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# **Supporting Information**

# Modelling the Critical and Phase Equilibrium Properties of Pure Fluids and Mixtures with the Crossover Cubic-Plus-Association Equation of State

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In this work, the critical point calculations using the crossover *CPA* EoS were done by applying Michelsen's<sup>1</sup> modification of Heideman and Khalil's algorithm.<sup>2</sup> The method is used to solve the stability and criticality conditions given by:

$$\left(\frac{\partial^2 A}{\partial n_i \partial n_j}\right)_{T, V, n_{i \neq i, j}} = 0 \tag{1}$$

$$\left(\frac{\partial^3 A}{\partial n_i \partial n_j \partial n_k}\right)_{T,V,n_{i \neq i,j,k}} = 0 \tag{2}$$

The algorithm is schematically represented by Figure S1. The efficient computation of the stability condition is based on the approach that, for a point to lie on the limit of stability, the matrix Q with elements  $\left(\frac{\partial^2 A}{\partial n_i \partial n_j}\right)$  should have a zero determinant:

$$det(Q) = 0 (3)$$

If this condition is not respected, the temperature is changed until convergence is reached. Besides, with the change of *T* it

is required to re-evaluate  $f_n$ , the free energy density. If the previous condition holds, there should be a vector  $\Delta n$  that satisfy the expression:

$$Q \cdot \Delta n = 0 \tag{4}$$

and if a non-zero vector  $\Delta n$  is found, the criticality condition has to be considered. Michelsen<sup>1</sup> proposed the following condition to replace eq 2:

$$C = \Delta n \cdot \left[ \frac{\partial q_{ij}(n + s\Delta n)}{\partial s} \right] \cdot \Delta n = 0$$
 (5)

where  $\frac{\partial q_{ij}(n+s\Delta n)}{\partial s}$  is the derivative of  $q_{ij}$  with respect to the directional increment s. Consequently, if eq 5 is greater than a predetermined tolerance, the volume of the system is changed and the free energy density  $(f_n)$  needs to be re-evaluated. The critical point is reached at a certain T and v when both det(Q) and C are close to zero. The critical pressure is calculated using the crossover  $CPA\ EoS$ .

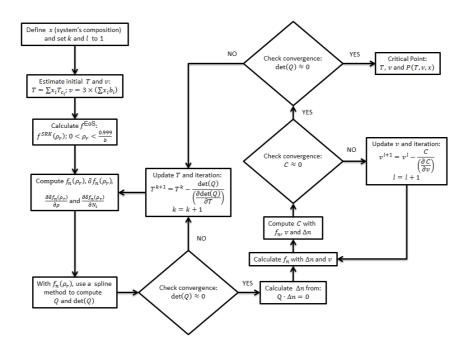


Figure S1: Flowchart for calculating the critical point of a system using a crossover EoS.

### References

- (1) Michlsen, M. L. Calculation of Phase Envelopes and Critical Points for Multicomponent Mixtures. *Fluid Phase Equilib.*, **1980**, 4, 1-10
- (2) Heidemann, R. A.; Khalil, A. M. The Calculation of Critical Poins. *AIChE Journal*, **1980**, 26, 769-779