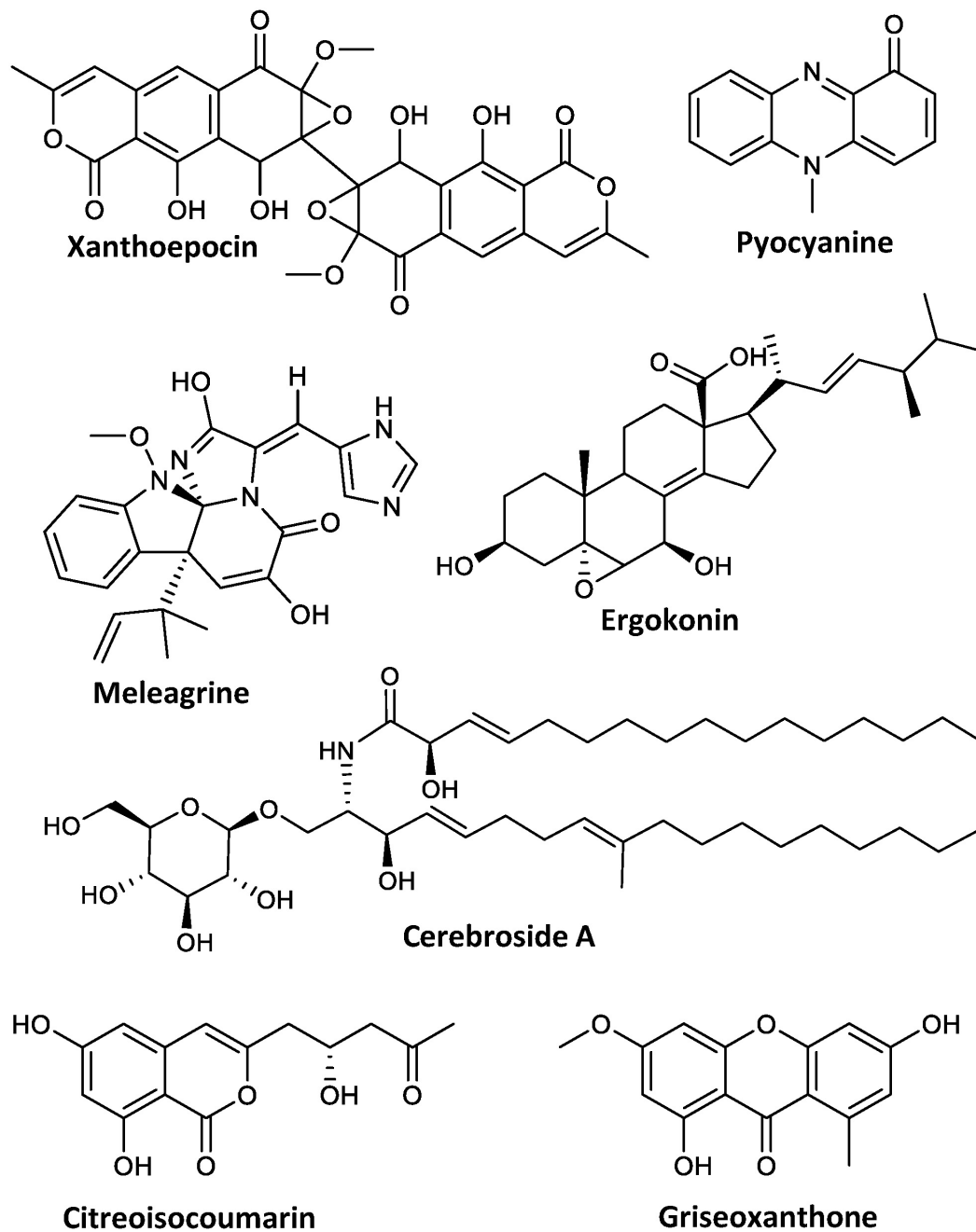


FIGURE 2 | Compounds produced by *P. astrolabium* based on the in-house database search.

using standard sequence pulses. Samples were analyzed in a 3 mm TCl cryoprobe using deuterated chloroform (CDCl_3) and referenced to the residual solvent signals $\delta_{\text{H}} = 7.26$ ppm and $\delta_{\text{C}} = 77.16$ ppm. J-couplings are reported in hertz (Hz) and chemical shifts (δ) in ppm. For 1D and 2D NMR data, see **Supplementary Tables 2–4** and **Supplementary Figures 1–5**.

Optical rotations were measured in chloroform (CHCl_3) on a PerkinElmer 341 Polarimeter (PerkinElmer, Waltham, MA, United States).

Cytotoxicity Assay

Compounds **1**, **1a**, **4**, **12**, and **13** were tested in triplicates against five cancer cell lines, i.e., human lung carcinoma A549 ATCC CCL-185, breast adenocarcinoma MCF7 ATCC HTB-22, human skin melanoma A2058 ATCC CRL-11147, hepatocyte carcinoma HepG2 ATCC HB-8065 and pancreas carcinoma MiaPaca-2 ATCC CRL-1420 following previously described methodology (Audoin et al., 2013; Lauritano et al., 2020).

