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1 Pulmonary toxicity of silver vapours, nanoparticles and fine
2 dusts: A review

3
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13 subchronic, chronic, inhalation, intratracheal instillation, argyria, occupational, dental, aerosol, genotoxicity.

14

15

16

17 **Abstract**

18 Silver is used in a wide range of products, and during their production and use, humans may be exposed
19 through inhalation. Therefore, it is critical to know the concentration levels at which adverse effects may
20 occur. In rodents, inhalation of silver nanoparticles has resulted in increased silver in the lungs, lymph nodes,
21 liver, kidney, spleen, ovaries, and testes. Reported excretion pathways of pulmonary silver are urinary and
22 faecal excretion. Acute effects in humans of the inhalation of silver include lung failure that involved
23 increased heart rate and decreased arterial blood oxygen pressure. Argyria—a blue-grey discoloration of skin
24 due to deposited silver—was observed after pulmonary exposure in 3 individuals; however, the presence of
25 silver in the discolorations was not tested. Argyria after inhalation seems to be less likely than after oral or
26 dermal exposure. Repeated inhalation findings in rodents have shown effects on lung function, pulmonary
27 inflammation, bile duct hyperplasia, and genotoxicity. In our evaluation, the range of NOAEC values was
28 0.11 to 0.75 mg/m³. Silver in the ionic form is likely more toxic than in the nanoparticle form but that
29 difference could reflect their different biokinetics. However, silver nanoparticles and ions have a similar
30 pattern of toxicity, probably reflecting that the effect of silver nanoparticles is primarily mediated by released
31 ions. Concerning genotoxicity studies, we evaluated silver to be positive based on studies in mammalian
32 cells *in vitro* and *in vivo* when considering various exposure routes. Carcinogenicity data are absent;
33 therefore, no conclusion can be provided on this endpoint.

34

35

36 1. Introduction

37 Occupational inhalation exposure to silver potentially occurs in industries that engage in the following
38 activities: manufacturing and using nanomaterials (Lee et al., 2011), silver smelting and refining and
39 preparation of silver salts (DiVincenzo et al., 1985), silver soldering (Kachru et al., 1989; Vance, 1960),
40 silver brazing (Gan et al., 1995; Mangold and Beckett, 1971), and recovery of silver from recycling sources
41 (Pifer et al., 1989). Another source of silver dusts is the trimming (grinding) of amalgam dies in dental
42 laboratories (Brune and Beltesbrekke, 1979). The occupation of silversmith has also been linked to silver
43 exposure (Aktepe et al., 2015), and silver nanoparticle-containing spray disinfectants are another source of
44 occupational and consumer exposure (Quadros and Marr, 2011; Rogers et al., 2018).

45 Physical forms of silver relevant to inhalation are ions, particles, flakes, and fibres; the latter 3 includes
46 nanoparticles (up to 100 nm in diameter) and larger particles, nanobelts, and wires. Ions—when present as
47 salts—may act as particle entities when brought into an aerosol in the workplace. Respirable sizes of
48 particles and fibres are up to 5 μm in diameter; thus, powders below this size are most important in the
49 context of alveolar endpoints. However, materials of a larger size can exert toxicity in the upper airways if
50 deposited there, or at sites of deposition upon removal from the upper airways, for example, the
51 gastrointestinal tract. In addition, the potential transfer of silver through the nose to deposit in the brain must
52 be considered.

53 We have published reviews on oral (Hadrup and Lam, 2014) and skin (Hadrup et al., 2018) toxicity of
54 silver exposure. In this work, we review the adverse effects of silver after pulmonary exposures. We have
55 evaluated studies describing the biokinetics (absorption-distribution-metabolism-excretion; ADME) and
56 toxicity, including genotoxicity and carcinogenicity. Concerning genotoxicity, we evaluated all exposure
57 pathways, and *in vitro* studies because genotoxicity and carcinogenicity are mostly incited as mechanisms
58 inside single cells irrespective of exposure pathway. Notably, the section on genotoxicity and carcinogenicity
59 is an update based on our recent published evaluation (Hadrup et al., 2018). Overall, the objective of this
60 work was to review the kinetics and adverse effects of silver after pulmonary exposure.

61

62 2. Methods

63 Concerning the literature search strategy, we identified all relevant articles in the PubMed database (Pubmed,
64 2020) by using combinations of the following search terms: ‘silver’, ‘toxicity’, ‘pulmonary’, ‘inhalation’,
65 ‘genotoxicity’, and ‘carcinogenicity’. This search yielded 485 articles. We also evaluated reference lists of
66 the retrieved articles to identify possible additional references missed in the aforementioned searches. The
67 inclusion criteria were that the articles included data on kinetics (absorption, distribution, metabolism and
68 excretion), or toxicity after pulmonary exposure of any form of silver. As aforementioned, for genotoxicity
69 and carcinogenicity: all exposure pathways as well as *in vitro* studies were included. Article languages
70 considered were English, German, Dutch, and French. In total, upon an initial screening of abstracts, we
71 evaluated 102 studies; 74 of those studies were included in this article.

72

73 3. Absorption, distribution, metabolism and excretion (ADME)

74 3.1. Absorption and distribution – following pulmonary exposure

75 *Levels of silver in the general population*

76 In 26 individuals living in the Melbourne metropolitan area and having no known occupational exposure to
77 silver, the blood levels were less than 1 µg silver/L (Wan et al., 1991). No silver was detected in the blood of
78 control populations with a detection limit of 6 µg silver/L (Rosenman et al., 1979) or 5 µg silver/L
79 (DiVincenzo et al., 1985).

80

81 *Levels of silver in workers who worked with metals*

82 Levels of silver in the body of workers reflect absorption from inhalation and from other pathways, for
83 example, through mucosal surfaces. Few studies provided estimates of exposure: Two men involved in the
84 manufacturing of silver nanomaterials were each exposed to 0.00035 and 0.00135 mg silver/m³, resulting in
85 blood levels of 0.34 and 0.30 µg silver/L, respectively (Lee et al., 2012). Two men worked with the recovery
86 of silver from X-rays and photographic films. One 42-year-old man was exposed to 0.085 mg silver/m³
87 (incineration area) and 1 mg/m³ (pulverising area) and had a blood level of 49 µg silver/L. The other, one 51-
88 year old man, with an estimated exposure between 0.03 and 0.17 mg silver/m³, had a blood level of 74 µg
89 silver/L (Williams and Gardner, 1995). The same 51-year-old man was monitored for 5 years, and over those
90 5 years, his initial silver of 74 µg/L decreased to approximately 11 µg/L (Williams, 1999).

