Exergy analysis of a combined heat and power plant with integrated lignocellulosic ethanol production

Lythcke-Jørgensen, Christoffer Ernst; Haglind, Fredrik; Clausen, Lasse Røngaard

Published in:
Proceedings of International Conference on Efficiency, Cost, Optimization, Simulation, and Environmental Impact of Energy Systems

Publication date:
2013

Citation (APA):
Exergy analysis of a combined heat and power plant with integrated lignocellulosic ethanol production

Christoffer Lythcke-Jørgensen a, Fredrik Haglind b and Lasse Røngaard Clausenc

a DTU Mechanical Engineering, Section for Thermal Energy, Copenhagen, Denmark, celjo@mek.dtu.dk
b DTU Mechanical Engineering, Section for Thermal Energy, Copenhagen, Denmark, frh@mek.dtu.dk
c DTU Mechanical Engineering, Section for Thermal Energy, Copenhagen, Denmark, lrc@mek.dtu.dk

Abstract:
Integrating second generation bioethanol production in combined heat and power units is expected to increase system energy efficiencies while producing sustainable fuel for the transportation sector at a competitive price. By applying exergy analysis, this study assessed the efficiency of an integrated system in which steam extracted from an existing combined heat and power unit is used for covering the heating demand of a lignocellulosic ethanol production facility. The integration solution was designed and optimized using already existing steam extraction points in the combined heat and power unit solely. The exergy flows inside the ethanol facility were determined, and the exergy in steam flows into and out of the system components were determined by combining exergy analysis with pinch analysis and setting a constant heat exchange temperature difference of 10K. The ethanol facility produces ethanol, solid biofuel, molasses, and is able to produce district heating hot water. Considering all products equally valuable, the exergy efficiency of the ethanol facility was found to be 0.790 during integrated operation with zero district heating production, and 0.852 during integrated operation with full district heating production. During separate operation, the exergy efficiency dropped to 0.564 with zero district heating production and 0.583 with full district heating production. The ratio of exergy losses and destruction in the heat integration network to the total exergy destruction and losses in the system was in the range of 0.46-0.87 depending on the system operation. This study suggests that a well-designed heat integration network can increase the exergy efficiency of the integrated system markedly.

Keywords:
Exergy analysis, lignocellulosic ethanol production, polygeneration system, plant operation.

1. Introduction
Lignocellulosic ethanol can be mixed with gasoline in order to reduce the greenhouse gas emissions from the transportation sector while decreasing the oil import dependency in countries without domestic oil reservoirs. At the same time, the production does not couple the food and energy prices in contrast to starch-based ethanol production. Lignocellulosic ethanol is therefore considered a promising second-generation biofuel by the Commission of the European Communities [1].

Due to the energy intensive nature of lignocellulosic ethanol production, production facilities are often assumed integrated in polygeneration systems. Several studies have treated systems with integrated production of power, heat, lignocellulosic ethanol, and synthetic natural gas (SNG) [2-6]. For such systems, Daianova et al. [2] and Ilic et al. [3] both report better energy economy for the integrated system compared to stand-alone systems, while Bösch et al. [4] reports a potential increase in both first law energy efficiency and exergy efficiency when integrating the processes. Modarresi et al. [5] applied pinch analysis to improve the heat integration of the system, which yielded an integrated exergy efficiency of 88% for the ethanol process. Furthermore, Gassner and Maréchal [6] has investigated process integration in the polygeneration system and concludes that both first and second law energy efficiencies for a combined heat and power (CHP) plant are increased significantly by integrating lignocellulosic ethanol and SNG production. Furthermore, a case study by Starfelt et al. [7] reports higher first-law energy efficiency for integrating lignocellulosic ethanol production in an existing CHP plant compared to a scenario with separate
production. These results all explain the industrial interest in retrofitting existing CHP units to obtain the mentioned polygeneration system benefits.

Analogue to the studies mentioned above, a previous case study by the authors [8] investigated the energy economy of a polygeneration system in which the Danish CHP unit Avedøreværket 1 (AVV1) is retrofitted by integrating a lignocellulosic ethanol production facility based on IBUS (Integrated Biomass Utilization System) technology [9-11]. It was found that the exergy flow from the CHP unit to the IBUS facility had a significant negative impact on the system economy, mainly due to the reduced power production capacity of the system. Integration design optimization was limited by stating that steam had to be extracted from existing steam extraction points only. This paper investigates the impact of the non-ideal integration solution, which is not considered in other studies of similar systems but is likely to be applied in real life plant retrofitting.

