

# Foaming properties of whey and soy protein isolates in mixed systems before and after heat treatment

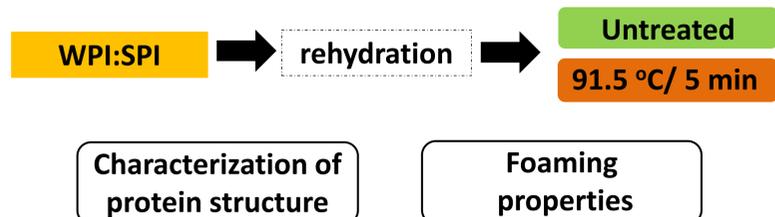
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## INTRODUCTION AND OBJECTIVE

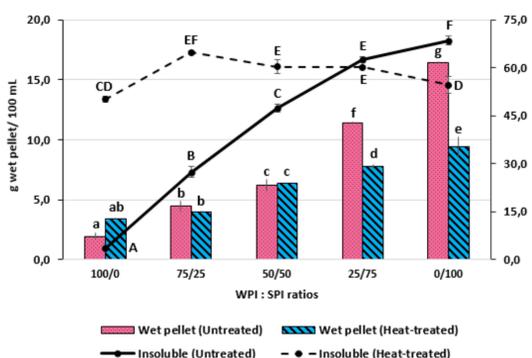
- Whey protein isolate (WPI):
  - β-lactoglobulin (β-Lg) – 18.3 kDa. Major proteins
  - α-lactalbumin (α-La) – 14.2 kDa. Major proteins
- Soy protein isolate (SPI):
  - β-conglycinin – 150-200 kDa. Major proteins
  - Glycinin – 300-380 kDa. Major proteins
- Heat treatment:
  - Changes on protein structure and on their techno-functional properties.

Do WPI and SPI mixed systems have synergic foaming properties?

## METHODOLOGY



## RESULTS AND DISCUSSION



**Solubility**

- WP proteins – decrease Thermo-induced protein aggregation;
- SP proteins – increase Thermo-induced dissociation of insoluble hydrophobic aggregates.

Fig. 1 Solubility of the samples. Different letters mean a significant difference ( $p < 0.05$ ). Lower-case letters were used for wet pellet data and upper-case letters were used for dry-basis insoluble material data.

### Total free sulfhydryl groups (SH)

- WPI:SPI (100:0) → disulfide bonds on stabilizing the thermo-induced WP aggregates;
- WPI:SPI (25:75) → formation or interchange of disulfide bonds promoted by highly reactive basic chain of Glycinin, in presence of β-LG.

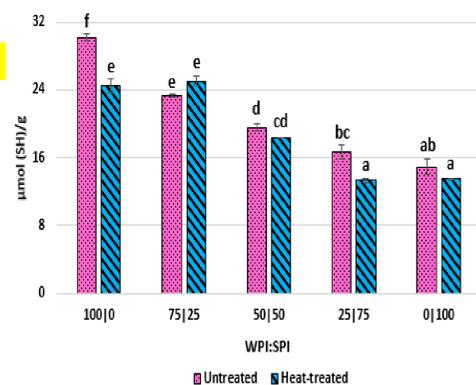


Fig. 2 Total free -SH groups of untreated and heat-treated samples containing different WPI:SPI ratios. Different letters mean significant difference ( $p < 0.05$ ) by One-way ANOVA with Tukey's post-test.

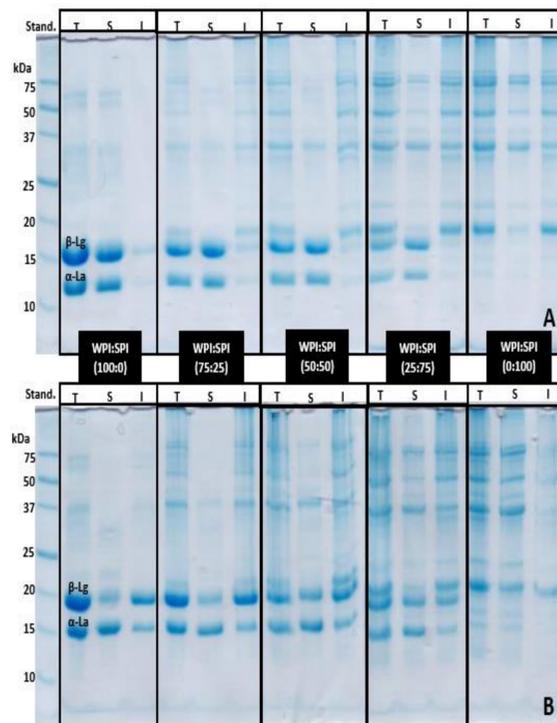


Fig. 3 SDS-PAGE in reduced condition of total, soluble and insoluble fractions of samples containing WPI:SPI mass ratios before (A) and after (B) heat treatment.

### Accessible hydrophobic patches of proteins in soluble fractions

**Heat treatment:** increase of the intensity of fluorescence: increase of the concentration of soluble SP → dissociation of SP hydrophobic aggregates

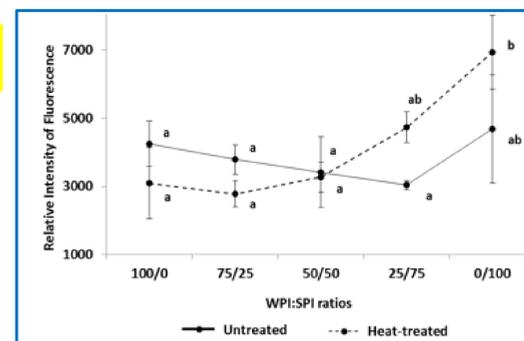


Fig. 4 Relative intensity of fluorescence of ANS (excitation: 365 nm and emission: 484 nm) added to the soluble fraction of samples containing different WPI:SPI ratios before and after heat treatment. Different letters mean significant difference ( $p < 0.05$ ) by One-way ANOVA with Tukey post-test.