91 Concerning studies in which exposure levels were not estimated: In 30 workers exposed to silver
92 nitrate or silver oxide, 12 had a measurable blood level of silver (>6 µg/L) (median level: 19.5 µg/L; range:
93 11 to 84 µg/L). In 25 workers manufacturing silver powders, the mean blood value was 10 µg silver/L (range
94 0.5 to 62) (Rosenman et al., 1987). In 21 silver reclamation workers, the mean level was also 10 µg silver/L
95 (Pifer et al., 1989). In another study, 98 workers had blood levels of silver between 0.1 and 23 µg/L, silver
96 reclaimers had on average 6.8 µg/L, workers refining silver for bullion coins had 2.5 µg/L, and workers in
97 jewellery production had 1.2 µg/L (Armitage et al., 1996). In 37 American workers involved in processes
98 such as smelting and refining silver and manufacturing silver salts, the mean blood level was 11 µg silver/L
99 (DiVincenzo et al., 1985). A 27-year-old man occupationally exposed to aerosolised silver had a serum level
100 of 154.4 µg silver/L (Cho et al., 2008). In an older study on silver finishers, they had silver in the lung, liver,
101 heart, and spleen (Barrie and Harding, 1947).

102

103 *Studies with rodent inhalation of silver nanoparticles*

104 Most reports on the distribution of silver after pulmonary exposure to nanoparticles are on the localisation of
105 elemental silver in organs; however, some studies have described the localisation of particles in the body.
106 The question is whether these reports reflect that silver is distributed as particles or whether this is caused by

107 dissolved silver nanoparticles that have reformed into secondary particles in the tissues. The latter has been
108 demonstrated after oral exposure e.g. in (Aaseth et al., 1981; Juling et al., 2016; Loeschner et al., 2011).

109 Distribution of silver (measured as elemental silver) after pulmonary exposure occurs in several
110 organs. In 1 study, inhalation of silver nanoparticles (14–15 nm) at 0.05, 0.12, and 0.38 mg/m³ (6 h/day, 5
111 day/week for 12 weeks) resulted—at all dose levels—in elevated silver in rat lung, whereas dose-dependent
112 elevations occurred in the liver, kidneys, spleen, testes, blood, and eye, but not the brain. The levels
113 decreased over a recovery period, but silver remained present in the liver, spleen, and eyes at the highest
114 dose at 12 weeks post exposure (Song et al., 2013). In another study, rats were exposed to silver
115 nanoparticles (18–19 nm, at 0.049, 0.133 or 0.515 mg/m³ 6 h/day, 5 days/week for 13 weeks); and silver
116 increased in the lungs liver, kidneys, brain, and olfactory bulb. Blood levels were increased to 0.7 (low dose),
117 1.8 (mid dose), and 4.3 µg silver/kg tissue wet weight (high dose) compared with 0.09 µg in controls (Sung
118 et al., 2009). In rats exposed to 15 nm silver nanoparticles (6 h at 0.133 mg/m³), immediately after exposure,
119 silver was detected in substantial amounts in the lung, nasal cavities, lung-associated lymph nodes, and
120 blood; and at low levels in the liver, kidney, spleen, brain, and heart. In the kidney, heart, and lymph nodes,
121 silver was observed 1 day later. The amount of silver in these organs normalised over a recovery period of 7
122 days (Takenaka et al., 2001). Rats were exposed to silver nanoparticles (18.1–19.6 nm) at mass
123 concentrations of 0.031, 0.082, or 0.116 ,g/m³, 6 hours/day, 5 days/week for 4 weeks. Silver levels in lungs
124 after 1 day of recovery of 14.7, 6.4 and 1.6 were initially cleared with half-lives of 2 to 4 days, followed by
125 a slow clearance phase with half-lives of 60 to 100 days (Jo et al., 2020). Other studies in rodents
126 demonstrated deposition in the lungs and liver (Braakhuis et al., 2016), heart, spleen, and testes (Kwon et al.,
127 2012).

128

129 *Other pulmonary exposure pathways*

130 A sufficient number of inhalation studies have directly represented the occupational pulmonary exposure
131 pathway and describe the absorption and distribution of silver nanoparticles. Nevertheless, to describe the

132 differential distribution of silver ions and nanoparticles, and on nanowires/nanofibers, we also included
133 aspiration and intratracheal instillation studies. Concerning the nanowire form of silver, silver nanowires 2 or
134 20 μm in length, dosed by intratracheal instillation, were in the lung, localised and enclosed in granulomas or
135 surrounded by only a few macrophages in the terminal bronchiole-alveolar duct junction. No further organs
136 were investigated (Silva et al., 2014).

137 Concerning the comparison of nanoparticles and ionic silver, mice were administered 20 or 110 nm
138 silver nanoparticles by oropharyngeal aspiration of a single bolus of 0.1, 0.5, or 1 mg/kg body weight (bw);
139 or to 1 mg/kg bw silver ions in the form of AgNO_3 . After 40 hours of recovery, the deposition of silver in
140 lung tissue was higher with exposure to particles compared with an equal mass amount of ionic silver. By
141 contrast, the amount of silver in the liver increased only after the ionic silver. At 21 days of recovery, silver
142 remained present in the lungs of particle-administered animals, whereas it was absent in the animals dosed
143 with ionic silver (Wang et al., 2014). A similar lung/liver deposition pattern was observed after the
144 intratracheal instillation of 20 nm silver nanoparticles (10.3 μg silver/mouse \sim 0.5 mg/kg bw) or ionic silver
145 in the form of AgNO_3 (7.5 μg silver/mouse \sim 0.38 mg/kg bw) (Arai et al., 2015). The biodistribution of silver
146 after the dosage with agglomerated 4–10 nm silver nanoparticles (agglomerated particles $>$ 100 nm
147 dominant, but also ultrafine particles present) was compared with that of ionic silver (AgNO_3) after
148 intratracheal instillation (50 μg of agglomerated silver nanoparticles/rat, or 7 μg AgNO_3 / rat equal to 4.4 μg
149 silver/rat). The amount of silver deposited in the lungs seemed to be more prolonged after the dosage with
150 nanoparticles compared with a more rapid clearance from this organ when silver was dosed as ions
151 (Takenaka et al., 2001). Taken together, the data from these 3 studies suggest that ionic silver is rapidly
152 distributed to the liver, kidney, and spleen, whereas silver in nanoparticles may remain in the lungs for a
153 longer duration, perhaps reflecting a slow release of silver ions from the surface.