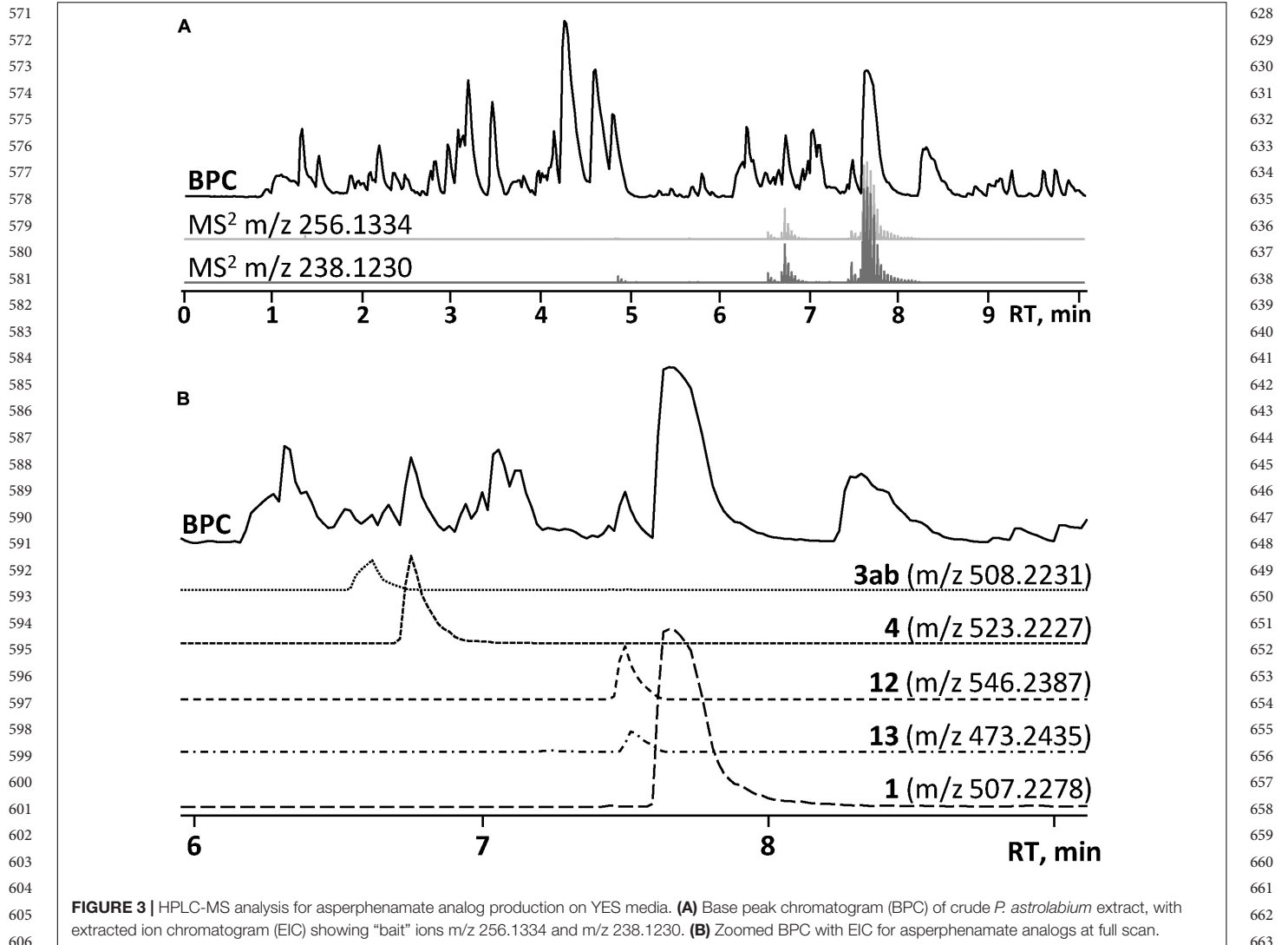


FIGURE 3 | HPLC-MS analysis for asperphenamate analog production on YES media. **(A)** Base peak chromatogram (BPC) of crude *P. astrolabium* extract, with extracted ion chromatogram (EIC) showing “bait” ions m/z 256.1334 and m/z 238.1230. **(B)** Zoomed BPC with EIC for asperphenamate analogs at full scan.

RESULTS

Chemical Profile of *Penicillium astrolabium*

To investigate the secondary metabolite profile from *P. astrolabium*, the fungus was inoculated on three media (MEA, CYA, and YES) and incubated at 20 and 25°C for 7, 10, and 14 days. The resulting 18 extracts were analyzed by UHPLC-DAD-QTOF-MS and used for automated in-house library search of fungal secondary metabolites (Kildgaard et al., 2014). In addition to previously reported asperphenamate (1), N-benzoylphenylalanine (2) and xantopocin (Perrone et al., 2015), all 18 extracts also revealed the presence of meleagrine and its biosynthetic intermediates neoxaline, glandicoline B, roquefortine C and histidyltryptophanyldiketopiperazine (Ali et al., 2013), as well as cerebroside A. Other notable secondary metabolites include ergokonin B and pyocyanine detected on YES and CYA media extracts; citreoisocoumarin detected only in YES extracts; griseoxanthone C mainly seen

on CYA 20°C (Figure 2). Additionally, a series of di- and tetracyclopeptides with varying AA composition, depending on growth conditions, were produced.

Targetted Asperphenamate Daughter Ion Search Reveals Novel Analogs

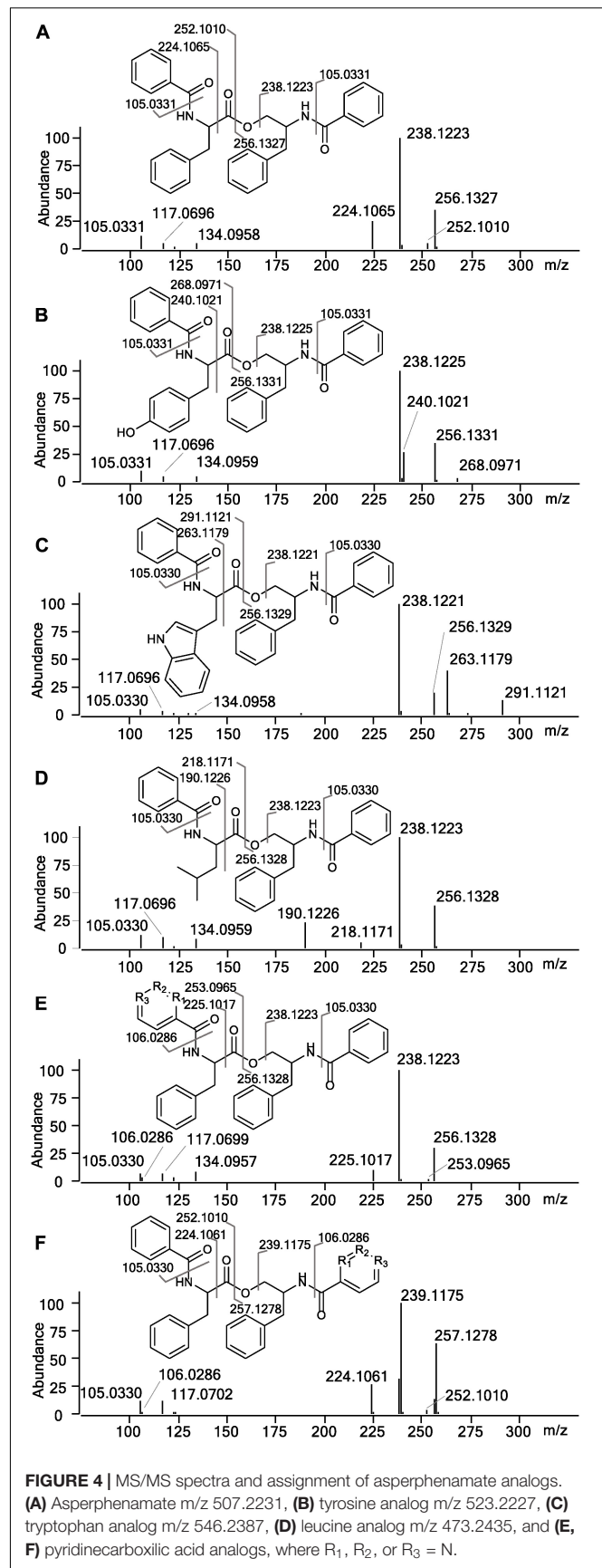
In the HRMS/MS analysis, asperphenamate (1) and reported fungal analogs (4–6) share two major fragment ions m/z 256.1334 and m/z 238.1230, corresponding to ester bond cleavage to result in a N-benzoylphenylalaninol protonated ion $[C_{16}H_{18}NO_2]^+$, followed by water loss on the same moiety to get $[C_{16}H_{16}NO]^+$. In addition, a minor fragment of m/z 105.0335 $[C_7H_5O]^+$ corresponding to a benzoyl loss was also observed. To screen for potential asperphenamate analogs, the two major fragment ions were used as “bait” (Figure 3A), resulting in five major peaks (Figure 3B): asperphenamate ($[M+H]^+$ m/z 507.2277, $C_{32}H_{30}N_2O_4$), the most abundant analog with an extra oxygen atom ($[M+H]^+$ m/z 523.2227, $C_{32}H_{30}N_2O_5$), an analog indicating a single carbon-nitrogen