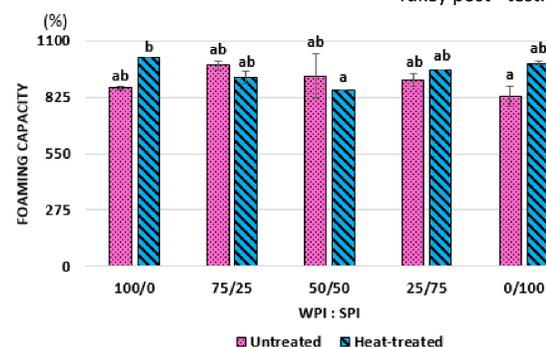


Fig. 5 Foaming capacity of untreated and heat-treated samples. Different letters mean significant difference ( $p < 0.05$ ) by One-way ANOVA with Tukey's post-test.

### SDS-PAGE

**WPI:SPI (100:0)**  
 - Untreated → complete soluble;  
 - Heat-treated → insoluble fraction: protein aggregation.

**WPI:SPI (0:100)**  
 - Untreated → Soluble and insoluble fractions;  
 - Heat-treated → predominance of soluble fraction;

**WPI:SPI (75:25 and 50:50)** → bands of SP were in the insoluble fractions;

**WPI:SPI (25:75)** → more equilibrated distribution of SP in the soluble and insoluble fractions.

### Foaming Capacity

- No significant differences of foaming capacity before and after heat treatment.
- All samples were able to entrap air in the same magnitude to produce foams.

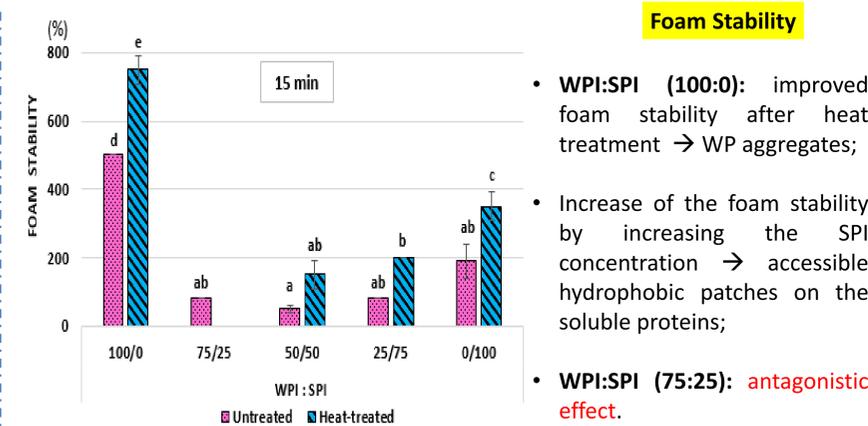


Fig. 6 Foam stability after 15 min on untreated and heat-treated samples. Different letters mean significant difference ( $p < 0.05$ ) by One-way ANOVA with Tukey's post-test.

### Hypothesis

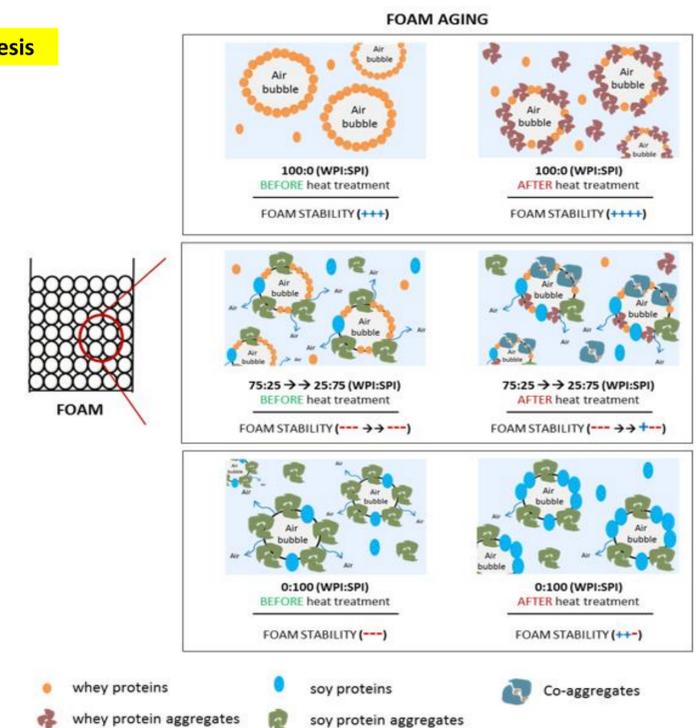


Fig. 7 Schematic representation of the foam stability of the different samples.

- Before heat treatment:** small amount the SP aggregates may sterically prevent the fixation of WP at the air-water interface, resulting in a less cohesive interfacial film.
- After heat treatment:** intense co-aggregation between WP and SP by hydrophobic interactions may reduce the susceptibility of such aggregates to act at the air-water interface.

## CONCLUSION

- Mixing WPI and SPI had a negligible effect on the FC of the mixtures, but it had an antagonistic effect on the FS of the samples;
- Before heat treatment SP insoluble aggregates negatively impacted the FS of samples containing SPI;
- After heat treatment, the formation of insoluble WP and SP co-aggregates, contributed to reduce the flexibility of the proteins and their susceptibility to act for reinforcing the interfacial film.