154

155 3.2. Metabolism of silver formulations following pulmonary exposure

156 Oral and dermal exposure to silver, in ionic or nanoparticle form, have been demonstrated to lead to the
157 formation of particles within various human and animal tissues—particles that, in addition to silver, contain
158 other elements such as selenium and sulphur (Aaseth et al., 1981; Hadrup et al., 2018; Hadrup and Lam,
159 2014; Loeschner et al., 2011). However, evidence from pulmonary studies has been scarce and inconclusive.
160 The presence of sulphur in the aforementioned newly formed particles has been explained by the high
161 affinity of silver and sulphur, which could result in a combination of silver ions with biological structures
162 and constituents with high sulphur content. A similar silver and sulphur complex has also been observed in
163 the lungs after a pulmonary exposure by using micro X-ray absorption near edge structure (μ XANES). Mice
164 were exposed to a heterogeneous composition of 25 nm spherical and 80–90 nm rod-shaped silver
165 nanoparticles (100 μ g/mouse or 4 mg/kg bw) by a single oropharyngeal aspiration. In lungs, silver
166 nanoparticles were mostly present in macrophages, either partially or totally dissolved and chelated by thiol-
167 containing ligands such as cysteine, glutathione, or metallothionein (Smulders et al., 2015). However, the
168 formation of new nanoparticles was not described.

169 Pregnant and non-pregnant mice were exposed to 18–20 nm silver nanoparticles at 0.54 mg/m³ for 1 to 4
170 hours. This exposure resulted in the location of silver-containing nanoparticles in the lungs, spleen, liver and
171 placenta, and the head region of the foetus. In the foetus, the detected silver was almost entirely in the ionic
172 form or as nanoparticles <13 nm. In the placenta, silver nanoparticles were approximately 6% of the total
173 silver, and in the liver and spleen, the percentage was approximately 12% and 14%, respectively. By
174 contrast, 21% of the total silver was still in the nanoparticulate form in the lung at the time of the
175 measurements. Thus, a conclusion was that the silver translocating from the lung was mainly ionic or in the
176 form of small, readily dissolving nanoparticles. This case was also observed for silver translocating through
177 barriers, for example, through the placenta and to the foetus. The authors could not rule out the possibility
178 that some of the nanoparticles identified in tissues were not the original nanoparticles that translocated the
179 lung barrier but instead particles newly formed after interaction with sulphur groups of proteins, and/or
180 selenium or chloride present in tissues (Campagnolo et al., 2017).

181 Rats inhaled 15 nm silver nanoparticles (0.179 mg/m^3) 6 hours per day for 4 consecutive days.
182 Electron-dense structures were observed in the vesicles and nuclei of lung cells on the first day of recovery
183 after employing the silver enhancement technique. The authors did not attempt to prove the presence of
184 silver within the particles. From the size of the silver-enhanced structures (15-20 nm), the authors concluded
185 that the original size of the silver nanoparticles must have been less than 5 nm in diameter. This size was
186 interpreted as an indication of partial dissolution of the nanoparticles (Braakhuis et al., 2014). Davidson and
187 colleagues exposed rats to 20 or 110 nm silver nanoparticles for 6 hours by inhalation. Metallic silver was
188 identified by X-ray absorption spectroscopy as the dominating silver species in the lung throughout a 7-day
189 period. The method could not exclude the existence of small amounts of silver atoms or ions. The results
190 further indicated that the original nanoparticles were transformed to other forms of metallic silver
191 nanomaterials, namely, much smaller silver nanoparticles or a highly porous, zeolite-like nanomaterial, 7
192 days post exposure. Dissolution of the inhaled silver nanoparticles was considered the most plausible
193 explanation for this observation (Davidson et al., 2015).

194 3.3. Excretion – following pulmonary exposure

195 *Urinary and faecal excretion of silver* Excretion of silver has been described to occur through urine and
196 faeces after oral, dermal, and mucosal surface exposure (Hadrup et al., 2018; Hadrup and Lam, 2014; Skare
197 and Engqvist, 1994). Concerning pulmonary exposure, many human reports of urinary excretion of silver
198 have been provided. Silver was elevated above normal ($1.9 \text{ }\mu\text{g/L}$) in urine in 26 of 27 workers involved in
199 the manufacturing of precious metal powders. The mean value was $11.3 \text{ }\mu\text{g/L}$ and the range was 0.5 to 52.0
200 $\mu\text{g/L}$ (Rosenman et al., 1987). In a man occupationally exposed to aerosolised silver, the urinary level was
201 $243.2 \text{ }\mu\text{g silver/L}$ (Cho et al., 2008). A man who had for 7 years been working with the manufacturing of
202 silver nanomaterials had a urine level of $0.43 \text{ }\mu\text{g/L}$ (Lee et al., 2012). However, other studies have
203 investigated the faecal pathway. Silver was detected in urine in only 1 of 27 silver reclamation employees
204 ($\geq 5 \text{ }\mu\text{g/L}$ urine), whereas silver was detected in all faecal samples ($n=18$) with a mean value of $16.8 \text{ }\mu\text{g/g}$. In
205 controls, the mean value in faeces was $1.5 \text{ }\mu\text{g/g}$ (Pifer et al., 1989). In 37 workers exposed to silver and

206 having a mean blood level of 11 $\mu\text{g/L}$, urine content was reported to be less than 0.005 $\mu\text{g silver/g}$, whereas
207 the faecal content was 15 $\mu\text{g silver/g}$. In controls, the faecal level was 1.5 $\mu\text{g/g}$ (DiVincenzo et al., 1985).
208 These data support that both the urinary and faecal pathways are pathways of excretion in humans; of these
209 the faecal one seems to be most pronounced.

210 *Timeframe for the elimination of silver* Rats inhaled 20 or 110 nm silver nanoparticles for 6 hours at 7.2
211 or 5.4 mg/m^3 , respectively. Although there was a reduction in silver in the lungs over a 56-day recovery
212 period, 33% of the delivered dose was still present at this time point (both particle sizes) (Anderson et al.,
213 2015a). Mice inhaled 5 nm silver nanoparticles at 3.3 mg/m^3 for 4 hours/day for 10 days. Silver in the lungs
214 was found to be 31 $\mu\text{g/g lung}$ (median value; dry weight) immediately after exposure, decreased to 10 $\mu\text{g/g}$
215 after a 3-week recovery period (Stebounova et al., 2011). Song *et al.* found that after the inhalation of silver
216 nanoparticles (0.05, 0.12, or 0.38 mg/m^3 , 6 h/day, 5 days/week for 12 weeks) the amount of silver in rat
217 lungs decreased over a 12-week recovery period; but was still different from controls at this time point (e.g.
218 following 12 weeks of inhalation at the low dose, the level in lungs was $\sim 100 \mu\text{g/kg}$ decreasing to $\sim 5 \mu\text{g/kg}$
219 after 12 weeks of recovery). Additionally, silver was still present in the liver, spleen, and eyes (in these
220 organs $< 1 \mu\text{g/kg}$) after the 12-week recovery period (Song et al., 2013). Studies in rodents that used
221 intratracheal instillation or aspiration have also demonstrated that after the dosage of silver nanoparticles,
222 silver is only slowly eliminated from the lungs (Anderson et al., 2015b; Smulders et al., 2015; Takenaka et
223 al., 2001; Wang et al., 2014). By contrast, after the dosage of ionic silver, a more rapid elimination occurs in
224 the lungs, accompanied by distribution to the liver (Arai et al., 2015; Wang et al., 2014).