685 exchange ($[M+H]^+$ m/z 508.2232, $C_{31}H_{29}N_3O_4$), and two
 686 other analogs with a significant mass differences, one 34Da
 687 lower ($[M+H]^+$ m/z 473.2436, $C_{29}H_{32}N_2O_4$) and the other
 688 39Da higher ($[M+H]^+$ m/z 546.2385, $C_{34}H_{31}N_3O_4$) to that
 689 of asperphenamate. Subsequent MS/MS analysis revealed, that
 690 two asperphenamate-specific fragments m/z 252.1010 and m/z
 691 224.1064, corresponding to ester cleavage and subsequent CO
 692 loss of an N-benzoylphenylalanine moiety (**Figure 4A**), have been
 693 exchanged for fragments 16Da higher, namely m/z 268.0971 and
 694 m/z 240.1021 (**Figure 4B**), for compound with m/z 523.2227.
 695 Therefore, with the fragment of m/z 105.0331 being present
 696 in both of compounds, and no other major differences in
 697 fragmentation patterns observed, a phenylalanine exchange for
 698 tyrosine in a non-reduced N-benzoyl AA moiety could be
 699 proposed, resulting in **4**. The asperphenamate analog with
 700 m/z 546.2385 produced unique fragments of m/z 291.1121
 701 $[C_{18}H_{15}N_2O_2]^+$ and m/z 263.1179 $[C_{17}H_{15}N_2O]^+$ (**Figure 4C**).
 702 Taking into account the presence of a benzoyl ion $[C_7H_5O]^+$,
 703 the rest of m/z 291.1121 fragment suggest molecular formula
 704 $C_{11}H_{10}N_2O$, corresponding to phenylalanine exchange for
 705 tryptophan in a non-reduced N-benzoyl AA moiety. Similarly,
 706 the differences between the unique fragments in m/z 473.2436
 707 and the benzoyl ion led to proposal of leucine containing analog
 708 (**Figure 4D**). Finally, a compound with m/z 508.2232 showed
 709 similar fragmentation patterns to those of asperphenamate,
 710 however, fragments corresponding to fragmentation of non-
 711 reduced AA moiety showed a fragment mass increase by 1Da,
 712 with fragment ion m/z 106.0286 $[C_6H_4NO]^+$ indicating a
 713 pyridinecarboxylic acid incorporation (**Figure 4E**). Moreover,
 714 MS/MS data revealed trace amounts of a coeluting isomer, with
 715 the two major ion fragments weighing 1 Da higher, namely
 716 m/z 257.1278 and m/z 239.1175, hence indicating that the
 717 pyridinecarboxylic acid can also be incorporated into the reduced
 718 AA part of the molecule (**Figure 4F**).

720 NMR Confirms Phenylalanine Exchange 721 for Other Amino Acids in the 722 Non-reduced N-Benzoyl Amino Acid 723 Moiety

725 To confirm the structures proposed by HRMS/MS fragmentation
 726 patterns, a large scale of 200 agar plates was grown for
 727 targeted isolation of asperphenamate (**1**), and tyrosine (**4**),
 728 tryptophan (**12**) and leucine (**13**) analogs, as well as one of
 729 the pyridinecarboxylic acid analogs (**1a**). 1H and ^{13}C NMR
 730 data shown in **Table 1**, with full assignment table and spectra
 731 available in supplementary material (**Supplementary Tables 2–**
 732 **4** and **Supplementary Figures 1–5**). Data for asperphenamate
 733 (**1**) and the tyrosine analog (**4**) fit with previously published
 734 data (Catalán et al., 2003; Liu et al., 2018). 1H and ^{13}C NMR
 735 shifts of N-benzoylphenylalaninol and the N-benzoyl part of
 736 non-reduced AA moiety was in agreement within all four
 737 compounds, further supported by COSY and HMBC correlations
 738 for tryptophan (**12**) and leucine (**13**) analogs (**Figure 5** and
 739 **Supplementary Tables 3–4**).

740 The rest of the shifts corresponding to **12** showed three spin
 741 systems, with the first comprised of an amino group at $NH-3'$



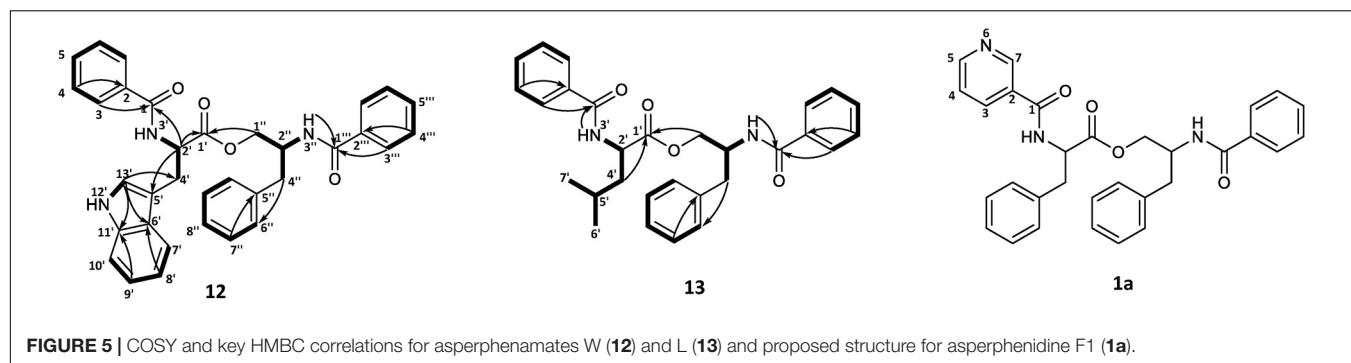


FIGURE 5 | COSY and key HMBC correlations for asperphenamates W (**12**) and L (**13**) and proposed structure for asperphenidine F1 (**1a**).