The objectives of the study are to determine the potential impact of a non-ideal integration solution in a polygeneration system, and to identify critical design challenges faced when retrofitting existing CHP units. To achieve this, the integration design of the polygeneration system previously treated by the authors [8] is addressed. Exergy analysis [12] is applied to determine irreversibilities in the heat integration network and the IBUS facility, and to identify ways of reducing the exergy flow from the CHP unit to the IBUS facility.

In this paper, the polygeneration system model and the exergy analysis approach are described in Section 2. The study results are presented in Section 3 and discussed in Section 4, while a conclusion of the study is given in Section 5.

2. Methodology

2.1. Polygeneration system model

The polygeneration system was modelled previously by the authors [8]. In this section, the basic design of the system and the data necessary to carry out an exergy analysis of the IBUS facility and the integration solution are presented. A more detailed explanation of the system modelling can be found in a previous study by the authors [8].

![Component layout of the modelled IBUS facility. Flow sizes are shown per kg/s straw processed. Pressure levels are only indicated for processes at elevated pressures](image)
The component layout of the modelled IBUS facility is presented in Fig. 1. The flows of yeast and enzymes are insignificant for the overall mass and energy balances of the system and were therefore not included in the model.

In the integrated polygeneration system, steam extracted from the Danish CHP unit AVV1 is used in the hydrothermal pretreatment and for covering the heating demands of the IBUS facility. A simplified component layout of the polygeneration system is presented in Fig. 2.

A rigid approach was taken for designing the steam extraction solution, stating that steam could only be extracted from already existing steam extraction points in the turbine system of AVV1. Three steam extraction points that meets the requirements for steam temperature and pressure, points (A), (B), and (C), were identified. The thermodynamic states of steam in the three points are summarized in Table 1. The value ranges reflect the load variation of the system.

<table>
<thead>
<tr>
<th>Point</th>
<th>Temperature range [C]</th>
<th>Saturation temperature range [C]</th>
<th>Pressure range</th>
</tr>
</thead>
<tbody>
<tr>
<td>(A)</td>
<td>431-467</td>
<td>199.8-241.2</td>
<td>15.5-34.2</td>
</tr>
<tr>
<td>(B)</td>
<td>359-392</td>
<td>176.7-213.6</td>
<td>9.3-20.5</td>
</tr>
<tr>
<td>(C)</td>
<td>257-289</td>
<td>145.4-176.1</td>
<td>4.2-9.2</td>
</tr>
</tbody>
</table>

To minimize exergy destruction in the heat integration network of the IBUS facility, steam was extracted at the lowest possible temperature and pressure which still allowed for the required heat transfer. Pinch analysis was used to calculate the required steam extraction flows that minimized the exergy flow into the IBUS facility. The method is described by Lythcke-Jørgensen et al. [8].

### 2.2. System operation

The thought behind the polygeneration system is to allow for the IBUS facility to run on excess capacity in the CHP unit boiler. As the IBUS facility has zero production flexibility, a back-up heat source must be used to maintain production when the CHP unit uses its full boiler capacity. This leads to two basic operation modes:

**Integrated operation**: The heat demand of the IBUS facility is covered by steam extracted from the CHP unit

**Separate operation**: The heat demand of the IBUS facility is covered by a natural gas boiler
In this study, four types of system operation schemes were investigated:

i) Integrated operation with full load condensation mode operation in AVV1 and zero district heating (DH) production in the IBUS facility

ii) Integrated operation with full load back pressure mode operation in AVV1 and full DH production in the IBUS facility

iii) Separate operation with zero DH production in the IBUS facility

iv) Separate operation with full DH production in the IBUS facility

During separate operation, the heat demand in the IBUS facility is covered by steam from a gas boiler with 96% heating efficiency. Chemical properties and specific flow sizes of gas per kg/s straw processed at various operation modes are summarized in Table 2. Further details about the operational patterns of the polygeneration system are given in Lythcke-Jørgensen et al. [8].

Table 2 – Chemical properties of natural gas and specific consumption during separate operation with zero and full IBUS district heating production. Chemical properties for natural gas have been taken from [13].