225

226 4. Toxicity observed in humans

227 4.1. Acute toxicity

228 A 27-year-old man, whose job was to melt silver ingots, accidentally inhaled massive silver-containing
229 vapours. Fourteen hours later, he developed a headache, shortness of breath, and difficulty breathing

230 (moderate dyspnoea). As the dyspnoea intensified, he was hospitalised. His breathing rate, heart rate, and
231 arterial pressure increased, and the oxygen pressure of the capillary blood was depressed. He had markedly
232 livid lips (dark bluish in colour, perhaps reflecting deposition of silver), numerous crackles in the lungs,
233 leucocytosis, and increased transaminase—the latter is an indicator of liver damage. Despite the treatment
234 with oxygen and pharmacological agents, the respiratory inefficiency intensified to include cyanosis (low
235 oxygen saturation leading to bluish colour). There was no verbal contact, and he had narrow, even pupils. He
236 was given artificial ventilation; X-ray photographs suggested *shock lung* (lung failure). He was given further
237 pharmacological treatment, and the next day, he exhibited improvement. On the 14th day of admission, he no
238 longer needed artificial ventilation, and afterwards, complete recovery was observed (Forycki et al., 1983).

239

240 4.2. Chronic toxicity

241 Argyria is a blue-grey discoloration of the skin (and other organs) due to the deposition of silver-containing
242 granules. Argyria can be localised at the exposed area or be generalised covering larger parts of the body—in
243 particular, the parts exposed to sunlight. Argyria has foremost been demonstrated after oral and dermal
244 exposure to silver (Hadrup et al., 2018; Hadrup and Lam, 2014). Notably, some studies have suggested the
245 occurrence of argyria also following pulmonary exposure; however, without testing for silver in the
246 discolorations: A 27-year-old man was exposed to an aerosol reportedly containing silver, alcohol, and
247 acetone. A blue-grey discoloration of the face, eye (sclerae and conjunctiva), and oral mucosa had
248 progressed over 4 months. In a biopsy specimen from the face, granules—proposed to contain silver—were
249 observed in the epidermal basal layer. Granules were also found located in the basement membrane zone of
250 sweat glands. His serum level of silver was 154.4 µg/L (Cho et al., 2008). Of 30 workers exposed to
251 silver nitrate and silver oxide, 6 were reported to have argyria. Based on personal air sampling, time-
252 weighed-average exposure was between 0.039 and 0.378 mg silver/m³. In addition to argyria, the workers
253 reported other symptoms, for example, the majority reported eye and upper and lower respiratory tract
254 irritation. The presence of silver in the blood was found to be associated with complaints of abdominal pain

255 (Rosenman et al., 1979). A male silver-cleaner aged 37 years was diagnosed with argyria localised to the
256 nasopharyngeal mucosa with sub-epithelial accumulation of black granules (Ferrara et al., 2018).

257 The presence of silver in granules was demonstrated in 1 study but only in the lungs: A 63-year-old man
258 who had worked as a silver finisher for almost his whole working life died of a blood clot in the heart. The
259 pleura covering the right lung was described as being blue-black with a few small denser grey spots. Small
260 areas of fibrosis contained high amounts of pigment; moreover, this was observed in some of the alveoli and
261 in non-fibrotic perivascular aggregates. The pigment contained iron and silver, as demonstrated by chemical
262 analysis of ash. However, the silver content was low (0.036%) compared with iron (3.5%); thus, the
263 discoloration could be due to iron and not silver. In a comparison of the samples, the controls contained iron
264 but no silver. Other signs of toxicity were lung emphysema and a fibrous thickening over 1 lung (Harding,
265 1948). Barrie and Harding reported 4 cases in which silver finishers had excess silver and iron deposition in
266 tissues. By autopsy, all the men were observed to have emphysema and detectable levels of silver in the
267 lungs; in 1 case, detectable levels of silver were observed in the liver and spleen; in another case, detectable
268 levels of silver were observed in the heart (no normal silver level reported). Notably, the men's iron levels in
269 the lung were increased compared with normal values (Barrie and Harding, 1947).

270 Argyria may also manifest in the eyes¹. Notably, although this condition in theory may develop
271 secondary to pulmonary silver exposure, this condition more likely occurs after direct exposure of the eyes.
272 One case of ocular argyria was in a 51-year-old male who for 7 years had worked at silver refinery and had a
273 blood level of 740 µg silver/L. Over an 18-month follow up period, the blood level of silver decreased to 60
274 µg/L, but the argyria remained. Silver was not chemically proven in the eyes (Williams, 1999; Williams and
275 Gardner, 1995). Thirty workers were exposed to silver nitrate and silver oxide after working for more than 2
276 years at a plant that manufactured precious metal powders, 20 of the workers were reported to have
277 deposition of silver in the eyes (Rosenman et al., 1979).

278

¹ Ocular silver deposition is sometimes referred to as 'argyrosis' (Pifer et al., 1989).

279 4.3. Human studies in which no toxicity was observed

280 Twenty-one out of 27 silver reclamation employees had measurable (above detection limit) silver levels in
281 the blood (mean level: 10 µg/L). All 27 controls (matched on sex, age, and race) had silver levels below the
282 detection limit (5 µg/L). No cases of generalised argyria were observed based on the colour of the face and
283 electron microscopy of skin biopsies. Additionally, no general signs of adverse effects of the silver were
284 observed (Pifer et al., 1989). In 2 workers recovering silver from X-rays and photographic films, and having
285 blood levels of silver of 49 and 74 µg/L, no signs or symptoms of toxicity were observed (Williams and
286 Gardner, 1995).