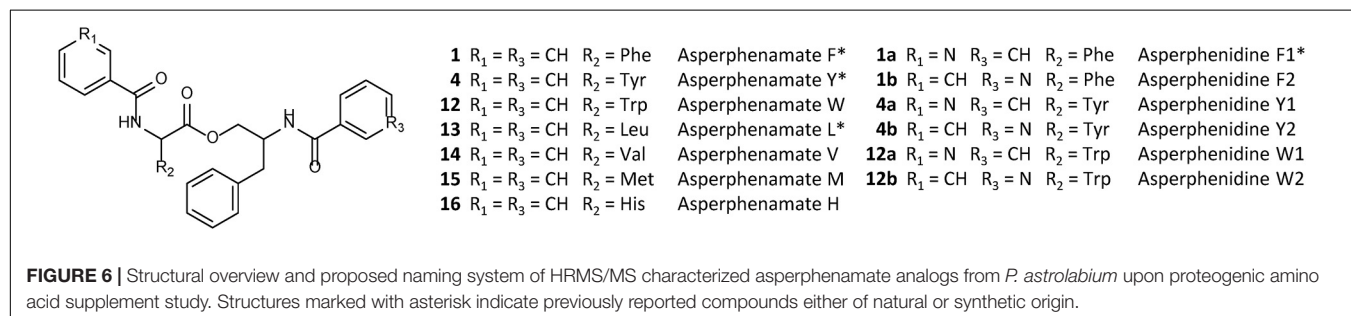


FIGURE 6 | Structural overview and proposed naming system of HRMS/MS characterized asperphenamate analogs from *P. astrolabium* upon proteogenic amino acid supplement study. Structures marked with asterisk indicate previously reported compounds either of natural or synthetic origin.

(δ_{H} 6.69), a methine at H-2' (δ_{H} 5.04) and a methylene at H-4' (δ_{H} 3.43), the second consisted of four aromatic methines at H-7' (δ_{H} 7.64), H-8' (δ_{H} 7.12), H-9' (δ_{H} 7.20) and H-10' (δ_{H} 7.33), and the third one included aromatic amino and methine groups, NH-12' (δ_{H} 8.06) and H-13' (δ_{H} 7.06), respectively. The HMBC correlations of the last two spin systems from H-7' and H-13' to C-5' (δ_{C} 110.1), H-8' and H-13' to C-6' (δ_{C} 127.5), and H-9' and H-13' to C-11' (δ_{C} 136.4), revealed the presence of indole, which was connected to the first spin system by H-2' to C-5' (δ_{C} 110.1) and H-4' to C-13' (δ_{C} 123.1), to confirm presence of tryptophan. The HMBC correlations from H-3 (δ_{H} 7.63) and H-2' to C-1 (δ_{C} 167.6) and H-2' and H-1'' (δ_{H} 4.46/4.06) to C-1' (δ_{C} 172.5), connected tryptophan moiety to the benzoyl and N-benzoylphenylalaninol parts of the molecule (Figure 5).

For **13**, the rest of the shifts comprised a single spin system of amino group NH-3' (δ_{H} 6.46), two methines at H-2' (δ_{H} 4.71) and H-5' (δ_{H} 1.75), a diastereotopic methylene at H-4' (δ_{H} 1.79/1.69), and two methyl groups at H-6' (δ_{H} 0.99) and H-7' (δ_{H} 1.02), to give a leucine backbone. The spin system was connected to the rest of the structure by H-3 (δ_{H} 7.72) and H-2' to C-1 (δ_{C} 167.8) and H-2' and H-1' (δ_{H} 4.59/4.08) to C-1' (δ_{C} 173.2) (Figure 5). The NMR data was eventually found to be in agreement with the commonly overlooked lichen secondary metabolite hypothallin (Huneck et al., 1992).

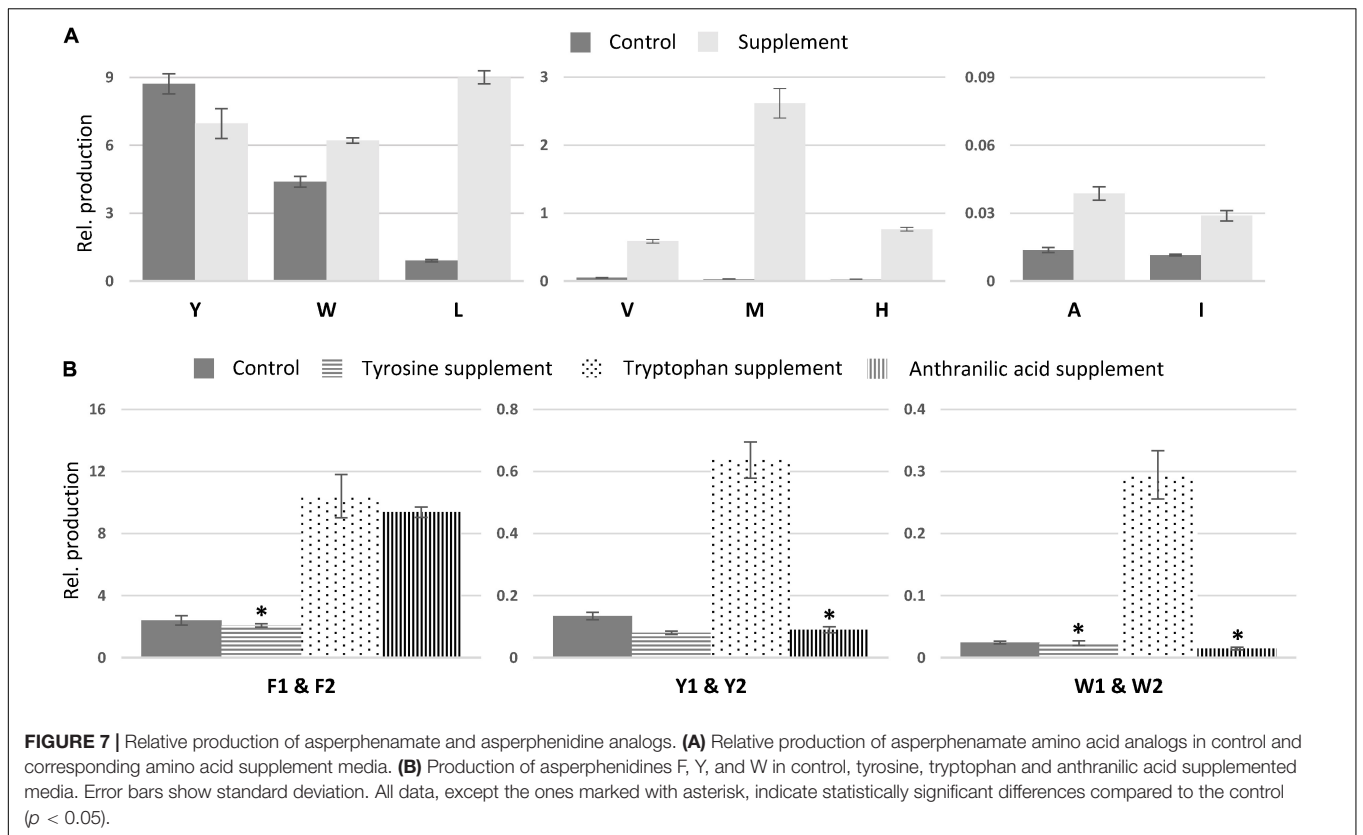
Compound **1a** was purified in trace amounts (0.2 mg) and only ¹H NMR was acquired (Supplementary Figure S5). In comparison to asperphenamate, all the proton shifts in the aliphatic range for **3a** were of same multiplicity and similar shift values, as well as shifts for both amino groups. In the aromatic range, three asperphenamate shifts at H-3 (δ_{H} 7.70), H-4 (δ_{H} 7.39) and H-5 (δ_{H} 7.50) were swapped for more downfield shifts