<table>
<thead>
<tr>
<th>Fuel</th>
<th>Lower heating value [kJ/kg]</th>
<th>Chemical exergy content [kJ/kg]</th>
<th>Gas flow, zero IBUS DH production [kg/s]</th>
<th>Gas flow, full IBUS DH production [kg/s]</th>
</tr>
</thead>
<tbody>
<tr>
<td>Natural Gas</td>
<td>41,426</td>
<td>43,497</td>
<td>0.270</td>
<td>0.325</td>
</tr>
</tbody>
</table>

2.3. Exergy Analysis Calculations

Exergy analysis was applied to the IBUS facility and the heat integration network to identify irreversibilities, and thereby ways for reducing the exergy flow from the CHP unit to the IBUS facility. The reference point for all exergy calculations was set to $T_0 = 298.15K$ and $p_0 = 1bar$.

The total specific exergy of a material stream, $e_x$, is given by the following expression:

$$e_x = e_{x_{phys}} + e_{x_{kin}} + e_{x_{pot}} + e_{x_{chem}}.$$  (1)

Here, $e_{x_{phys}}$ is the specific physical exergy, $e_{x_{kin}}$ is the specific kinetic exergy, $e_{x_{pot}}$ is the specific potential exergy, and $e_{x_{chem}}$ is the specific chemical exergy. $e_{x_{kin}}$ and $e_{x_{pot}}$ were not considered in the analysis as they are negligible in magnitude for the given material streams [4].

Exergy flows related to mass flows in the system, $\dot{E}X$, were calculated as the total specific exergy of the material stream $e_x$ times the mass flow of the stream, $\dot{m}$:

$$\dot{E}X = e_x \cdot \dot{m}.$$  (2)

Considering all useful material streams out of a component as valuable, the exergy efficiency $\eta_{EX}$ of a system component was calculated as the exergy content in product flows, $\Sigma \dot{E}X_{products}$, divided by the exergy content in inlet flows, $\Sigma \dot{E}X_{in}$:

$$\eta_{EX} = \frac{\Sigma \dot{E}X_{products}}{\Sigma \dot{E}X_{in}}.$$  (3)

In this study, the difference between the exergy content in inlet mass flows and in product mass flows was caused partly by exergy destruction within the component, partly by exergy content in unused material flows from the component. For simplicity, these fractions were merged in a term called exergy losses and destruction, $\dot{E}X_{L&D}$, which is calculated as:

$$\dot{E}X_{L&D} = \Sigma \dot{E}X_{in} - \Sigma \dot{E}X_{products}.$$  (4)

As the main product of the IBUS facility is lignocellulosic ethanol and solid fuel for combustion, the exergy efficiency for the fuel production, $\eta_{EX, fuels}$, is evaluated for the whole system as well:
\[ \eta_{EX, fuels} = \frac{EX_{ethanol} + EX_{solid fuel}}{\sum EX_{in, system}}. \] (5)

### 2.3.1. Chemical Exergy

The chemical exergy of multi-component material streams, \( ex_{chem} \), was calculated as the sum of the chemical exergy content of the individual components \( ex_{chem,i} \) multiplied by their weight fraction \( x_i \):

\[ ex_{chem} = \sum_i ex_{chem,i} x_i. \] (6)

The chemical exergy for the material components that occur in the IBUS facility are summarized in Table 3. The values have been taken from a study by Bösch et al. [4].

### 2.3.2. Physical Exergy

As material flows in the system occur at reference pressure, the mechanical part of the physical exergy is zero. As suggested by Bösch et al. [4], heat capacities of biomass material streams were assumed constant over the relevant temperature ranges present. The physical exergy of multi-component biomass materials was calculated according to the specific heat capacity of the material, \( C_p \), and the temperature of the material, \( T \), using the following equation [12].

\[ ex_{phys,biomass} = C_p \left( T - T_0 - \ln \left( \frac{T}{T_0} \right) \right). \] (7)

The heat capacity of multi-component materials were calculated as the sum of the individual components’ heat capacities \( C_{pi} \) multiplied by their weight fraction \( x_i \). Used values are presented in Table 3.

\[ C_p = \sum_i C_{pi} x_i. \] (8)

### Table 3 – Chemical properties of material components in the IBUS facility. Heat capacities for biomass components are assumed constant over the relevant temperature range. Heat capacities for water and ethanol have been taken from the software ‘Engineering Equation Solver’[14]. All other values have been taken from ref. [4]