287

288 5. Toxicity findings in animal studies

289 5.1. Lung function endpoints

290 Decreased tidal and minute volumes were observed in rats during inhalation of 18 nm silver nanoparticles
291 ($LOAEC_{lung\ function} 0.05\ mg/m^3$, 6h/day for 13 weeks) (Sung et al., 2008). The same endpoints and peak
292 expiratory flow were decreased in rats that inhaled 14–15 nm silver nanoparticles (6 h/day, 5 days/week for
293 12 weeks) at 3 mass concentrations: 0.05, 0.12 and 0.38 mg/m³. The effects were persistently observed
294 during exposure and over 12 weeks of recovery at all dose levels ($LOAEC_{lung\ function} 0.05\ mg/m^3$) (Song et al.,
295 2013). Inhalation of 13–16 nm silver particles increased tissue elastance in 1 of 2 rat stains ($LOAEC_{elastance}:$
296 0.6 mg/m³ for 12 h) (Seiffert et al., 2016). Increased pulmonary elastance was also observed in mice 1 day
297 after intratracheal instillation of 0.05 mg/kg bw silver nanoparticles 20 and 110 nm) (Botelho et al., 2015).
298 Additionally, following intratracheal instillation of 15 nm silver nanoparticles (0.5 mg/kg bw), a decrease in
299 tissue stiffness and resistance was accompanied by increased surfactant protein D in mice (Botelho et al.,
300 2018), and increased pulmonary resistance and decreased dynamic compliance was observed in rats (Seiffert
301 et al., 2015). Concerning negative studies, no effect on tidal volume and minute volume was observed after 4
302 hours of inhalation of silver nanoparticles (18–20 nm) at up to 0.75 mg/m³ (Sung et al., 2011).

303

304 5.2. Inflammation and cytotoxicity measured in bronchoalveolar lavage fluid

305 *Inhalation studies* Rats that inhaled silver nanoparticles of various sizes had increased neutrophil
306 numbers, total protein, and LDH in bronchoalveolar lavage (BAL) fluid at all exposure periods (34 min–6
307 h/day for 4 consecutive days, 1 dose level per particle). The following dose descriptors were observed: 18
308 nm particle: LOAEC 0.055 mg/m³; 34 nm: NOAEC: 0.041 mg/m³; 60 nm: LOAEC: 0.043 mg/m³; 160 nm:
309 NOAEC: 0.555 mg/m³ (Braakhuis et al., 2016). In the same animal species, inhalation of 15 nm silver
310 nanoparticles was compared with inhalation of 410 nm particles (0.179 mg/m³ and 0.167 mg/m³, 6 h/day for
311 4 days). Exposure to 15 nm silver nanoparticles resulted at 1 day of recovery in increased neutrophil,
312 lymphocyte, and monocyte numbers and increased cellular damage markers in the lungs. Only increased
313 BAL protein was observed in response to exposure to the 410 nm silver particle (15 nm silver
314 LOAEC_{neutrophils}: 0.18 mg/m³; 410 nm silver NOAEC_{neutrophils}: 0.17 mg/m³)(Braakhuis et al., 2014). The
315 inhalation of 13–16 nm silver nanoparticles (0.6 to 0.8 mg/m³, 3 or 12 h) resulted in increased numbers of
316 neutrophils in rat BAL fluid and changes in IL-1 β , KC, IL-17, CCL2 and CCL3, phospholipid levels,
317 surfactant protein D across doses, and time points (LOAEC_{neutrophils}: 0.62 mg/m³, 12 h) (Seiffert et al., 2016).
318 A NOAEC_{neutrophils} in BAL fluid was 0.082 mg/m³ in rats exposed to silver nanoparticles 18.1–19.6 nm in
319 diameter (6 hours/day, 5 days/week for 4 weeks) (Jo et al., 2020). Mice were exposed to 5 nm silver
320 nanoparticles (3.3 mg/m³, 4 h/day for 10 days), and increased neutrophils in BAL fluid were observed both at
321 exposure end and at 3 weeks of recovery. No effects were observed in the other endpoints (LOAEC_{neutrophils}:
322 3.3 mg/m³) (Stebounova et al., 2011).

323 Other studies have shown no effect on neutrophil numbers. Neutrophil numbers were not increased in
324 rats exposed to silver nanoparticles (0.01 or 0.1 mg/m³; 5 h). The mean aerodynamic diameters were 33 to 39
325 nm (NOAEC_{neutrophils}: 1 mg/m³) (Roberts et al., 2013). Rats were exposed to 18 nm silver nanoparticles at
326 0.05, 0.13 mg/m³, or 0.5 mg/m³. The duration was 6 hours/day 5 days/week for 90 days. There were no

327 differences in polymorphonuclear cells² in BAL (NOAEC_{polymorphonuclear cells}: 0.5 mg/m³) (Sung et al., 2008).
328 Mice were exposed to 20 nm silver nanoparticles at 2.9 mg/ for 6 hours. There were no changes in neutrophil
329 numbers in BAL fluid. Total protein in BAL increased immediately after exposure but normalised after 24
330 hours of recovery (NOAEC_{neutrophils}: 2.9 mg/m³) (Kwon et al., 2012).

331 *Intratracheal instillation or oropharyngeal aspiration studies* A substantial number of studies have
332 investigated pulmonary inflammation following inhalation of silver nanoparticles, and a less important role
333 for intratracheal instillation/aspiration studies in the hazard assessment is suggested. However, these studies
334 are crucial for the comparison between silver nanoparticles and ionic silver and nanowires/fibres, as these
335 have not been investigated by inhalation. Concerning nanowires, rats were instilled with 2 or 20 µm silver
336 nanowires (0.1, 0.5, or 1.0 mg/kg bw). After 1 day of recovery, exposure to the short wire was associated—at
337 the 2 highest doses—with increases in polymorphonuclear cells in BAL. The long wire only increased
338 inflammation and eosinophilia at the highest dose. Protein in BAL was increased at all doses and both
339 lengths at Day 1. Frustrated phagocytosis was reported for the long nanowires. After exposure to the 20 µm
340 nanowires, alveolar and bronchial inflammation was observed by histopathology (2 µm nanowires
341 NOAEL_{neutrophils}: 0.1 mg/kg bw; 20 µm nanowires NOAEL_{neutrophils}: 0.5 mg/kg bw (Silva et al., 2014).

342 Concerning intratracheal instillation studies comparing silver nanoparticles to ionic silver: Mice were
343 exposed through single oropharyngeal aspiration to silver nanoparticles 20 or 110 nm, coated with citrate or
344 polyvinylpyrrolidone (0.1, 0.5 or 1 mg/kg bw) or silver ions (AgNO₃ at 1 mg/kg bw). Exposure to the 20 nm
345 particle resulted in increased neutrophils in BAL at all doses, whereas the 110 nm particle only had an effect
346 at the highest dose. AgNO₃ also increased neutrophil numbers after 40 hours of recovery. At 21 days of
347 recovery, there was still increased neutrophil numbers in BAL at the highest dose of the 110 nm particle, but
348 not after exposure to the 20 nm particle or AgNO₃ (20 nm particle LOAEL_{neutrophils}: 0.1 mg/kg bw; 110 nm
349 particle NOAEL_{neutrophils}: 1 mg/kg bw; AgNO₃ LOAEL_{neutrophils}: 1 mg/kg bw) (Wang et al., 2014). Mice were
350 exposed to 20 nm silver nanoparticles (10.3 µg silver/mouse ~0.5 mg/kg bw) or ionic silver in the form of

² Polymorphonuclear cells (PMN) is the collective term for neutrophils, basophils, and eosinophils.