at δ_{H} 7.94 (m), δ_{H} 8.72 (dd), and δ_{H} 8.87 (d). Based on the two latter shifts and their multiplets, they were assigned as H-5 and H-7, respectively, with δ_{H} 7.94 (m) assigned at H-3, and the H-4 shift assigned to the general aromatic region at 7.18–7.35, led to confirmation of pyridinecarboxylic acid moiety as nicotinic acid. This fits with the published NMR data for nicotinic acid (Chen et al., 1999) and the corresponding synthetic asperphenamate analog (Liu et al., 2016).

Herein we propose a new asperphenamate analog naming system using one letter AA abbreviation to denote a specific AA incorporation, based on similar azaphilone pigment naming system proposed by Isbrandt et al. (2020). Compounds **12** was named asperphenamate W, whereas compounds **4** and **13** will be referred to as asperphenamate Y and L, respectively. Compound **1a** was named asperphenidine F1, to signify it being an asperphenamate analog, with phenylalanine incorporation and benzoic acid exchange for nicotinic acid at the non-reduced part of the molecule.

Amino acid Enriched Media Induces Phenylalanine Exchange in the N-Benzoylphenylalanine Moiety

To investigate if higher AA availability can induce AA exchange in asperphenamate biosynthesis, the fungus was incubated on CZ media supplemented with one of each of the 20 proteogenic AAs. Subsequently, targeted MS analysis was performed by search of masses corresponding to phenylalanine exchange for one AA moiety within the asperphenamate backbone (Supplementary Table S1). In addition tyrosine (**4**), tryptophan (**12**), and leucine (**13**) analogs, the novel valine (**14**), methionine (**15**), histidine (**16**), alanine and isoleucine



analogues could also be observed, and AA exchange in the non-reduced AA part of the molecule was confirmed for compounds **14–16** by HRMS/MS fragmentation patterns (**Figure 6** and **Supplementary Figure S6**). The new analogues were accordingly named as asperphenamates V, M, and H.

In comparison to non-fed control cultures of previously characterized compounds, only small changes in production were observed for asperphenamate Y and W production, a slight decrease and increase, respectively (**Figure 7A**). However, upon leucine supplement, asperphenamate L production increased 10-fold in comparison to the control. A similar increase pattern was also observed in the valine supplement experiment, whereas the production of histidine and methionine analogues was drastically boosted upon respective AA supplement, with 33-fold increase for asperphenamate H and more than a 100-fold increase for asperphenamate M. Although, the production of alanine and isoleucine analogues increased by three-fold for each, the relative amounts were still marginally lower in comparison to other uptake experiments, and were not sufficient for MS/MS data acquisition and assignment of structures (**Figure 7A**).

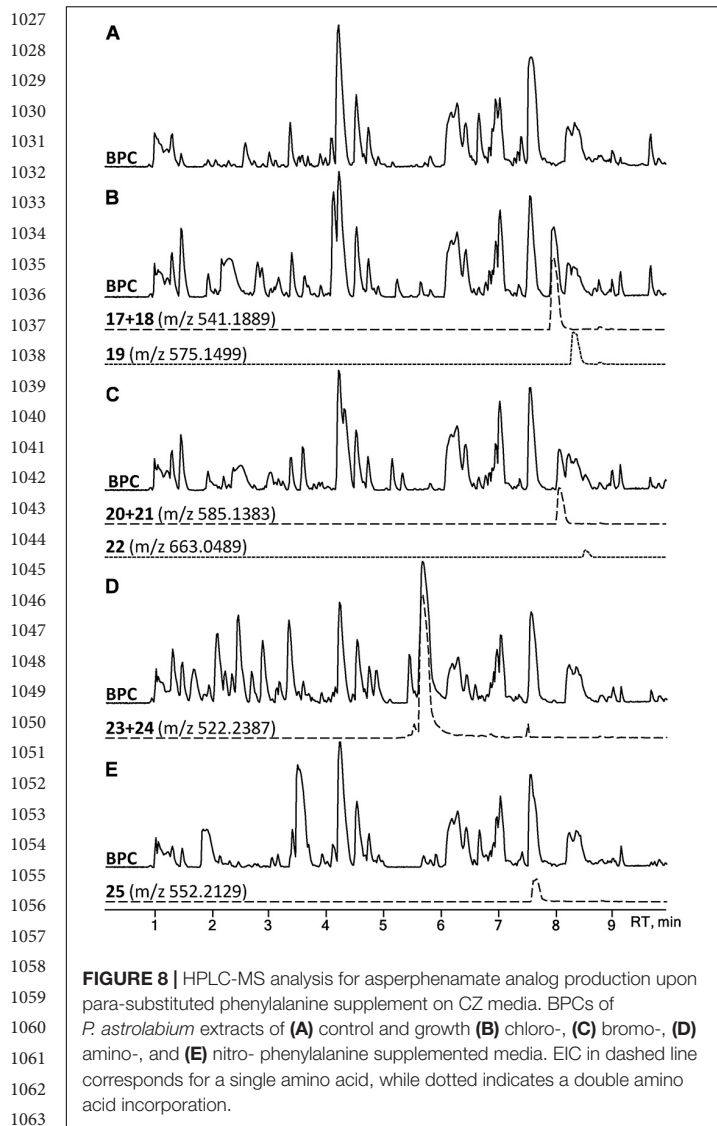
Additionally, MS analysis targeting reduced and non-reduced AA N-benzoyl precursors, and single or double AA exchange in asperphenamate, patriscabratine, aurantiamide and aurantiamide acetate backbones, was performed to result in the discovery of an additional double leucine asperphenamate analogue characterized by HRMS/MS (**Supplementary Figure S6**). No other AA analogues or analogues