<table>
<thead>
<tr>
<th>Material component</th>
<th>( C_p ) [kJ/kg-K]</th>
<th>( ex_{chem} ) [kJ/kg]</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cellulose</td>
<td>1.28</td>
<td>18,808</td>
</tr>
<tr>
<td>Hemicellulose</td>
<td>1.28</td>
<td>18,808</td>
</tr>
<tr>
<td>Lignin</td>
<td>1.29</td>
<td>25,648</td>
</tr>
<tr>
<td>Monomers</td>
<td>1.15</td>
<td>16,687</td>
</tr>
<tr>
<td>Proteins</td>
<td>1.30</td>
<td>24,488</td>
</tr>
<tr>
<td>Ash</td>
<td>0.70</td>
<td>1,006</td>
</tr>
<tr>
<td>Glucose</td>
<td>1.15</td>
<td>16,687</td>
</tr>
<tr>
<td>Ethanol</td>
<td>2.53</td>
<td>29,532</td>
</tr>
<tr>
<td>Water</td>
<td>4.18</td>
<td>51</td>
</tr>
</tbody>
</table>

### 2.3.3. Exergy in Heat Flows

The exergy content of heat flows in the system are associated with the exergy content in the heat exchange media used. In the IBUS facility, water was assumed used as heat exchange media. Heat is provided in the form of steam from AVV1 or a gas boiler, and cooling is provided by district heating water and sea water at the given location.

To illustrate the best possible heat exchange conditions, the entire span of any heat exchange between heat exchange media and system components have been set to occur at a temperature difference equal to the pinch point temperature difference, which was set to 10K as suggested by Modarresi et al. [5]. The concept is illustrated in Fig. 3.
The reference pressure of the heat exchange media was set to 1 bar, but for phase change heat exchange, the pressure has been modified to maintain the 10 K temperature difference over the entire phase change of the hot and cold flows. A list of all heat exchange processes in the system, described by the heat flow transferred, \( Q \), the temperature interval, the necessary mass flow, \( m \), and pressure, \( p \), of the heat exchange media is given in Table 4. Pumping in the heat integration network was not considered. Further details about heat flows and pinch analysis of the integrated system are given in Lythcke-Jørgensen et al. [8].

![Fig. 3 – Illustration of best-possible heat transfer between a hot stream and heat exchange media in the IBUS facility](image)

**Table 4 – Hot and cold heat exchange media streams in the IBUS facility**

<table>
<thead>
<tr>
<th>Component</th>
<th>Type</th>
<th>( Q ) [kJ/kg]</th>
<th>( T_{in} ) [C]</th>
<th>( T_{out} ) [C]</th>
<th>( p ) [bar]</th>
<th>( m ) [kg]</th>
</tr>
</thead>
<tbody>
<tr>
<td>Pretreatment</td>
<td>Hot</td>
<td>5,466</td>
<td>195</td>
<td>-</td>
<td>13</td>
<td>2.00</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>341</td>
<td>90</td>
<td>180</td>
<td>0.7018</td>
<td>1.91</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>4,287</td>
<td>90</td>
<td>91</td>
<td>0.7018</td>
<td>1.88</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>92</td>
<td>70</td>
<td>90</td>
<td>1</td>
<td>1.09</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>247</td>
<td>40</td>
<td>90</td>
<td>1</td>
<td>1.76</td>
</tr>
<tr>
<td>Liquefaction</td>
<td>Cold</td>
<td>60</td>
<td>25</td>
<td>40</td>
<td>1</td>
<td>0.96</td>
</tr>
<tr>
<td>Distillation</td>
<td>Hot</td>
<td>1,631</td>
<td>111</td>
<td>110</td>
<td>1.431</td>
<td>0.73</td>
</tr>
<tr>
<td></td>
<td>Hot</td>
<td>21</td>
<td>47</td>
<td>43</td>
<td>1.431</td>
<td>1.26</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>1,044</td>
<td>68</td>
<td>69</td>
<td>0.286</td>
<td>0.45</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>1</td>
<td>25</td>
<td>90</td>
<td>1</td>
<td>0.10</td>
</tr>
<tr>
<td>Separation</td>
<td>Hot</td>
<td>4,368</td>
<td>111</td>
<td>110</td>
<td>1.431</td>
<td>1.96</td>
</tr>
<tr>
<td></td>
<td>Hot</td>
<td>100</td>
<td>110</td>
<td>90</td>
<td>1.431</td>
<td>1.18</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>4,368</td>
<td>90</td>
<td>91</td>
<td>0.7018</td>
<td>1.91</td>
</tr>
<tr>
<td></td>
<td>Cold</td>
<td>807</td>
<td>25</td>
<td>90</td>
<td>1</td>
<td>2.97</td>
</tr>
<tr>
<td>District Heating Water*</td>
<td>Cold</td>
<td>13,071</td>
<td>50</td>
<td>100</td>
<td>20</td>
<td>62.4</td>
</tr>
</tbody>
</table>