351 AgNO₃ (7.5 µg silver/mouse ~0.38 mg/kg bw) by intratracheal instillation. Exposures to both forms of silver
352 were associated with increases in neutrophil numbers at 4 and 24 hours of recovery. However, the AgNO₃
353 exposed animals had higher neutrophil numbers at 24 hours compared with silver nanoparticles. The
354 interleukin-1β concentration in BAL fluid was, by contrast, higher for silver nanoparticles than for ionic
355 silver (20 nm particle LOAEL_{neutrophils}: 0.5 mg/kg bw; AgNO₃ LOAEL_{neutrophils}: 0.38 mg silver/kg bw) (Arai et
356 al., 2015).

357 Concerning other intratracheal studies, and investigating only nanoparticles, the NOAEL/LOAEL
358 values of increased neutrophil/polymorphonuclear-cell numbers in BAL fluid are as follows. In rats,
359 NOAELs of 0.1 mg/kg bw (Seiffert et al., 2015) (Silva et al., 2015) to 0.2 mg/kg bw (Haberl et al., 2013)
360 were observed. In mice, NOAELs of 0.05 and 0.5 mg/kg bw (Botelho et al., 2018, 2015) and of 6.4 mg/kg
361 bw (Gosens et al., 2015) have been reported. LOAEL values of 0.25 and 4 mg/kg bw have been reported by
362 (Scoville et al., 2017; Smulders et al., 2015, 2014).

363 5.3. Inflammation evaluated by histology

364 Increased alveolar accumulation of macrophages and alveolar chronic inflammation were observed in rats
365 after inhalation of 14–15 nm silver nanoparticles (NOAEC: 0.12 mg/m³ 6 h/day, 5 days/week for 12 weeks)³
366 (Song et al., 2013). Abnormal inflammatory cell infiltrates, chronic alveolar inflammation, and small
367 granulomatous lesions were observed in rats exposed to silver nanoparticles (18–19 nm) (NOAEC_{histology} 0.13
368 mg/m³ for 390 h) (Sung et al., 2009). In addition, intratracheal instillation studies with silver nanoparticles
369 produced LOAEL values of 0.1 mg/kg bw (Seiffert et al., 2015) and 1 mg/kg bw (Silva et al., 2015). After
370 instillation of 20 µm silver nanowires, histological scores showed that 0.5 and 1.0 mg/kg bw produced
371 inflammation (NOAEL 0.25 mg/kg bw). Additionally, frustrated phagocytosis was observed—the
372 insufficient ingestion of long fibres into BAL fluid cells (Silva et al., 2014).

³ No statistics were reported, but we applied Fisher's exact test

373 5.4. Respiratory sensitisation and irritation

374 Eye irritation and skin sensitisation after silver exposure have been described in some human case studies
375 (Hadrup et al., 2018). However, following pulmonary exposure, we found no studies in the open peer-
376 reviewed literature describing a sensitisation or irritation caused by silver.

377 5.5. Other toxicity endpoints

378 *Liver toxicity* Bile duct hyperplasia was observed in rats exposed to silver nanoparticles (18-19 nm) at 0.5
379 mg/m³ for 13 weeks (390 h) in a study conducted under the OECD 413 guideline (NOAEC 0.133 mg/m³)
380 (Sung et al., 2009).

381 *Renal toxicity* Creatinine clearance was lower in workers involved in the manufacturing of precious
382 metal powders and exposed to silver (26 of 27 workers had elevated urinary silver concentrations). However,
383 no firm conclusion could be drawn on the role of silver because the effect of cadmium could not be excluded
384 (Rosenman et al., 1987).

385 *Cardiovascular effects* Decreased stimulation-induced dilation of the tail artery and elevated heart rate
386 were observed in rats after inhalation of 0.1 mg/m³ silver nanoparticles (5 h), but not at a higher exposure
387 level (1 mg/m³). Other pulmonary and vascular endpoints were not affected by silver (Roberts et al., 2013).
388 Concerning blood coagulation, activated partial thrombin time was decreased at all dose levels in rats
389 exposed to silver nanoparticles (18.1–19.6 nm) at mass concentrations of 0.031, 0.082, or 0.116 mg/m³ (6
390 hours/day, 5 days/week for 4 weeks) (Jo et al., 2020).

391 *Reproductive toxicity* Pregnant and non-pregnant mice were exposed to 18–20 nm silver nanoparticles
392 at 0.54 mg/m³. Four hours of exposure resulted in an increased number of foetal resorptions. The oestradiol
393 serum level was decreased and a range of cytokines was increased after exposure to silver nanoparticles
394 (LOAEC_{foetal resorptions}: 0.54 mg/m³) (Campagnolo et al., 2017).

395 *Animal studies in which no toxicity was observed* We have described studies in which no effect on lung
396 function or inflammation occurred. In addition, 1 study in which rats were exposed to a low dose of silver
397 nanoparticles did not cause any effects on body weight, or biochemical or haematological parameters
398 (NOAEC: 0.061 mg/m³, 120 h of exposure) (Ji et al., 2007).

399

400 6. Genotoxicity and carcinogenicity

401 Concerning genotoxicity and carcinogenicity, all pathways of exposure and also *in vitro* incubation have
402 often been described in articles otherwise with a focus on 1 specific exposure pathway. We have also done
403 that for the oral and dermal/mucosal surface pathways of silver exposure (Hadrup et al., 2018; Hadrup and
404 Lam, 2014). The reason was the severity of this endpoint, combined with genotoxicity and carcinogenicity,
405 often developing through mechanisms inside single cells. The result of our previous review was that the
406 genotoxic potential of silver seemed likely, but the data was insufficient to provide a firm conclusion on the
407 carcinogenic potential (Hadrup et al., 2018). However, since our previous review, new studies have been
408 published, and we present these in the following section. Notably, carcinogenicity studies of silver were not
409 observed. First, we describe genotoxicity studies with the exposure pathway of inhalation.

