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## Vanadium in Al-ore (bauxite) from mines of central Greece

P.N. GAMALETSOS<sup>1,2\*</sup>, A. GODELITSAS<sup>2</sup>, T. KASAMA<sup>1</sup>,  
E.M. FIORDALISO<sup>1</sup>, J. GÖTTLICHER<sup>3</sup>, R. STEININGER<sup>3</sup>

<sup>1</sup>Center for Electron Nanoscopy, Technical University of Denmark, Denmark (\*correspondence: plagka@dtu.dk)

<sup>2</sup>School of Science, N&K University of Athens, Greece

<sup>3</sup>ANKA Synchrotron Radiation Facility, Karlsruhe Institute of Technology, Eggenstein-Leopoldshafen, Germany

Vanadium, either as critical metal extracted as by-product of other metal ores (e.g., [1]) or as potential hazardous element due to its involvement in red mud (RM) accident in Hungary (e.g., [2]), is an element of increased importance in aluminum mining and metallurgical industry. Recently, seawater-leaching experiments indicated significant V release from Greek RM and negligible for its parent Al-ore (bauxite) [3]. Acetic acid-leaching tests with the RM also indicated V release, as in the case of seawater [3]. While there is much debate on V in Hungarian, Greek and Australian RM [2-4], there are no advances concerning its partitioning and speciation in the mineral phases of bauxite. Commercial bauxite samples from mines of central Greece have been investigated -in bulk- by PXRD (incl. the clay fraction of 2-0.2 $\mu$ m) and ICP-MS, as well as, at the micro-/nano-scale using electron microscopy (SEM-WDS and TEM/HRTEM, after FIB-SEM -in selected areas previously checked by WDS- and Ar ion milling for microstructure observation) along with Synchrotron radiation (SR  $\mu$ -XRF and  $\mu$ -XAFS).

The measurements showed that V is mainly intercorrelated to Si, Ca and Mg (not to Al, Ti, Fe) in micro-areas between pisoliths, following the Si/Al ratio. The hosting phase is most likely a kaolinite-type clay (V up to 1400 ppm), and less probably boehmite. It predominantly occurs as V<sup>5+</sup> (although minor contribution of V<sup>4+</sup> cannot be excluded), indicating the existence of adsorbed V-anions, belonging to the series of metavanadate (VO<sub>3</sub><sup>-</sup>) to decavanadate (V<sub>10</sub>O<sub>28</sub><sup>6-</sup>), rather than V-cations of lower oxidation state (V<sup>4+</sup>/vanadyl/oxovanadium or even V<sup>3+</sup>) substituting Al<sup>3+</sup> into the octahedral sheet of kaolinite-type structure [5,6]. It is therefore demonstrated that V is not related to Fe- and Fe-Ti-oxide minerals, despite their abundance in Al-ores, in contrast to its behaviour in other common basic metal ores [6].

[1] Mudd *et al.*, (2016 – in press) *Ore Geol. Rev.*; [2] Burke *et al.*, (2013) *Environ. Sci. Technol.*, **47**, 6527; [3] Gamaletsos *et al.*, (2016) *Sci. Rep.*, **6** (21737), 1; [4] Gräfe *et al.*, (2011) *J. Environ. Quality*, **40**, 767; [5] Gehring *et al.*, (1993) *Clays Clay Mineral.*, **41**(6), 662; [6] Huang *et al.*, (2015) *Chem. Geol.*, **417**, 68.