* District heating water is only produced when necessary in the IBUS facility.
3. Results

Using the method described previously, the exergy flows of the integrated system were calculated. A Grassmann diagram of the exergy flows within the IBUS facility is presented in Fig. 4. The diagram shows how the exergy content of the inlet wheat straw passes through the various processes in the facility until it ends up in the final energy products: ethanol, solid fuel and molasses. The exergy flows into the system from heat transfer and power are also illustrated. Exergy losses and destruction (L&D), which cover exergy destruction in components and exergy content of discarded material streams, are illustrated as well. Exergy destruction in the system is in general related to heat transfer over a temperature difference, frictional exergy destruction, and material degradation.

The largest exergy L&D in the IBUS facility processes occurs in the Simultaneous Saccharification and Fermentation (SSF) component where the exothermal fermentation process takes place. Heat released from the fermentation is used to maintain an elevated temperature in the component, explaining the lack of heat flow into the component. L&D in this component are associated with heat losses, product degradation from the fermentation process, and exergy content of the CO2 which is discarded at an elevated temperature.

![Grassmann diagram illustrating the exergy flows inside the IBUS facility per kg/s straw processed. Exergy losses and destruction (L&D) is illustrated as well](image)

In the heat integration network, the heat exchange media is conditioned and distributed to meet the heating and cooling demands of the IBUS facility. Heating is provided by steam from AVV1 or a gas boiler, and recovered heat from the processes in the IBUS facility. Cooling is provided by seawater and by district heating water when district heating production is active in the IBUS facility.
The Grassmann diagrams in Fig. 5 show the exergy flows in the heat integration network during the following operation schemes:

i) Integrated operation, full load condensation mode operation in AVV1, zero IBUS DH production

ii) Integrated operation, full load back pressure mode operation in AVV1, full IBUS DH production

iii) Separate operation, zero IBUS DH production

iv) Separate operation, full IBUS DH production

Exergy L&D in the heat integration network are related to exergy destruction from heat transfer over a temperature difference, and heat losses. Pumping work and frictional losses were not considered.

Fig. 5 – Grassmann diagrams of the heat integration network during (i) full load condensation mode operation in AVV1 with zero IBUS DH production, (ii) full load back pressure operation in AVV1 and full IBUS DH production, (iii) separate operation, zero IBUS DH production, and (iv) separate operation, full IBUS DH production. Exergy flows are presented per kg/s straw treated in the IBUS facility.
The calculated exergy efficiencies of the IBUS facility components and the heat integration network are presented in Table 5. The IBUS facility alone was found to have an exergy efficiency of 0.913, while the exergy efficiency of the heat integration network was found to be highly dependent on operation mode.

The lignocellulosic ethanol production system was found to have the highest exergy efficiencies during integrated operation. When the CHP unit was operated at full load in condensation mode, an exergy efficiency of 0.790 was obtained, while it was increased to 0.852 during full load back pressure operation in the CHP unit and with DH production in the IBUS facility. The reason for this increase is the fact that DH hot water was considered a product in the analysis. For a similar system with lignocellulosic ethanol production integrated in a CHP unit, Modarresi et al. [5] reported an exergy of 0.793 for the lignocellulosic ethanol system.

Markedly lower exergy efficiencies are obtained for the system during separate operation. Without DH production in the IBUS facility, the system was found to have an exergy efficiency of 0.564, while it was increased to 0.583 when DH production was included. This clearly illustrates the benefits of integrated operation for the system in terms of exergy efficiency.

When looking at the exergy efficiency of the fuel production solely, it is seen that integrated operation yields higher exergy efficiencies as well, while operation with DH production in the IBUS facility gives lower exergy efficiencies than for operation without DH production. The reason is that DH hot water is not included as a valuable product in the fuel production exergy efficiency, but the DH production increases the heating demand of the system slightly.