410 Two studies report genotoxicity following inhalation. One study assessed DNA strand breaks in lungs in
411 male rats after 12 weeks of inhalation of silver nanoparticles (0.05, 0.12, or 0.38 mg/m³, 6h/day). An effect
412 was observed only at the highest dose level (Cho et al. 2013). The other study investigated micronuclei
413 formation after rats inhaled silver nanoparticles at 0.7×10^6 , 1.4×10^6 , or 2.9×10^6 particles/cm³, 6
414 hours/day for 90 days. The exposure to silver did not induce micronuclei, however, the treatment schedule
415 (90 days) is unusual for the *in vivo* micronucleus assay, and the calculation of the dose into mass units is not
416 readily performed (Kim et al. 2011). Only 1 study reported genotoxicity after intratracheal instillation.
417 Double strand breaks (γ -H2AX) were induced in rat lungs after a dosage of polyvinylpyrrolidone-coated silver
418 nanoparticles (300 μ g/rat, ~1.2 mg/kg bw) (Wiemann et al. 2017).

419 Concerning other exposure pathways, 2 oral studies that investigate if silver nanoparticles induce
420 micronucleus in bone marrow have reported contradictory results. The first study observed no effect in rats
421 after 28 days of exposure up to 1000 mg/kg bw/day (Kim et al. 2008). The other study reported an effect of
422 both coated and uncoated particles at the highest tested dose of 250 mg/kg bw/day in both male and female
423 mice after 28 days of exposure (Wang et al. 2009). After the intraperitoneal exposure of nanoparticles,
424 micronuclei formation was induced in bone marrow cells of mice, whereas DNA strand breaks results in the
425 same cell type were negative (Ghosh et al. 2012). No induction of sister chromatic exchanges were induced
426 in lymphocytes in the peritoneal cavity in mice after intraperitoneal injection of silver iodide (up to 100
427 mg/kg bw, 48 h) (Eliopoulos and Mourelatos, 1998). Intravenous exposure of nanoparticles has induced
428 different effects: oxidised damage to DNA (8-Oxo-2'-deoxyguanosine; 8-OHdG) in pregnant rats and
429 foetuses, DNA strand breaks (rabbit liver and mice spleen), and micronuclei formation in rat bone marrow
430 (Salim et al. 2019; Kim et al. 2019; Dobrzynska et al. 2014; Ordzhonikidze et al. 2009). Silver iodide also
431 induced DNA strand breaks in mice after iv exposure (Ordzhonikidze et al. 2009). In erythrocytes from the
432 African sharptooth catfish (*Clarias gariepinus*) with water as the exposure route, micronuclei formation was
433 induced by nanoparticles (Ogunsuyi et al. 2019). One study investigated *in vivo* genotoxic effects of silver
434 nitrate. DNA strand breaks were induced in fruit fly (*drosophila melanogaster*) haemocytes by both silver
435 nitrate and nanoparticles (Alaraby et al. 2019). One study investigated jewellery workers exposed to metallic
436 silver through inhalation and skin contact, and DNA strand breaks were observed in circulating mononuclear
437 leukocytes (Aktepe et al., 2015).

438 The deposition of silver nanoparticles in the nuclei of lung cells has been demonstrated in lungs of rats
439 exposed to 15 nm silver nanoparticles (Braakhuis et al., 2014). Whether the silver entered the nucleus as
440 whole nanoparticles or as ions is unknown. Whether the silver can interact directly with DNA to induce
441 genotoxicity is also unknown.

442 We reviewed a range of *in vitro* assays and present the details in Tables S1 and S2. The general
443 summary of all the *in vitro* studies is that silver does not induce mutations in bacteria but induces DNA
444 strand breaks and chromosomal aberrations in different mammalian cell lines. The addition of new *in vivo*

445 studies, compared with (Hadrup et al., 2018; Hadrup and Lam, 2014), does indicate that silver is genotoxic
446 by different exposure routes. However, we still find no data on carcinogenicity and therefore cannot provide
447 a conclusion on this endpoint. The references presented in tables S1 and S2 are as follows: (Akram et al.,
448 2013; Castro-Gamboa et al., 2019; Clark, 1953; Demerec et al., 1951; Eliopoulos and Mourelatos, 1998;
449 Foldbjerg et al., 2011; Guo et al., 2016; Hackenberg et al., 2011; Kanematsu et al., 1980; H. R. Kim et al.,
450 2013; J. S. Kim et al., 2013; Li et al., 2012; Mei et al., 2012; Nishioka, 1975; Park et al., 2011; Rossman and
451 Molina, 1986; Roszak et al., 2017; Wang et al., 2019); And *in vivo*: (Aktepe et al., 2015; Alaraby et al.,
452 2019; Cho et al., 2013; Dobrzyńska et al., 2014; Eliopoulos and Mourelatos, 1998; Ghosh et al., 2012;
453 Katsnelson et al., 2013; Kim et al., 2011, 2019, 2008; Ogunsuyi et al., 2019; Ordzhonikidze et al., 2009;
454 Salim et al., 2019; Wang et al., 2019).

455 *Conclusion* The addition of new studies indicates that silver is genotoxic in mammalian cells, and *in*
456 *vivo* when considering various exposure routes. Carcinogenicity data are absent; therefore, no conclusion can
457 be provided on this endpoint.

458

459 7. Comparison of ionic and nanoparticulate silver

460 A question of silver toxicity is whether the effect is mediated by the particle size and shape or by ions
461 released from the particle surface. In oral and dermal studies, it was difficult to distinguish a nanoparticle-
462 specific effect of the particle form because there were only limited differential effects and differential organ
463 deposition (Hadrup et al., 2018; Hadrup and Lam, 2014). Only a few studies on pulmonary exposure have
464 compared effects of silver nanoparticle preparations directly to those of free ions. Ionic silver was rapidly
465 distributed from the lung to the liver compared with silver nanoparticles for which silver was retained in the
466 lung for a longer period (Arai et al., 2015; Takenaka et al., 2001; Wang et al., 2014). Arai *et al.* found that
467 silver ions induced a higher effect on neutrophil cell numbers in BAL compared with silver nanoparticles;
468 this, taken together with the difference in deposition of equal doses, suggests that silver nanoparticles do not
469 rapidly dissolve in the lungs (Arai et al., 2015). Wang and colleagues found neutrophil numbers increased

470 after similar doses of both 20 nm silver nanoparticles and AgNO₃ but not after 110 nm silver nanoparticles,
 471 also suggesting a similar toxic mechanism of silver ions and nanoparticles in the lungs (Wang et al., 2014).
 472 Coccini *et al.* investigated changes in mRNA levels in rat liver and testes after a silver nanoparticle or silver
 473 nitrate instillation. The mRNA profiles differed between the 2 formulations, suggesting dissimilar
 474 mechanisms (Coccini, 2014). By contrast, in another study by the same group, silver nanoparticles at 50 µg
 475 and a dose of 7 µg AgNO₃ (4.4 µg Ag)/rat) given by intratracheal instillation exhibited similar effects in the
 476 kidney, including histopathological effects and dilation of Bowman's space (Roda et al., 2017).