for benzoic acid exchange for 4-hydroxybenzoic or anthranilic acids were observed.

Tryptophan Induces Nicotinic Acid Incorporation

In contrast to the expected asperphenamate W (**12**) being one of the major metabolites upon growth on tryptophan supplemented media, both asperphenidines F1 and F2 (**1ab**), with nicotinic acid exchange on either the non-reduced or reduced part of the molecule, respectively, showed the most drastic increase in the relative amount in comparison to the control. With the similar behavior observed in anthranilic acid supplement, targeted MS analysis of all extracts was performed with masses corresponding to the benzoic acid exchange for nicotinic acid for novel AA analogues described above (**Supplementary Table S1**). The MS profile indicated the potential for nicotinic acid incorporation in all eight AA analogues, however, only tyrosine (**4ab**) and tryptophan (**12ab**) analogues, with nicotinic acid incorporation in either of two possible positions, could be confirmed by MS/MS (**Figure 6** and **Supplementary Figure S7**). With individual analogue peaks strongly overlapping in the MS profile, MS/MS of two coeluting analogues as well as combined peak area were used for further structure assignment and relative quantification, respectively.

The relative amounts of the most common AA and nicotinic acid analogues, asperphenidines Y1–Y2 (**4ab**), and asperphenidines W1–W2 (**12ab**), were further compared to those of the asperphenidines F1–F2 (**1ab**) (**Figure 7B**).

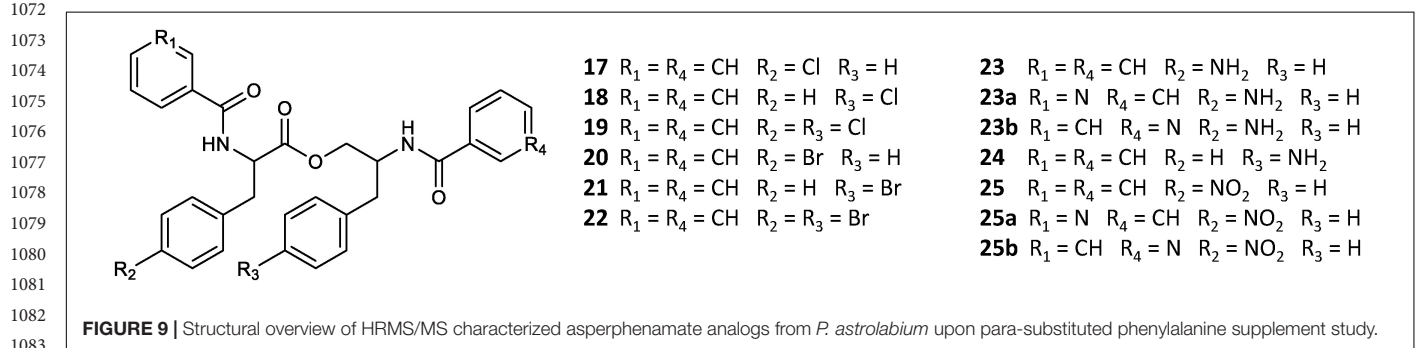


The relative production of asperphenidines followed the same pattern as observed in asperphenamate production: asperphenidines F1-F2 being the major nicotinic acid analogs, followed by asperphenidines Y1-Y2 and W1-W2. Moreover, production of all three upon growth on tryptophan supplemented

media was drastically higher in comparison to control, with four- and five-fold increase in asperphenidines F1-F2 and Y1-Y2, and 12-fold increase for asperphenidine W1-W2 production. However, upon anthranilic acid supplement, a significant increase was observed only in asperphenidine F1-F2 production. Additionally, production of all asperphenamates and asperphenidines was slightly lower to that of the control. Additional analysis of other AA supplement cases or additional inorganic nitrogen supplement experiments did not trigger similar nicotinic acid incorporation response.

Para-Substituted Phenylalanines Are Incorporated in Either of N-Benzoyl Amino Acid Moieties

A set of four *para*-substituted, namely chloro-, bromo-, amino- and nitro-, phenylalanines were used to investigate the uptake of non-natural phenylalanines in asperphenamate biosynthesis. The targeted MS analysis was performed by search of masses corresponding to a single or double AA exchange in both asperphenamate and asperphenidine backbones (Supplementary Table S1). This revealed, that incorporation of single *para*-substituted AA was successful in all four supplement cases, however, a double *para*-substituted AA exchange was also observed in the halogenated phenylalanine supplement experiments (Figure 8). Moreover, small amounts of asperphenidine derivatives for single amino- and nitro-phenylalanine incorporation analogs were also detected. Subsequent MS/MS analysis revealed, that halogenated *para*-substituted phenylalanine can be incorporated at either or both reduced or non-reduced parts of the molecule, resulting in three analogs each for chloro- (17-19) and bromo- (20-22) asperphenamates (Figure 9 and Supplementary Figure S8). In case of amino- and nitro- phenylalanine exchange, the single AA incorporation was clearly preferred at the non-reduced part of the molecule, with only trace amounts of substituted AA incorporation at the reduced part of the molecule detected (23-25). As a result, asperphenidine analogs were detected only with AA exchange at the non-reduced part of the molecule, resulting in two analogs each for amino- (23ab) and nitro- (25ab) asperphenidines (Figure 9 and Supplementary Figure S9). Additionally, non-reduced and reduced pathway intermediates, containing *para*-substituted phenylalanines, were also detected in each of the supplement study cases.