Table 5 – Exergy destruction and exergy efficiency in the IBUS facility, its processes and the heat integration network

<table>
<thead>
<tr>
<th>System</th>
<th>Exergy losses and destruction per kg/s of straw processed [kW]</th>
<th>(\eta_{\text{EX}} [-])</th>
<th>(\eta_{\text{EX,fuels}} [-])</th>
</tr>
</thead>
<tbody>
<tr>
<td>IBUS facility</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Hydrothermal pretreatment</td>
<td>205</td>
<td>0.990</td>
<td>-</td>
</tr>
<tr>
<td>Liquefaction</td>
<td>73</td>
<td>0.994</td>
<td>-</td>
</tr>
<tr>
<td>SSF</td>
<td>710</td>
<td>0.939</td>
<td>-</td>
</tr>
<tr>
<td>Distillation</td>
<td>201</td>
<td>0.986</td>
<td>-</td>
</tr>
<tr>
<td>Separation</td>
<td>570</td>
<td>0.954</td>
<td>-</td>
</tr>
<tr>
<td>IBUS facility, total</td>
<td>1,759</td>
<td>0.913</td>
<td>0.600</td>
</tr>
<tr>
<td>Heat Integration Network</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Operation mode (i)</td>
<td>2,642</td>
<td>0.550</td>
<td>-</td>
</tr>
<tr>
<td>Operation mode (ii)</td>
<td>1,509</td>
<td>0.725</td>
<td>-</td>
</tr>
<tr>
<td>Operation mode (iii)</td>
<td>11,057</td>
<td>0.217</td>
<td>-</td>
</tr>
<tr>
<td>Operation mode (iv)</td>
<td>11,712</td>
<td>0.312</td>
<td>-</td>
</tr>
<tr>
<td>Integrated System, total</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Operation mode (i)</td>
<td>4,401</td>
<td>0.790</td>
<td>0.579</td>
</tr>
<tr>
<td>Operation mode (ii)</td>
<td>3,268</td>
<td>0.852</td>
<td>0.560</td>
</tr>
<tr>
<td>Operation mode (iii)</td>
<td>12,816</td>
<td>0.564</td>
<td>0.385</td>
</tr>
<tr>
<td>Operation mode (iv)</td>
<td>13,471</td>
<td>0.583</td>
<td>0.357</td>
</tr>
</tbody>
</table>

Depending on the operation mode, 46% to 87% of the exergy L&D in the integrated system occurs in the heat integration network. This leaves room for a significant exergy efficiency increase by optimizing this part of the system. One way would be to extract steam at conditions closer to those
needed in the heat integration network. Another way to optimize the average exergy efficiency of the system over the lifetime is to minimize the duration of separate operation in the polygeneration system. Comparing these results with the outcomes of a thermo-economic analysis of the system [8], it is suggested that a carefully designed heat integration network combined with well-placed steam extraction points can increase the production economy of the treated polygeneration system.

4. Discussion

The outcomes of this study suggest that for the polygeneration system treated, the rigid approach for steam extraction when designing the heat integration network caused the exergy efficiency of the lignocellulosic ethanol production system to decrease significantly. It is therefore suggested that a more flexible approach is taken as extraction of steam at conditions closer to those that are optimal for the heat exchange will increase both the exergy efficiency and production economy of the polygeneration system. The theoretic optimal efficiency might be increased further if the pinch point temperature difference of 10°C could be lowered. However, the designed heat integration network is optimal in the sense that all heat exchange occurs at a temperature difference equal to the pinch point temperature difference. This would hardly be the case in real life, and it is therefore not reasonable to assume that the exergy losses and destruction in the heat integration network calculated in this study can be eliminated completely.

There are several other limitations to the efficiency increase that can be gained through the design of the heat integration network. First of all, it is not certain that steam is available at the ideal conditions for the heat exchange with the IBUS facility. The CHP unit is designed to fit the steam conditions for turbine expansion, meaning that the steam is superheated until it reaches the lower pressure levels. Opposed to this, saturated steam is optimal for heat exchange with the IBUS facility, and they might therefore not be available in the CHP unit.