477 Overall, we evaluate that the effects of silver nanoparticles after pulmonary dosage exert effects similar
 478 to those of silver ions/salts, suggesting that the effect of the nanoparticles is mediated by ions released from
 479 their surface. Concerning kinetics, there seems to be a more rapid distribution of silver from salts compared
 480 with silver from nanoparticles, likely reflecting the time necessary to release silver ions from the particle
 481 surface.

482

483 8. Hazard assessment

484 The dose descriptors of animal studies we deemed most relevant are summarised in Table 1. Based on this
 485 overview, we could suggest that a NOAEC value be set in the range of 0.11 to 0.75 mg/m³. A previous
 486 suggestion was (Weldon et al., 2016) that the bile duct hyperplasia observed in a subchronic study (Sung et
 487 al., 2009) was a critical effect, suggesting a NOAEC of 0.133 mg/m³, and this is in accordance with our
 488 aforementioned range.

Endpoint	NOAEC	LOAEC
Lung function	18–20 nm particles did not affect pulmonary tidal or minute volume at up to 0.75 mg/m ³ (Sung et al., 2011) Not available for studies showing an effect at the lowest tested concentration.	14–15 nm particle 0.05 mg/m ³ (360 h) (Song et al., 2013) 18 nm particle, 0.05 mg/m ³ (390 h) (Sung et al., 2008)

Inflammation (BAL neutrophils)	18 nm particle, rat, 0.5 mg/m ³ (390 h) (highest tested dose level) (Sung et al., 2008) 34 nm particle, rat, 0.041 mg/m ³ (24 h) (only dose level tested) (Braakhuis et al., 2016) 160 nm particle, rat, 0.555 mg/m ³ (24 h) (only dose level tested) (Braakhuis et al., 2016) 410 nm particle, rat, 0.17 mg/m ³ (24 h) (Braakhuis et al., 2014)	15 nm particle, rat, 0.18 mg/m ³ (24 h) (Braakhuis et al., 2014) 18 nm particle, rat 0.055 mg/m ³ (24 h) (only dose level tested) (Braakhuis et al., 2016) 60 nm particle, rat, 0.043 mg/m ³ (24 h) (only dose level tested) (Braakhuis et al., 2016)
Inflammation (histology)	14–15 nm particle, rat, 0.117 mg/m ³ (420 h) (Song et al., 2013) 18–9 nm, rat, 0.133 mg/m ³ (390 h) (Sung et al., 2009)	14–15 nm particle, rat, 0.381 mg/m ³ (420 h) (Song et al., 2013) 18–19 nm, rat, 0.515 mg/m ³ (390 h) (Sung et al., 2009)
Bile duct hyperplasia	18–19 nm particle, rat, 0.133 mg/m ³ (390 h) (Sung et al., 2009)	18.2 nm particle, rat, 0.515 mg/m ³ (390 h) (Sung et al., 2009)
Reproductive toxicity (foetal resorptions)	Not available	18–20 nm particle, mouse, 0.54 mg/m ³ (4 h) (Campagnolo et al., 2017)

489 **Table 1 Suggested dose descriptors from inhalation studies in rodents.** No descriptors were reported for
490 cardiovascular effects because the data were deemed to be scarce. Genotoxicity was not included in this table because it is
491 not an endpoint but a mechanism for cancer and reproductive effects.

492

493 We identified no inhalation studies in animals using silver ions (e.g. as silver nitrate); however, in a direct
494 comparison by intratracheal instillation, there is an indication on neutrophil numbers in BAL that ionic silver
495 has a higher effect compared with that of silver nanoparticles (Arai et al., 2015). This finding indicates that
496 lower exposure limits should be considered if also covering silver salts/ions.

497 Concerning particle size, if the limit is set at 100 nm, some studies compare particles that are smaller
498 and larger than this. Neutrophil numbers were increased in BAL, and a 160 nm particle had a higher NOAEC
499 of 0.56 mg/m³ compared with the LOAECs and NOAECs of 18–60 nm particles (0.05 mg/m³) (agglomerated
500 sizes) (Braakhuis et al., 2016). In another study, a 15 nm particle resulted in a LOAEC of 0.18 mg/m³,
501 whereas this level was a NOAEC for a 410 nm particle (agglomerated sizes), and a similar picture was
502 observed after intratracheal instillation (Braakhuis et al., 2014). These comparisons indicate that fine dust

503 (larger than nanoparticles but below 2.5 μm) silver particles are less toxic compared with silver
504 nanoparticles.

505 Finally, we evaluated that silver is genotoxic in mammalian cells and *in vivo*. However, there was
506 insufficient data to determine whether silver has a carcinogenic potential, warranting that additional data on
507 this endpoint be produced.

508

509 9. Summary

510 In 1 acute toxicity case, respiratory inefficiency and lung failure were reported, but the patient recovered.
511 Concerning chronic effects in humans, argyria was only observed in 3 cases in which silver was not proven
512 in the skin, and in 1 case, argyria was observed localised to the lungs but not the skin of 1 patient. The
513 occurrence of argyria after pulmonary exposure seems less likely than after dermal or oral exposure probably
514 due to lower total exposure levels of silver. In animal studies using inhalation, based on our evaluation, the
515 NOAEC values should be set in the range of 0.11 to 0.75 mg/m^3 . Ions are likely more toxic than
516 nanoparticles, but that could reflect that differences in biokinetics; and silver nanoparticles have a similar
517 pattern of toxicity to that of silver ions, probably reflecting that the effect of silver nanoparticles is mediated
518 by released ions. We evaluated silver to be genotoxic *in vitro* in mammalian cells, and *in vivo* when
519 considering various exposure routes. Carcinogenicity data are absent; therefore, no conclusion can be
520 provided on this endpoint.

521

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527

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Highlights

Inhaled silver is distributed to several organs. Few cases of argyria are reported

Excretion of silver includes faecal and to a lesser extent urinary pathways

Most affected endpoints in rodents are lung function and pulmonary inflammation

Ions show a higher effect, but the toxicity pattern is similar to nanoparticles

Silver was evaluated to be genotoxic based on *in vitro* and *in vivo* studies

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Declaration of interests

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

The authors declare the following financial interests/personal relationships which may be considered as potential competing interests:

On behalf of all co-authors,

Yours Sincerely

Niels Hadrup
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