1141 **TABLE 2** | Cytotoxic activities of 1, 1a, 4, 12, and 13.

Compound	EC ₅₀ (μg/mL)				
	A549	MCF7	A2058	HepG2	MiaPaca
1	>46	>46	1.1	28.5	>46
4	>46	23	24.8	21.6	>46
12	>46	>46	16.6	>46	>46
13	>46	>46	>46	>46	>46
1a	>46	>46	13.3	>46	13.3

1151 A549, lung adenocarcinoma; MCF7, breast carcinoma; A2058, skin melanoma;
 1152 HepG2, hepatocellular carcinoma; MiaPaca, pancreas carcinoma.

1153

1154 Asperphenamate Amino Acid Exchange 1155 Is Also Observed in Other Section 1156 *Brevicompecta* strains

1157 To compare the production of asperphenamate analogs among
 1158 section *Brevicompecta*, the chemical profile of *P. astrolabium*
 1159 was compared to three other section species, namely *P. olsonii*,
 1160 *P. bialowiezense*, and *P. brevicompactum* (Houbraken et al.,
 1161 2020). Targeted MS search based on 5 readily observed
 1162 asperphenamates in *P. astrolabium* (asperphenamates F, Y, W,
 1163 L, and asperphenidine F1) revealed, that all the other strains
 1164 were also able to exchange phenylalanine in the non-reduced
 1165 AA moiety (**Supplementary Figure S10**). Additionally, all three
 1166 strains were producing isoleucine analog in similar or higher
 1167 amount in comparison to asperphenamate L and were able to
 1168 produce compound 5, which was not detected in *P. astrolabium*.
 1169 Moreover, *P. brevicompactum* revealed two peaks corresponding
 1170 to the value of Asperphenamate W protonated adduct (m/z
 1171 546.2385), both with the same fragmentation patterns, suggesting
 1172 them being structural isomers.
 1173

1174 Amino Acid Exchange Affects the 1175 Asperphenamate Analog Cytotoxicity

1176 Asperphenamates F, Y, W, and L (**1**, **4**, **12** and **13**) as
 1177 well as asperphenidine F1 (**1a**) were tested for their cytotoxic
 1178 activities against five cancer cell lines, i.e., lung carcinoma
 1179 A549, breast adenocarcinoma MCF7, skin melanoma A2058,
 1180 hepatocyte carcinoma HepG2 and pancreas carcinoma MiaPaca
 1181 (**Table 2**). Asperphenamate Y and Asperphenidine F1 were the
 1182 only compounds exhibiting moderate cytotoxic activities against
 1183 MCF7 and MiaPaca cell lines, respectively. Asperphenamates
 1184 F and Y exhibited moderate activity against HepG2 cell lines,
 1185 whereas all but asperphenamate L showed activity toward A2058
 1186 cell line, with asperphenamate F exhibiting the strongest activity.
 1187 Asperphenamate L did not show activity against any of the
 1188 cell lines at the tested concentrations. None of the compounds
 1189 exhibited activity against A549 cell line at tested concentrations.
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1193 DISCUSSION

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1195 Only a handful of asperphenamate analogs from fungal sources,
 1196 including endophytic and parasitic fungi, have been isolated to
 1197 date (Jia et al., 2009; Frisvad et al., 2013; Houbraken et al.,

2020). In this study we have demonstrated yet another powerful
 application of a HRMS/MS guided discovery approach, for
 detection and structural characterization of novel peptide natural
 products via HRMS/MS fragmentation pattern analysis, which
 is in line with current HRMS/MS based peptide detections and
 characterization approaches (Mohimani et al., 2017; Jarmusch
 et al., 2020; Ricart et al., 2020). Moreover, we report that the
 choice of complex growth medium and/or a simple growth media
 supplement with selected building blocks, such as proteogenic
 and non-proteogenic AAs, *P. astrolabium* and related species can
 produce a series of novel asperphenamate analogs, which can be
 readily characterized by HRMS/MS.

Incubation of *P. astrolabium* IBT 28865 on complex
 media revealed the strain being readily capable of exchanging
 phenylalanine for tyrosine, leucine or tryptophan in
 N-benzoylphenylalanine moiety, suggesting that the preferred
 AA substrate should be either aromatic or aliphatic. A subsequent
 proteogenic AA supplement study supported the hypothesis, with
 all but one AA incorporated being either aromatic or aliphatic.
 It can be speculated, that the incorporation of AA is dependent
 on the side chain size and conformational similarity to that of
 phenylalanine, since tyrosine, leucine and methionine have the
 highest production rate in the AA supplement studies, whereas
 the smaller alanine and isoleucine analogs are produced at the
 lower rate. Lastly, histidine was the only non-hydrophobic AA to
 be incorporated into the asperphenamate backbone, something
 that can be attributed to its similarity to the other aromatic AAs,
 thereby likely interacting via similar π - π interactions.

Although the relative production upon tyrosine supplement
 decreased in comparison to the control, it might be attributed
 to tyrosine being preferentially taken up by other pathways,
 such as di- or tetra-peptide biosynthesis. Nevertheless, upon
 growth on non-supplemented media, it was observed that
 tyrosine incorporation in general was preferred over any other
 AA incorporation. Subsequent supplement study with other
 synthetic *para*-substituted phenylalanines confirmed previous
 observation, with all four selected substrates being incorporated
 into the asperphenamate backbone irrespective of the size of the
para-moiety. In comparison to proteogenic AAs being mainly
 incorporated into the non-reduced part of asperphenamate
 backbone, *para*-substituted AAs were readily incorporated into
 either or both the reduced and the non-reduced part of
 the molecule. Additionally, no other pathway intermediates
 rather than N-benzoylphenylalanine, N-benzoylphenylalaninol
 and respective *para*-substituted phenylalanine analogs were
 observed. This suggests, that only intermediates with the highest
 similarity to phenylalanine intermediates can be recognized and
 released from the asperphenamate biosynthetic machinery.