Secondly, if points exist where the steam conditions are ideal for exchanging heat with the IBUS facility, it is not certain that steam can be extracted therefrom without having to replace system components, which not only poses an economical challenge for the plant owner but might in fact be infeasible as new equipment might not fit with other system components. And even if steam can be extracted from near-optimal points in the CHP unit it is not necessarily economically reasonable to do so. Investment in advanced heat integration network components and upgrading or replacement of several existing components in the CHP unit might not be justified by the increased production economy. Whether this is the case or not requires further investigations, but these points represent main challenges when retrofitting existing CHP units, and it urges a careful approach when making assumptions on benefits from retrofitting.

5. Conclusion

This study treated the integrated production of heat, power and lignocellulosic ethanol in a polygeneration system based on an existing CHP unit. A rigid approach was taken when designing and optimizing the system integration, stating that steam extracted from the CHP unit could only be extracted from already existing steam extraction points. The efficiency of the lignocellulosic ethanol production facility, called the IBUS facility, was evaluated using exergy analysis. At optimal heat exchange conditions, the results indicate an exergy efficiency of 0.913 for the IBUS facility. Calculations suggest an exergy efficiency of 0.790 for the integrated system with full load condensation mode operation in the CHP unit and zero DH production in the IBUS facility, and 0.852 with full load back pressure operation in the CHP unit and full DH production in the IBUS facility. During separate operation, the exergy efficiency of the lignocellulosic ethanol production decreases to 0.564 with zero DH production, and 0.583 with full DH production due to the combustion of natural gas for covering the heat demand. When only including ethanol and solid biofuel as products in the exergy analysis, the exergy efficiency of the system is suggested to be in the range 0.357-0.579 depending on operation mode. The results illustrate the benefits of maintaining integrated operation in the treated polygeneration system in terms of exergy efficiency,
but they also point out challenges associated with retrofitting existing CHP units as the heat integration network, which was optimized with respect to already existing steam extraction points in the CHP unit, was responsible for up to 87% of the exergy losses in the integrated lignocellulosic ethanol production.

Acknowledgments

The authors would like to thank Brian Elmegaard for allowing the use of his numerical model of the Danish combined heat and power unit Avedøreværket 1, and for his feedback on the method used for analysing exergy flows in the system.

Nomenclature

<table>
<thead>
<tr>
<th>Symbol</th>
<th>Description</th>
<th>Units</th>
</tr>
</thead>
<tbody>
<tr>
<td>$C_p$</td>
<td>specific heat capacity</td>
<td>kJ/(kg K)</td>
</tr>
<tr>
<td>$EX$</td>
<td>exergy flow</td>
<td>kW</td>
</tr>
<tr>
<td>$ex$</td>
<td>specific exergy content</td>
<td>kJ/kg</td>
</tr>
<tr>
<td>$ex_{chem}$</td>
<td>specific chemical exergy content</td>
<td>kJ/kg</td>
</tr>
<tr>
<td>$ex_{kin}$</td>
<td>specific kinetic exergy content</td>
<td>kJ/kg</td>
</tr>
<tr>
<td>$ex_{phys}$</td>
<td>specific physical exergy content</td>
<td>kJ/kg</td>
</tr>
<tr>
<td>$ex_{pot}$</td>
<td>specific potential exergy content</td>
<td>kJ/kg</td>
</tr>
<tr>
<td>$\dot{m}$</td>
<td>mass flow</td>
<td>kg/s</td>
</tr>
<tr>
<td>$p$</td>
<td>pressure</td>
<td>bar</td>
</tr>
<tr>
<td>$\dot{Q}$</td>
<td>heat flow</td>
<td>kJ/s</td>
</tr>
<tr>
<td>$T$</td>
<td>temperature</td>
<td>K</td>
</tr>
<tr>
<td>$x$</td>
<td>mass fraction</td>
<td>-</td>
</tr>
</tbody>
</table>

Abbreviations

<table>
<thead>
<tr>
<th>Abbreviation</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>AVV1</td>
<td>Avedøreværket 1</td>
</tr>
<tr>
<td>CHP</td>
<td>Combined Heat and Power</td>
</tr>
<tr>
<td>DH</td>
<td>District Heating</td>
</tr>
<tr>
<td>L&amp;D</td>
<td>(exergy) Losses and Destruction</td>
</tr>
<tr>
<td>SNG</td>
<td>Synthetic Natural Gas</td>
</tr>
<tr>
<td>SSSF</td>
<td>Simultaneous Saccharification and Fermentation</td>
</tr>
</tbody>
</table>

References