The production of nicotinic acid containing analogs,
 asperphenidines, was strongly correlated with production
 of the corresponding asperphenamates upon growth on
 complex media. However, upon proteogenic AA supplement
 study, asperphenidines F1-F2 and Y1-Y2 were produced
 in higher amounts when tryptophan was supplemented,
 and in case of asperphenidines F1-F2 also in anthranilic
 acid supplemented media. This suggest, that biosynthesis
 of asperphenamate directly intercepts primary metabolism,

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1255 since both anthranilic and nicotinic acids are intermediates of
 1256 tryptophan catabolism in nicotinamide adenine dinucleotide
 1257 (NAD) biosynthesis (Foster and Moat, 1980). This hypothesis
 1258 can be further substantiated by the fact that no other organic
 1259 and inorganic nitrogen source resulted in similar nicotinic
 1260 acid production and incorporation response. However, no
 1261 incorporation of directly supplemented benzoic acid derivatives,
 1262 namely 4-hydroxybenzoic and anthranilic acids, was observed.
 1263 This indicated, that incorporation of nicotinic acid is most
 1264 likely driven by the availability of the substrate, rather than
 1265 promiscuity of either of NRPS domains, since there are only a
 1266 minute structural difference among benzoic and nicotinic acids,
 1267 as well as no clear discrepancies among preference of nicotinic
 1268 acid over benzoic acid by either of the NRPS domains.

1269 In general natural product biosynthesis of small peptides
 1270 involves a very strict uptake of AAs controlled by the NRPS
 1271 adenylation domains leading to a conserved sequence of AAs
 1272 present in the final product (Fischbach and Walsh, 2006).
 1273 However, certain cyanobacteria have been reported to possess
 1274 adenylation domains capable of activation of two or more
 1275 chemically distant AA (Kaljunen et al., 2015; Meyer et al., 2016).
 1276 In contrast, our study revealed an unusually high flexibility,
 1277 rather than specificity of fungal adenylation domain toward the
 1278 uptake of structurally related natural AAs, as well as synthetic
 1279 *para*-substituted phenylalanine analogs. Such unusual NRPS
 1280 flexibility is rather uncommon, with only one recent similar case
 1281 observed in filamentous fungi (Hai et al., 2020). Recently, the
 1282 Tang lab demonstrated that the hybrid NRPS-NRPKS involved in
 1283 biosynthesis of α -pyrones in *Aspergillus niger* is also promiscuous
 1284 toward the uptake of tyrosine, leucine and a number of *para*-
 1285 substituted phenylalanines with small substitution groups (Hai
 1286 et al., 2020). However, with the higher variety of natural AA being
 1287 tolerated in asperphenamate biosynthesis, our results altogether
 1288 suggest an even more relaxed substrate specificity in comparison
 1289 to that of α -pyrone biosynthesis.

1290 Interestingly, three other related species from
 1291 section *Brevicompecta*, *P. olsonii*, *P. bialowiezense*, and
 1292 *P. brevicompactum*, were also found to be producers of the
 1293 same analogs as observed in *P. astrolabium* when grown
 1294 on complex media. In addition, detection of several other
 1295 asperphenamates, such as a 4-hydroxybenzoic acid containing
 1296 analog (5), indicates an even more relaxed substrate specificity
 1297 in comparison to that of *P. astrolabium*. Nonetheless, it might
 1298 be speculated that the aforementioned analogs are not observed
 1299 in *P. astrolabium* due to a lower growth rate in comparison
 1300 to the other three strains (Serra and Peterson, 2007; Perrone
 1301 et al., 2015). Moreover, the presence of two Asperphenamate
 1302 *W* stereoisomers in *P. brevicompactum* suggest the presence
 1303 of a biosynthetically unrelated enzymatic activity responsible
 1304 for epimerization of tryptophan. Similar enzymatic activity was
 1305 previously characterized in a single-module NRPS responsible
 1306 for specific stereoconversion of L-tryptophan to D-tryptophan in
 1307 *A. niger* (Hai et al., 2019).

1308 Asperphenamates F, Y, L, and W, as well as asperphenidine
 1309 F1 were tested against five cancer cell lines. Although
 1310 asperphenamate L did not exhibit activity against any of the
 1311 cell lines, the four other compounds revealed moderate activities

1312 against breast, skin, liver or pancreas cell lines. In particular,
 1313 asperphenamate Y was the only active compound against the
 1314 breast cell line, suggesting that the presence of tyrosine at the
 1315 non-reduced AA moiety might be essential for the observed
 1316 activity. Therefore, further investigations of asperphenamates
 1317 harboring a *para*-substituted phenylalanine could be of interest
 1318 for future cytotoxicity studies. Moreover, asperphenidine F1, was
 1319 the only active candidate against the pancreas cell line, suggesting
 1320 the nicotinic acid analogs being more active than the benzoic acid
 1321 analogs. Although our cytotoxicity results for asperphenamates F
 1322 and Y, as well as asperphenidine F1 are comparable to previously
 1323 published data, the natural analogs, none of them show improved
 1324 bioactivity compared to synthetic asperphenamate derivatives (Li
 1325 et al., 2012; Yuan et al., 2012, 2018, 2019, 2020; Liu et al., 2016,
 1326 2018).

1327 In conclusion, HRMS/MS based analysis and the use
 1328 of a targeted media supplement approach demonstrated
 1329 an extraordinary relaxed substrate specificity in the double
 1330 NRPS system responsible for asperphenamate production.
 1331 The proteogenic and non-proteogenic *para*-substituted
 1332 L-phenylalanine analog supplements led to biosynthesis of
 1333 22 new analogs, all of which could readily be characterized by
 1334 HRMS/MS. Here we proposed a standardized naming system for
 1335 asperphenamate and asperphenidine analogs denoting specific
 1336 amino acid incorporation. This strategy illustrates the potential
 1337 for future combinatorial biosynthesis of asperphenamate and
 1338 similar small NRPS products.

1340 DATA AVAILABILITY STATEMENT

1341 The original contributions presented in the study are included
 1342 in the article/**Supplementary Materials**, further inquiries can be
 1343 directed to the corresponding author/s. Q7

1347 AUTHOR CONTRIBUTIONS

1348 KS and TL designed the experiments. KS and JF performed
 1349 HRMS/MS library search. KS, XW, and FN performed
 1350 purification and structure elucidation, with the assistance
 1351 by CG. KS performed asperphenamate design study and
 1352 subsequent HRMS/MS the data analysis compounds, assisted by
 1353 TI. OG, FV, CR, and TM designed and performed the bioassay.
 1354 KS wrote the manuscript with contribution from all authors. Q8

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1363 SUPPLEMENTARY MATERIAL

1364 The Supplementary Material for this article can be found online
 1365 at: [https://www.frontiersin.org/articles/10.3389/fmicb.2020.
 1366 618730/full#supplementary-material](https://www.frontiersin.org/articles/10.3389/fmicb.2020.618730/full#supplementary-material) Q10

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	Conflict of Interest: The authors declare that the research was conducted in the absence of any commercial or financial relationships that could be construed as a potential conflict of interest.	
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